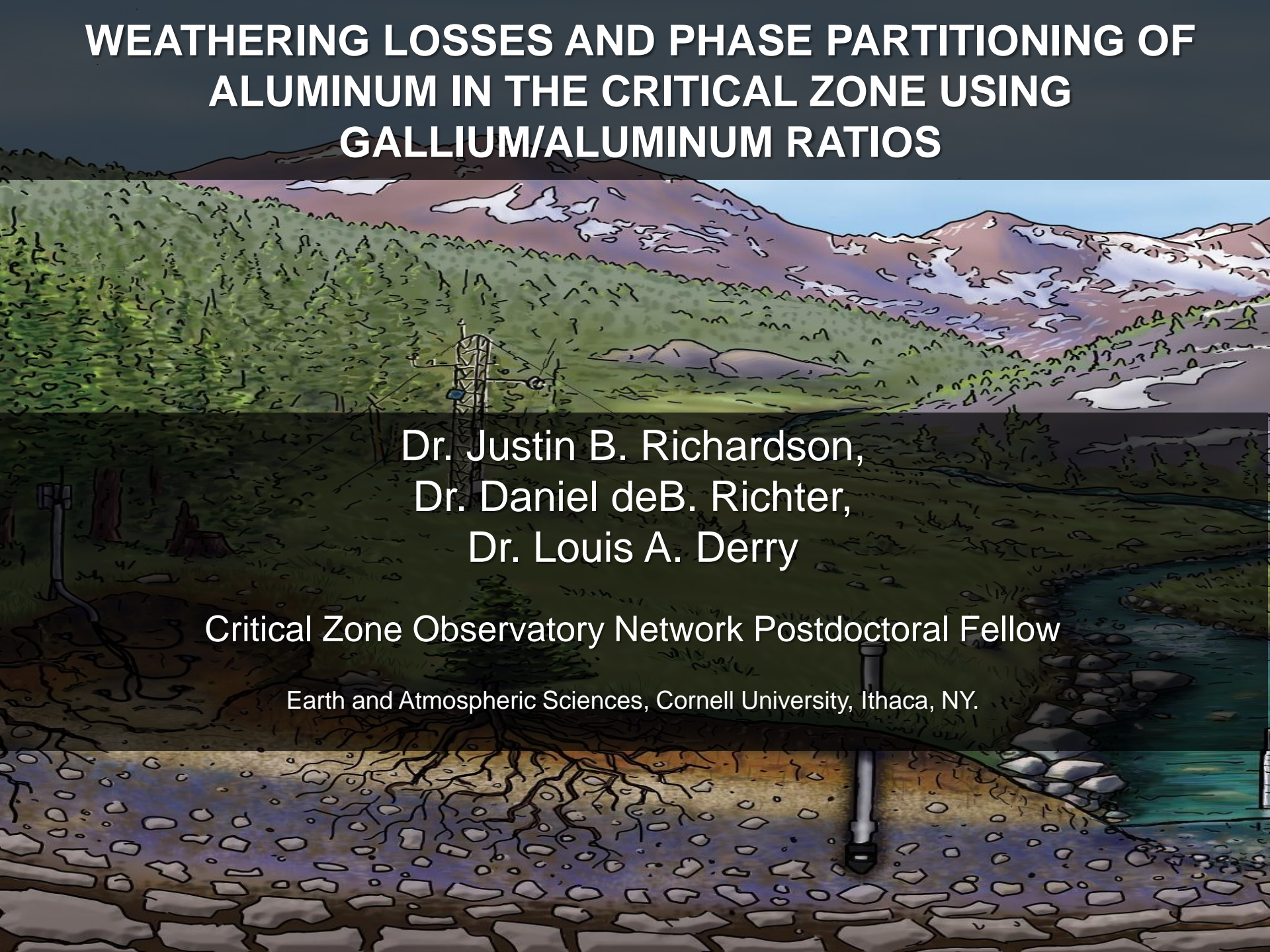


WEATHERING LOSSES AND PHASE PARTITIONING OF ALUMINUM IN THE CRITICAL ZONE USING GALLIUM/ALUMINUM RATIOS

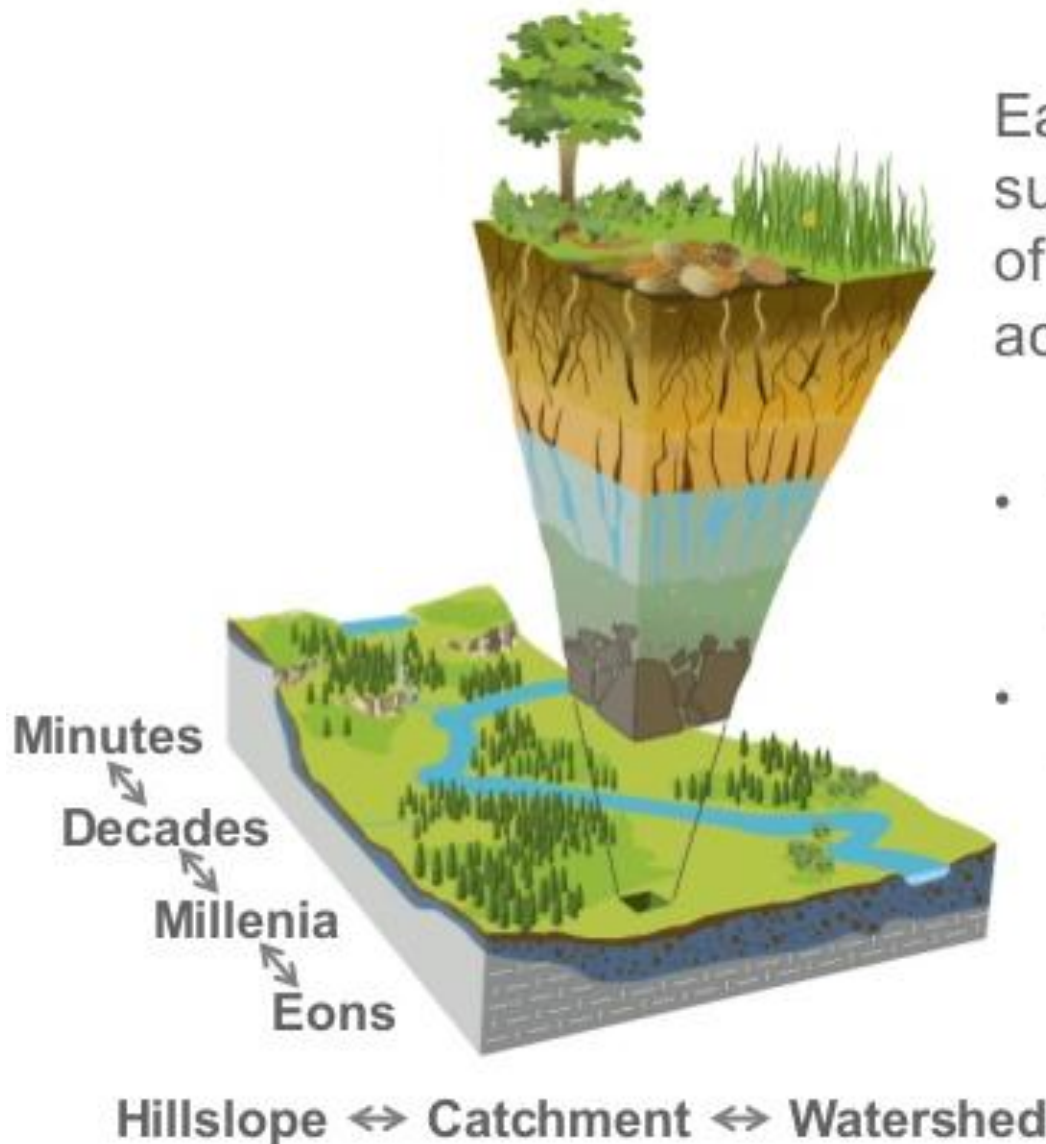


Dr. Justin B. Richardson,
Dr. Daniel deB. Richter,
Dr. Louis A. Derry

Critical Zone Observatory Network Postdoctoral Fellow

Earth and Atmospheric Sciences, Cornell University, Ithaca, NY.

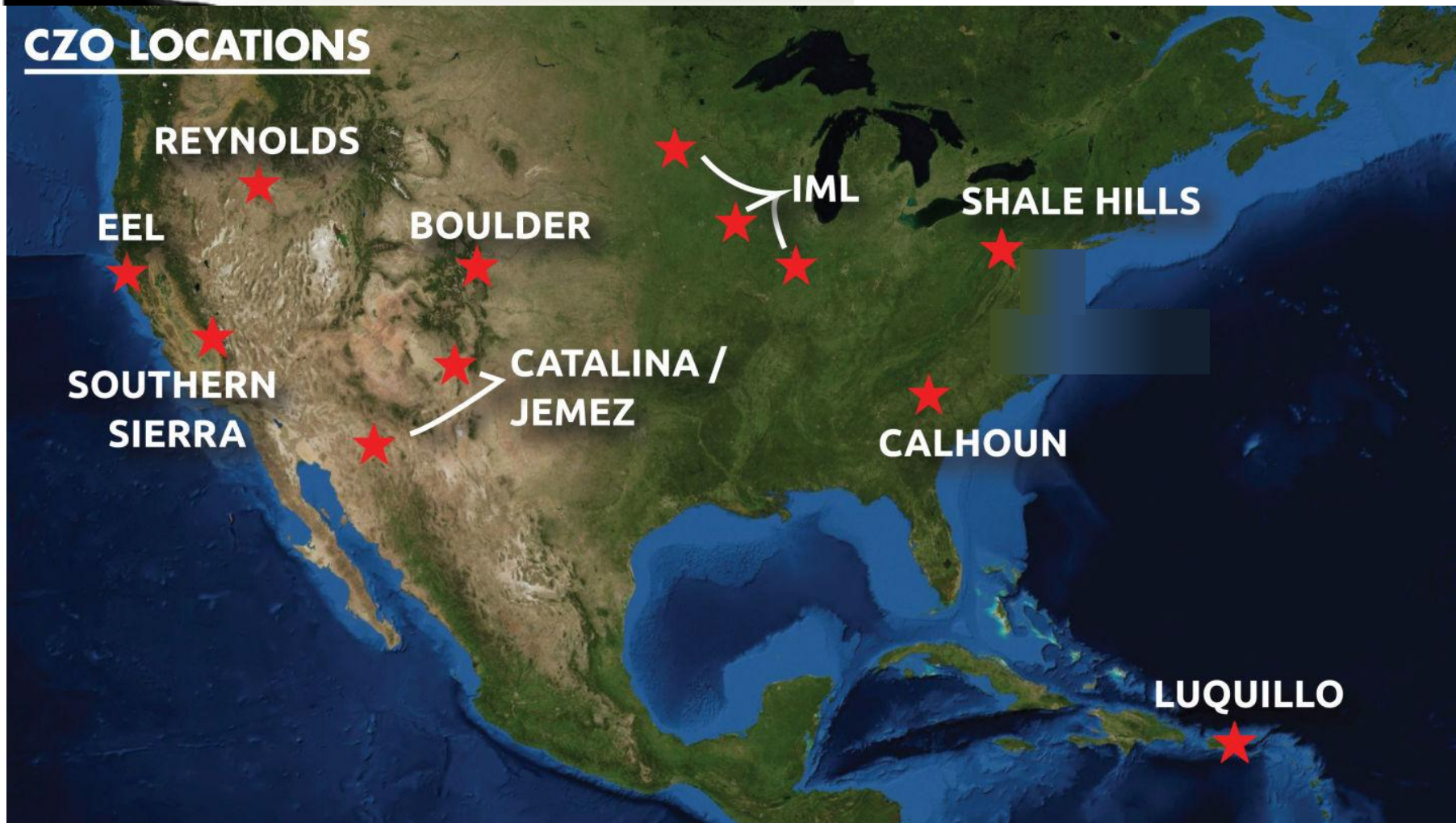
Critical Zone Science



Earth's permeable near-surface layer from the tops of the trees to the bottom of actively cycling groundwater.

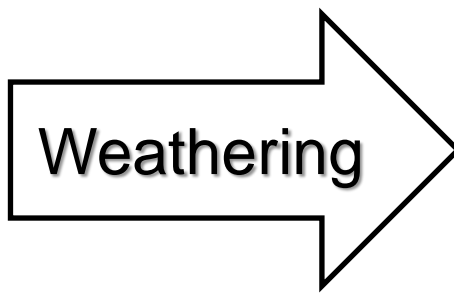
- Where rock, soil, water, air, and living organisms interact and shape the Earth's surface.
- Critical to sustaining the earth's sustaining services
 - Clean water
 - Productive soil
 - Balanced atmosphere

CZO LOCATIONS





**Unweathered
bedrock**



**Aluminum
retention or
mobilization**



**Weathered
soil**



**Al can be harmful
to plants and
organisms**



**Al causes acidity
in soils and water**

**Important for
secondary
mineral formation**



Developing the Ga/Al pseudo-isotope system

$$\delta^{13}\text{C}_{\text{Sample}} = \left\{ \begin{array}{l} \left(\frac{^{13}\text{C}}{^{12}\text{C}} \right)_{\text{Sample}} \\ \left(\frac{^{13}\text{C}}{^{12}\text{C}} \right)_{\text{Reference}} \end{array} \right.$$

