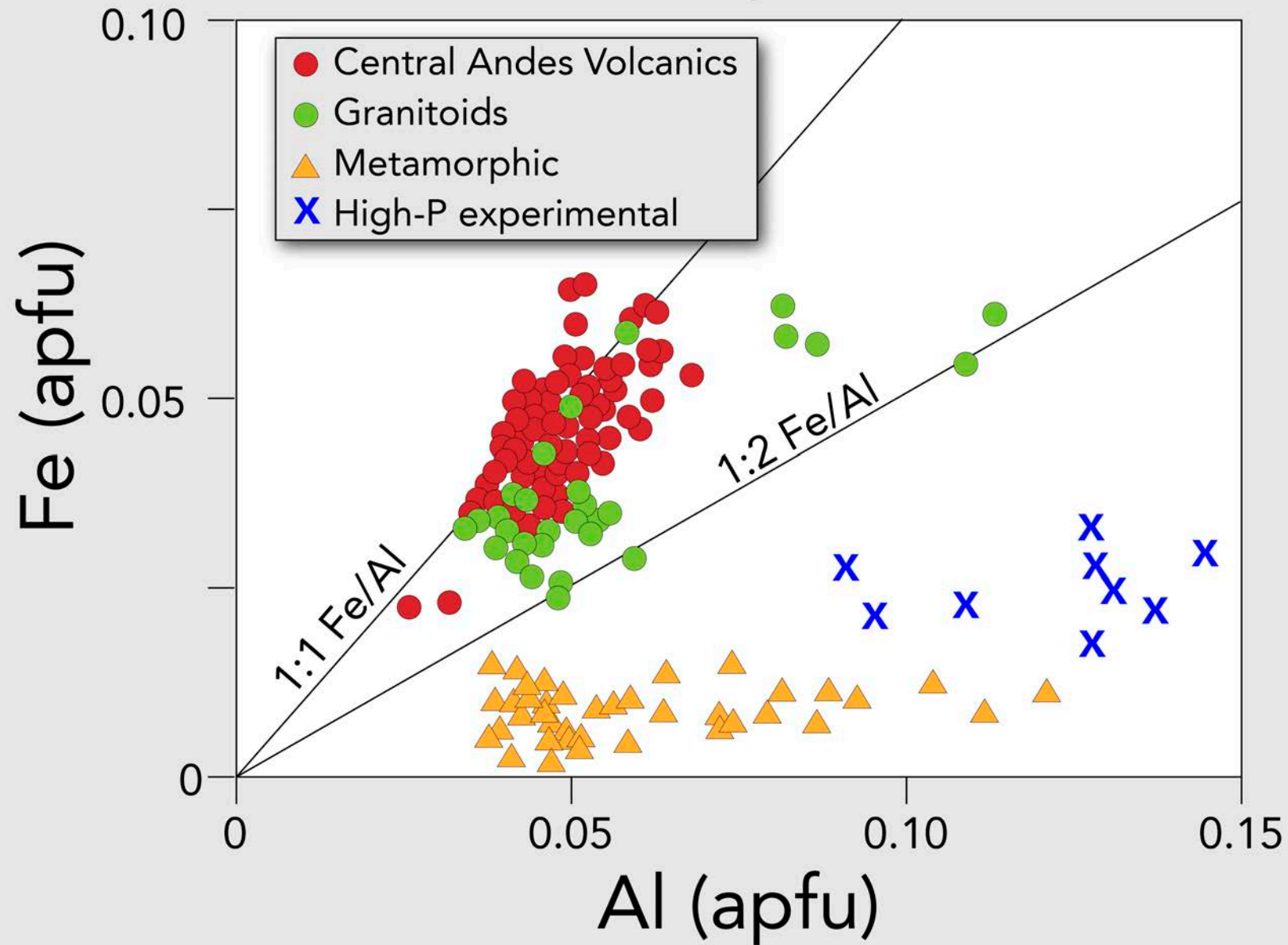


GSA Annual Meeting – 2018 Indianapolis

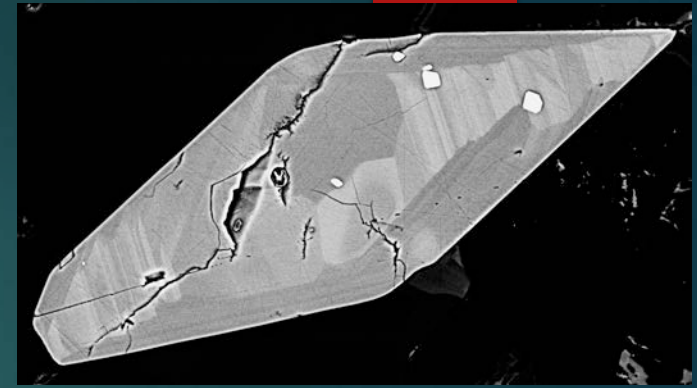
Compositional Variation of Fe, Al, & F in Titanite

Bart J. Kowallis, Eric H Christiansen, Michael J. Dorais, Tony Winkel,
Porter Henze, Lauren Franzen, and Haley Mosher – Brigham Young U.

Nakada, 1991



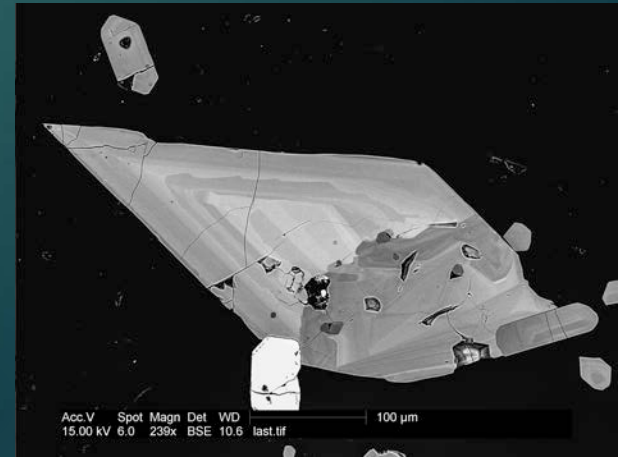
Conclusions – Fe/Al ratios



- ▶ The atomic ratio of Fe/Al in titanite from both volcanic and plutonic rocks is typically close to 1:1 and almost always $>1:2$.
- ▶ Volcanic titanite compositions typically cluster more tightly in terms Fe, Al, and F than do titanite compositions from any other environment.
- ▶ Fe/Al ratios in titanite from peralkaline silica-undersaturated volcanic and plutonic rocks are typically $>1:1$.
- ▶ Titanite from metamorphic, hydrothermal, and pegmatitic environments scatter widely in Fe/Al.
- ▶ Titanite from eclogite tends to have the lowest Fe/Al ratios, typically $<1:8$.

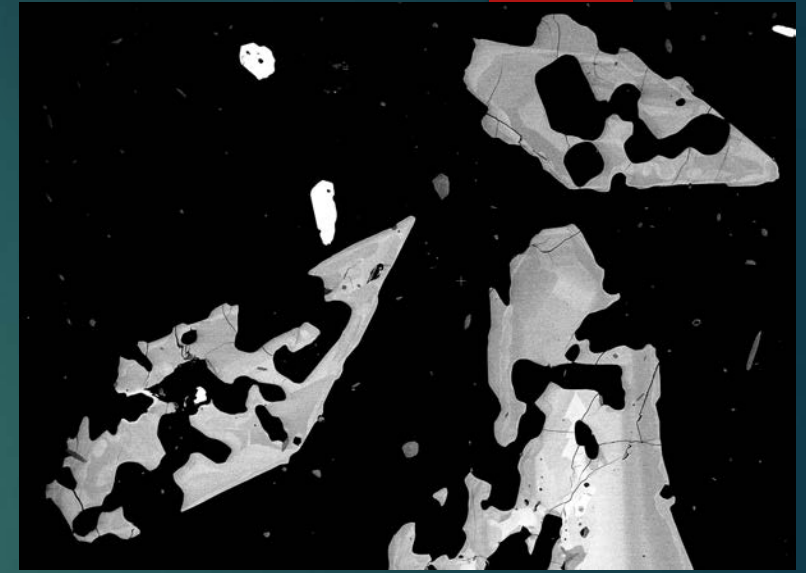
Conclusions – Charge Balancing

- ▶ Charge balance in metamorphic, hydrothermal, and pegmatitic titanite due to Fe^{+3} and Al^{+3} substitution into the Ti^{+4} site is largely accomplished by the coupled substitution of F^- for O^{-2} .
- ▶ However, in volcanic and plutonic titanite the charge imbalance due to Fe^{+3} and Al^{+3} substitution appears to be mainly coupled with REE^{+3} and Y^{+3} substitution into the Ca^{+2} site with a more minor contribution from F^- substitution.
- ▶ In Si-undersaturated rocks, substitution into the Ti^{+4} site by Nb^{+5} coupled with Fe^{+3} is a major factor in charge balancing.



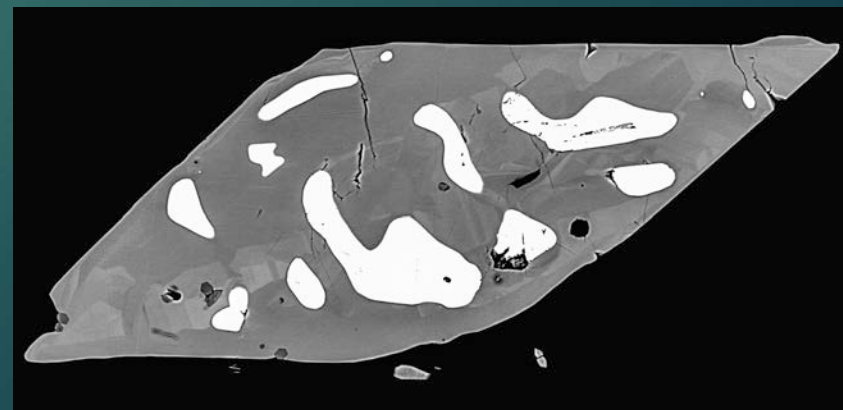
Data Base

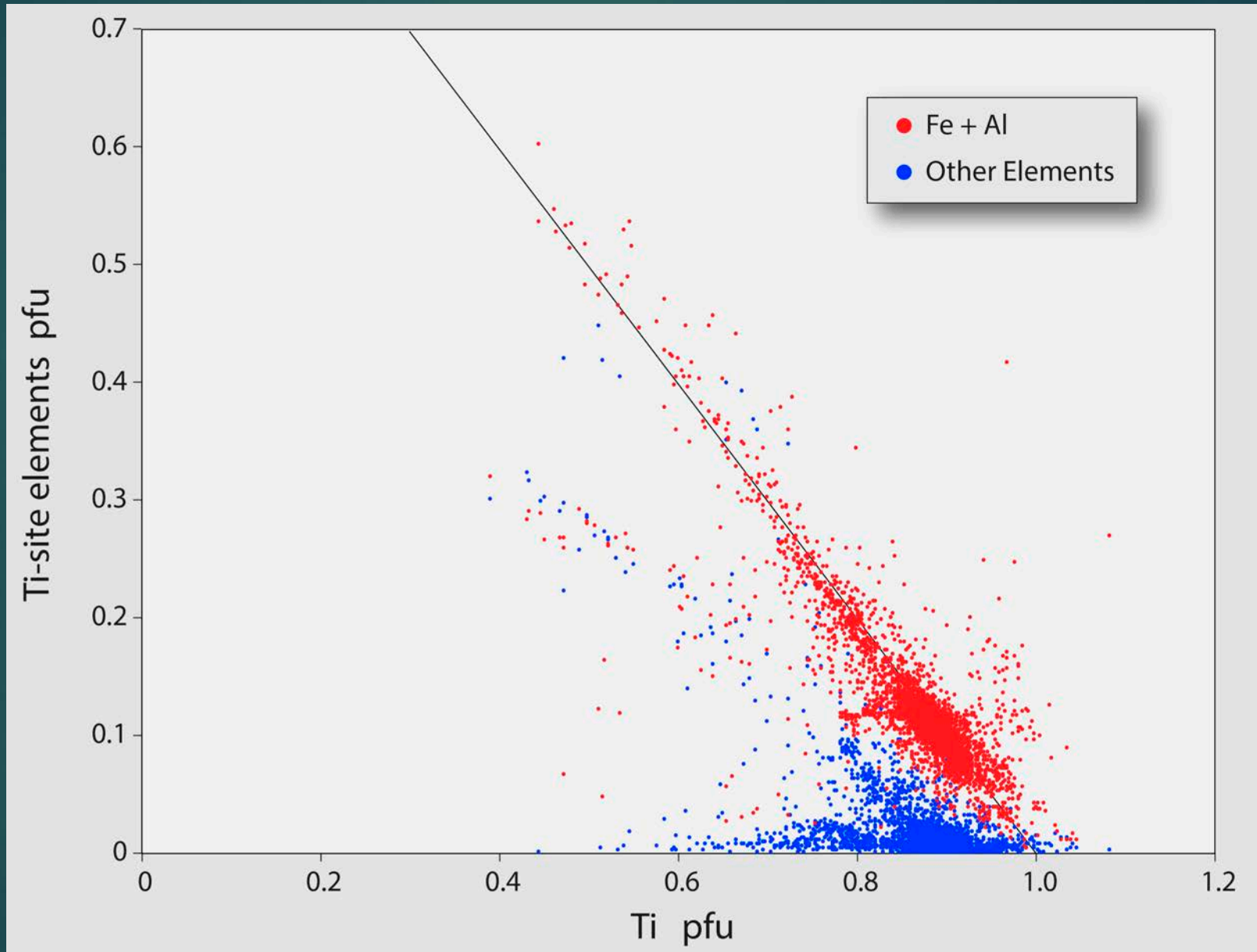
- ▶ 8,100+ titanite analyses
 - ▶ Most include Fe and Al
 - ▶ A large number also include F
- ▶ ~4,800 analyses are from Brigham Young Univ. and Univ. of Utah microprobe labs
- ▶ ~3,300 analyses from literature

