



SOVEREIGN CONSULTING INC.  
AN ENVIRONMENTAL SERVICES FIRM

# How Could the Gold King Mine Water be Passively Treated?

*Jim Gusek, P.E.  
Sovereign Consulting Inc.  
Lakewood, Colorado*

GEOLOGICAL SOCIETY OF AMERICA  
ANNUAL MEETING – DENVER, COLORADO  
GOLD KING MINE SPECIAL SESSION

# Outline

- Gold King Mine Thumbnail Sketch
- Passive Treatment Biogeochemistry
- Two Gold King Passive Treatment Design Concepts



# The Gold King Site

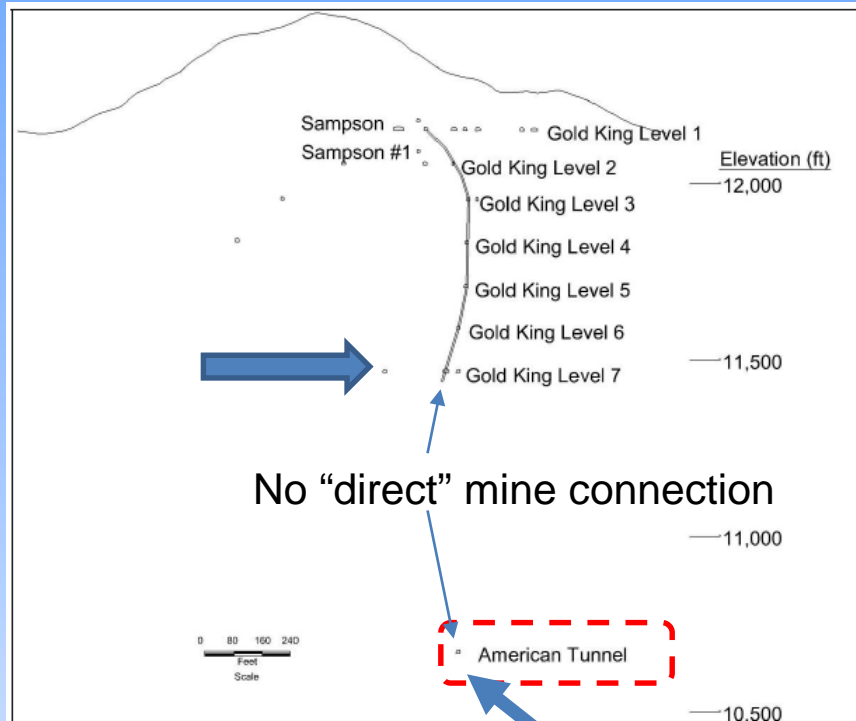
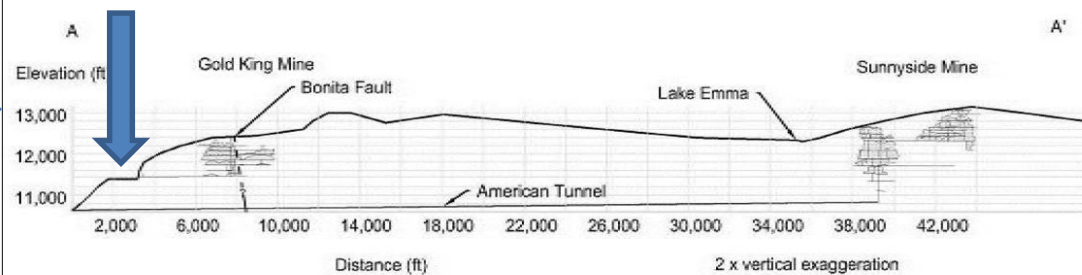


Figure 53.—Photograph showing the blowout at 10:56 a.m. on August 5, 2015 (photograph from EPA project files).

*US Bureau of Reclamation, 2015*



← 2 miles →

***American Tunnel Profile***



# “Natural” Attenuation

***Mother Nature*** is pretty talented; to remediate heavy metals situations, ***She*** uses:

***S***equential

***E***cological

***eX***traction

processes that have evolved over millennia (Thanks, *C. Darwin*)



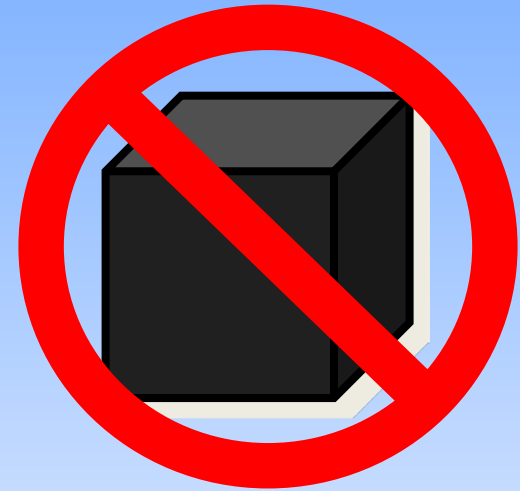
# What Is the Passive Treatment Process?

Passive Treatment of MIW  
involves the:

***S**equential*

***E**cological*

*e**X**traction*



Of metals in a man-made but  
naturalistic bio-system



# Iron Stromatolites & Fe/Mn Nodules

## Shark Bay, Australia



Stromatolites built by cyanobacteria/algae, a process over 1 billion yrs. old



## Fe/Mn Fossil Nodules

Courtesy of Nick Shearer, WV DEP





# Ferricrete Deposits

## Animas Basin, Colo.



Courtesy of USGS

Deposit of iron-cemented stream gravel (ferricrete) with embedded wood fragments



Courtesy of USGS

Natural Iron-rich Acidic Spring Flowing into Cement Creek



# Manganocrete Deposits

## Animas Basin, CO & Patagonia, AZ



Alluvial manganocrete  
near the former Lake  
Emma in Eureka Gulch



MnO<sub>2</sub>-cemented  
alluvium in Alum Creek





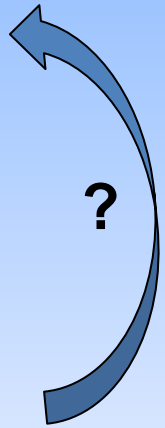
# Natural Metal Removal Mechanisms

## Major

- Sulfide and carbonate precipitation via sulfate reducing bacteria, et al.
- Hydroxide and oxide precipitation by *acidithiobacillus ferro-oxidans* bacteria, et al.
- Filtering of suspended materials and precipitates
- Carbonate dissolution/replacement
- Metal uptake into live roots, stems and leaves

