Multiple Approaches to Characterizing Pore Structure in Barnett Shale

(Max) Qinhong Hu^a, Zhiye Gao^a, Sheng Peng^a, Robert Ewing^b, Shoichiro Hamamoto^c, Stefan Dultz^d, and Beau Webber^e

^a University of Texas at Arlington, Arlington, TX, USA

^b Iowa State University, Ames, IA, USA

^c Saitama University, Saitama, Japan

^d Leibniz University of Hannover, Hannover, Germany

^e Lab-Tools Ltd, England



ТЕХАЅ

ARLINGTON

Saitama University



Source: Energy Information Administration based on data from various published studies Updated: May 9, 2011

http://www.eia.gov/pub/oil_gas/natural_gas/analysis_publications/maps/maps.htm

Average First Year Decline



Dan Jarvie, Worldwide Geochemistry

AAPG European Region ICE, 17-19 October 2010, Kiev, Ukraine

Pore Structure and Low Hydrocarbon Production



Pore Geometry and Topology



Intraparticle organic nanopores

Ar ionbeam milling and field emission gun SEM: resolve pores as small as 5 nm

Loucks et al. (2009)



NANOPORES IN SILICEOUS MUDSTONES

Where is the porosity?

FIG. 5.—Nanopores associated with organic matter in the Barnett Shale. A) Elliptical to complexly rounded nanopores in an organic grain. Darker materials are organics. BSE image. Blakely #1, 2,167.4 m. B) Angular nanopores in a grain of organic matter. SE image. Blakely #1, 2,167.4 m. Accelerating voltage = 10 kV; working distance = 6 mm. C) Rectangular nanopores occurring in aligned convoluted structures. SE image. T.P. Sims #2, \sim 2,324 m. Accelerating voltage = 2 kV; working distance = 3 mm. D) Nanopores associated with disseminated organic matter. Carbon-rich grains are dark gray; nanopores are black. SE image. T.P. Sims #2, \sim 2,324 m. Accelerating voltage = 2 kV; working distance = 2 mm.

Shale Gas Flow: Matrix "diffusion" vs. "Darcy" flow



2009 North American Shale Gas Overview NECA.pdf

Multiple Approaches to Studying Pore Structure

- Pore-scale network modeling (ISU)
- Imbibition with samples of different shapes (UTA-Gao)
- Edge-accessible porosity
- Liquid and gas diffusion
- Mercury injection porosimetry (UTA-Gao)
- N₂ adsorption isotherm (Saitama Univ.; Quantachrome)
- Water vapor adsorption isotherm
- Nuclear Magnetic Resonance Cryoporometry (Univ. Kent)
- SEM imaging after Wood's metal impregnation (Univ. Hannover; Swiss EPMA)
- Microtomography (high-resolution, synchrotron) (PNNL-EMSL; Swiss Light Source; Univ. Hannover; Saitama Univ.)
- Focused Ion Beam/SEM imaging (PNNL-EMSL)
- Small-Angle Neutron Scattering (SANS) (LANL)

(Spontaneous) Imbibition Test



Imbibition Results for **Barnett Shale** Samples

Depth	Sample dimension	Height/width	Imbibition slope
7,109 ft	1.33 cm L×1.76 cm W ×1.43 cm H (Vertical)	0.93	$0.214 \pm 0.059 (N=3)$
(2,167 m)	1.76 cm L×1.72 cm W ×1.32 cm H (Horizontal)	0.76	$0.291 \pm 0.027 (N=3)$
7,136 ft	1.38 cm L×1.71 cm W ×1.72 cm H (Vertical)	1.12	$0.269 \pm 0.0045 (N=3)$
(2,175 m)	1.73 cm L×1.73 cm W ×1.21 cm H (Horizontal)	0.70	$0.216 \pm 0.040 (N=3)$
7,169 ft	1.35 cm L×1.79 cm W ×1.81 cm H (Vertical)	1.16	$0.273 \pm 0.050 (N=3)$
(2,185 m)	1.24 cm L×1.78 cm W ×1.32 cm H (Horizontal)	0.87	$0.357 \pm 0.006 (N=3)$
7,199 ft	1.24 cm L×1.74 cm W ×1.67 cm H (Vertical)	1.12	$0.284 \pm 0.062 (N=3)$
(2,194 m)	1.74 cm L×1.72 cm W × 1.26 cm H (Horizontal)	0.67	$0.282 \pm 0.047 (N=3)$
7,219 ft	1.37 cm L×1.74 cm W × 1.95 cm H (Vertical)	1.25	$0.306 \pm 0.019 (N=3)$
(2,200 m)	1.69 cm L×1.71 cm W ×1.36 cm H (Horizontal)	0.80	$0.264 \pm 0.046 (N=3)$