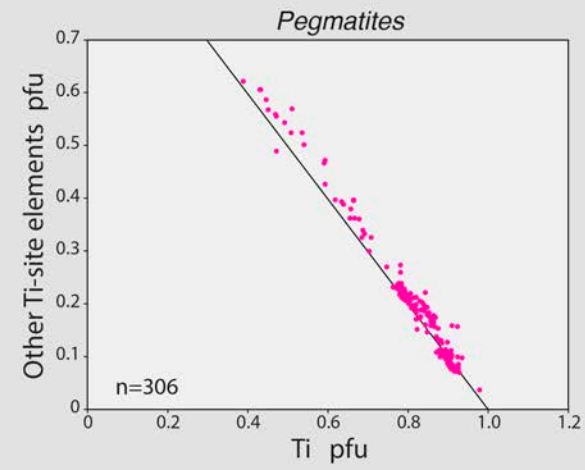
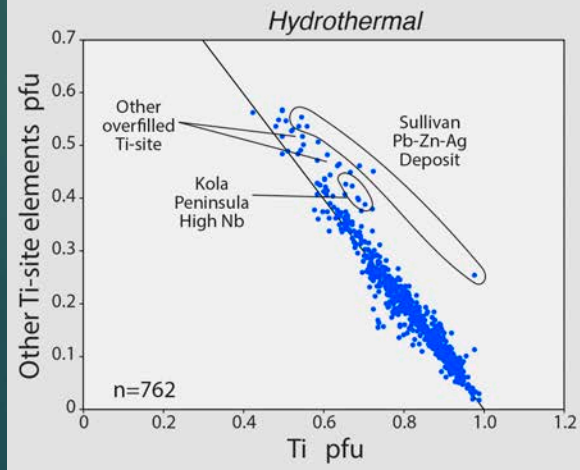
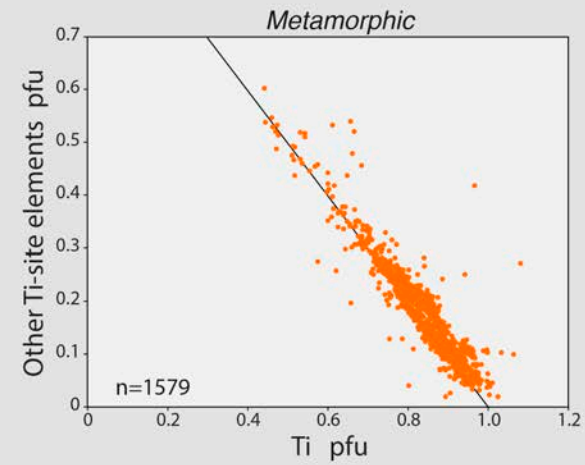
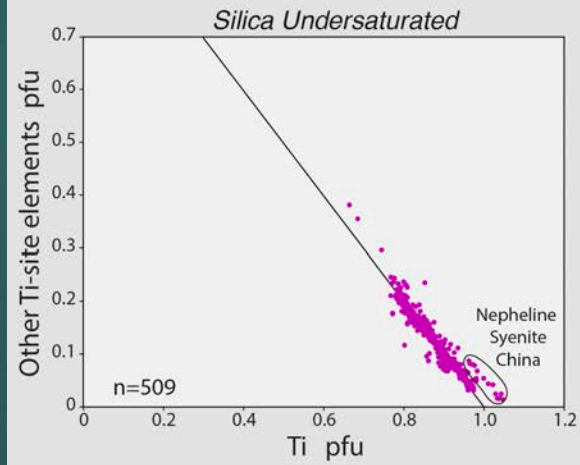
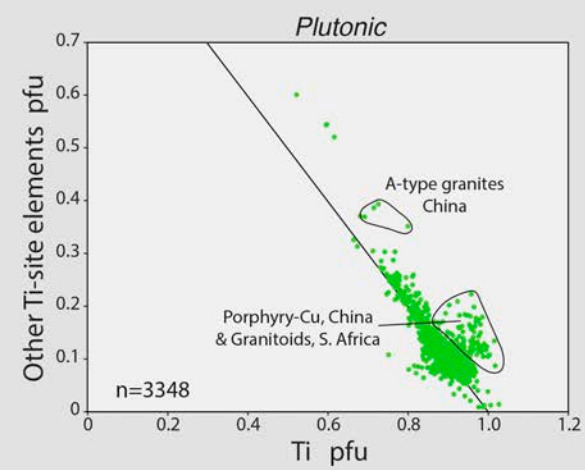
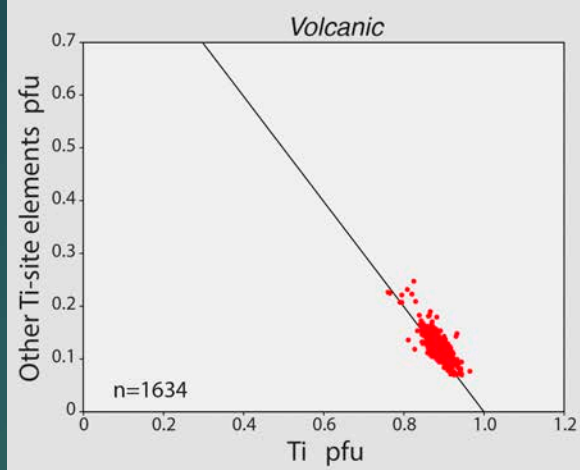


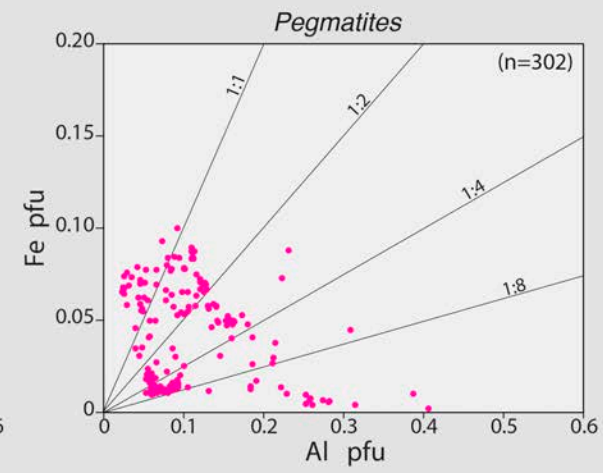
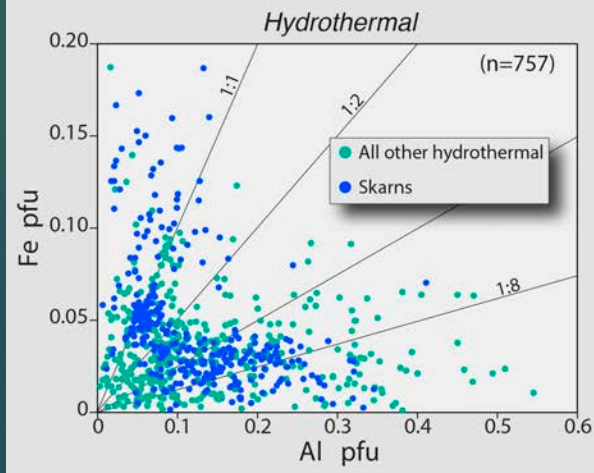
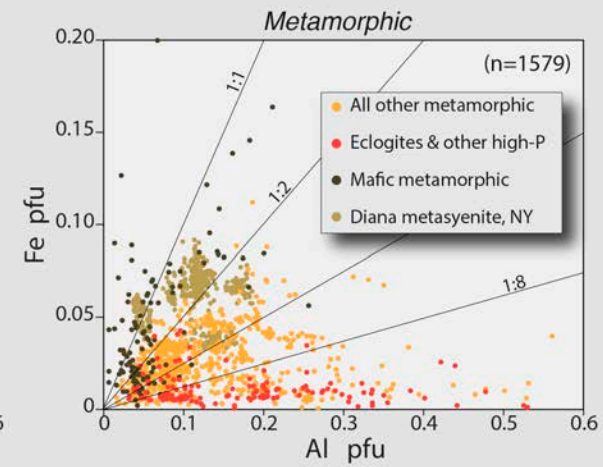
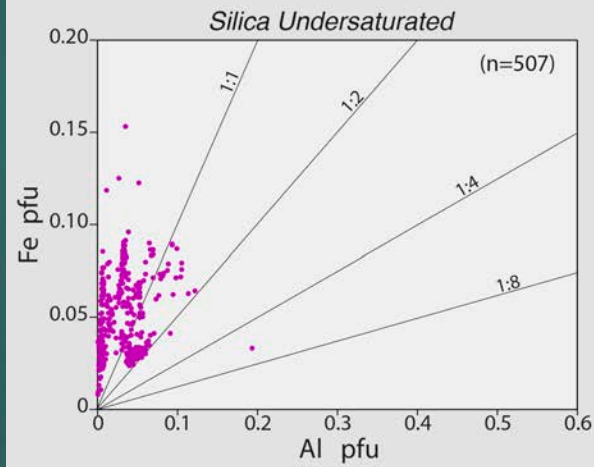
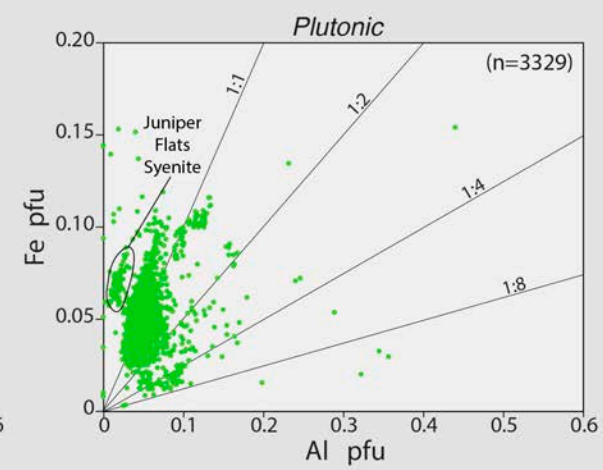
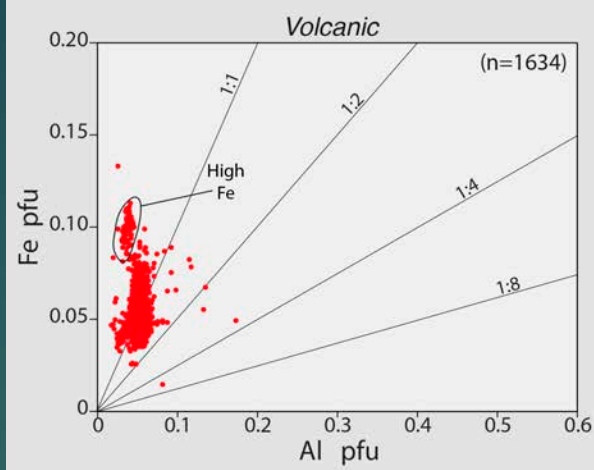
Element Substitutions

- ▶ Basic Formula = $\text{CaTi}(\text{SiO}_4)(\text{O}, \text{F}, \text{OH})$
- ▶ Ti^{4+} -site elements = Al^{3+} , Fe^{3+} , Nb^{5+} , Ta^{5+} , V^{5+} , Zr^{4+} , etc.
- ▶ Ca^{2+} -site elements = REE^{3+} , Y^{3+} , Mn^{2+} , etc.

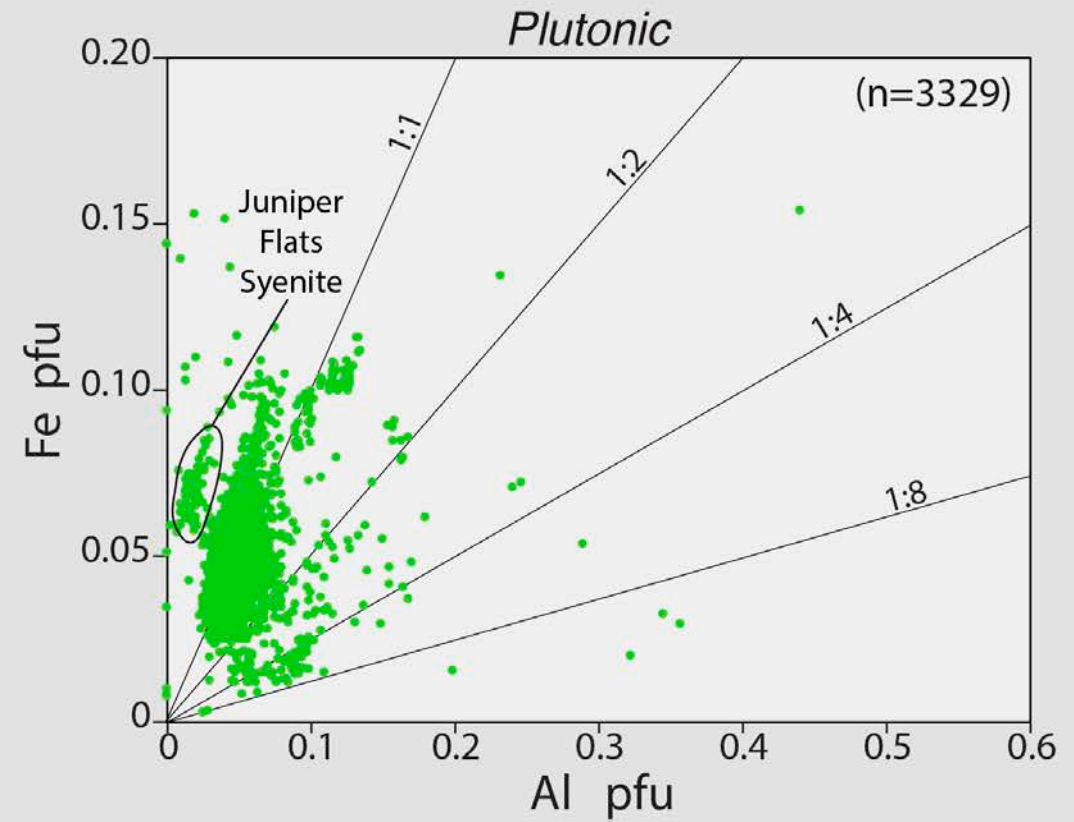
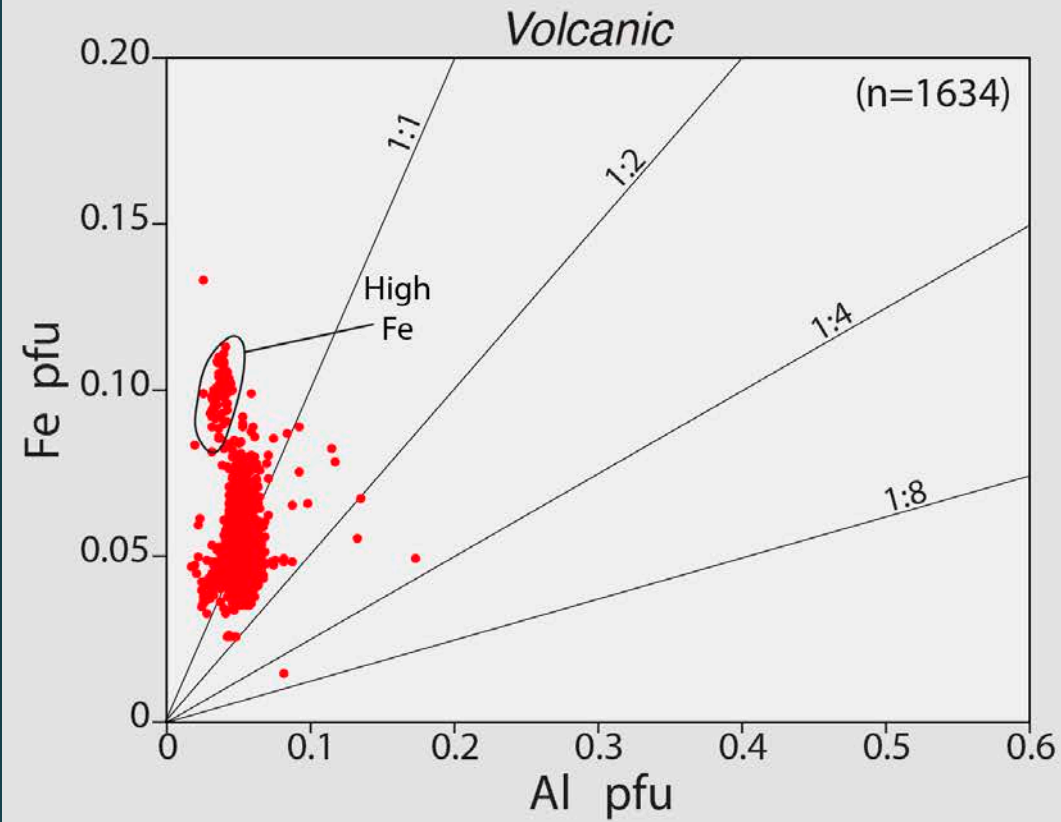