**Aluminum is
monoisotopic**

^{27}Al

$^{69,71}\text{Ga}$

To develop a low temperature pseudo-isotope system:

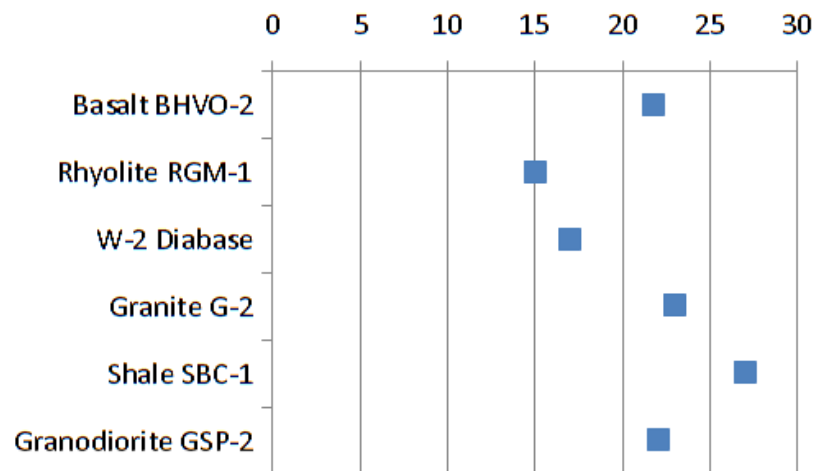
- (1) The elements must have similar geochemical properties.
- (2) Sources must have known ratios to quantify their respective contributions.
- (3) Processes that affect the element ratio (fractionation) must be known.

(1) Ga and Al have similar geochemical properties

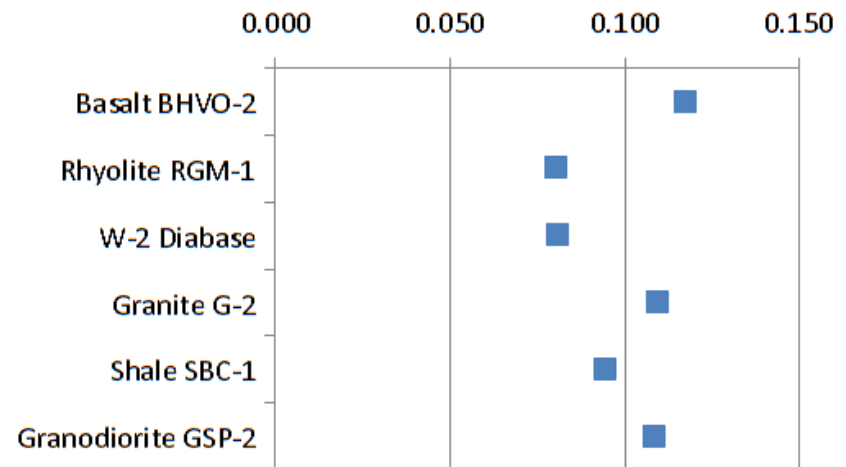
	Gallium	Aluminum
Charge	+3	+3
Isotopes	^{69}Ga , ^{71}Ga	^{27}Al
Atomic radii	62 pm	54 pm
Electronegativity	2.01	1.61
Redox sensitivity	None	None

(2) Gallium and Aluminum have known sources

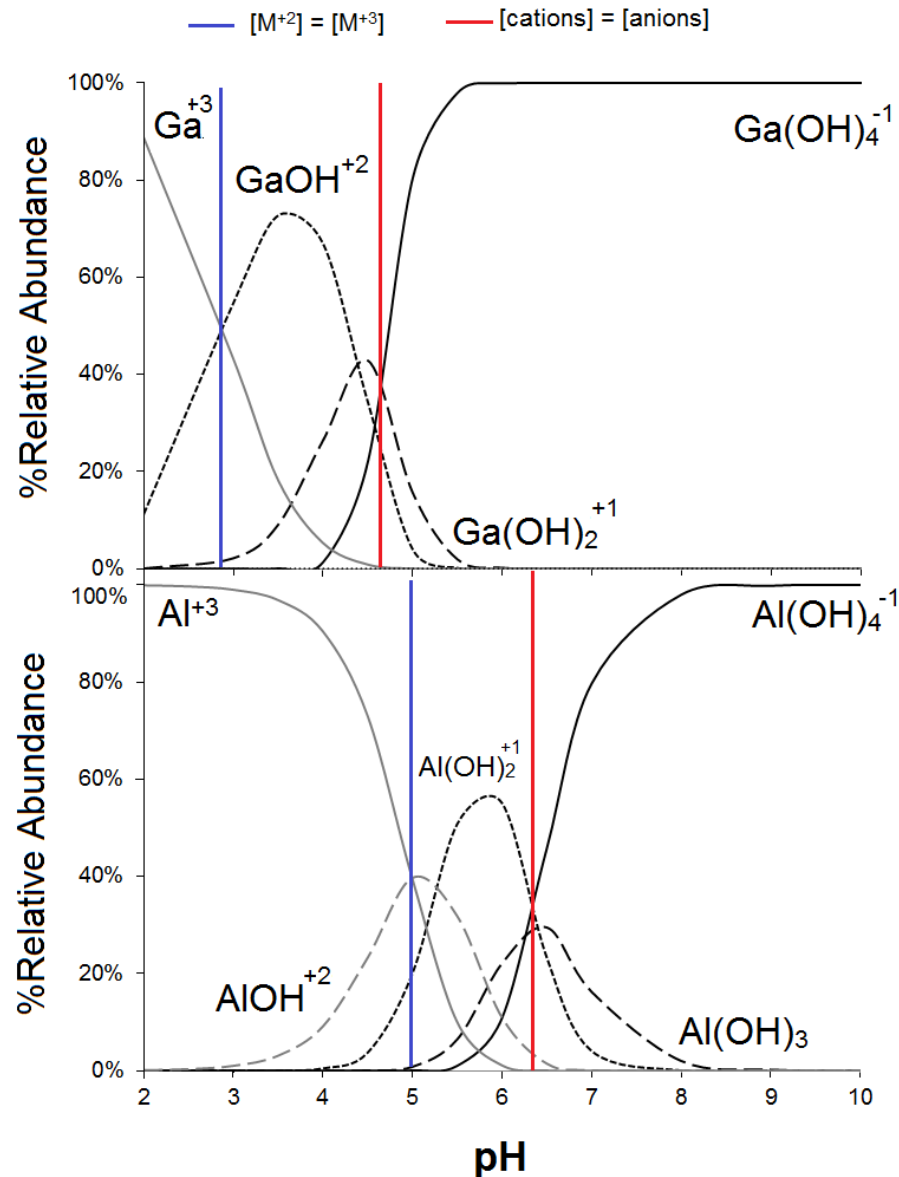
USGS Rock Standard Gallium
($\mu\text{g g}^{-1}$)