Percolation Theory

The mathematics of how macroscopic properties result from local (microscopic) connections



p is the local connection probability

percolation threshold $0.5 < p_c < 0.66$ (for 2D square lattice)



p = 0.66

p = 0.5





LA-ICP-MS instrumentation

-

3D Elemental Mapping: Edge-Accessible Porosity



Liquid Tracer Diffusion in Saturated Intact



- Gas molecule movement in shale on the order of 10 feet in the lifetime of a well - Dr. Mohan Kelcar, University of Tulsa.
- Gas molecule movement of about a meter/year modeled by Nexen's Unconventional Team, presented at Global Gas Shales Summit, Warsaw, Poland.
 - ~1 m/yr movement (advection vs. diffusion ?)
- Gas molecule movement of a few feet/year modeled by Dr. Chunlou Li, Shale Gas Technology Group.

$$\frac{C(x,t)}{C_0} = erfc(\frac{x}{2(D_e t)^{0.5}}) \qquad D_e = \frac{\delta D_0}{\tau}$$

For C/C₀=0.5 @ 1 m/y, $\tau = 613$ For C/C₀=0.01 @ 1 m/y, $\tau = 9,800$

LaFollette, R. 2010. Key Considerations for Hydraulic Fracturing of Gas Shales. Manager, Shale Gas Technology, BJ Services Company, September 9, 2010. www.pttc.org/aapg/lafollette.pdf

Gas Diffusion in Partially-Saturated Shale Powder



Multiple Approaches to Studying Pore Structure

- Pore-scale network modeling (ISU)
- Imbibition with samples of different shapes (UTA-Gao)
- Edge-accessible porosity
- Liquid and gas diffusion
- Mercury injection porosimetry (UTA-Gao)
- N₂ adsorption isotherm (Saitama Univ.; Quantachrome)
- Water vapor adsorption isotherm
- Nuclear Magnetic Resonance Cryoporometry (Univ. Kent)
- SEM imaging after Wood's metal impregnation (Univ. Hannover; Swiss EPMA)
- Microtomography (high-resolution, synchrotron) (PNNL-EMSL; Swiss Light Source; Univ. Hannover; Saitama Univ.)
- Focused Ion Beam/SEM imaging (PNNL-EMSL)
- Small-Angle Neutron Scattering (SANS) (LANL)

MIP: Pore-Throat Size Distribution



- **Mercury Injection Porosimetry (MIP)**
- volume Non-wetting Hg intrusion progressively intrude smaller Cumulative pores with increasing pressures
- Measurable pore diameter range: 3 nm to 360 μ m



Barnett Shale sample (~15 mm cube) in the penetrometer



MIP Intrusion Results: 6 Representative Rocks

Depth	Porosity (%)	Median pore- throat diameter (nm)	Permeability (µdarcy)	Tortuosity	
Berea Sandstone	22.9 ± 1.72	23,776±876	$(595 \pm 21.2) \times 10^3$	3.31 ± 0.33	
Indiana Sandstone	16.4 ± 0.4	19,963±2,932	$(221 \pm 40.8) \times 10^{3}$	4.68 ± 1.68	
Welded Tuff	10.0 ± 0.5	47 ± 7.1	0.83 ± 0.14	1,745 <u>+</u> 66	
Dolomite	9.15	873	409	38.3	
Barnett Shale (7,199')	5.97 ± 1.43	6.1 ± 0.3	$(4.96 \pm 1.42) \times 10^{-3}$	12,867±16,224	
NC Granite	1.05	970	12.4	38.2	
D 1^{11} T (1^{11}) $(400 < 400^{-1})$					

Permeability: Katz and Thompson (1986; 1987) Tortuosity: Hager (1998)

N₂ Sorption Isotherm



- Autosorb-IQ-MP by Quantachrome
- Pore size range:
 0.35–500 nm

Shoichiro Hamamoto (Saitama University)

- Physical adsorption of N₂ at cryogenic temperatures (77K, -196°C)
- Molecular sorption by van der Waals forces; monolayer coverage; multilayer formation; capillary condensation; total pore volume filling
- Various theory to estimate pore-size distribution





N, Sorption Isotherm: Hysteresis Loop



- Isotherm does not close for the Barnett Shale from extremely complex pore network effects
- CO₂ adsorption at 273.15K for micropore (0-2 nm) analysis indicates the presence of some volume of pores at $\sim 0.35-0.7$ nm



Drying CH₃COOK | K₂CO₃ | NaNO₂ | NaCl Na₂SO₄ **NaOH** KC H₂O **CaSO**₄ Wetting **RH (%)** 6.96 22.9 66 75.4 93 99 43.2 84.8 98 P_c (MPa) 38.5 114 56.5 22.6 9.88 1.37 363 202 3.52 Dia. of meniscus 1.45 2.54 5.13 7.55 12.9 29.4 0.80 106 212 curvature (nm)

Water Vapor Sorption with RH Chambers

Capillary Pressure Curve: Hysteresis Loop



NMR Cyroporometry (NMRC)

- Use melting curve to calculate the pore size distribution by Gibbs-Thomson equation
- Measureable pore diameter range:
 ~1 nm to 10 μm
- Sample size: NMR probe/tube 2.5 mm dia. × 12 mm (30 to 300 mg)
- Measurement
 time: a few hrs to
 >24 hrs

Pore Size Distribution: Method Comparison

(NMRC data from Beau Webber, University of Kent)



