



Effect of Calcium and Bicarbonate on Iron Removal During Groundwater Treatment

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Where is the Netherlands





Drinking water in the Netherlands

Key figures

• Population	17 mil
• Total production volume	1,126 mil m ³
• Network length	119,000 km
• Annual investments	€431 mil.
• NRW	5.5 %
• Connected	99.9 %

Sources

Ground water	2/3
Surface water	1/3

Treatment

Ground water: Aeration, filtration, softening

Surface water: Extensive treatment

Distribution No Chlorine



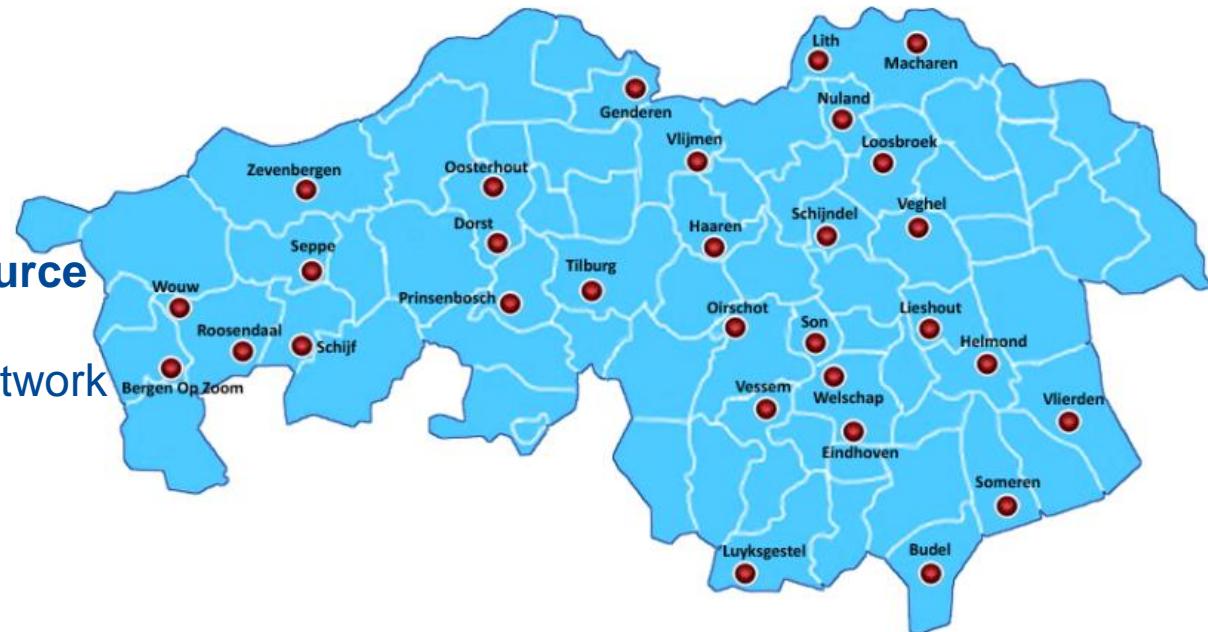


Drinking water company Brabant Water

BRABANT WATER N.V.

- 2.5 mil. inhabitants
- 1.1 mil. connections
- 180,000 regional industries
- Annual production 176 mil. m³
- 30 treatment plants
- **Groundwater as the main source**
- **No chlorine**
- 18,000 km main distribution network
- Non Revenue Water 2.5%
- 800 Staff
- Annual turnover €200 mil.

Province North-Brabant



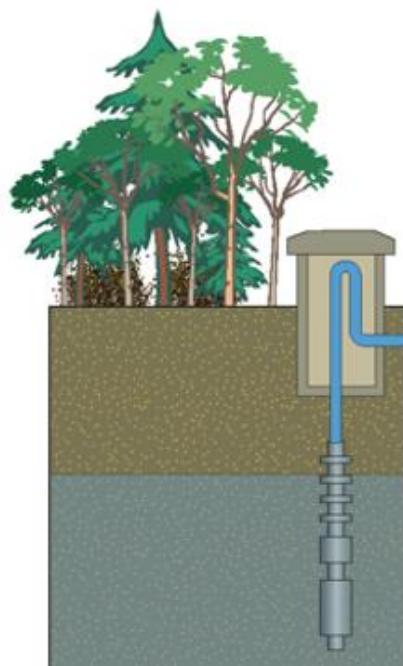
Water use 31.7 USgal (120 l) per person per day