USGS Rock Standard Ga/Al
ratio (mmol mol^{-1})

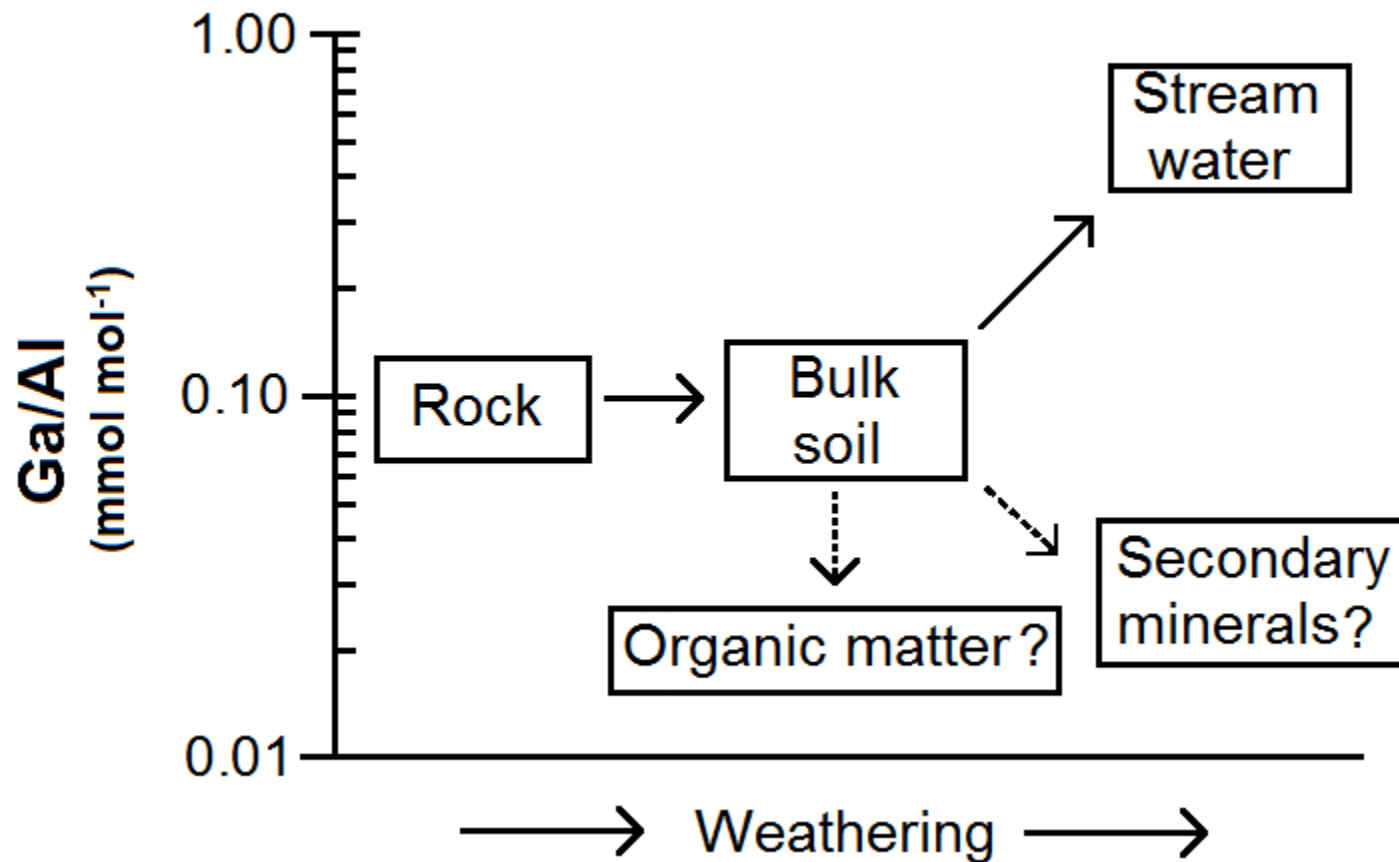


(3) Ga and Al must have known fractionation processes



Visual MINTEQ

(3) Ga and Al must have known fractionation processes



Based on data from Shiller and Frilot (1996)

Question

- Are sorption/precipitation reactions affecting Ga/Al ratio in Critical Zone during weathering?

Hypothesis

- Less strongly bound fractions will have a very different Ga/Al ratio due to sorption.