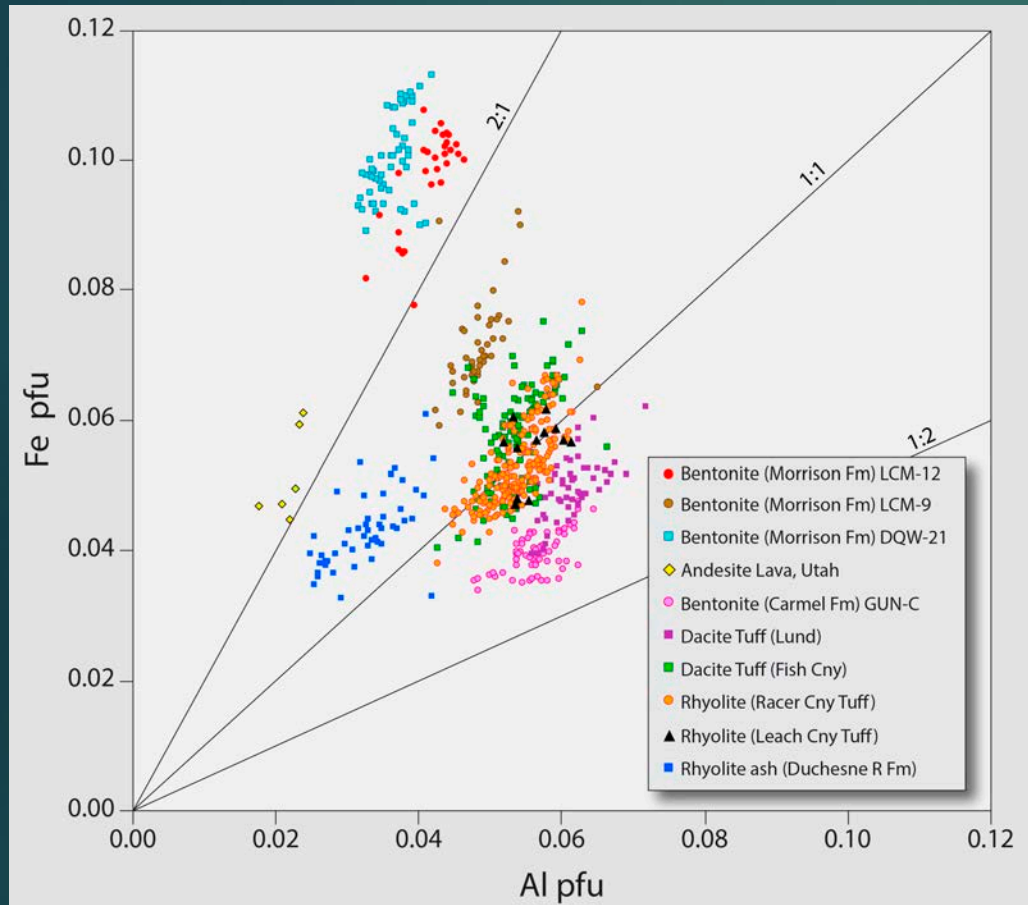




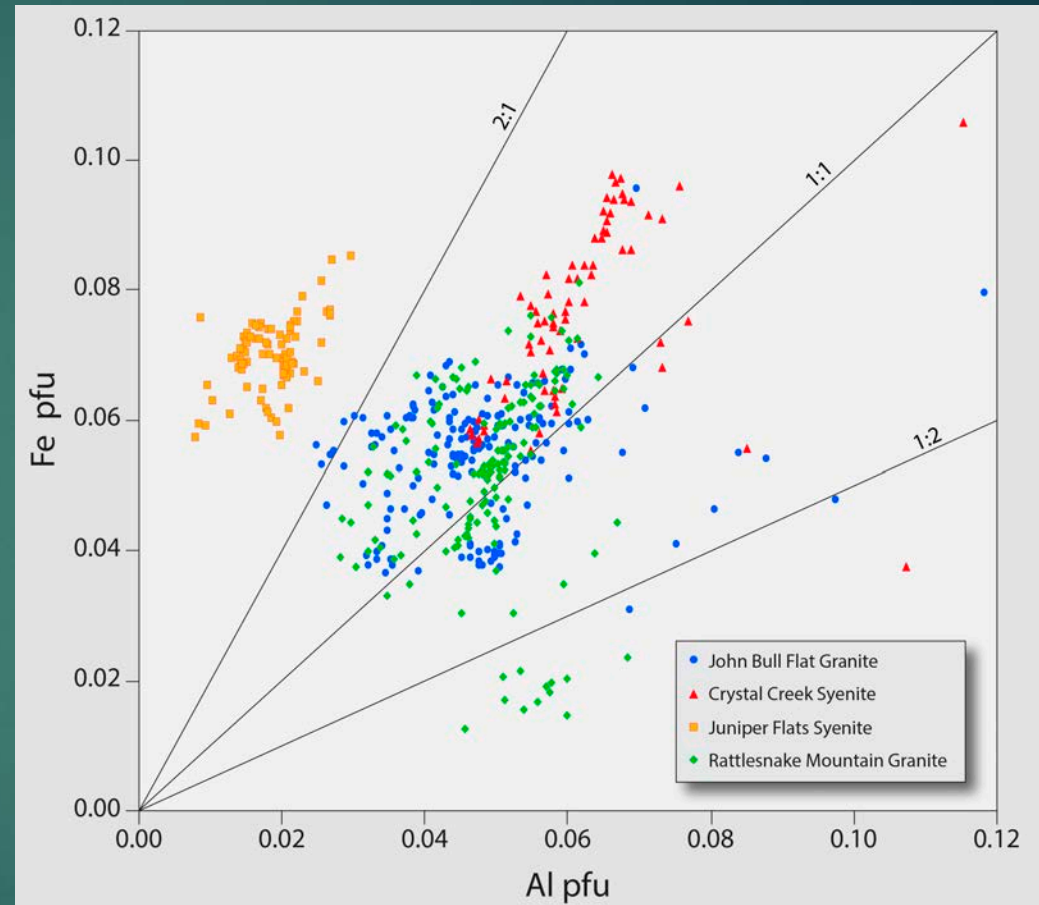
Fe versus Al in Titanite



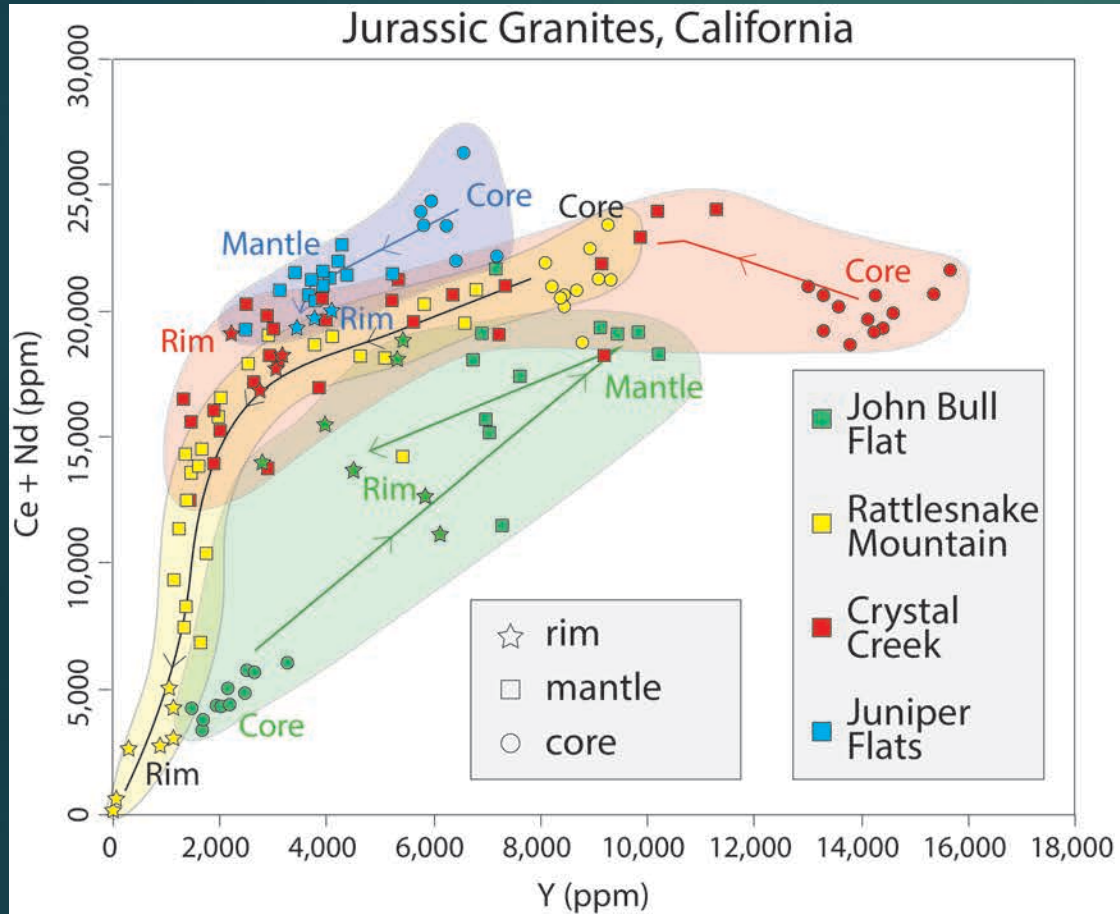
Volcanic Samples



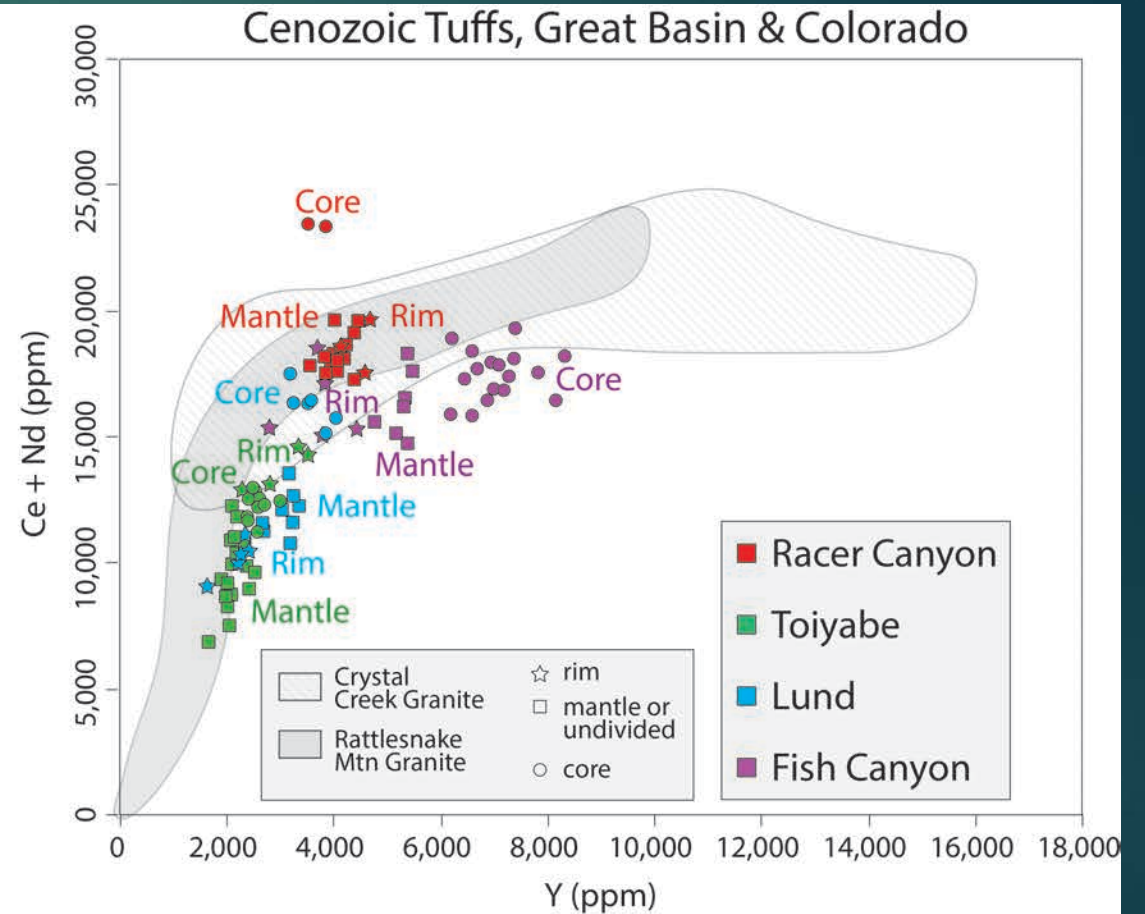
Plutonic Samples