## Minor

- Adsorption and exchange with plant, soil and other biological materials



# Periodic Table of Passive Treatment

1																		18			
1 H	2															13	14	15	16	17	He
Li	4 Be															B	6 C	7 N	8 O	9 F	Ne
11 Na	12 Mg	3	4	5	6	7	8	9	10	11	12	13 Al	14 Si	15 P	16 S	17 Cl	Ar				
19 K	20 Ca	Sc	Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	Ga	Ge	33 As	34 Se	Br	Kr				
Rb	Sr	Y	Zr	Nb	42 Mo	Tc	Ru	Rh	Pd	47 Ag	48 Cd	In	Sn	51 Sb	Te	I	Xe				
Cs	56 Ba	La*	Hf	Ta	W	Re	Os	Ir	Pt	79 Au	80 Hg	81 Tl	82 Pb	Bi	Po	At	Rn				
Fr	88 Ra	Ac~	Rf	Db	Sg	Bh	Hs	Mt	---	---	---										

Actinide Series

92  
U

**LEGEND**

Red - passive untreatable

Blue – anaerobic (BCR)

Orange – oxidizing (Aerobic Cell)

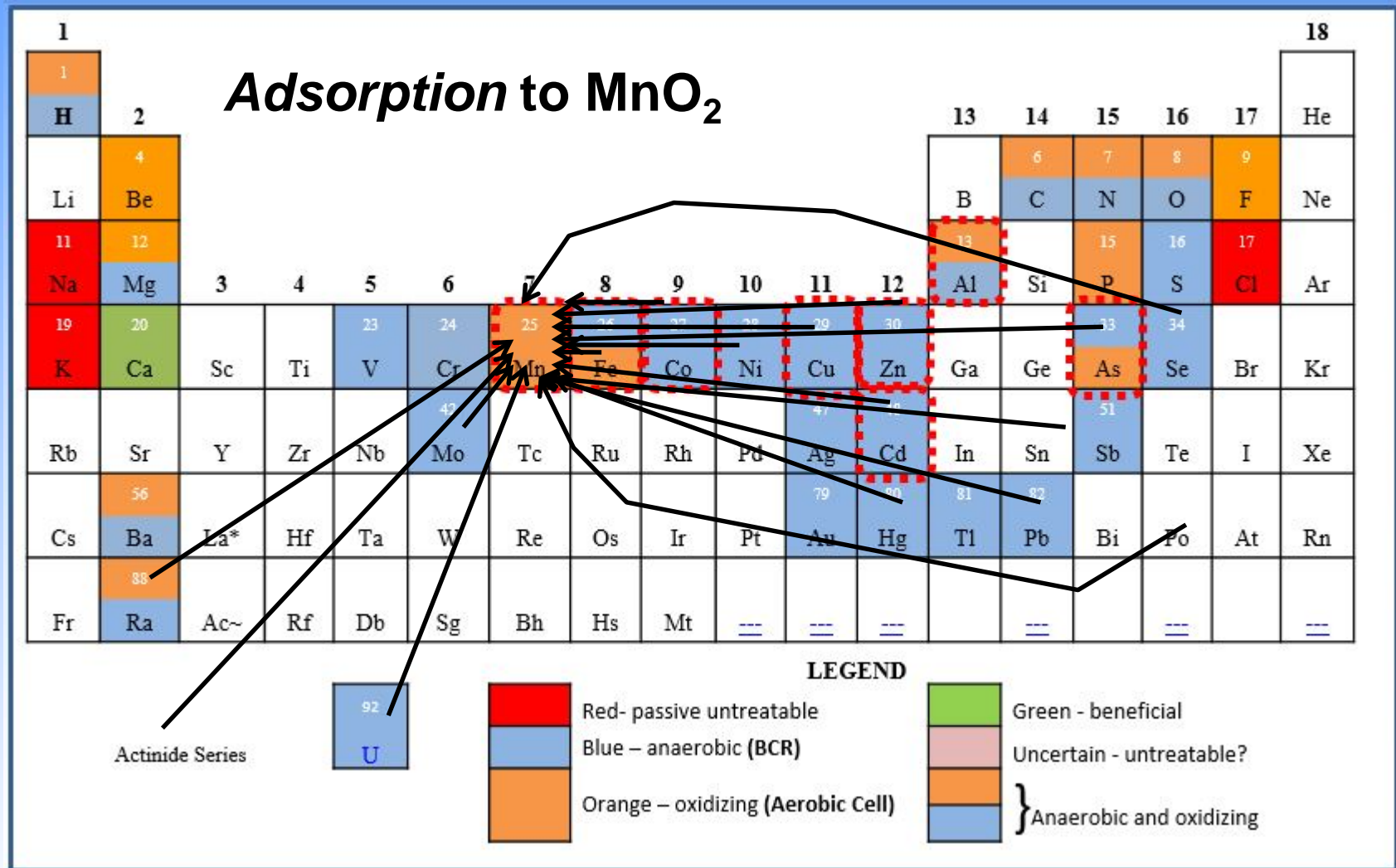
Green - beneficial

Uncertain - untreatable?