Boulder Creek CZO

Granitic

MAT: 5 °C MAP: 519 mm/yr

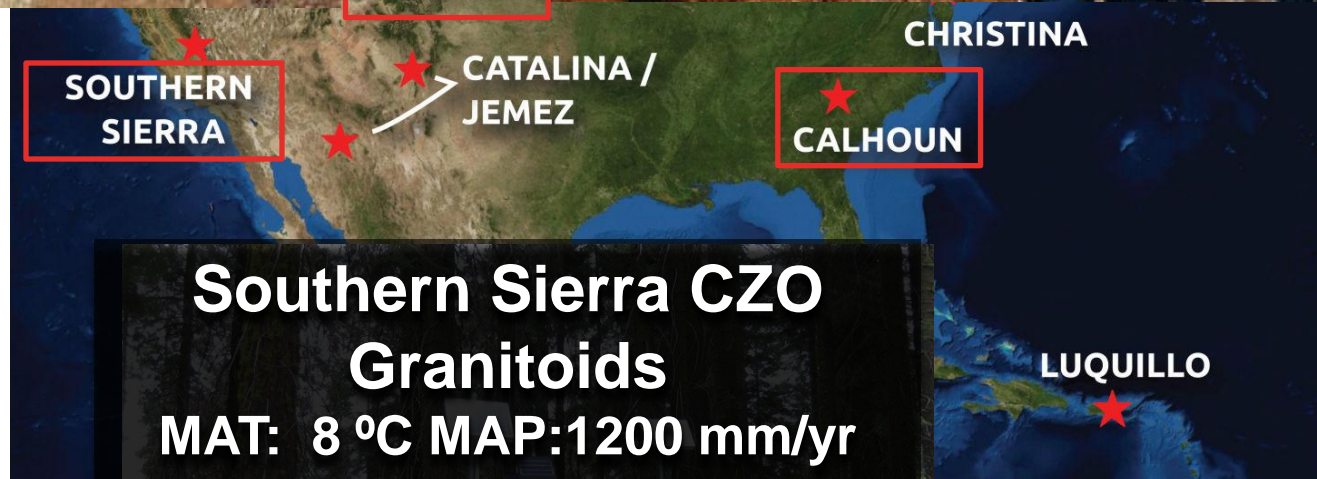


BOULDER

Calhoun CZO

Granitoids

MAT: 16°C MAP: 1250 mm/yr



Southern Sierra CZO

Granitoids

MAT: 8 °C MAP: 1200 mm/yr



Sequential extractions

**Organic matter/
extractable phase
(H_2O_2 + Acetic acid)**



**Oxide extraction
(Citrate-
biocarbonate-
dithionite)**



**Total digestion/
residual fraction
(HF + HClO_4)**

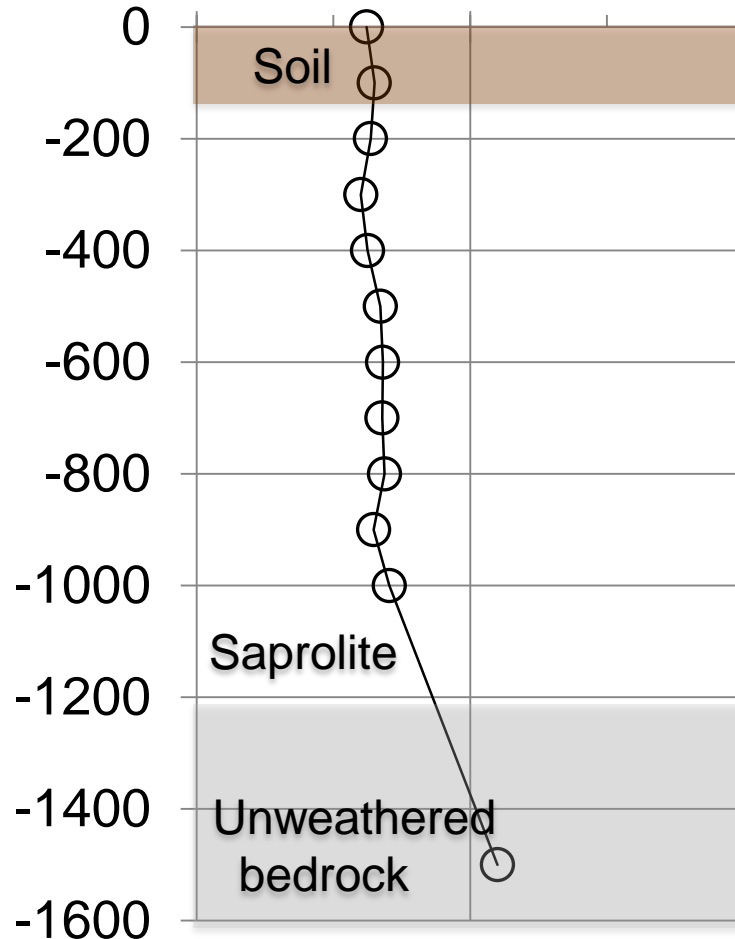


Southern Sierra CZO

Ga/Al (mmol mol⁻¹)

0.00 0.10 0.20

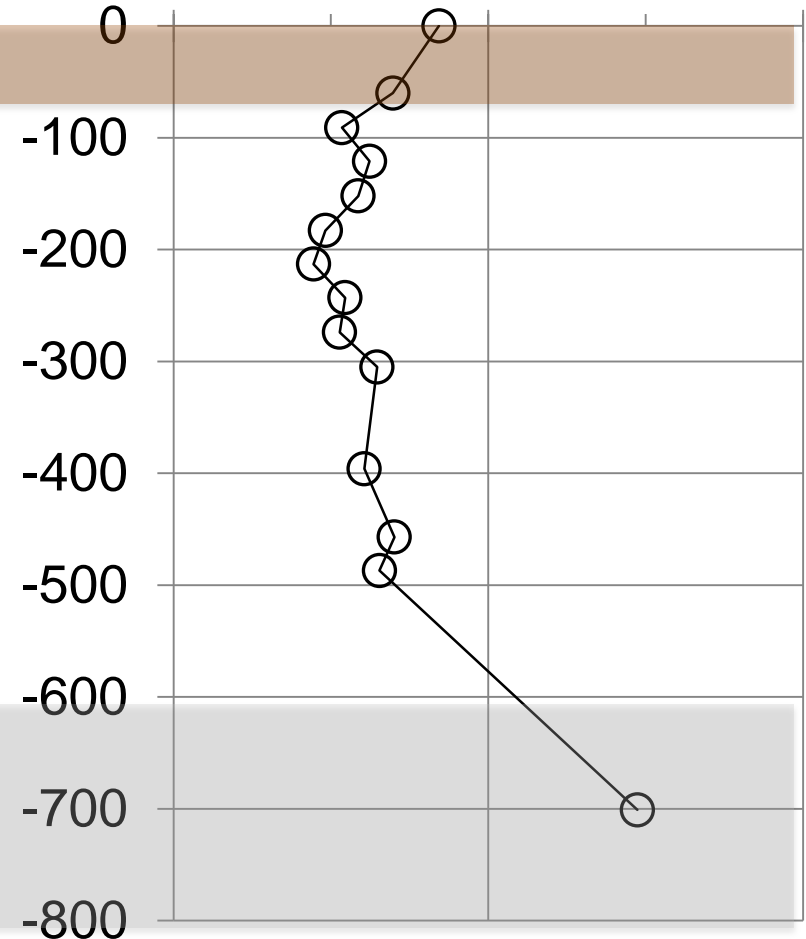
Depth (cm)