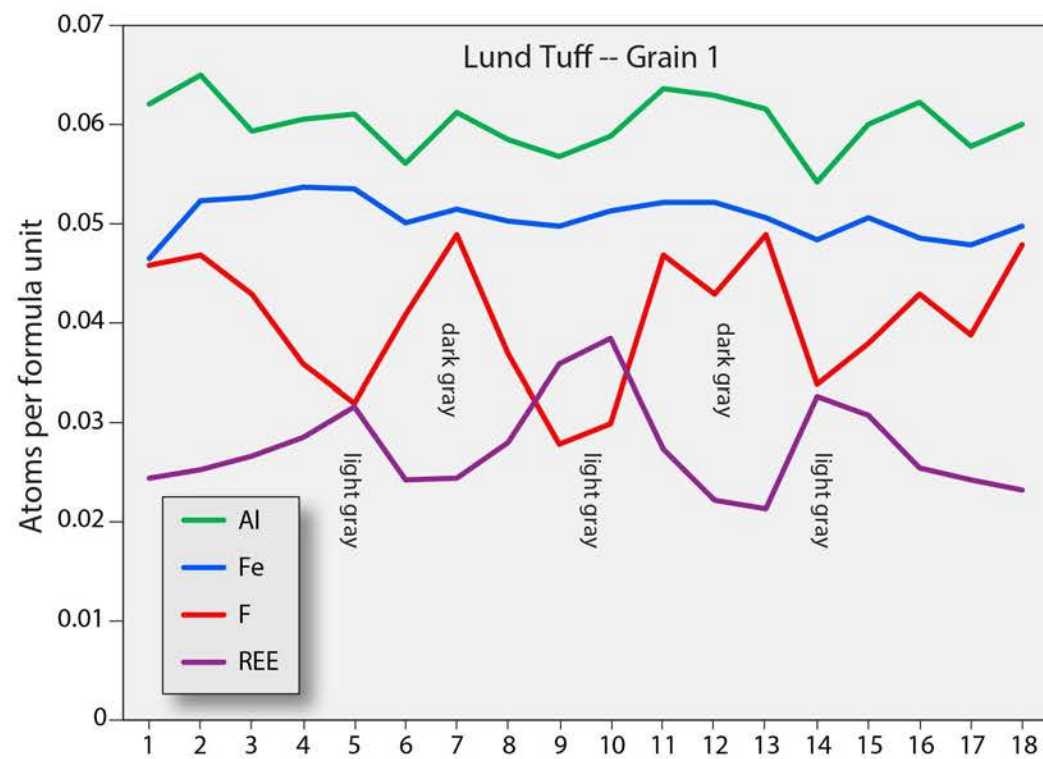
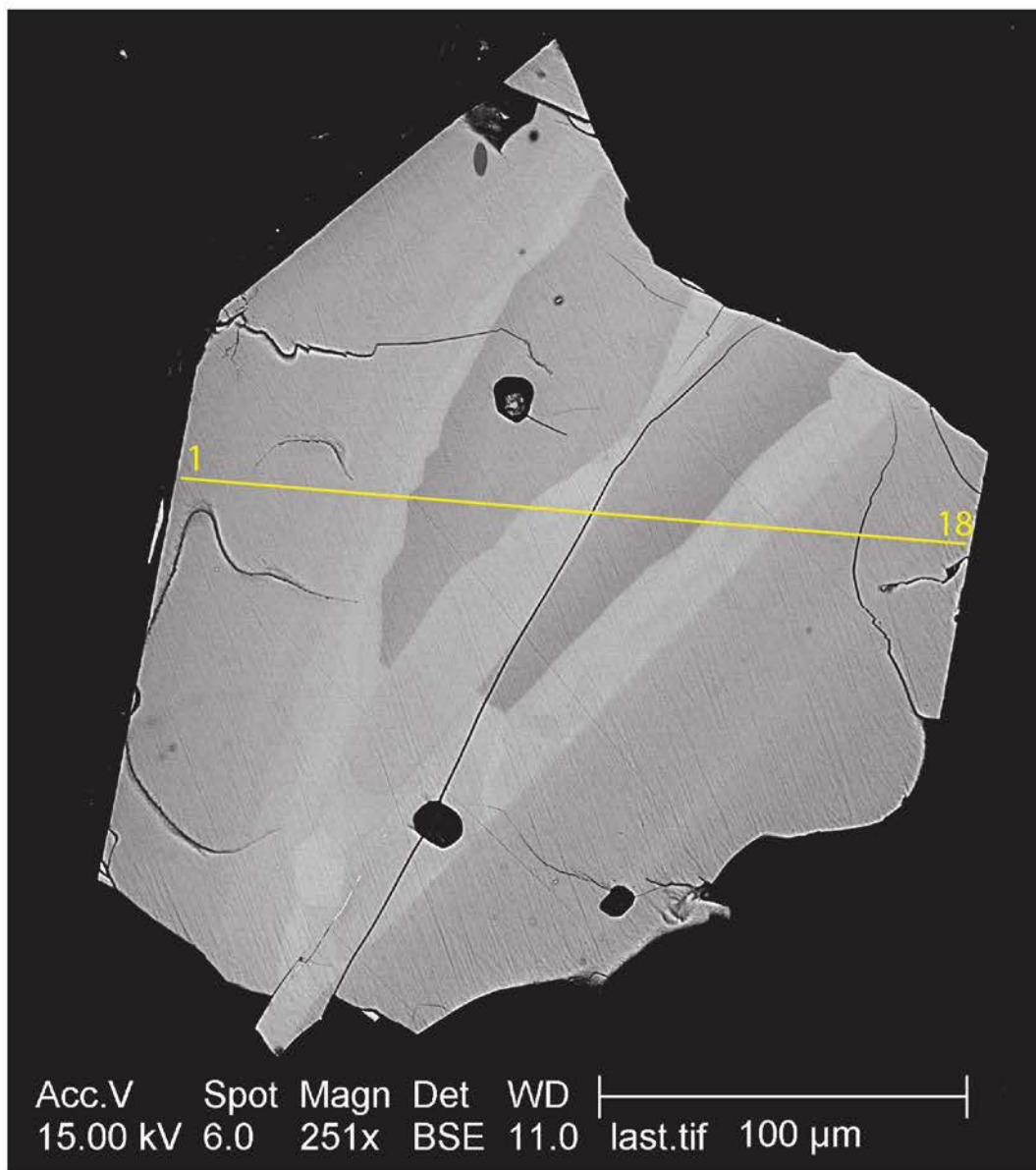


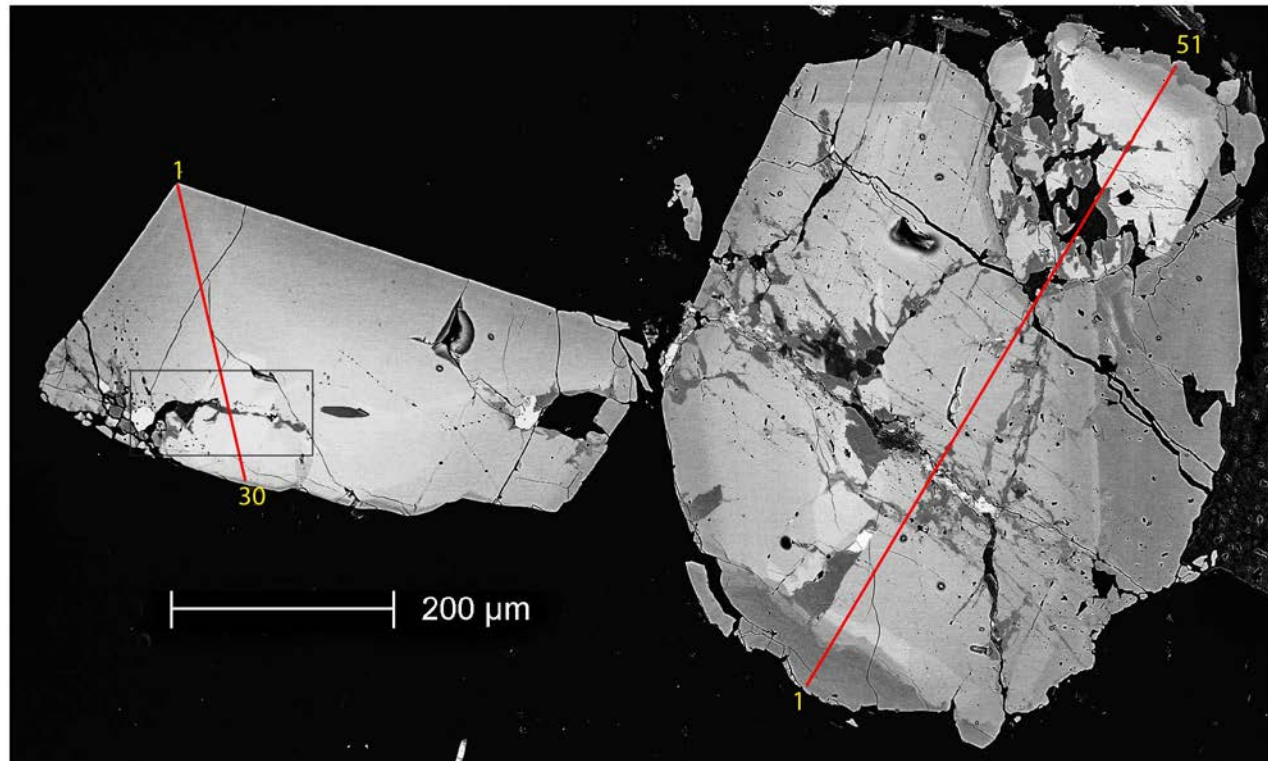
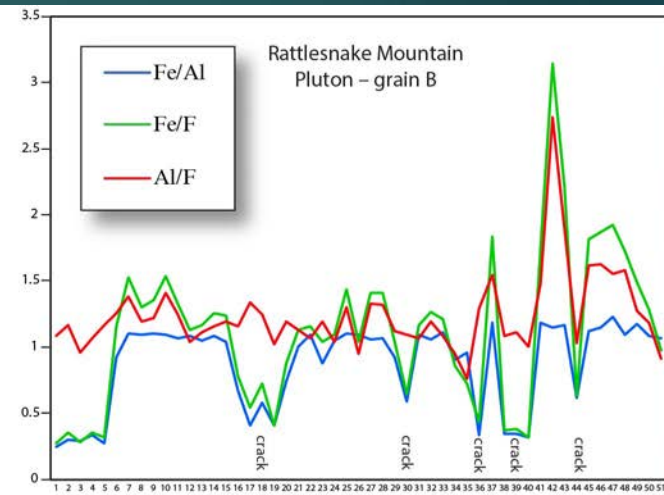
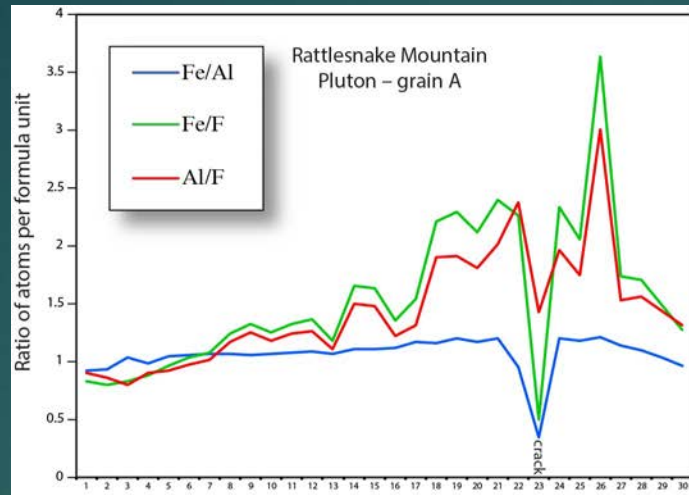
Plutonic Grains