Anaerobic and oxidizing



# Periodic Table of Passive Treatment Re-Visited

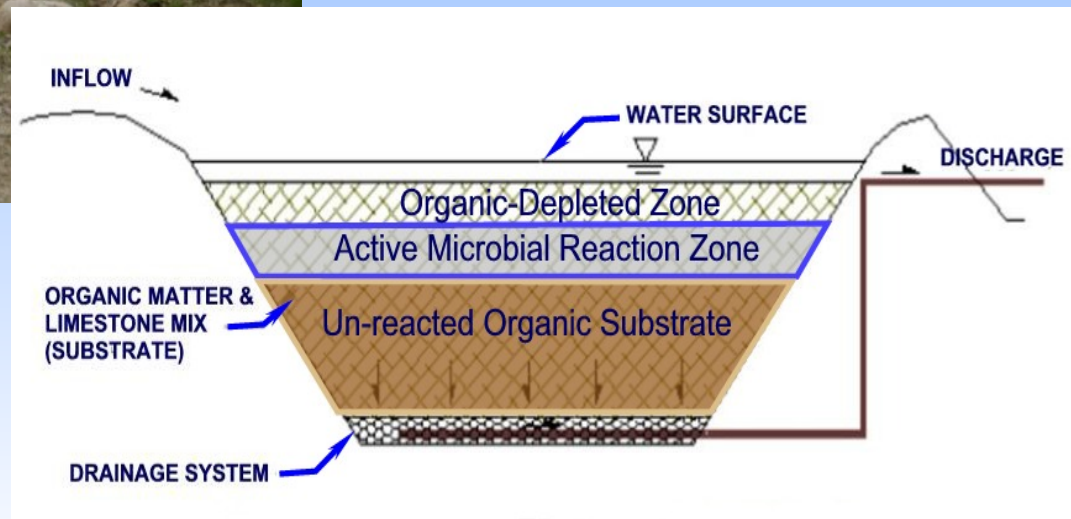


# Anaerobic Biochemical Reactors (BCRs)



Aluminum and heavy metal removal, selenium removal, de-nitrification, pH adjustment, alkalinity & hardness addition

AKA  
Vertical Flow Reactors  
or  
Sulfate Reducing  
Bioreactors (SRBRs)





# Aerobic Cells



**Fe, As,  
Biochemical  
Oxygen Demand  
(BOD), and Mn  
removal  
(plus adsorbed  
metals)**



**AKA Rock Filters**





# Iron Terraces – Aerobic Special Case

## Modern Aerobic Wetlands & Ferricrete/Iron Terraces

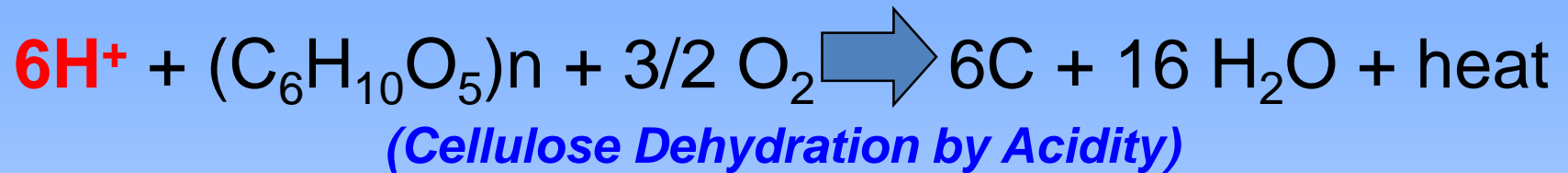


Some ferricrete deposits in the Animas Basin are 9,000 yrs. old!





## IRON TERRACE REACTIONS

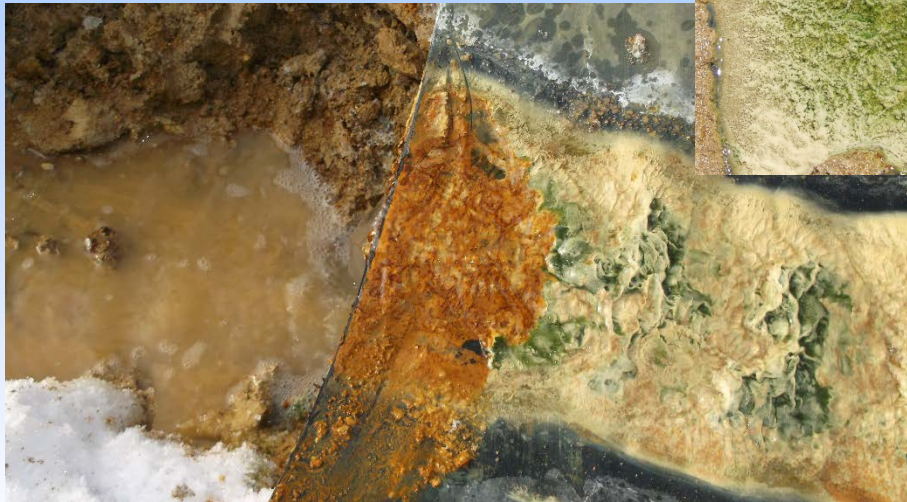


Reaction consumes  $\text{H}^+$   
(acid) & pH rises and iron  
or aluminum can drop out