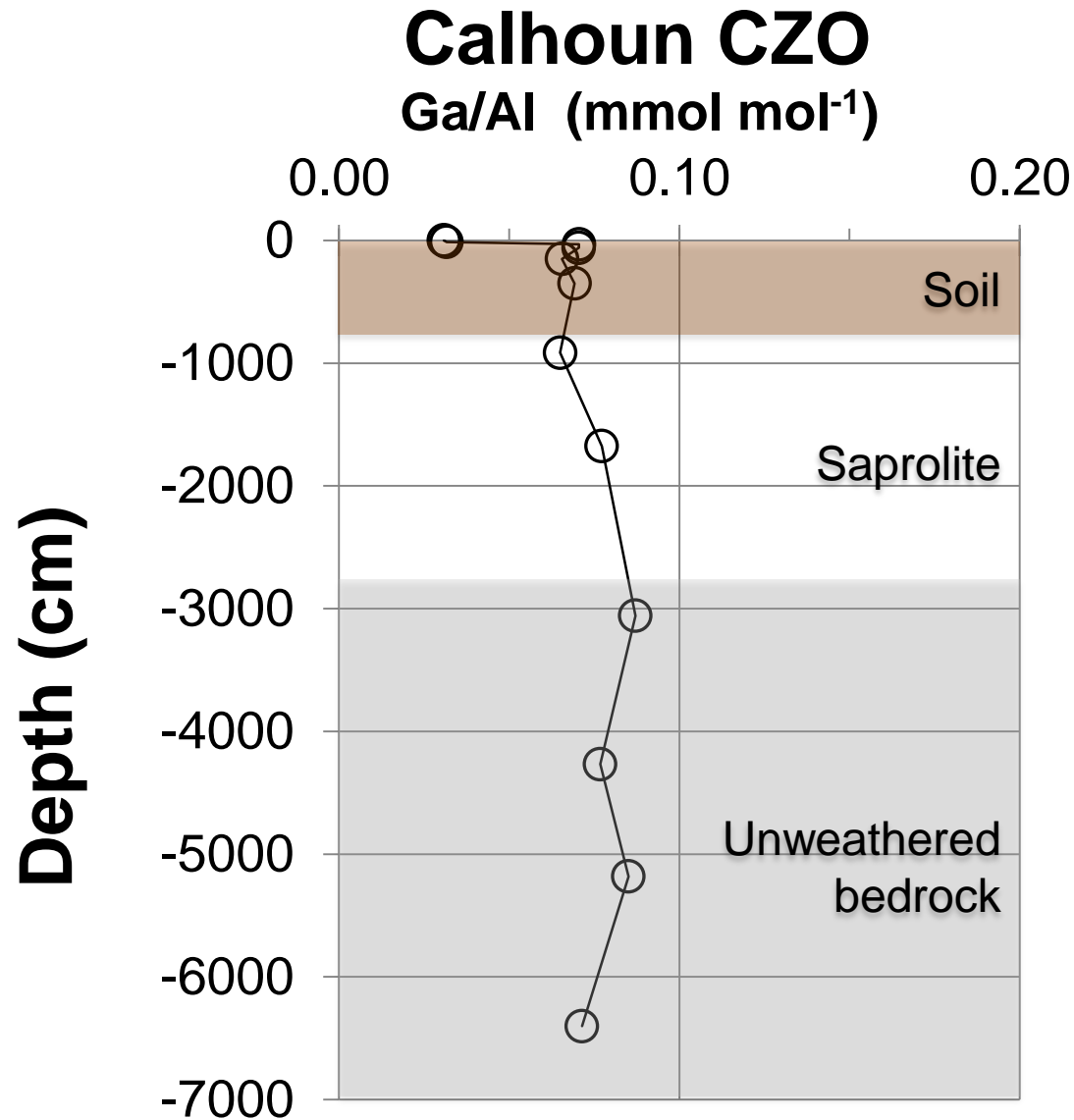


Boulder Creek CZO

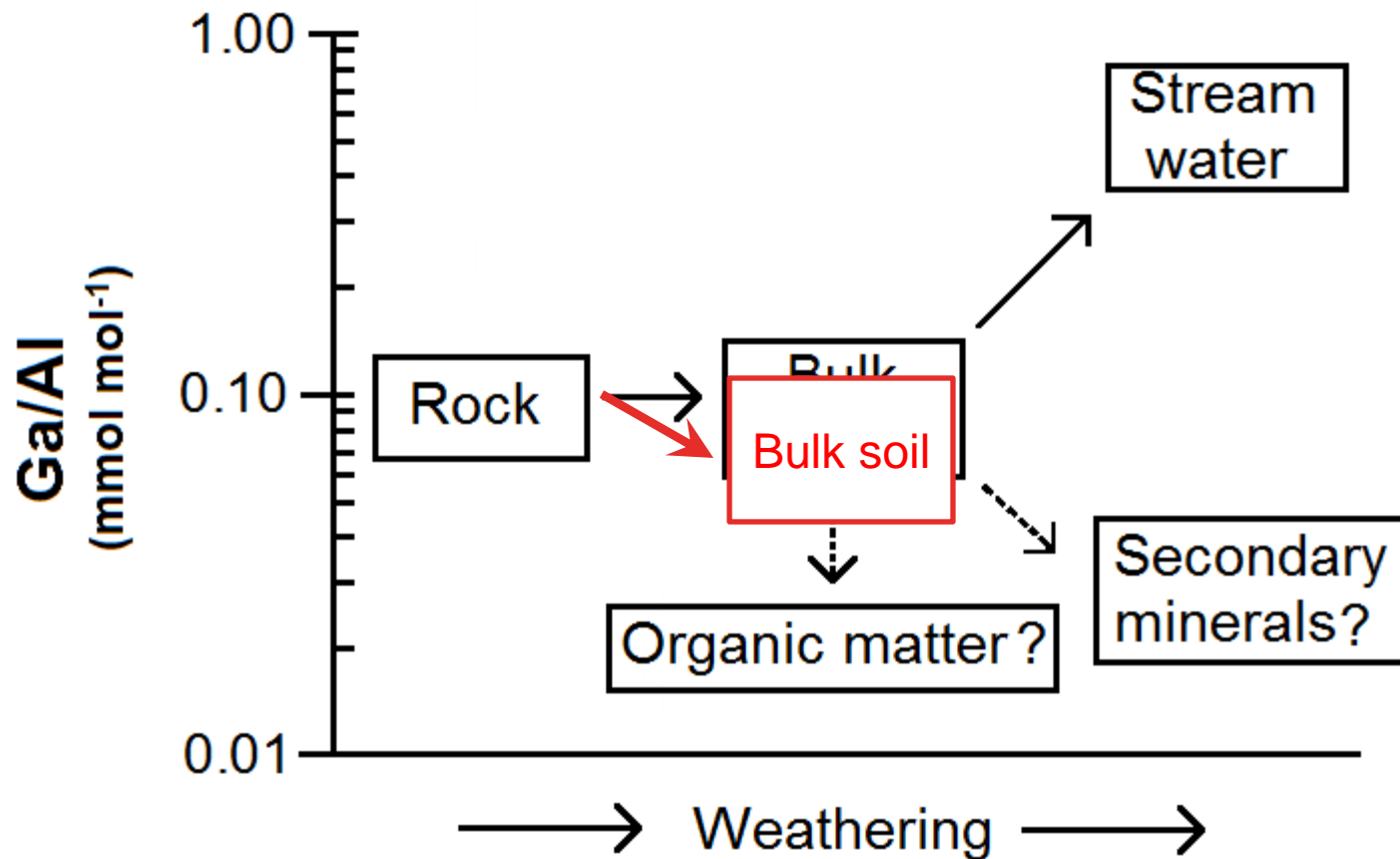
Ga/Al (mmol mol⁻¹)