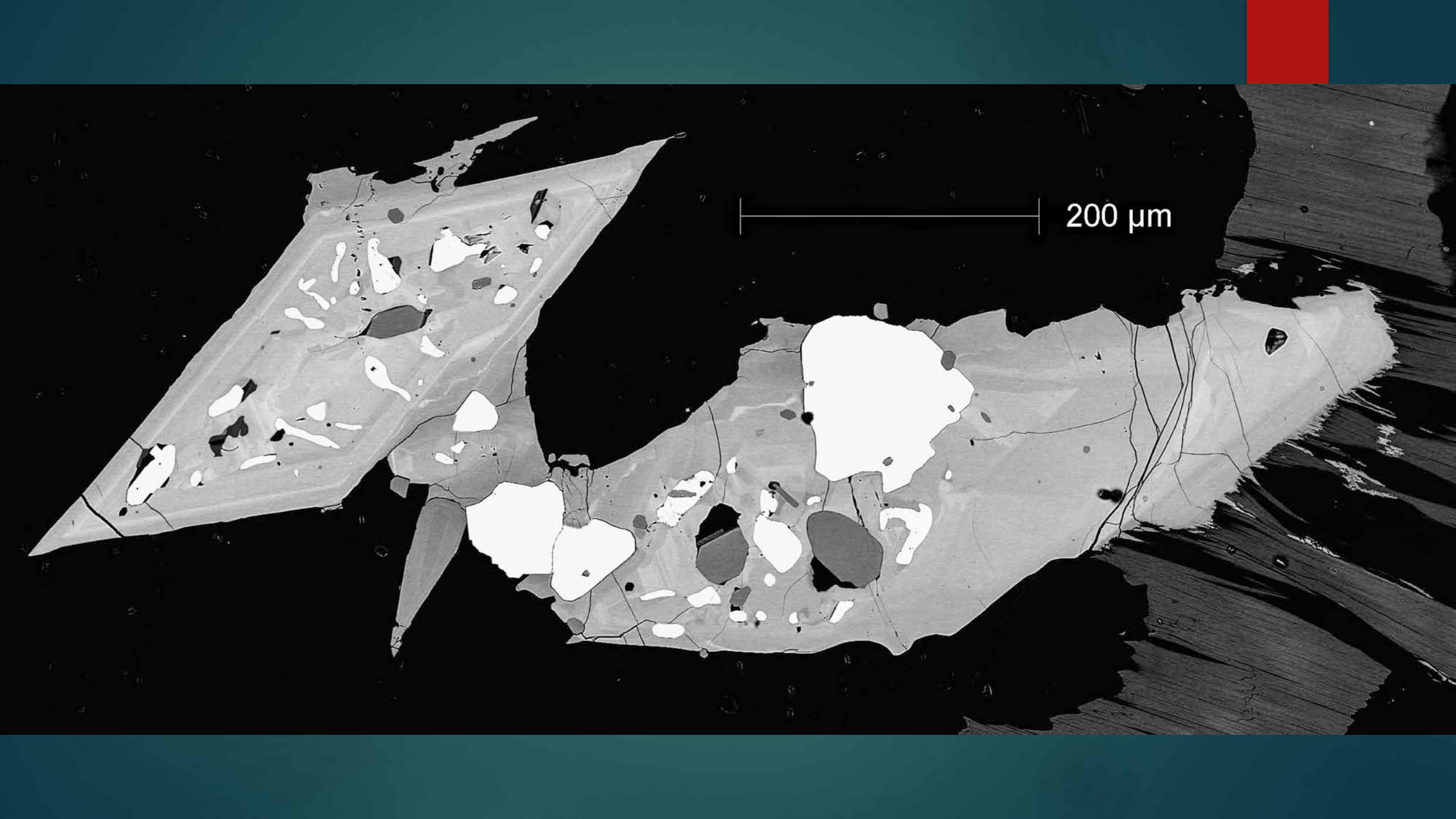


Volcanic Grains







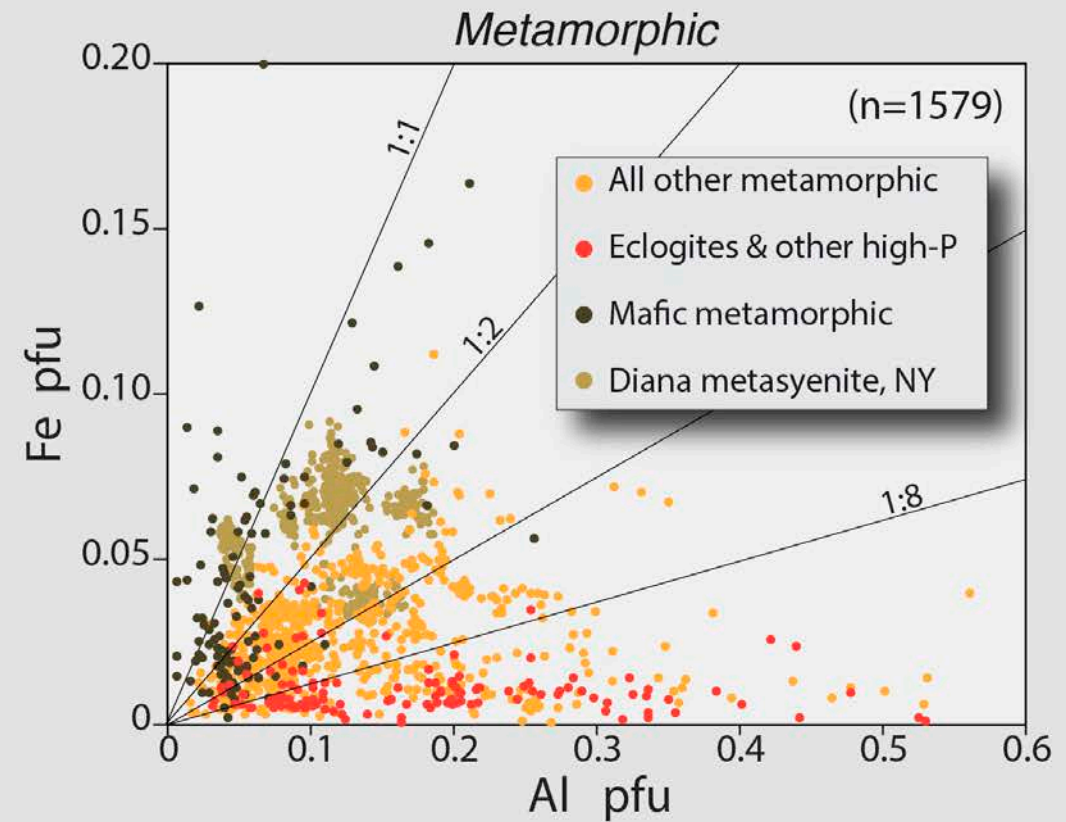
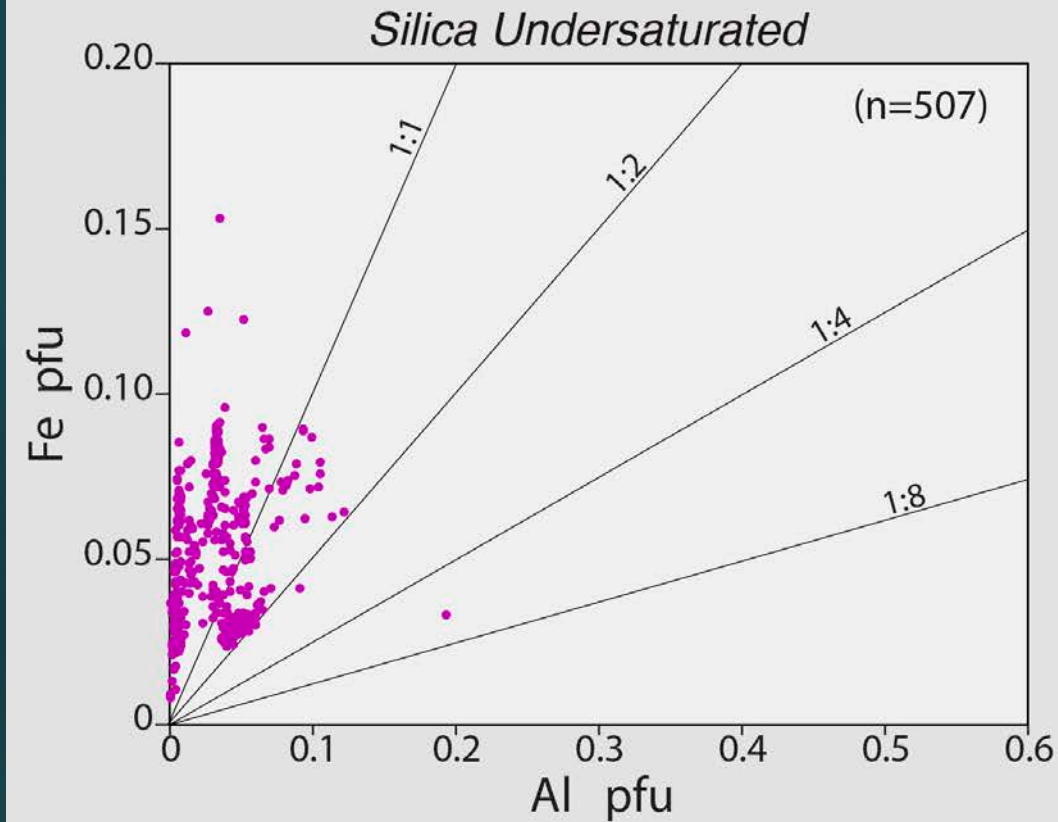


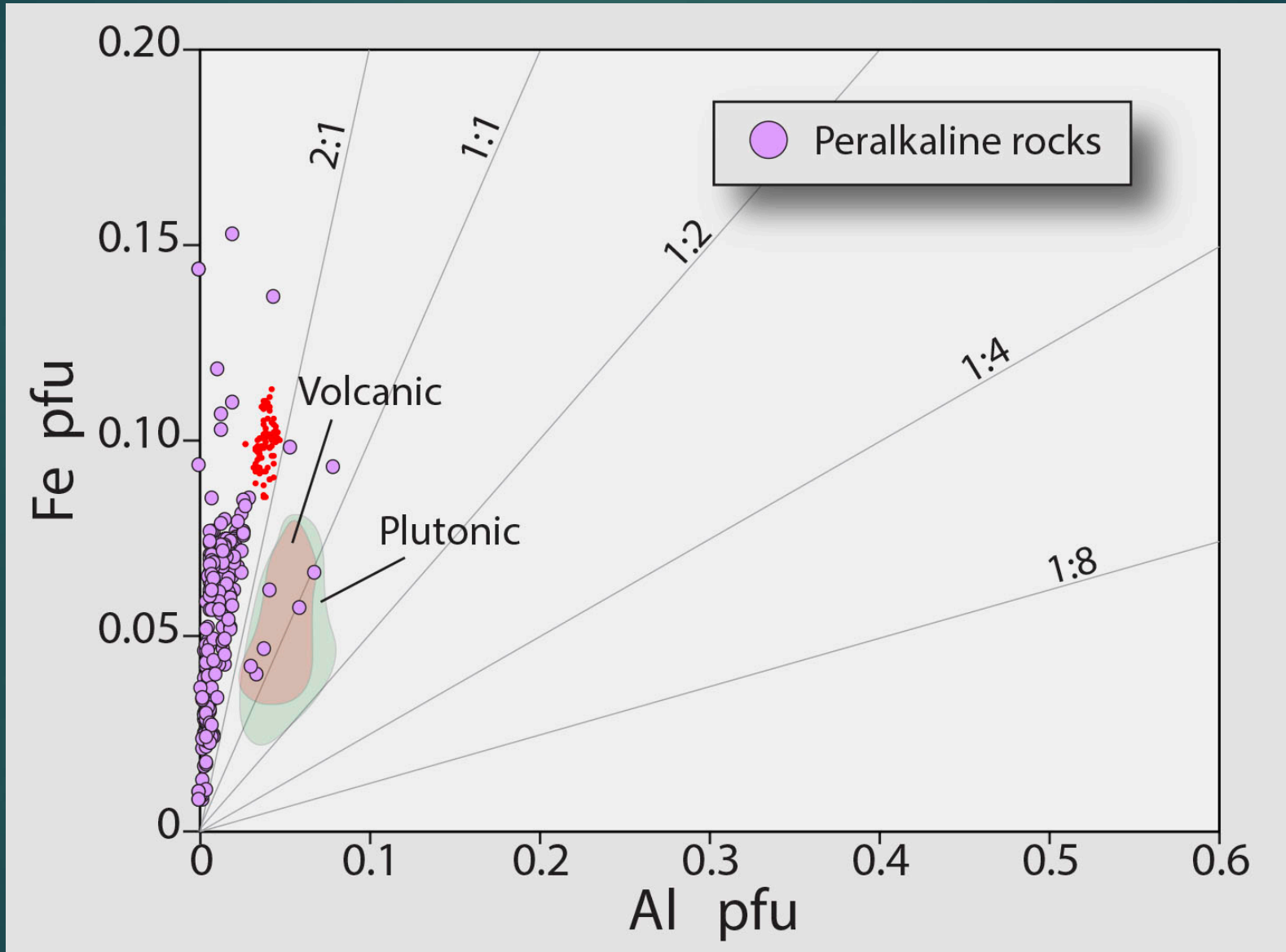


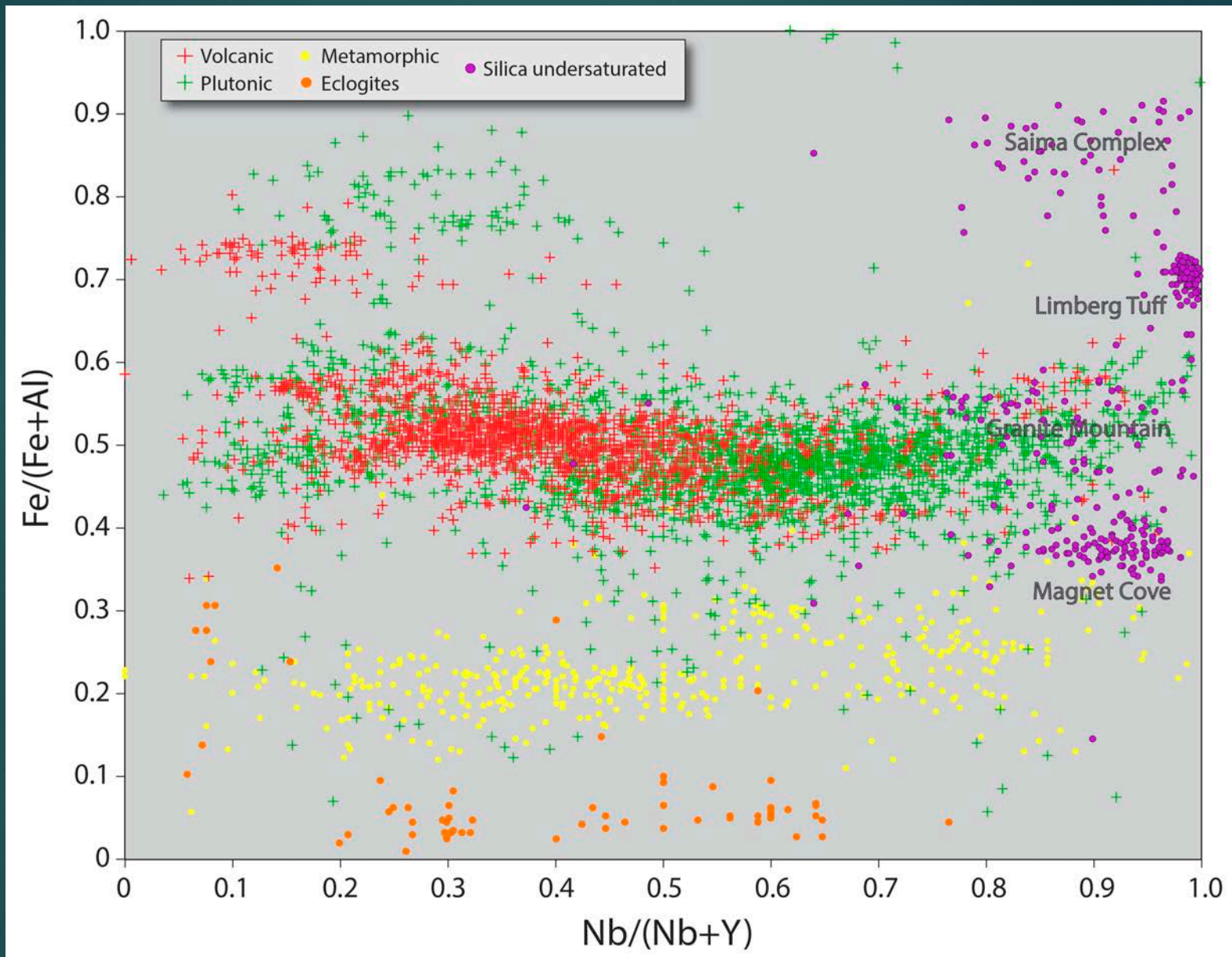
Acc.V	Spot	Magn	Det	WD	
15.00 kV	6.0	239x	BSE	10.6	100 μ m

last.tif

Fe versus Al in Titanite

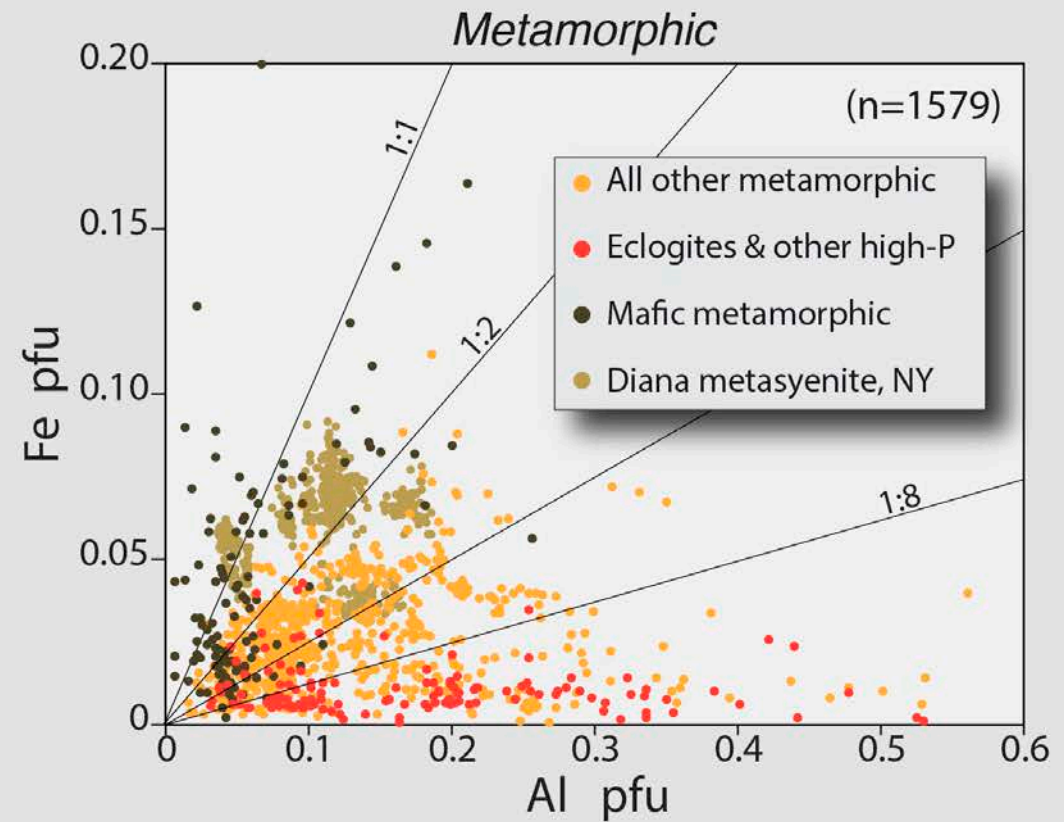
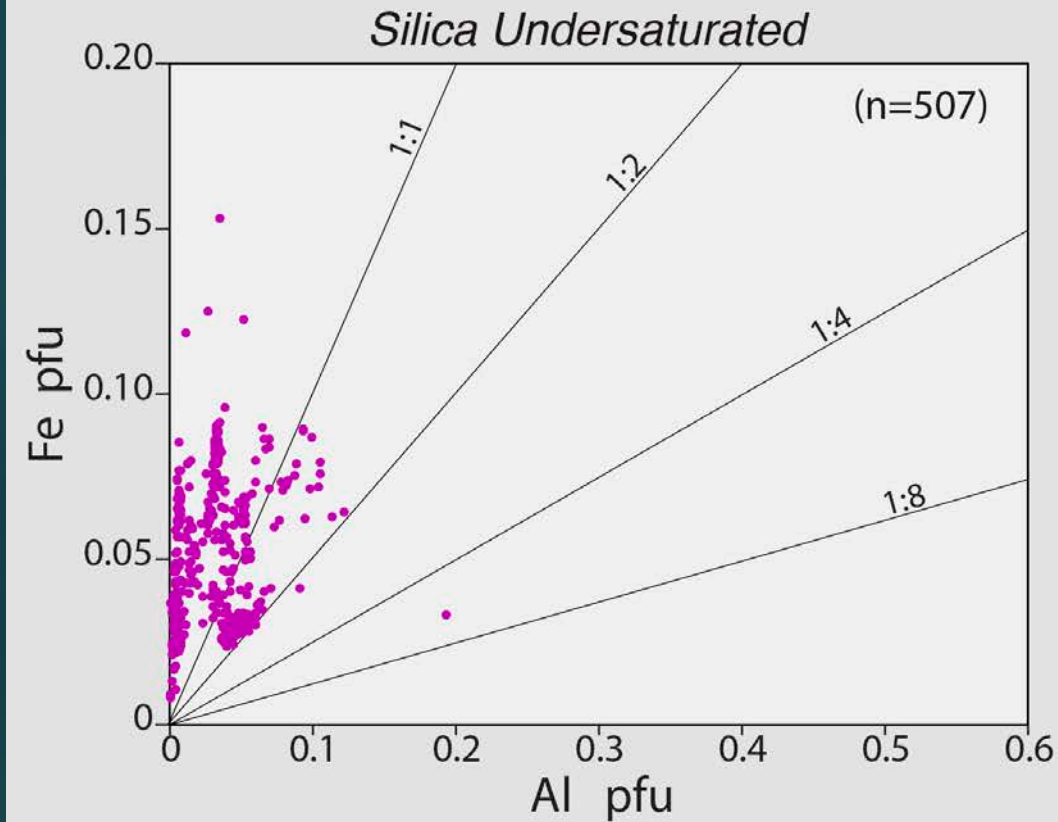