Multiple Approaches to Studying Pore Structure

- Pore-scale network modeling (ISU)
- Imbibition with samples of different shapes (UTA-Gao)
- Edge-accessible porosity
- Liquid and gas diffusion
- Mercury injection porosimetry (UTA-Gao)
- N₂ adsorption isotherm (Saitama Univ.; Quantachrome)
- Water vapor adsorption isotherm
- Nuclear Magnetic Resonance Cryoporometry (Univ. Kent)
- SEM imaging after Wood's metal impregnation (Univ. Hannover; Swiss EPMA)
- Microtomography (high-resolution, synchrotron) (PNNL-EMSL; Swiss Light Source; Univ. Hannover; Saitama Univ.)
- Focused Ion Beam/SEM imaging (PNNL-EMSL)
- Small-Angle Neutron Scattering (SANS) (LANL)

Wood's Metal Intrusion and Imaging

- Wood's metal (50% Bi, 25% Pb, 12.5% Zn, and 12.5% Cd) solidifies below 78°C without shrinking
- Heat the metal slowly (about 1 hr) above the melting point (120-150°C)
 Dultz at al. (2006)
- Inject molten metal into the connected pore spaces under high pressure; sample size (up to 5 mm dia. and 15 mm long)
- Image metal distribution in polished sections 150 µm thick



Kaufmann (2010)

Fig. 1. Apparatus Wood's metal intrusion.

600 bars used (invade 20 nm)

Wood's metal injection

Stefan Dultz (University of Hannover)

Sig

Det

Spot

15.0 kV

HFW

4.0 | SSD | BSE | 2.56 mm | 10.0 mm | 100x |

WD

Pressure

Mag

Image porosity: 21.4% Measured porosity: 21.0±0.147

sandstone

Berea

·──500.0µm── Berea Sandstone 600 bars used (invade 20 nm)

Wood's metal injection

Stefan Dultz (University of Hannover)

Natural fractures

Natural fracture (porosity: 1.85%)

pyrite framboids (2-10%)

Eagle Ford shale

HVSpotDetSigHFWWDMagPressure15.0 kV4.0SSDBSE0.51 mm9.9 mm500x---

1

<u>—100.0µm</u>— Shale

1,542 bars used (invade 9 nm in pore dia.) by Josef Kaufmann of EPMA Wood's metal injection

SEM-BSE by Stefan Dultz (University of Hannover)

Wood's metal roccupied crack and matrix pores connected to the sample surface

Barnett Shale 7,169 ft

Wood's metal accumulation at

the surface

Sig Spot Det HFW 20.0 kV 4.0 SSD BSE 1.02 mm 12.4 mm 250x

HV

WD Mag Pressure ·300.0µm 48.4 kN

Nano-Scale FIB-SEM Imaging



Nano-Scale FIB-SEM Imaging



Small-Angle Neutron Scattering (SANS)

- Developed and refined over the past 2 decades for structural characterization of various natural and engineered porous materials
- Non-destructive nature
- Record the scattering from all pores (connected and closed); closed pores are inaccessible to fluids and, therefore, immeasurable by other techniques
- Have the ability to investigate <u>pore structure at realistic (reservoir) P-T</u> <u>conditions</u> and changes in pore structure at variable P-T conditions
- BT-5 perfect crystal USANS at NIST Center for Neutron Research (NCNR); General-Purpose SANS instrument at Oak Ridge National Lab (ORNL); The Lujan Neutron Scattering Center at Los Alamos National Lab (LANL)
- Measurable pore diameter range: 0.5 to 200 nm (for SANS) and $\sim 10 \ \mu m$ (for ultra SANS or USANS)



Low gas		
_recovery		
in Barnett		Doro
Shale Question	Hypothesis	geometry and topology
Analysis	Complementa	ry
and	measurements	;
evaluation	Theory Mode	ling
Ongoing work: CH ₄ re crushed and inta	tention and transport ir act Barnett Shale	

Summary

- Steep 1st year decline and low overall hydrocarbon production observed in stimulated shales
- Shales show low pore connectivity, which reduces gas diffusion from matrix to stimulated fractured network
- Several complementary approaches are used to investigate pore structure in tight shales
 - Imbibition and diffusion: macroscopic method
 - Porosimetry and vapor condensation: indirect method
 - Imaging (Wood's metal, FIB/SEM, SANS): nanoscale tool

RPSEA Project Number: 09122-12

Funding for this project is provided by RPSEA through the "Ultra-Deepwater and Unconventional Natural Gas and Other Petroleum Resources" program authorized by the U.S. Energy Policy Act of 2005. RPSEA (<u>www.rpsea.org</u>) is a nonprofit corporation whose mission is to provide a stewardship role in ensuring the focused research, development and deployment of safe and environmentally responsible technology that can effectively deliver hydrocarbons from domestic resources to the citizens of the United States. RPSEA, operating as a consortium of premier U.S. energy research universities, industry, and independent research organizations, manages the program under a contract with the U.S. Department of Energy's National Energy Technology Laboratory.