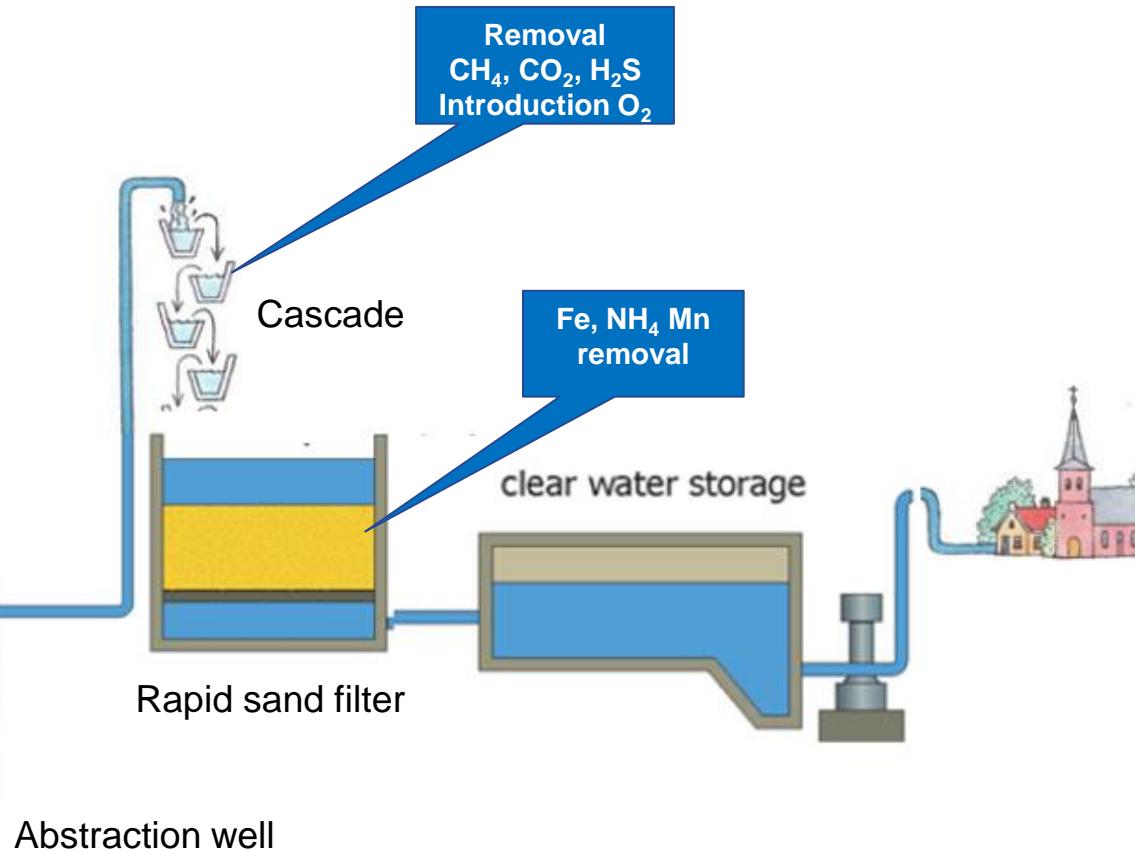
Typical treatment schematic

General design rapid sand filtration

- Filtration rate 3 – 14 m/h
- Filter media 0.71 – 1.25 mm
- Surface area 25 – 30 m²
- Bed height 2.0 m
- Running times 35 – 200 hours
- Backwash speed 35 m/h
- Water loss 1.5 – 4 %



Ground water treatment





The city of Eindhoven

Province North-Brabant





The city of Eindhoven



ASML



Treatment plant Eindhoven

Figures

- Founded 1904
- Inhabitants 420,000
- Annual production 19 mil m³ / year
- Max production 25 mil m³ / year
- Production capacity 4000 m³ / hour
- Distribution capacity 8200 m³ / hour
- Storage 27,000 m³