0.00 0.10 0.20





(3) Processes that fractionate Ga/Al ratio

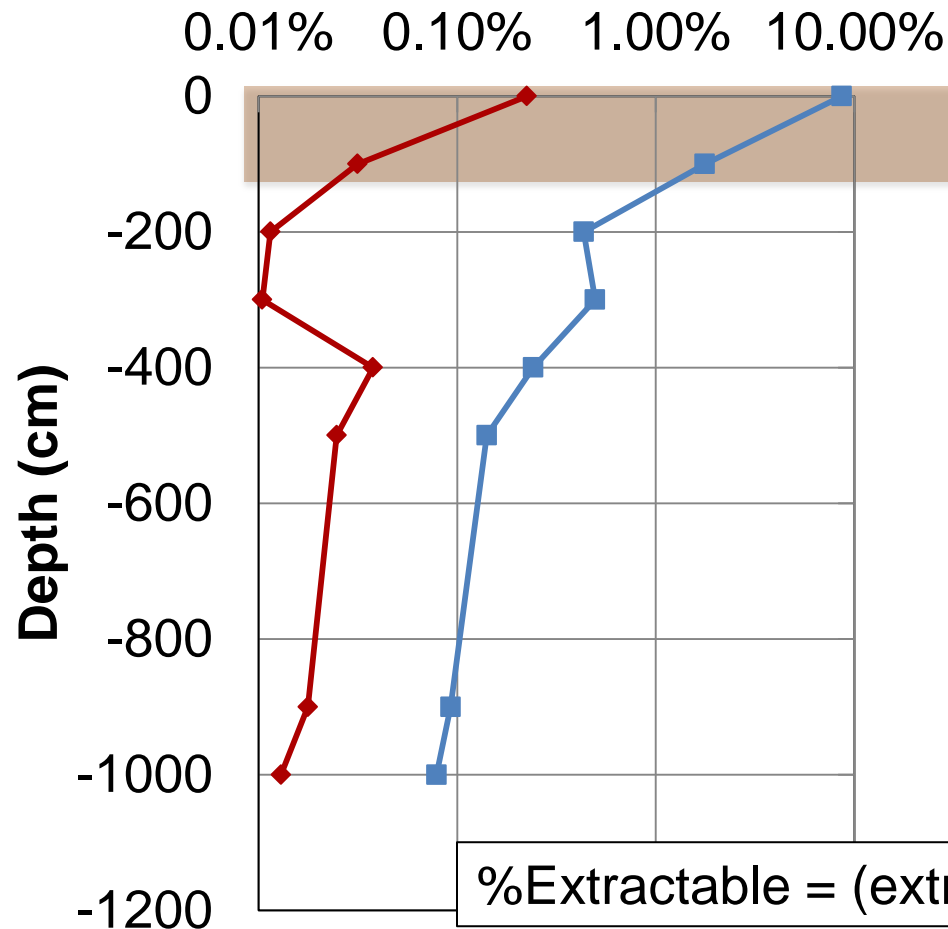


Based on data from Shiller and Frilot (1996)