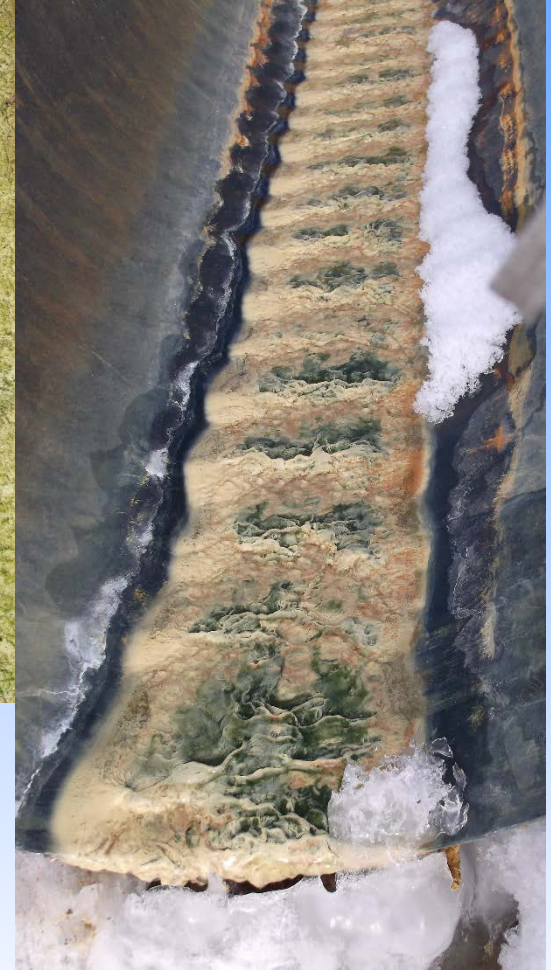




# Volunteer Aluminum Terrace Deposition Idaho



pH 2.5;  
Al 800 mg/L  
Fe 500 mg/L





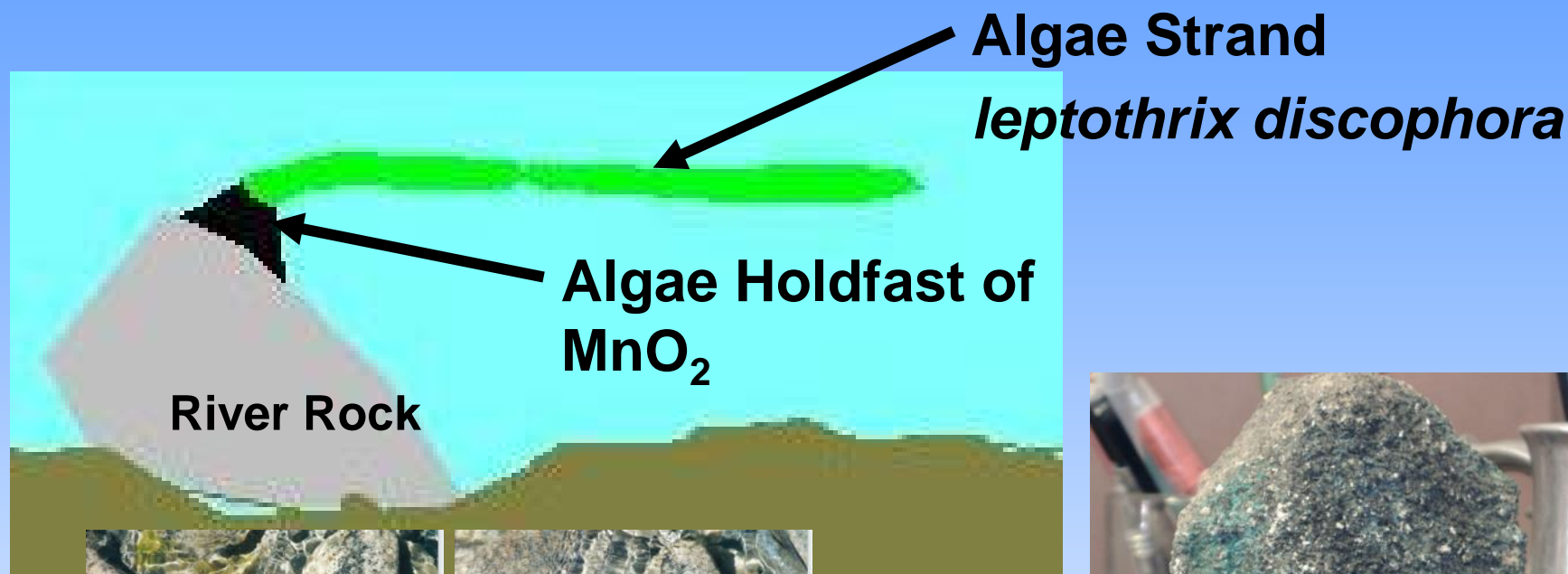
# Compare to Red and Bonita Mine Portal



US Bureau of Reclamation, 2015



# Manganese Oxidation at Neutral pH

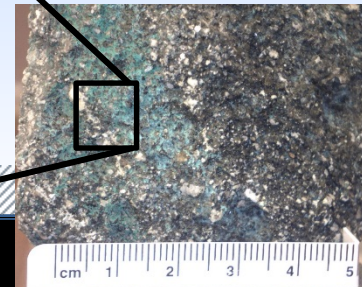
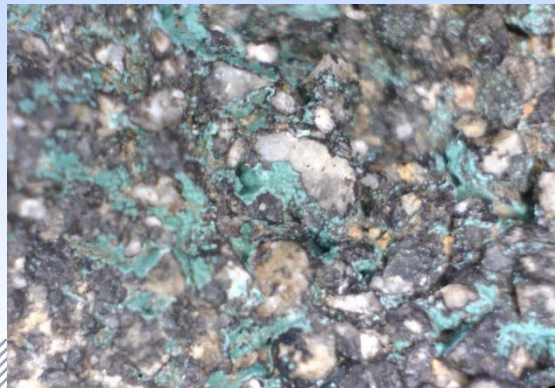
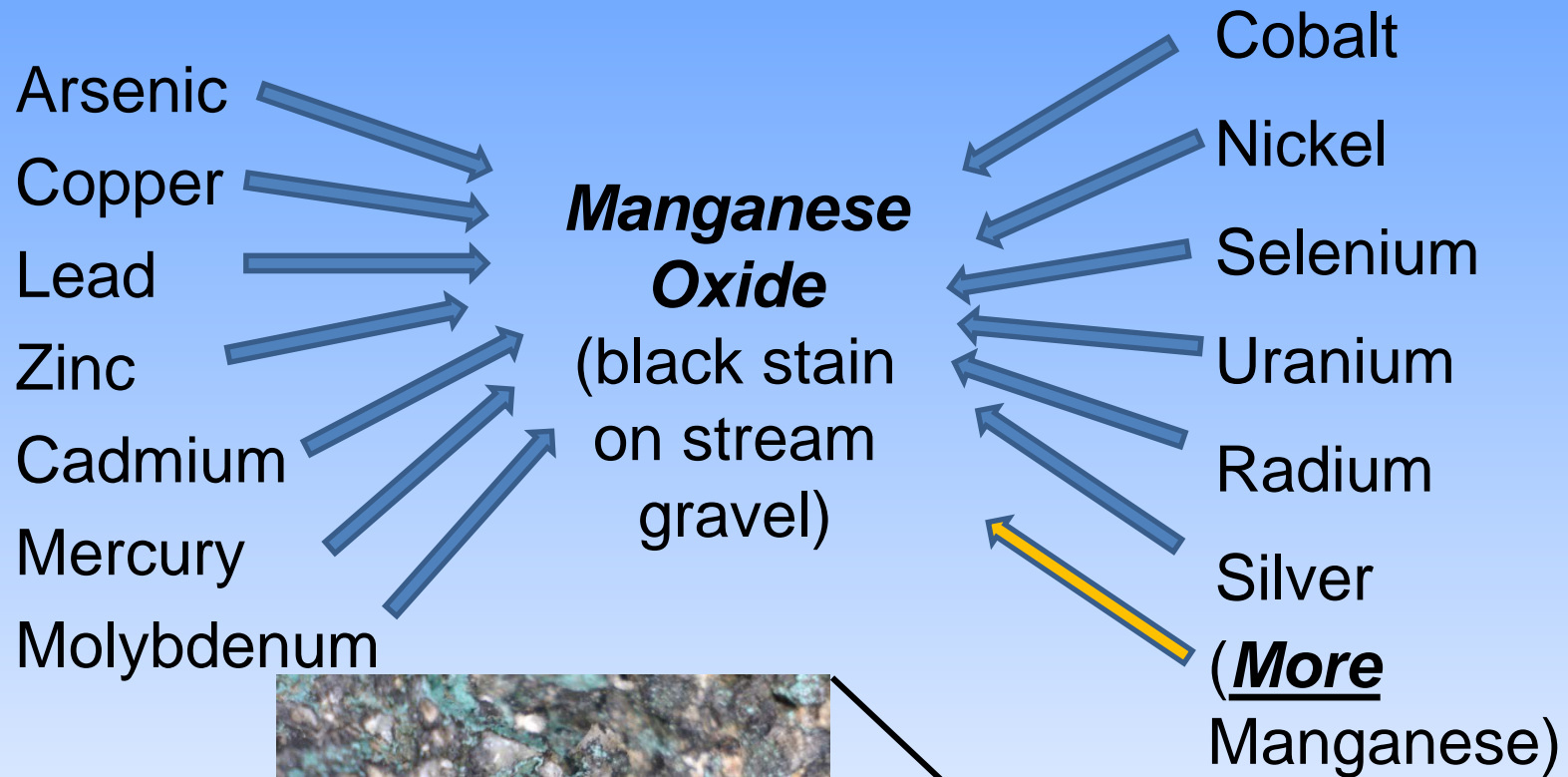


Manganocrete from paleo-channel near Prescott, AZ

Manganese/algae in outfall from Leadville Colorado (El. 10,000ft/3,050m) WTP in March



# Why is *manganese* removal so important?





# The Gold King PTS Design Conditions

Parameter	Value	Units	Assumed Condition
Flow	300	gpm	Spring freshet
Flow	200	gpm	Steady state
pH	3.0	S.U.	Steady state
Aluminum	25	mg/L	Steady state
Arsenic	22	µg/L	Steady state
Iron	126	mg/L	Steady state
Cadmium	75	µg/L	Steady state
Copper	6.0	mg/L	Steady state
Cobalt	111	µg/L	Steady state
Manganese	35	mg/L	Steady state
Sulfate	1,760	mg/L	Steady state
Zinc	26	mg/L	Steady state