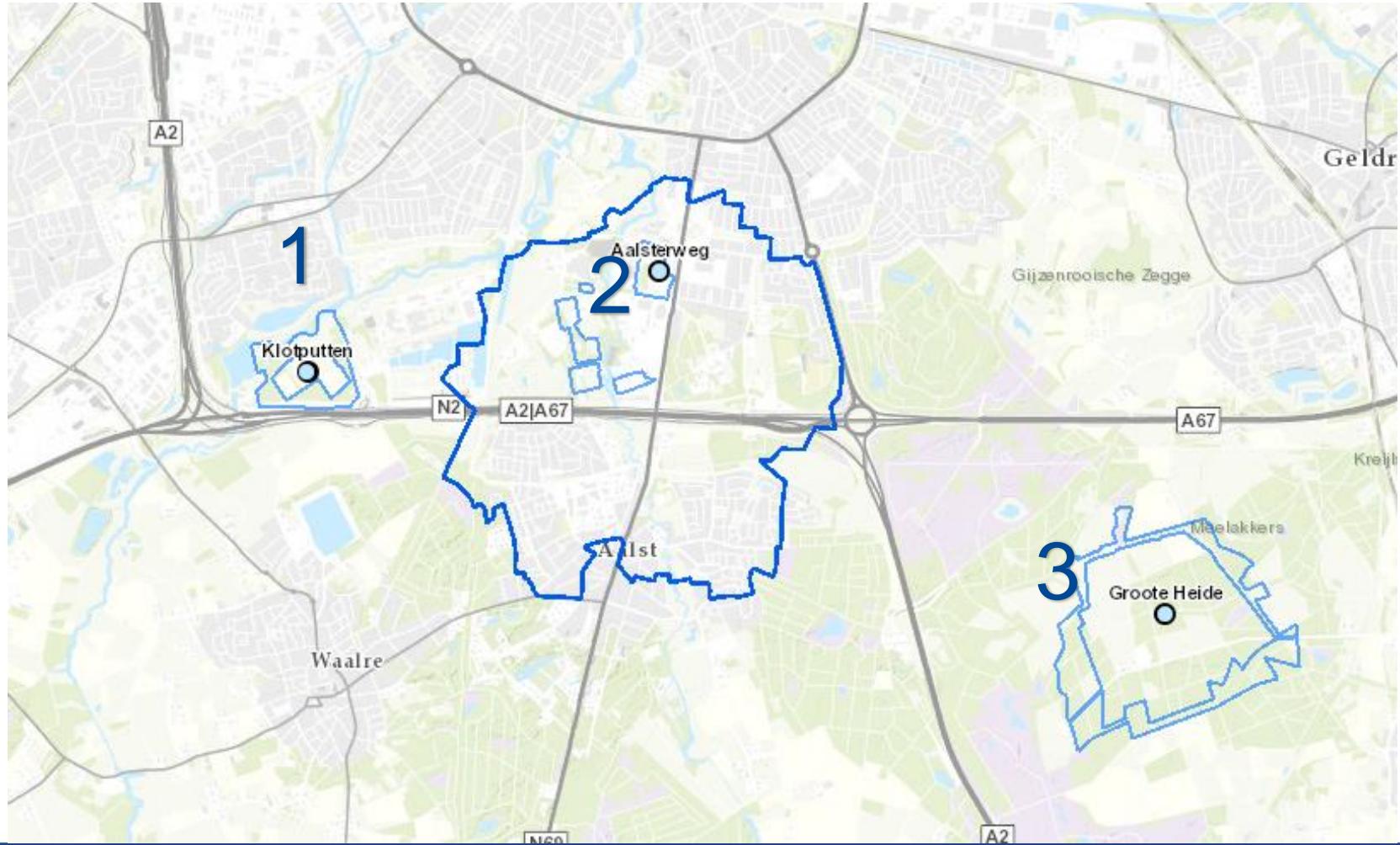


Eindhoven Water tower



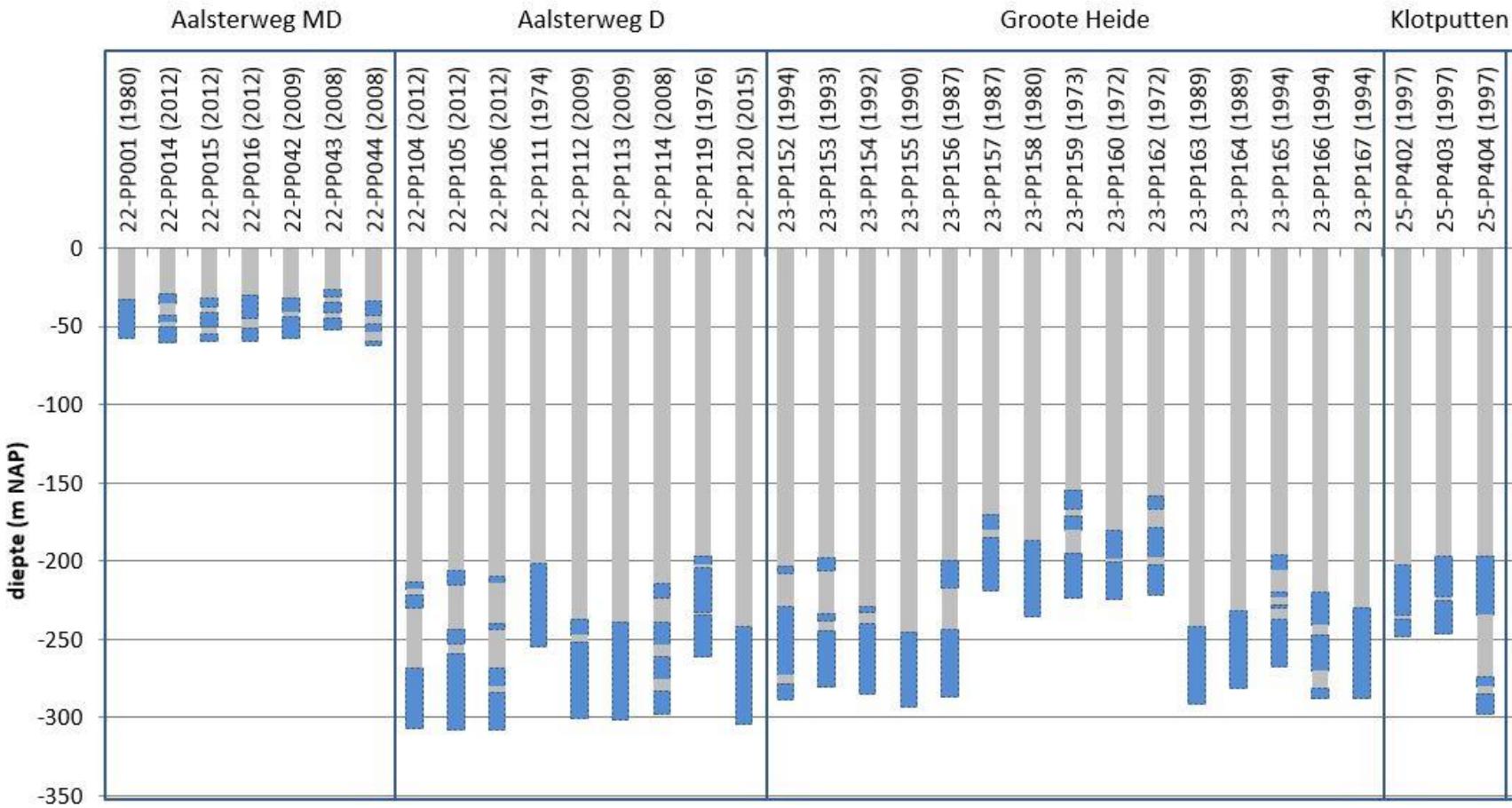


Three ground water borehole fields



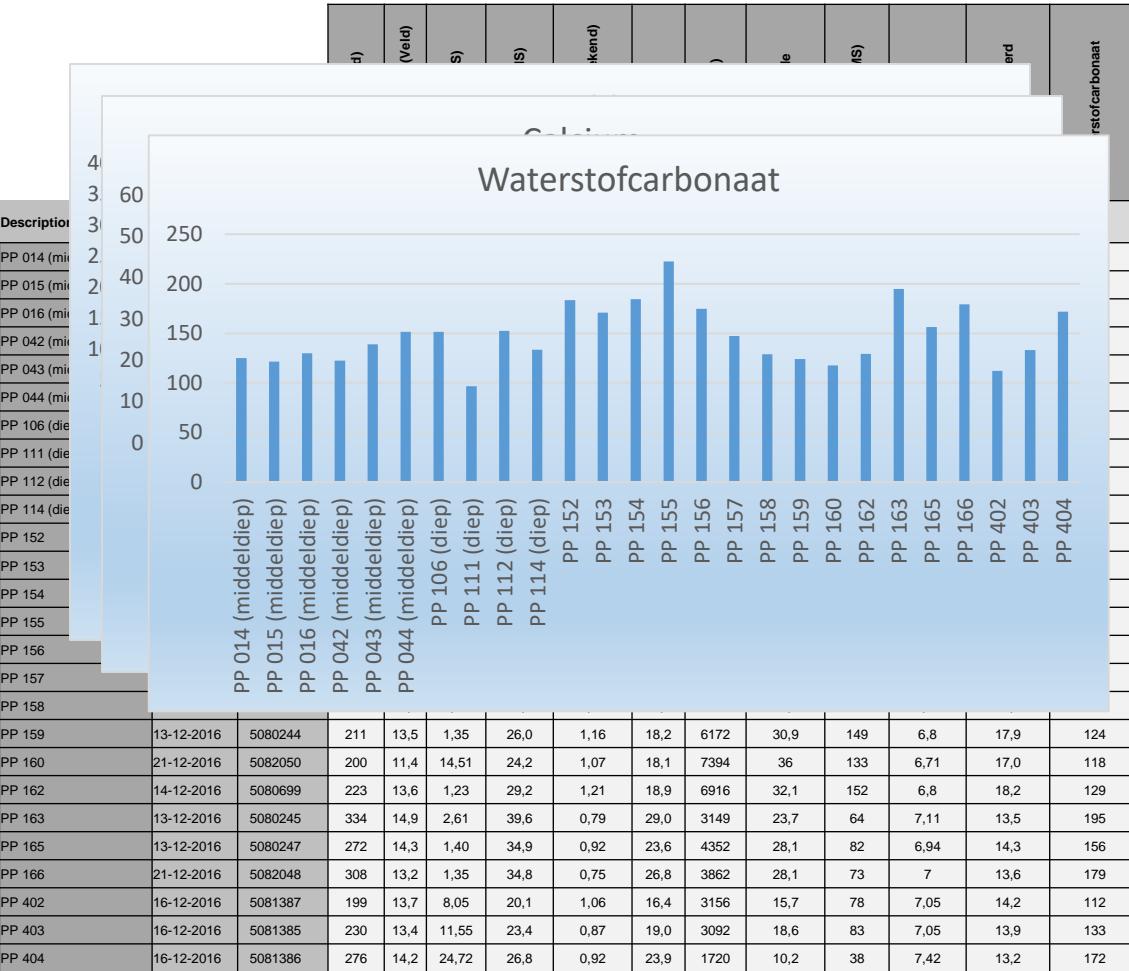


Abstraction depths



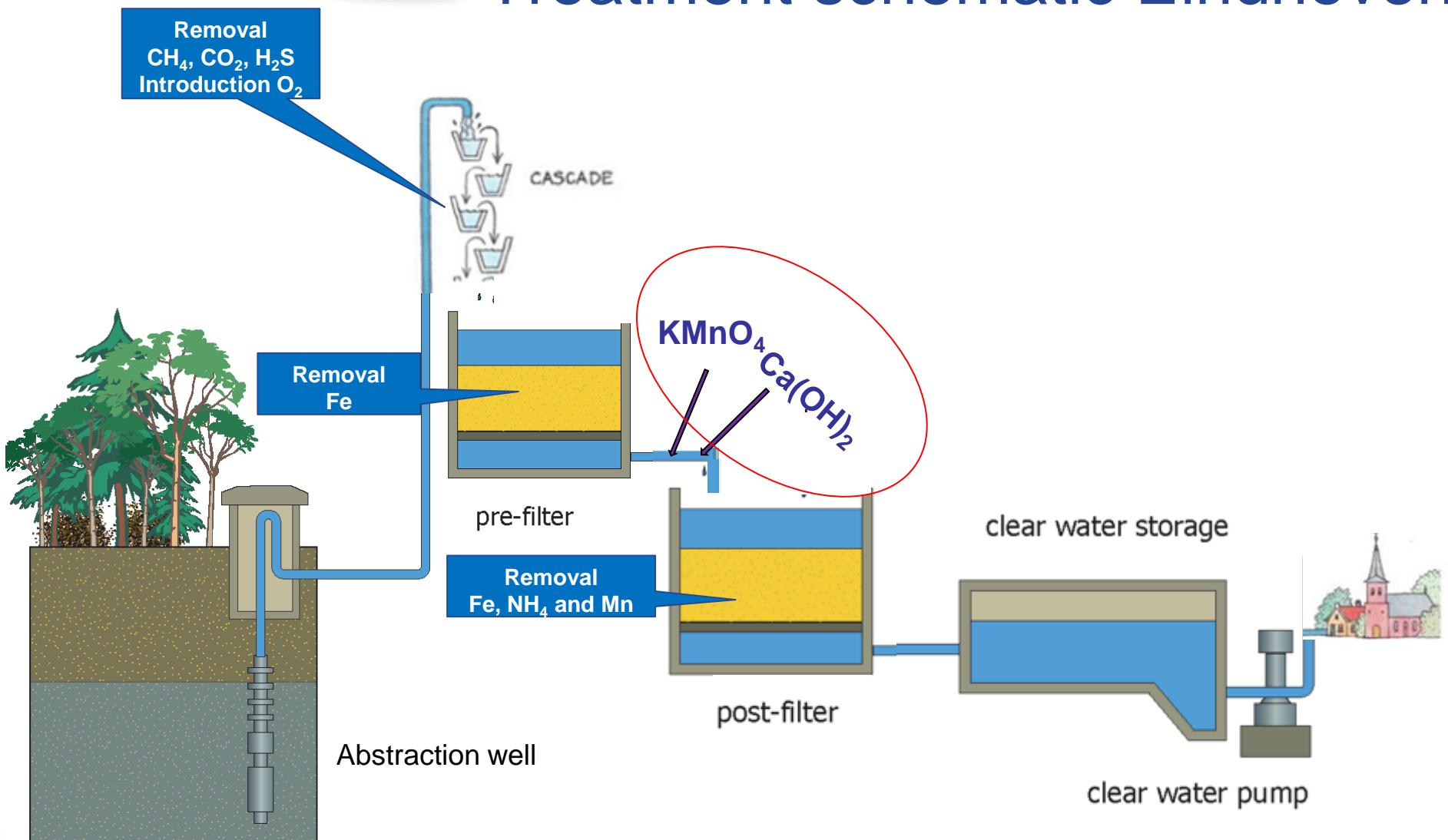


Variation in raw water quality





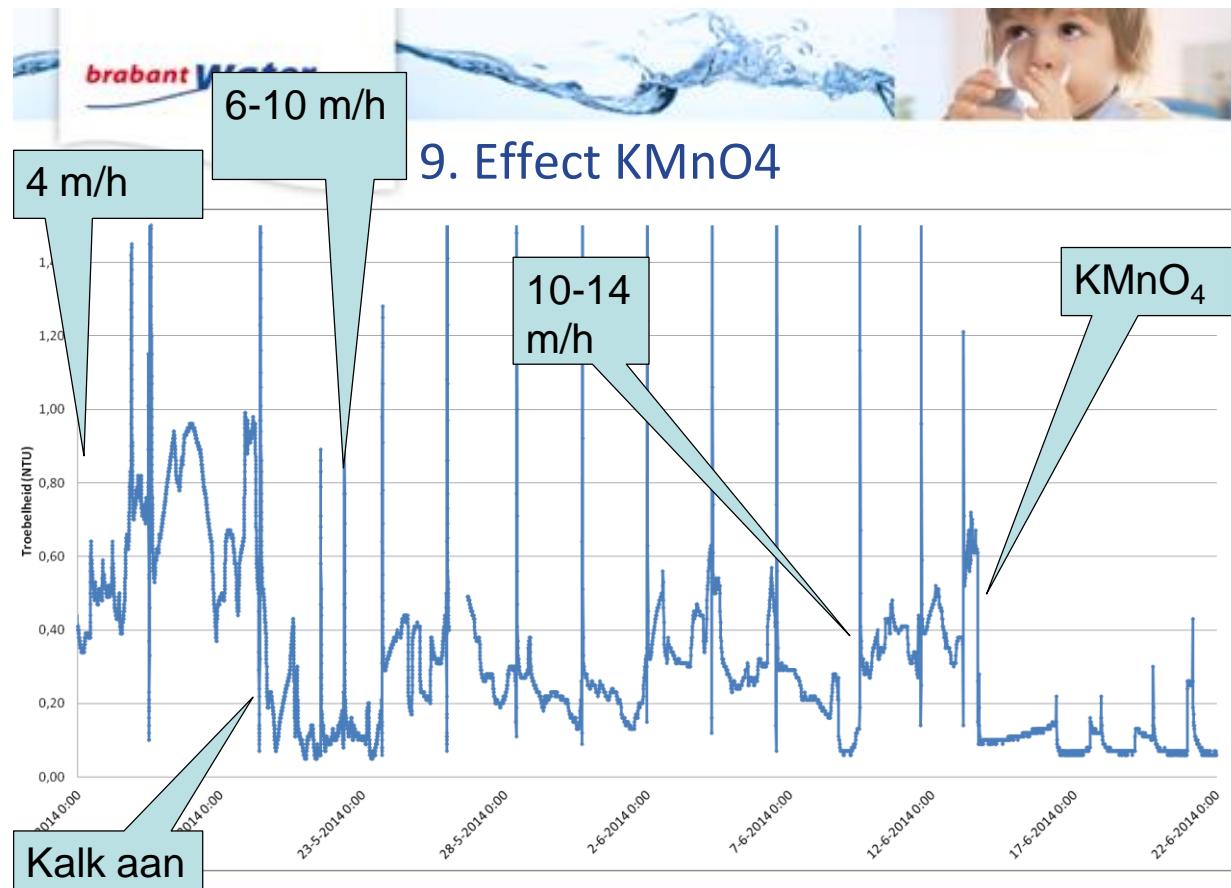