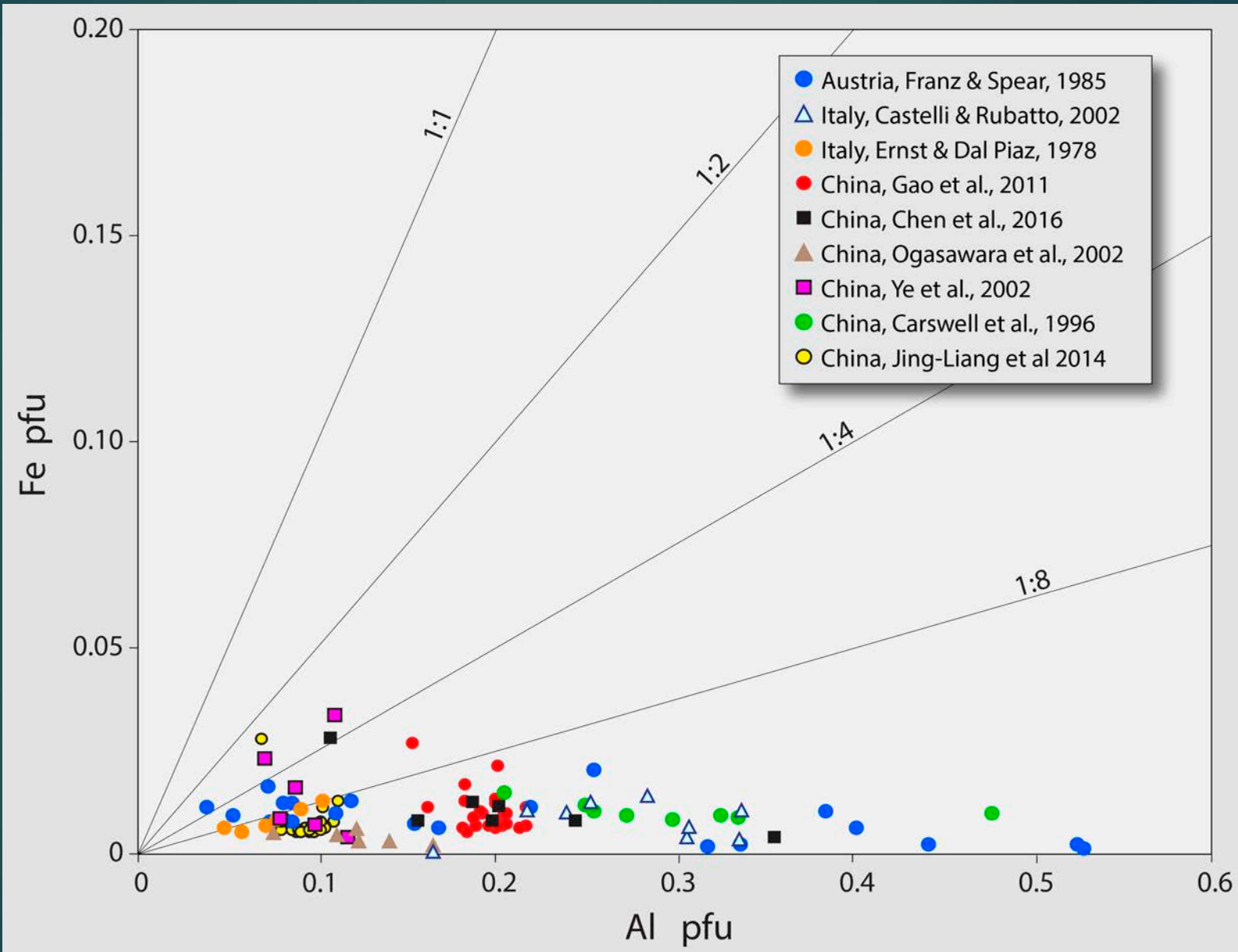


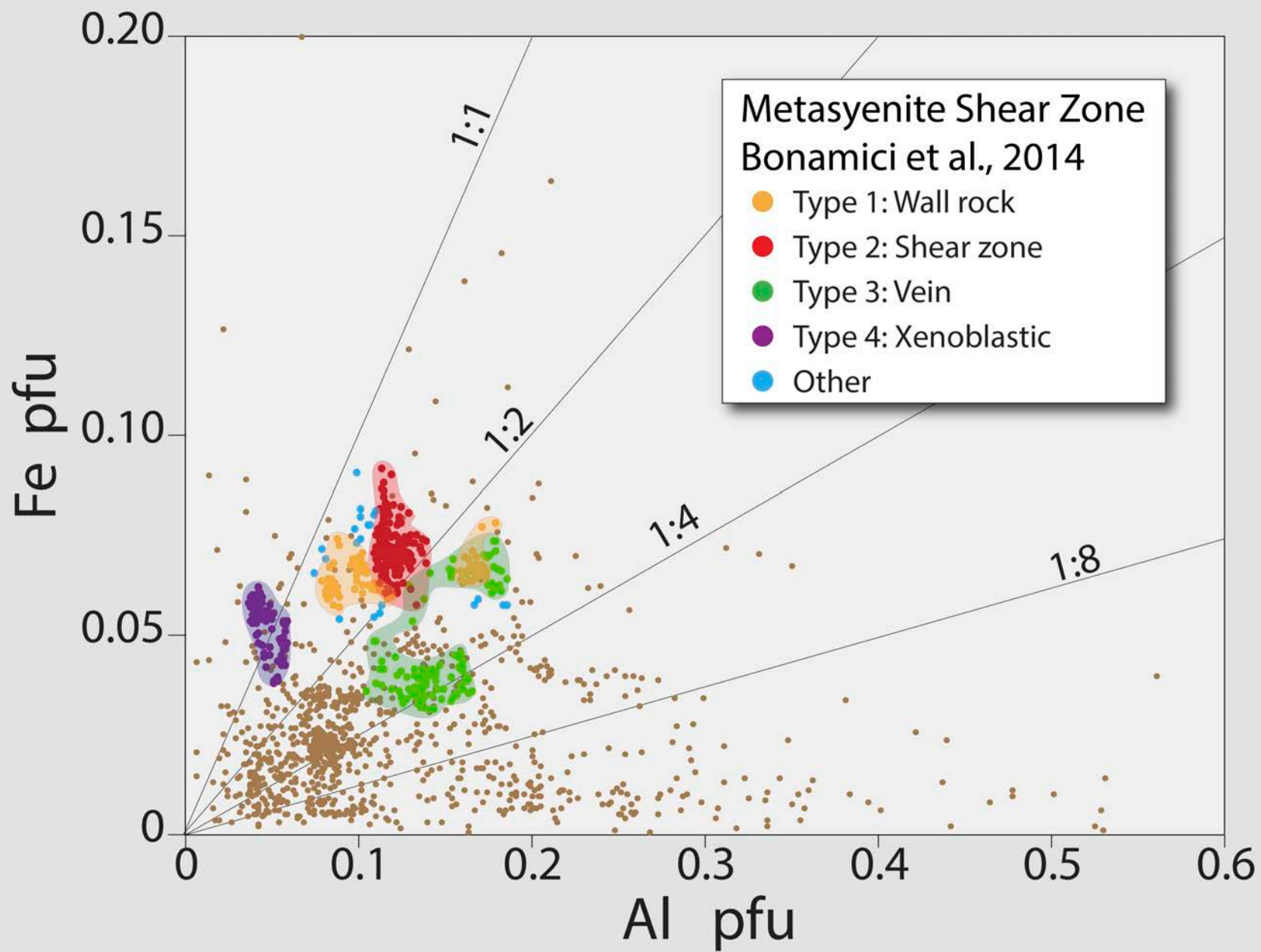


Saima alkaline complex, China (Wu et al., 2016), all other data from BYU.