Chemistry from Aug-Sept 2015

US EPA, 2016

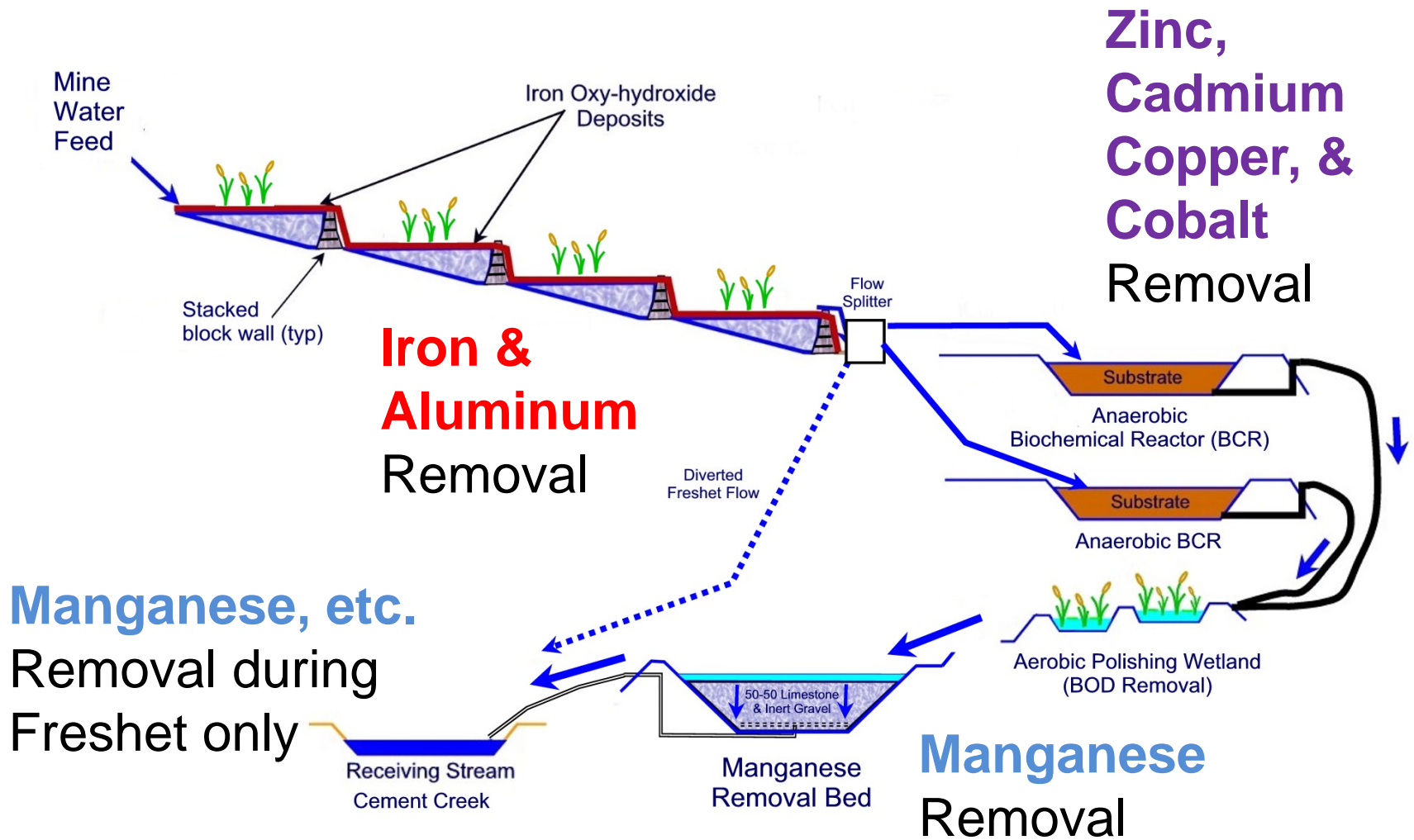




# Gold King Conceptual PTS Design #1

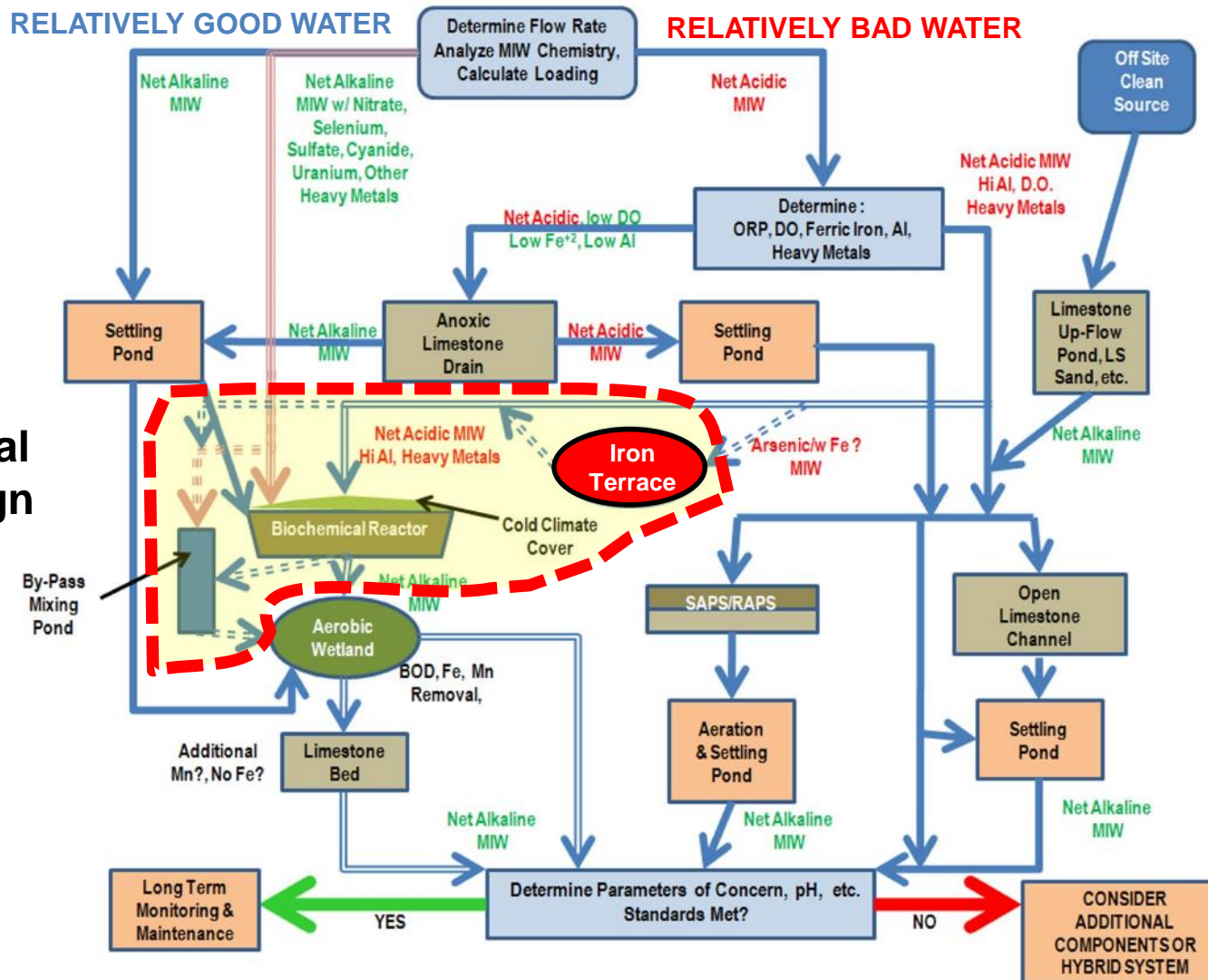


# Conceptual Gold King PTS #1

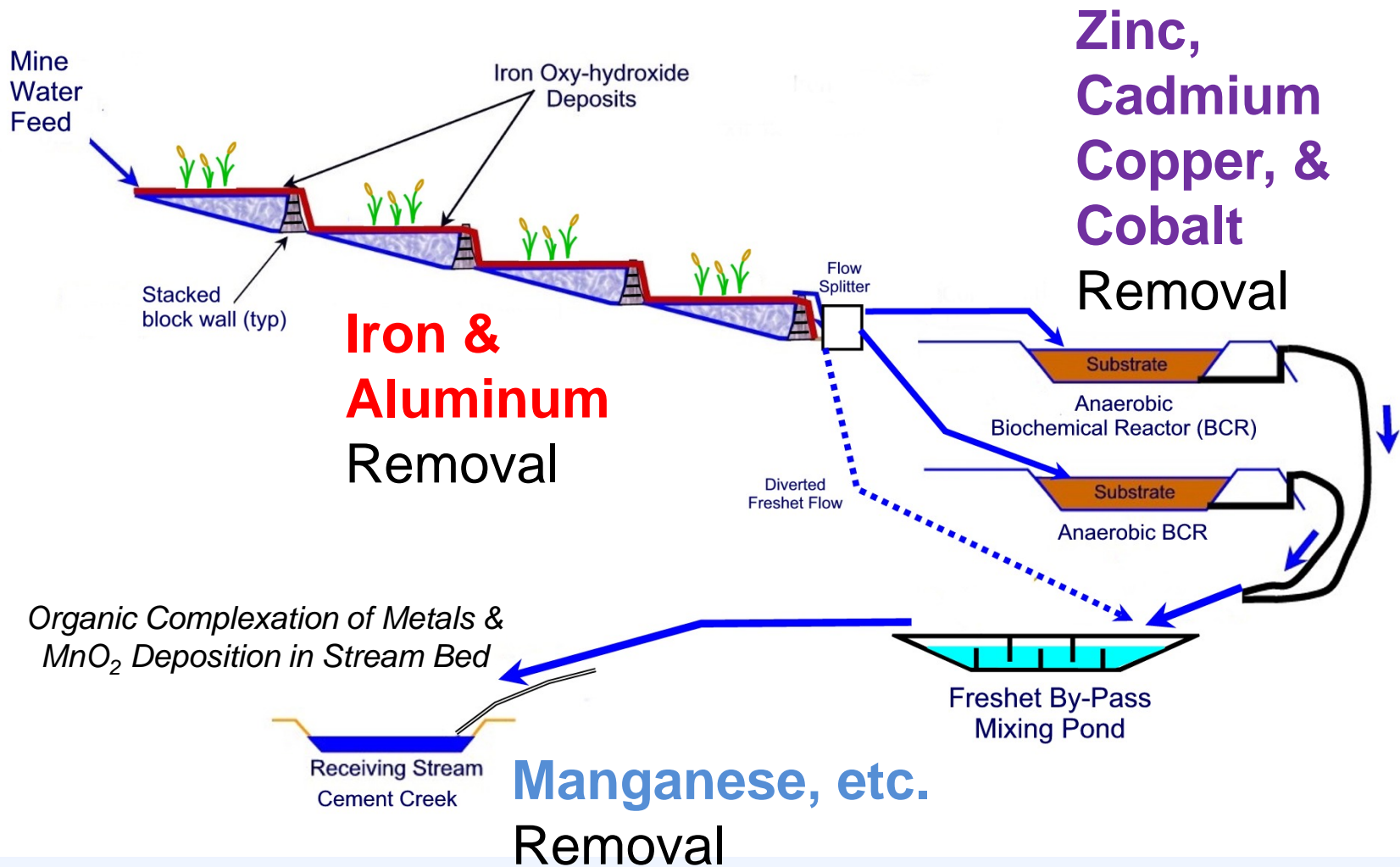


# Passive Treatment Decision Tree 2016

## Gold King Conceptual PTS Design #2

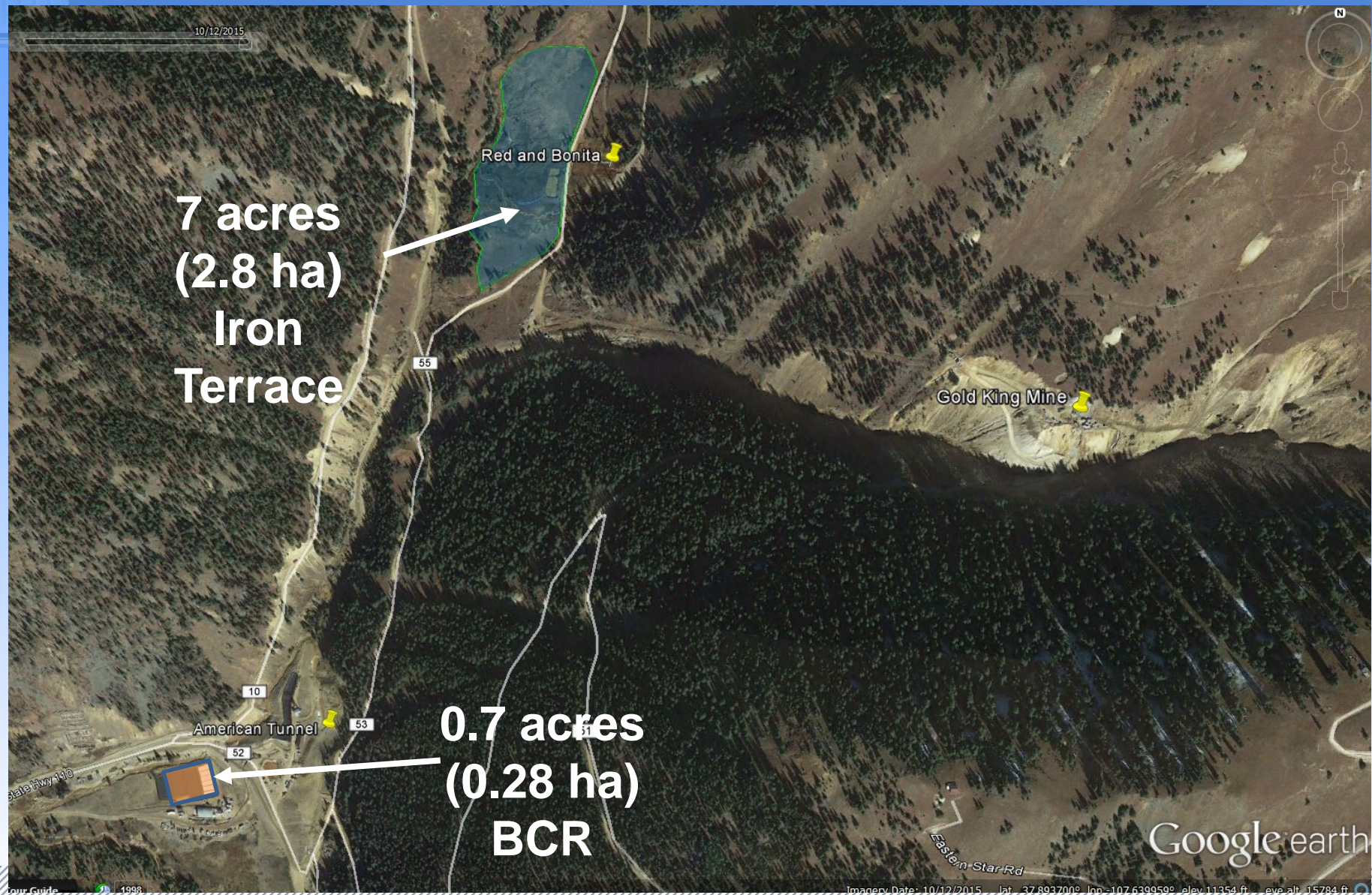


# Conceptual Gold King PTS #2





# Gold King PTS Sites?



## Tailings and Mine Waste Conference, Keystone Colorado Oct. 3-5, 2016

### *A Pathway to Walk-Away? – 