Treatment schematic Eindhoven





Existing treatment

Water quality without chemicals





Water quality Eindhoven 2015

Random raw water quality treatment

Parameter	Eenheid	Aantal metingen	Bedrijfsnorm	Gemiddeld	Min	Max
Troebelheid	NTU	44	<0.5	< 0.10	<0.10	0.37
IJzer	µg/l	44	<50	<10	<10	10
Mangaan	µg/l	44	<20	<10	<10	<10
Arseen	µg/l	16	<1	0.84	0.70	0.99
Ammonium	mg/l	44	<0.05	<0.03	<0.03	0.03
Nitriet	mg/l	44	<0.05	<0.01	<0.01	<0.01
Totale hardheid	mmol/l	4	< 2.0	1.16	1.10	1.23
Kol 22	kve/100 ml	44	<100	2	0	42
Aeromonas	kve/100 ml	44	<20	0	0	0
Zuurgraad	pH-eenheden	44	7,3 > pH < 9	8.03	7.92	8.15



Pilot research

- The renovation of the treatment in Eindhoven is planned for 2020
- Understanding role of chemicals existing treatment
- Looking for an alternative treatment
 - Reducing use of chemicals
 - Reducing maintenance treatment



Iron removal in general aeration – filtration treatment

- Design and operating mostly based on experience and rules of thumb
- pH is widely regarded as a major parameter that determines the iron removal efficiency
- The ionic composition of water may also strongly affect iron removal, especially during the flocculation phase



Research goal

Understanding the important role of calcium (Ca) and bicarbonate (HCO_3) in iron removal during aeration—rapid sand filtration