Organic matter/extractable fraction

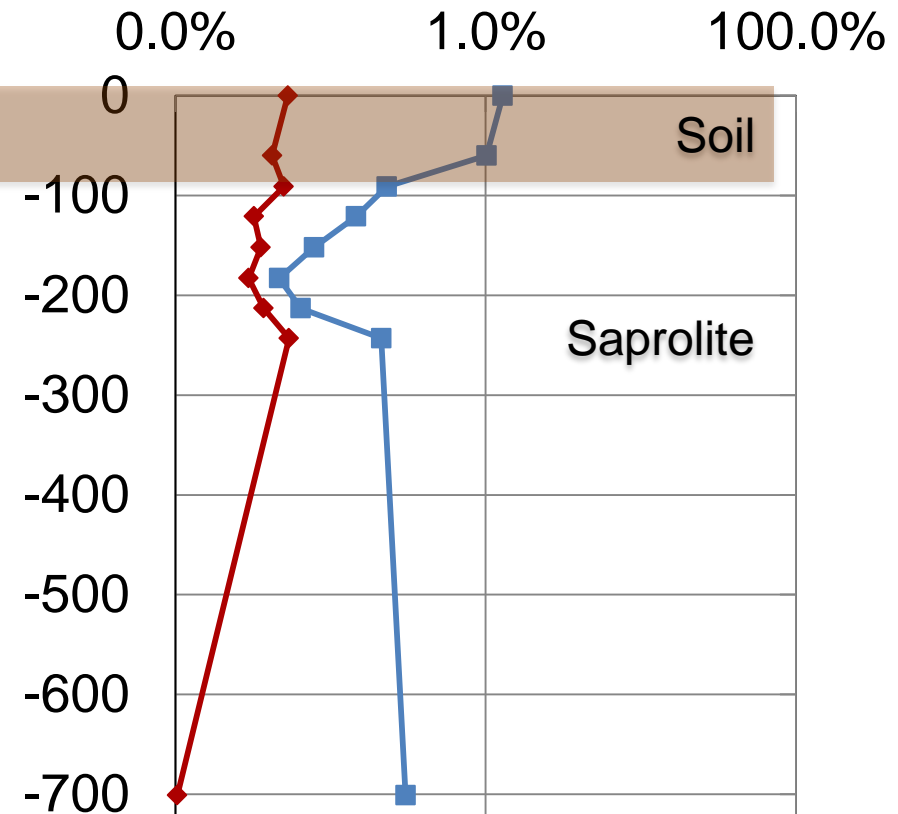
Southern Sierra CZO

◆ %Extractable Ga
■ %Extractable Al



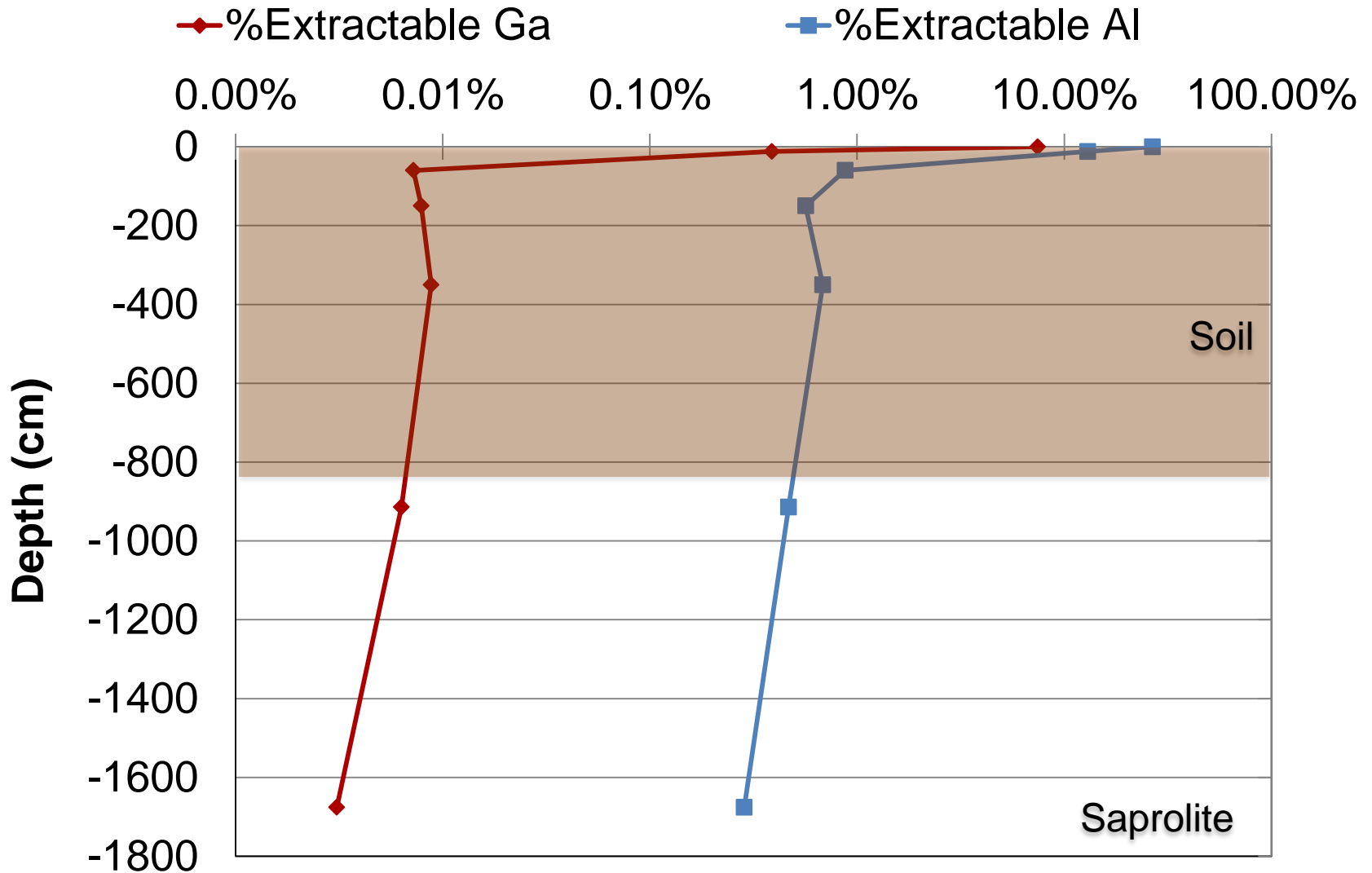
Boulder Creek CZO

◆ %Extractable Ga
■ %Extractable Al

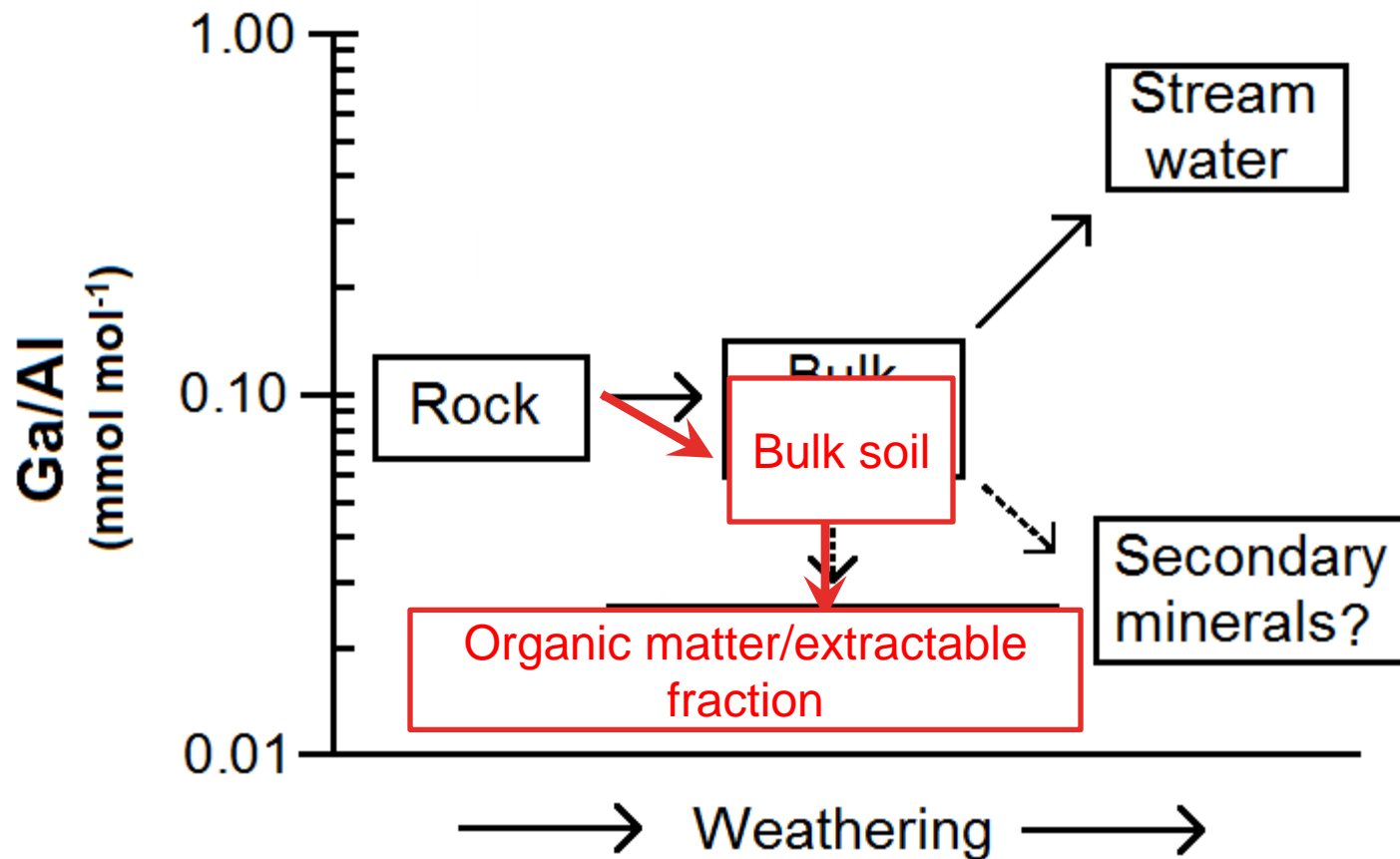


%Extractable = (extractable/total)

Calhoun CZO



(3) Processes that fractionate Ga/Al ratio



Based on data from Shiller and Frilot (1996)

• Summary

- Ga/Al ratio decreased during the bedrock-regolith-soil transition.
- Ga/Al in the organic matter/exchangeable fraction was very different.
- The Ga/Al ratio maybe useful for distinguishing colloidal and dissolved losses of Al.



Thank you!
Questions?