30 Year Old Technology to Suppress Acid Rock Drainage Revisited*

- Use of bactericides to suppress ARD
- Seven case history successes
- Merging old and new technologies



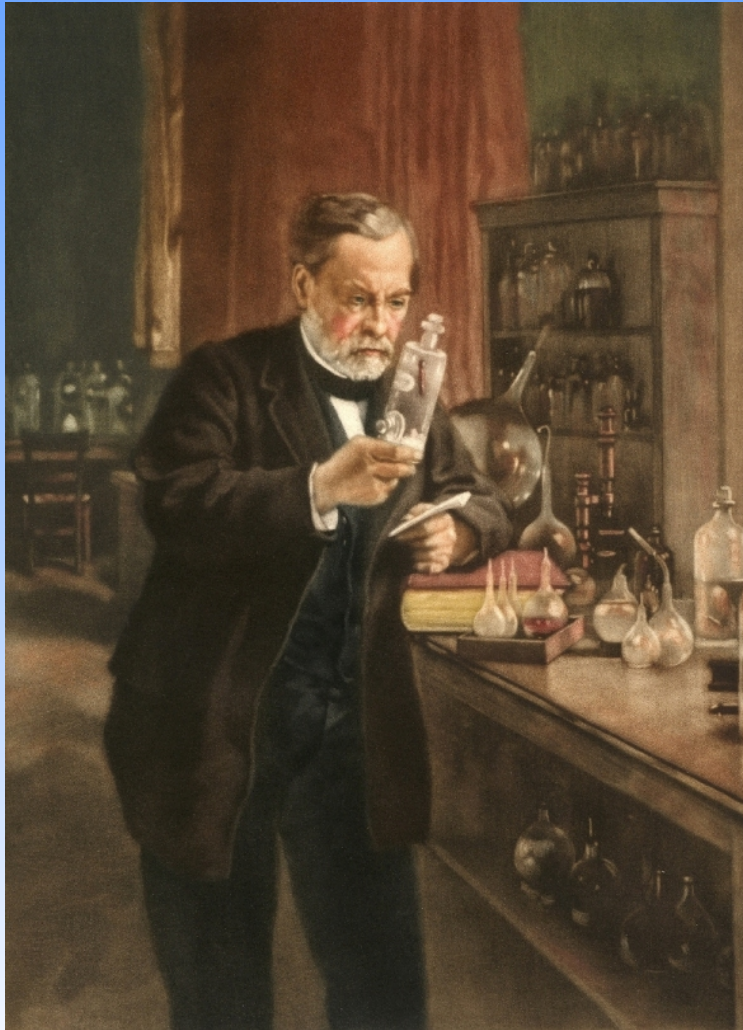


# REVIEW – Pathway to Walk-Away

- 1. Primary Source Control** to minimize flow, metals concentrations, and loading
- 2. Reclamation/Remediation** to sustain primary source control measures for the long term
- 3. Passively Treat** residual conditions
  - pH 2.5 to 8.5
  - Metals (**Fe**, **Cu**, Pb, **Zn**, **Cd**, Cr, **Mn**, Hg, Mo, **Al**, Se, **As**, U, **Co**, Tl)
  - Non-metals (CN, **SO<sub>4</sub>**, NO<sub>3</sub>, NH<sub>3</sub>, BOD<sub>5</sub>, P)
  - Temperatures (0 to 30 deg C)
  - Major processes are:
    - Chemical precipitation (usually facilitated by bugs) in aerobic and anaerobic conditions
    - Adsorption to MnO<sub>2</sub>, etc. (facilitated by algae)



# Thank You



***“In the fields of observation,  
chance favors only the  
prepared mind.”***

*L. Pasteur*

[jgusek@sovcon.com](mailto:jgusek@sovcon.com)