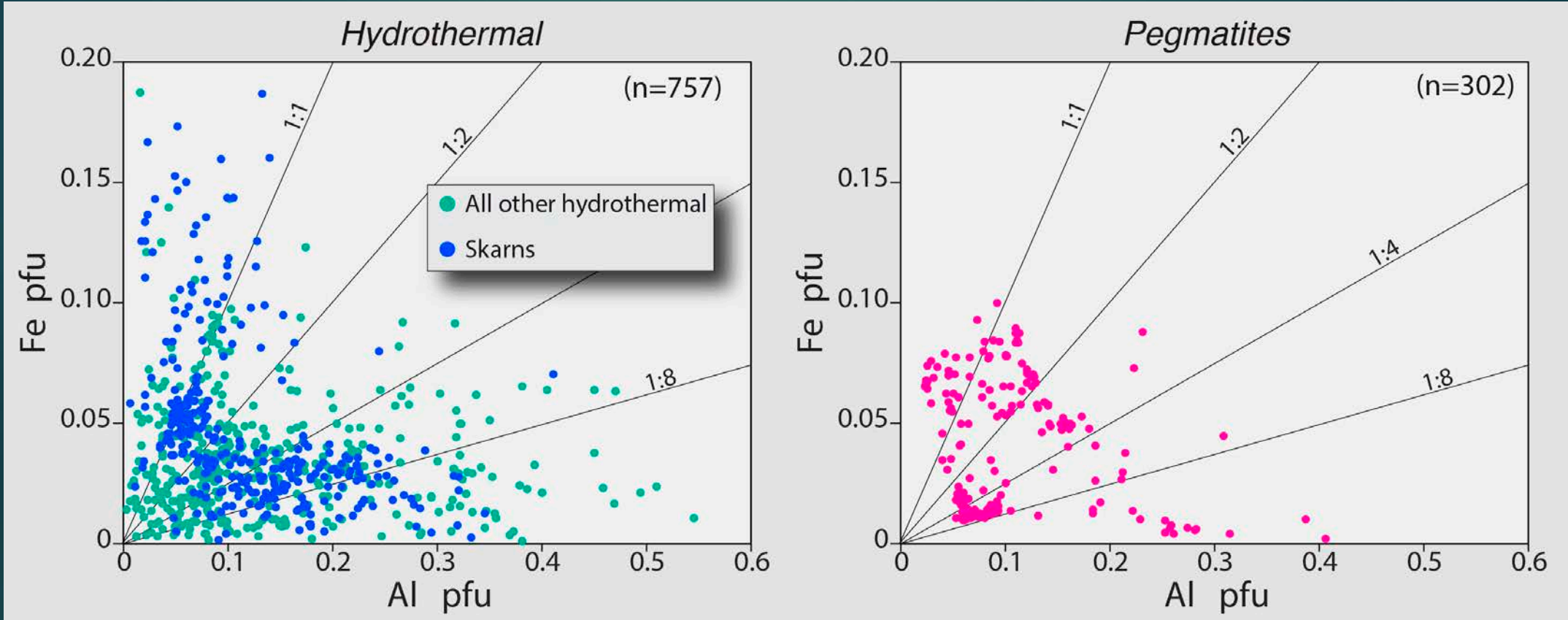
Fe versus Al in Titanite

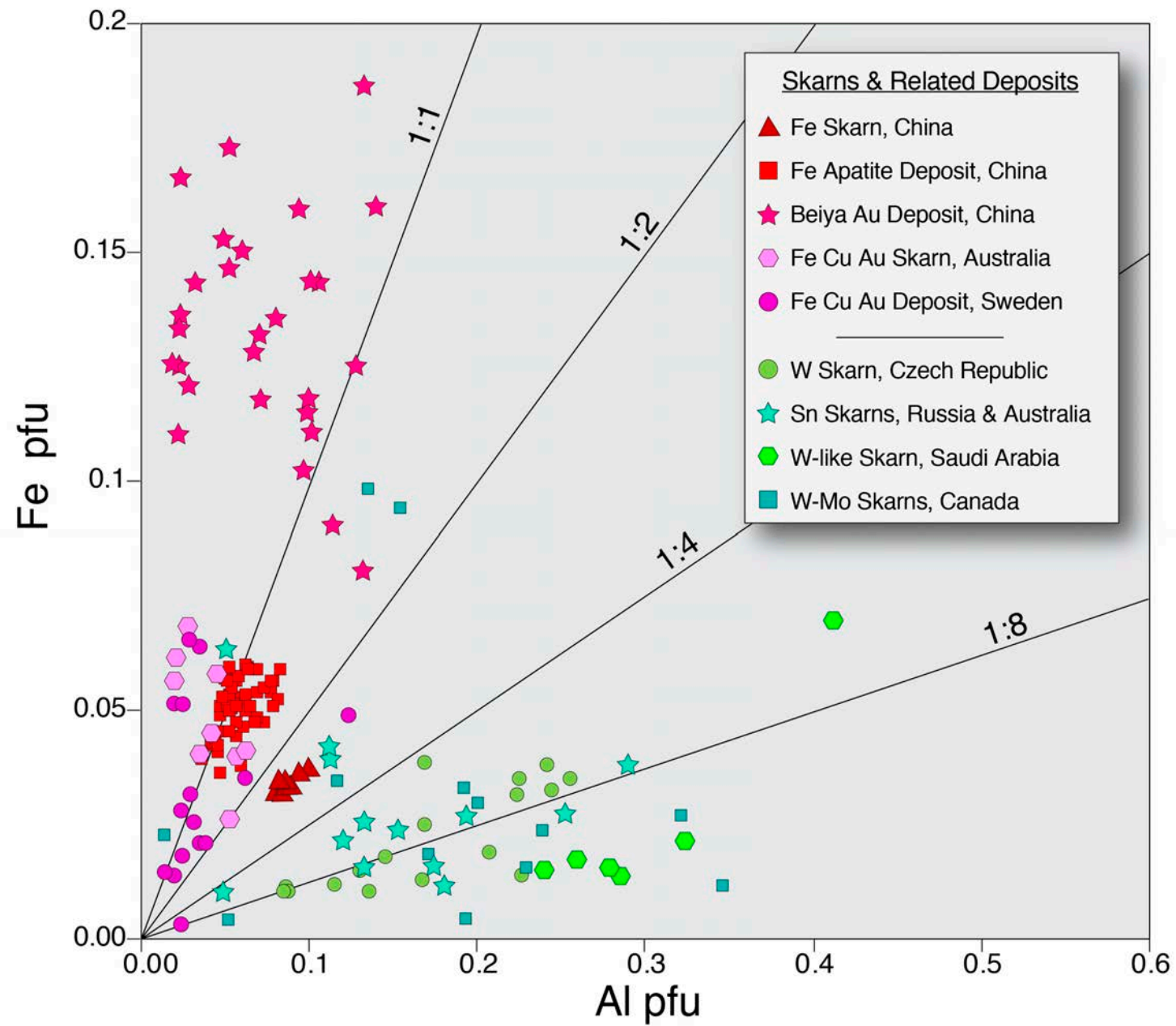


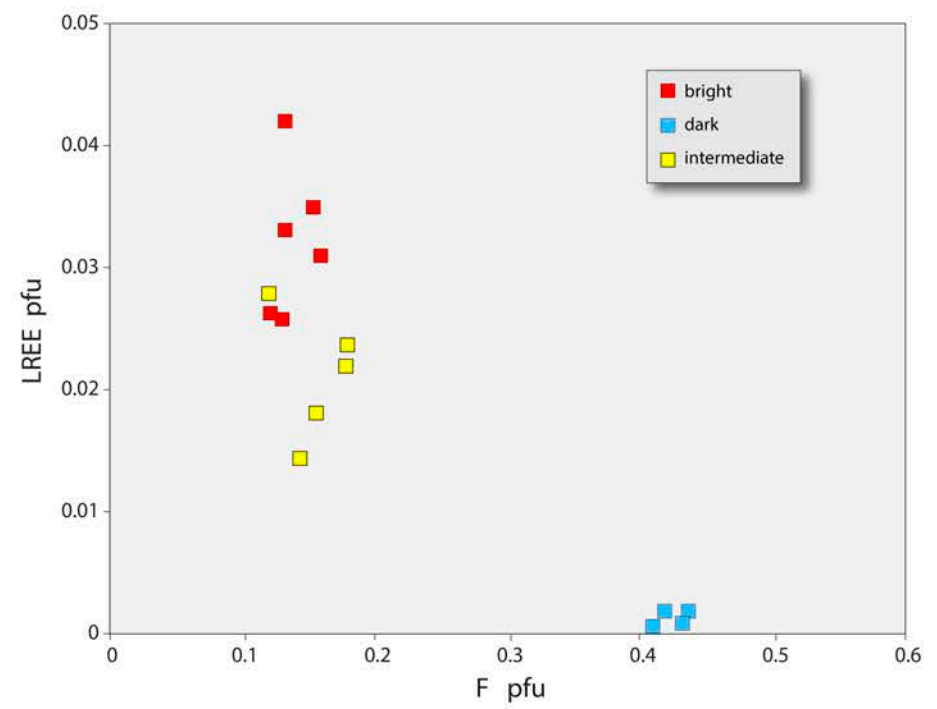
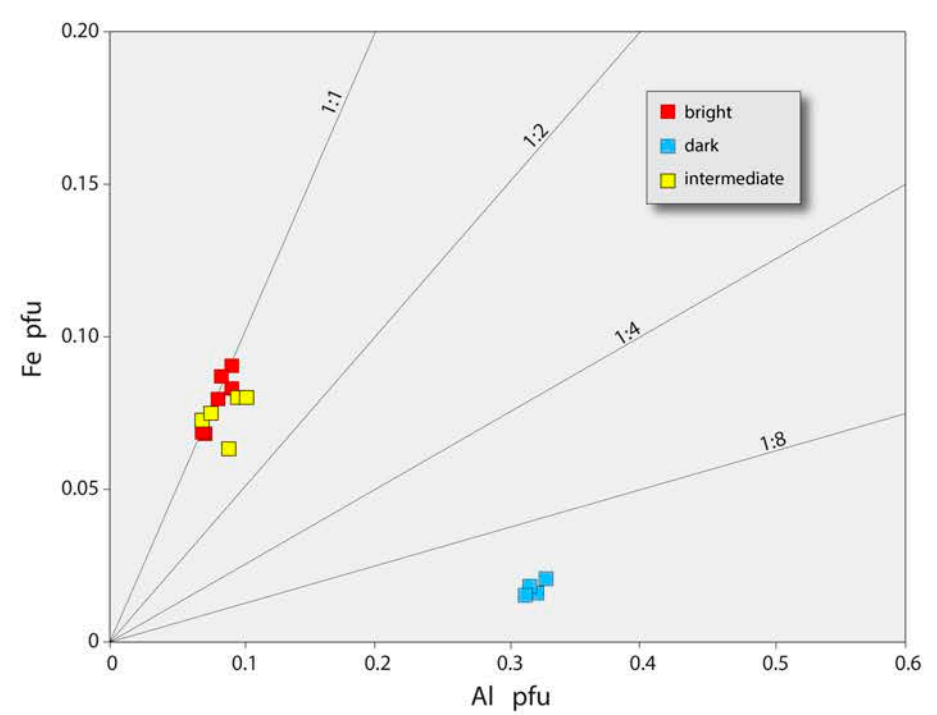
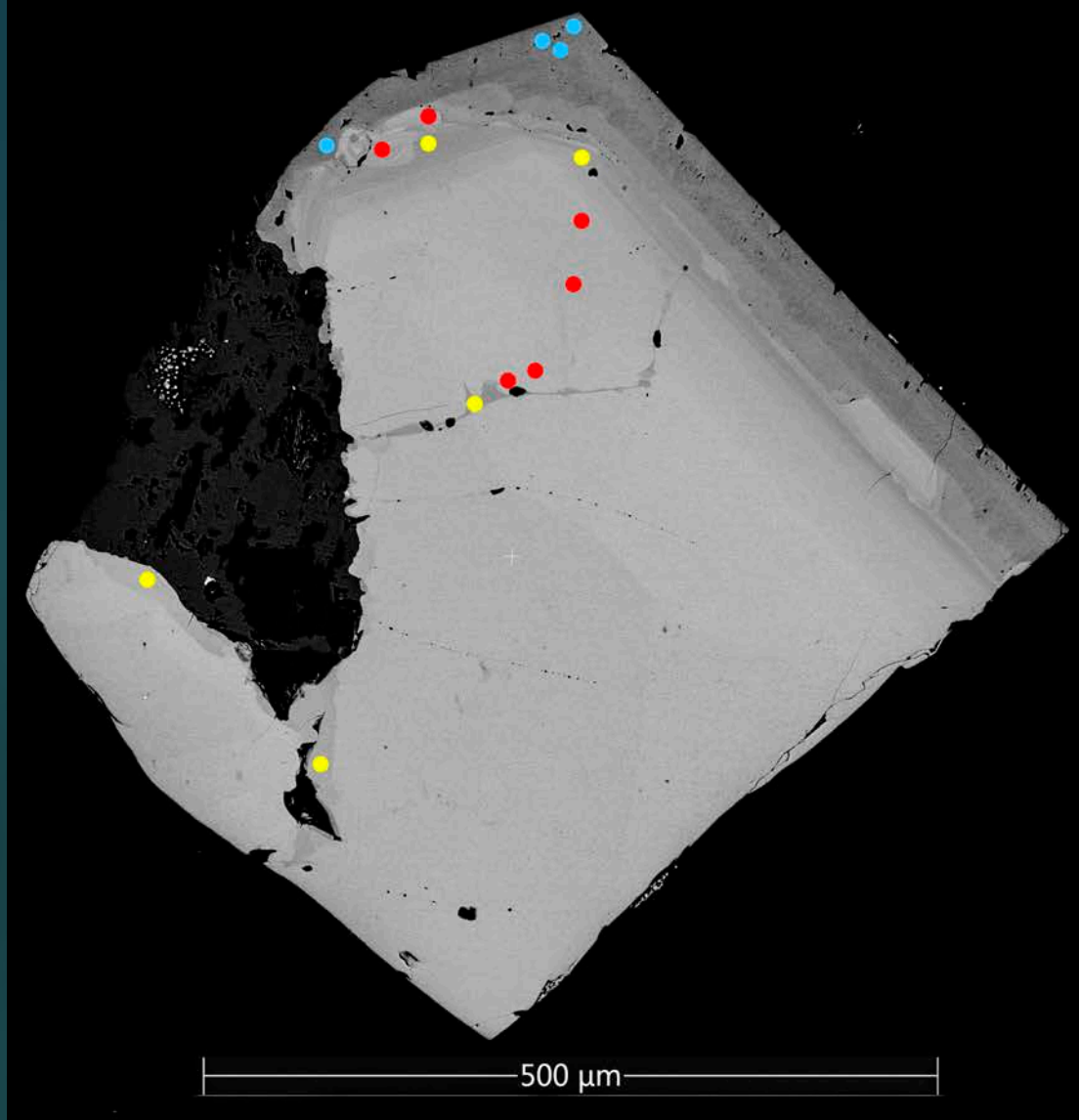


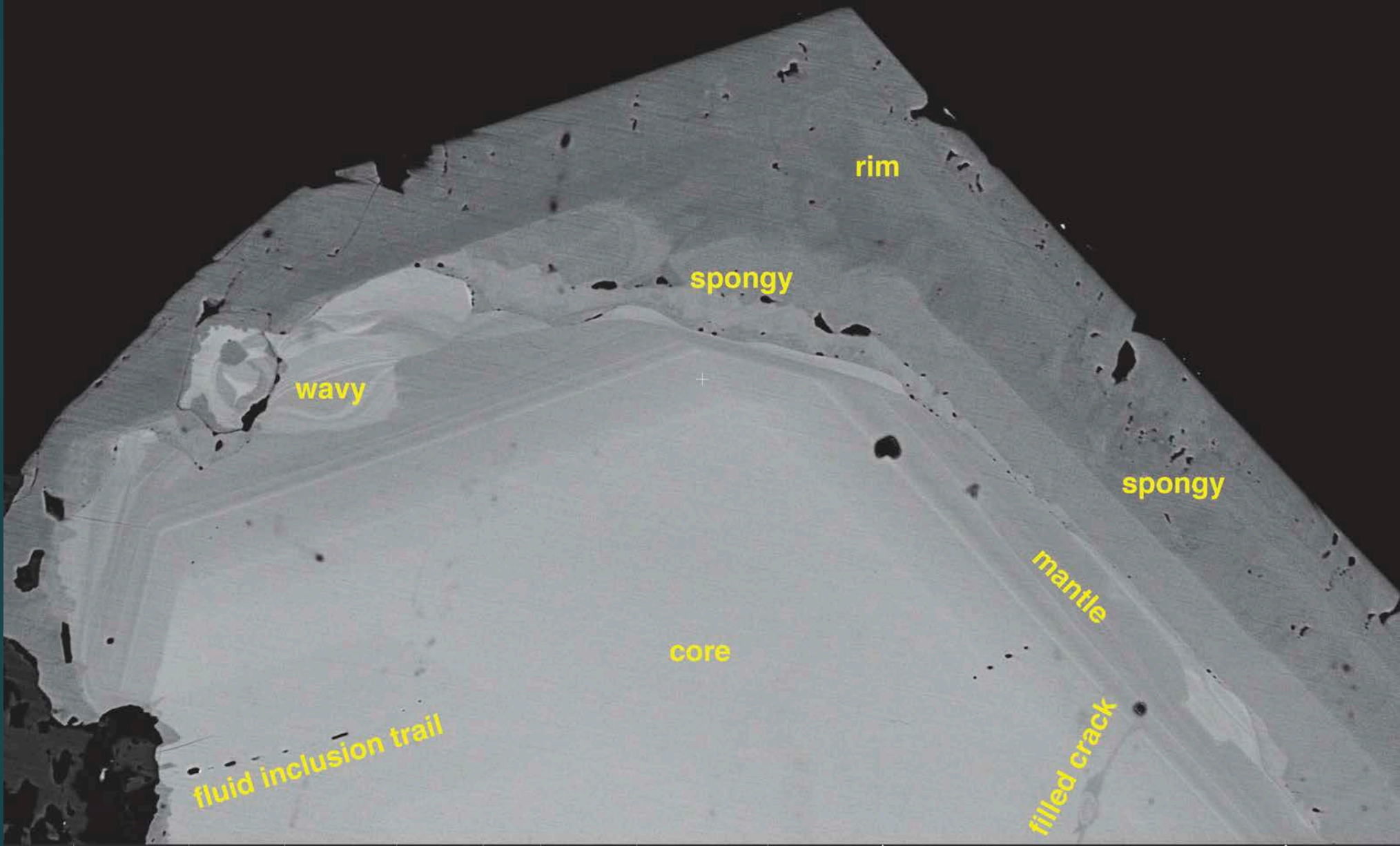


Fe versus Al in Titanite









use case
Standard

WD
9.8 mm

HV
15.00 kV

curr
3.2 nA

det
ABS

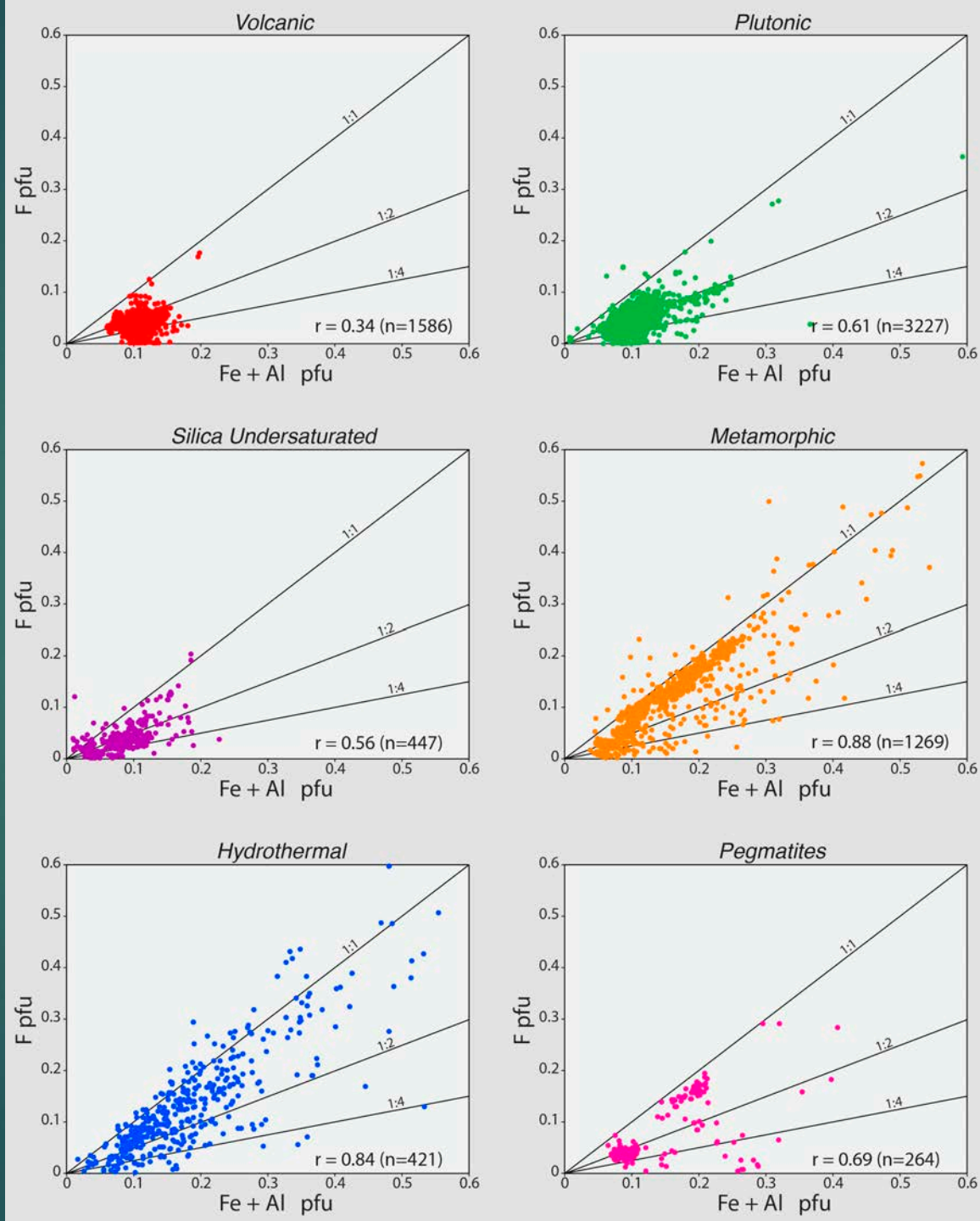
mag ☐
1 200 x

4/25/2018
9:41:10 AM

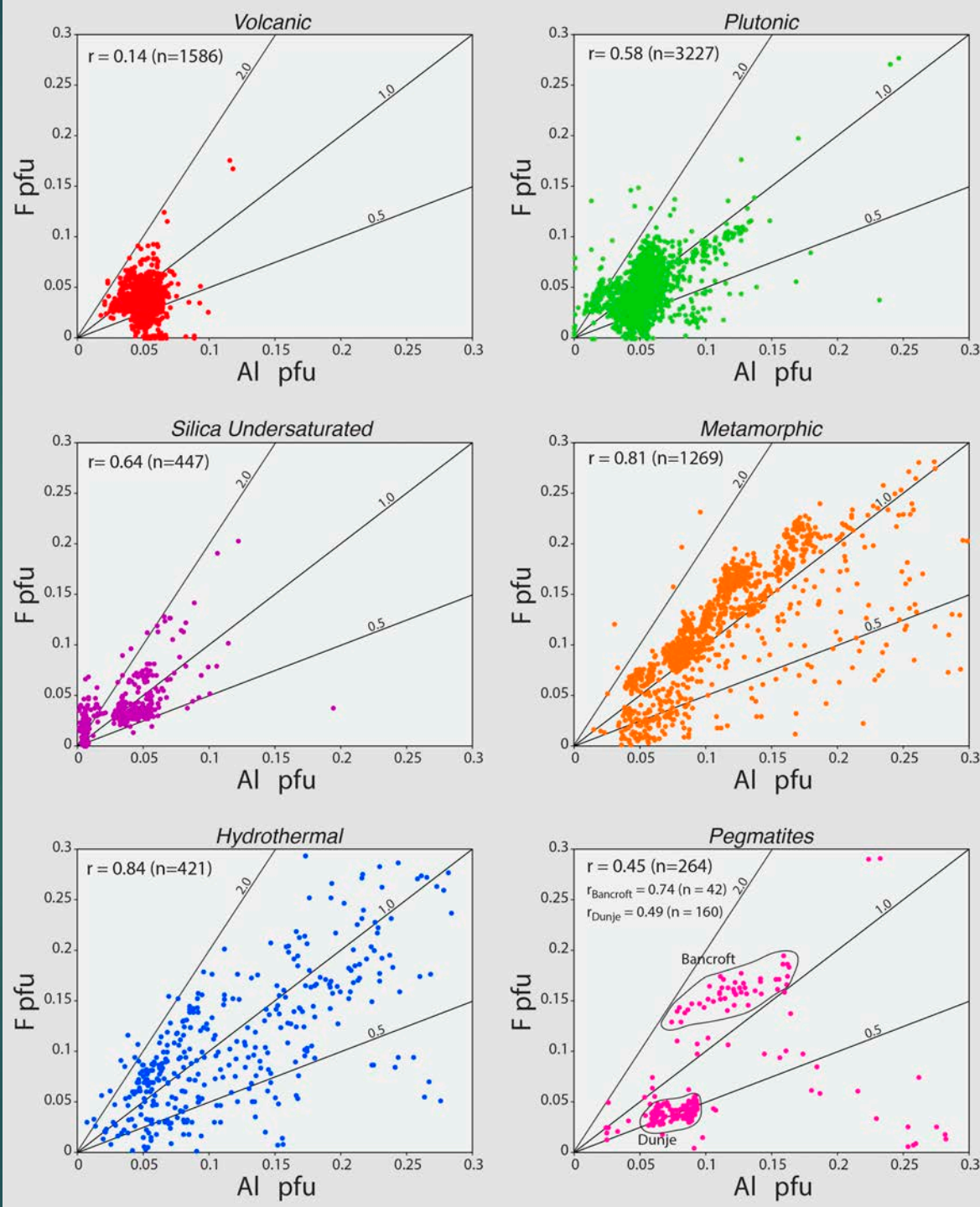
100 μm

Apereo

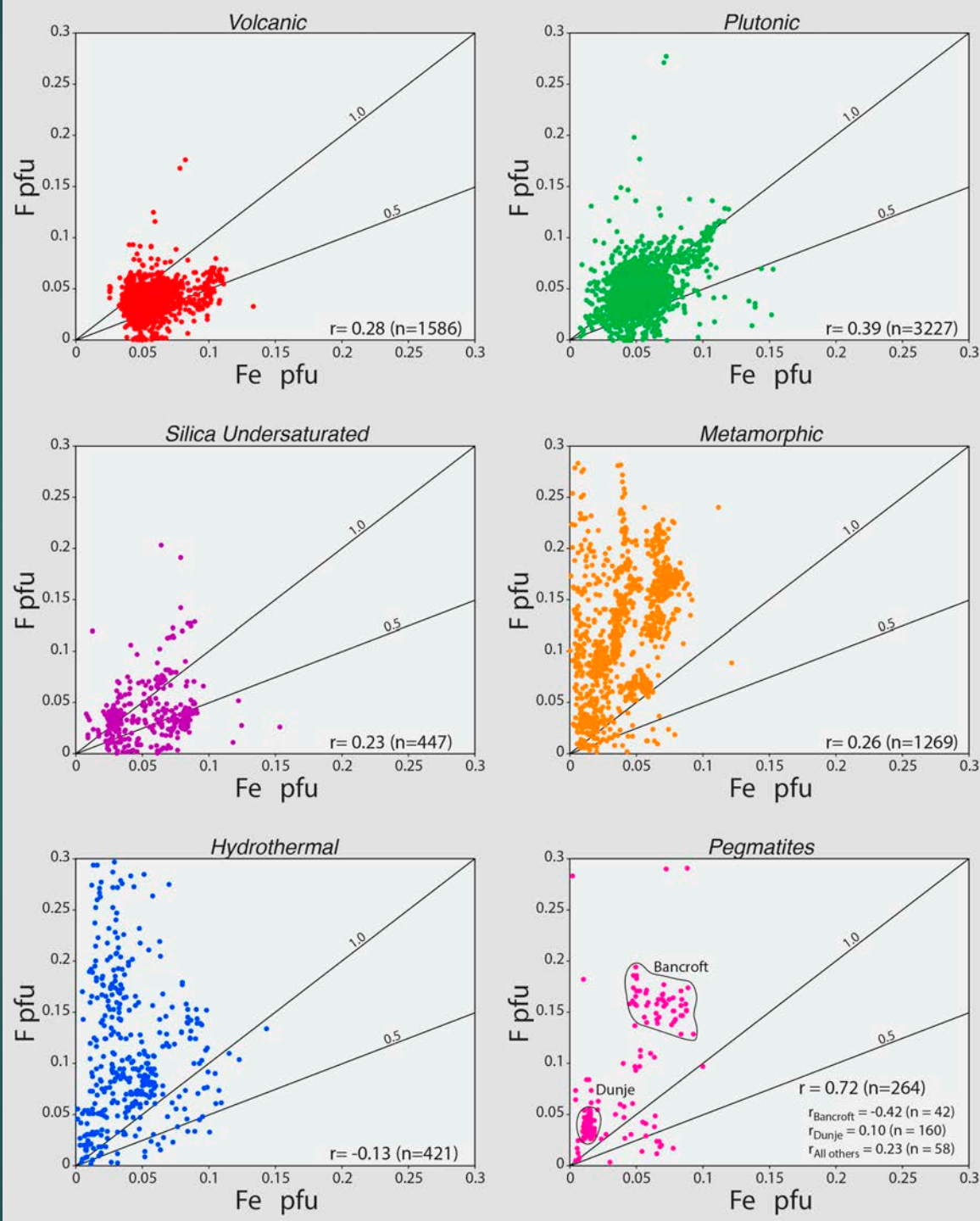
Fe + Al versus F in Titanite

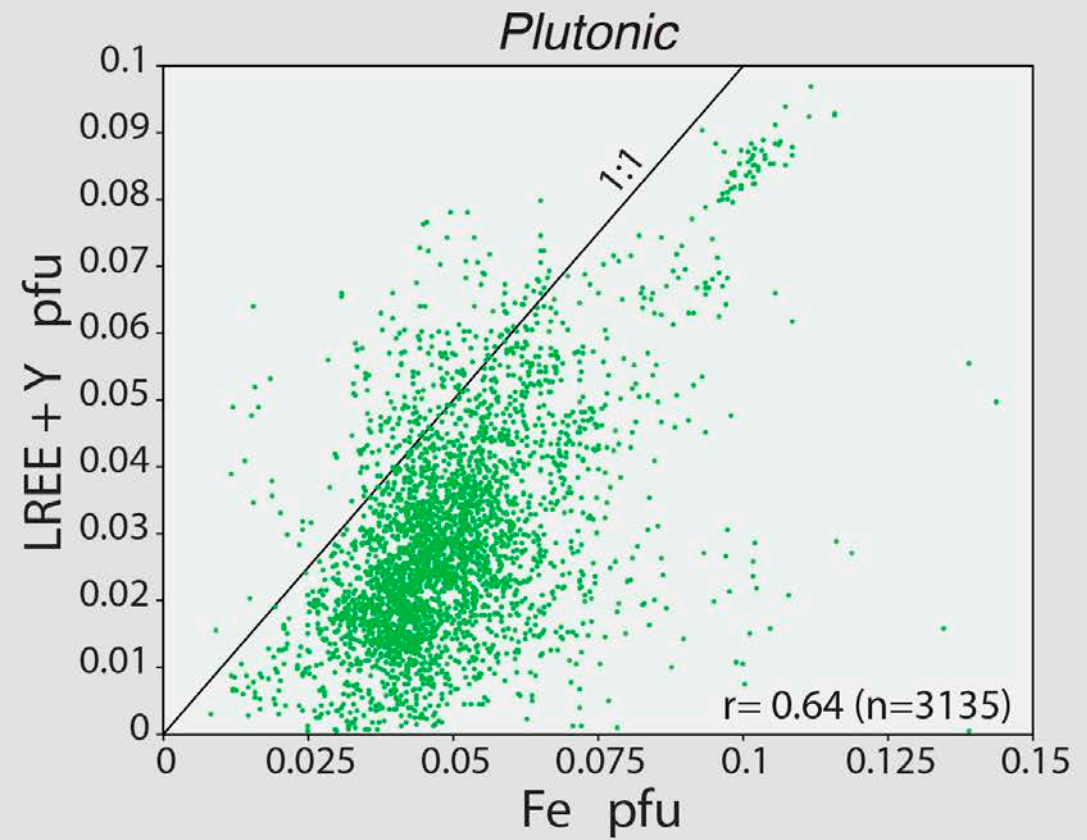
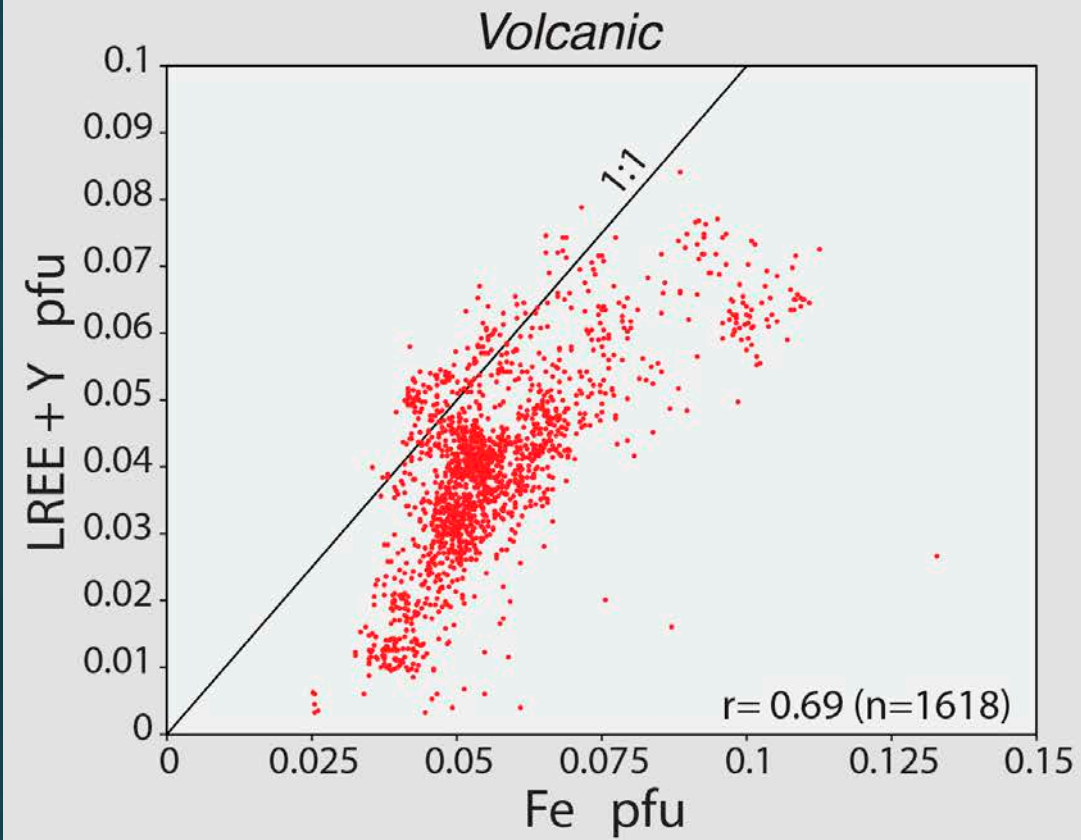


Al versus F in Titanite



Fe versus F in Titanite





Conclusions – Fe/Al ratios

- ▶ The atomic ratio of Fe/Al in titanite from both volcanic and plutonic rocks is typically close to 1:1 and almost always $>1:2$.
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