Pilot set up

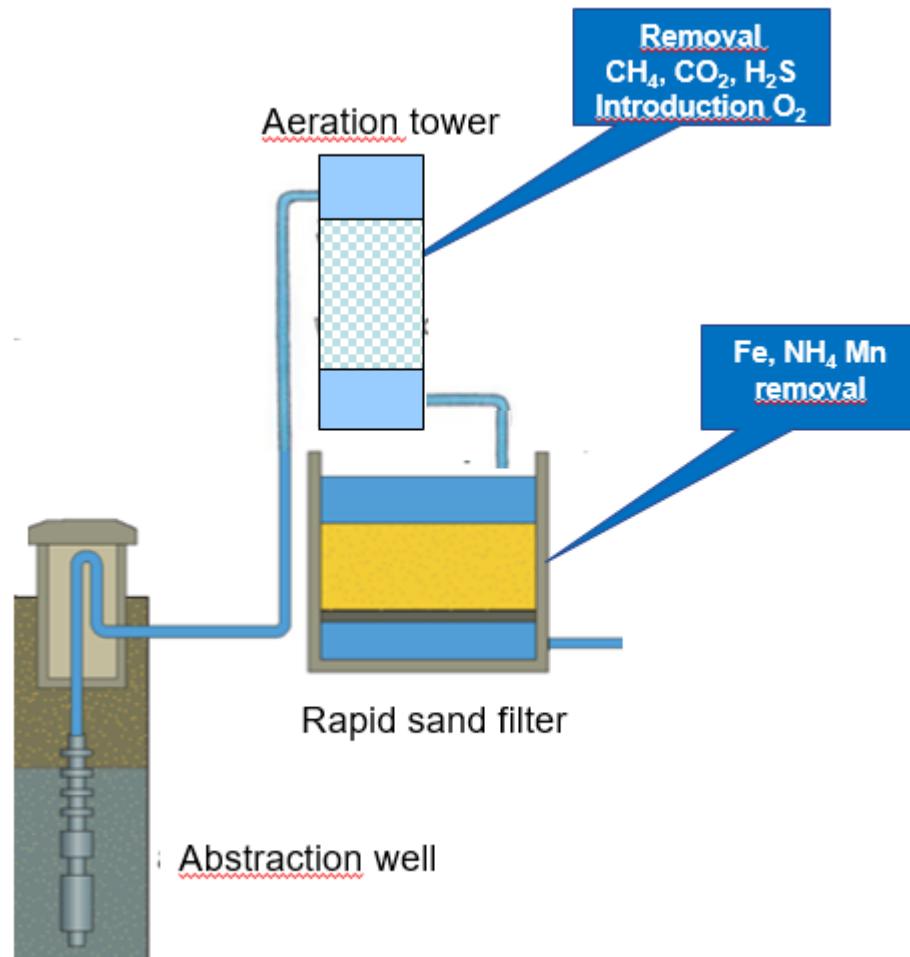
Aeration tower

- Filtration rate 60 – 120 m/h
- Filling material 25 mm plastic Pall ringen
- Bed height 2.0 m
- Surface area 0.27 m²
- Air / water ratio 3



Rapid sand filter

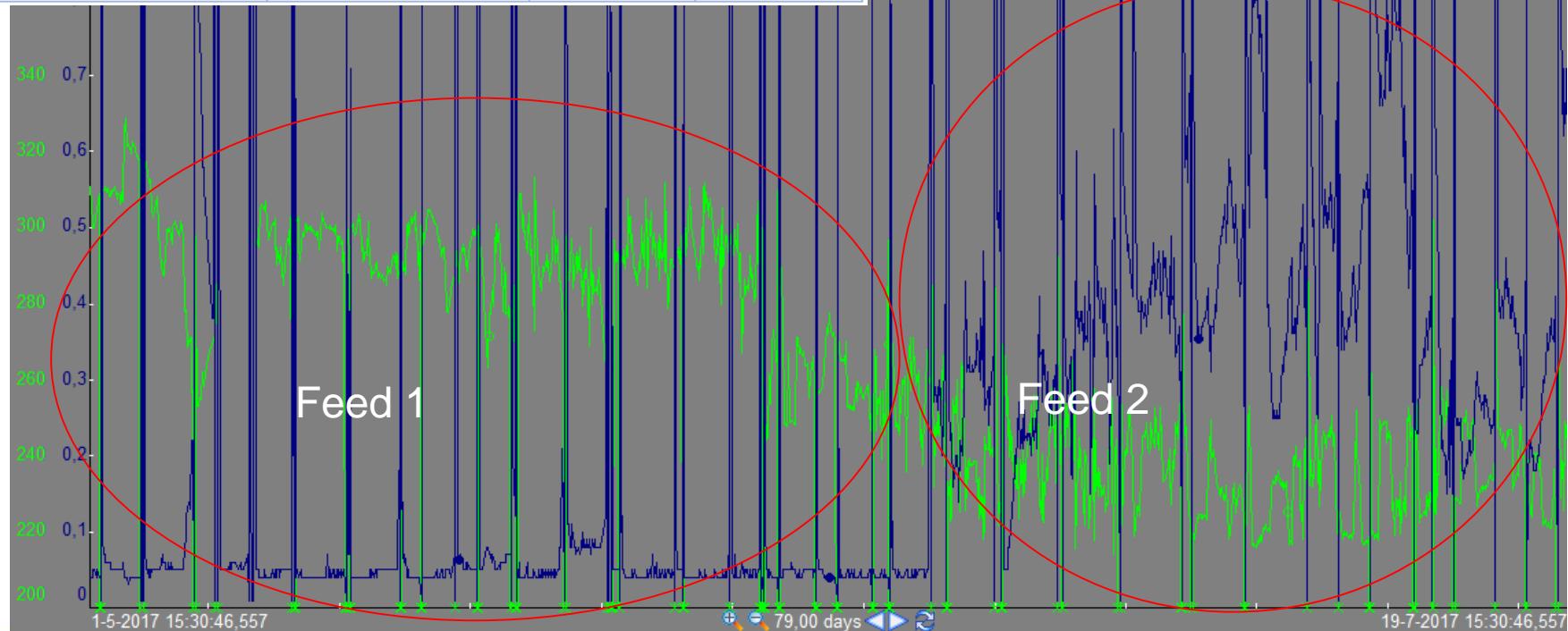
- Filtration rate 3 – 7.5 m/h
- Filter media 0.71 – 1.25 mm
- Surface area 0.8 m²
- Bed height 2.0 m
- Running times 35 – 120 hours
- Backwash speed 35 m/h
- Water loss 1,6%





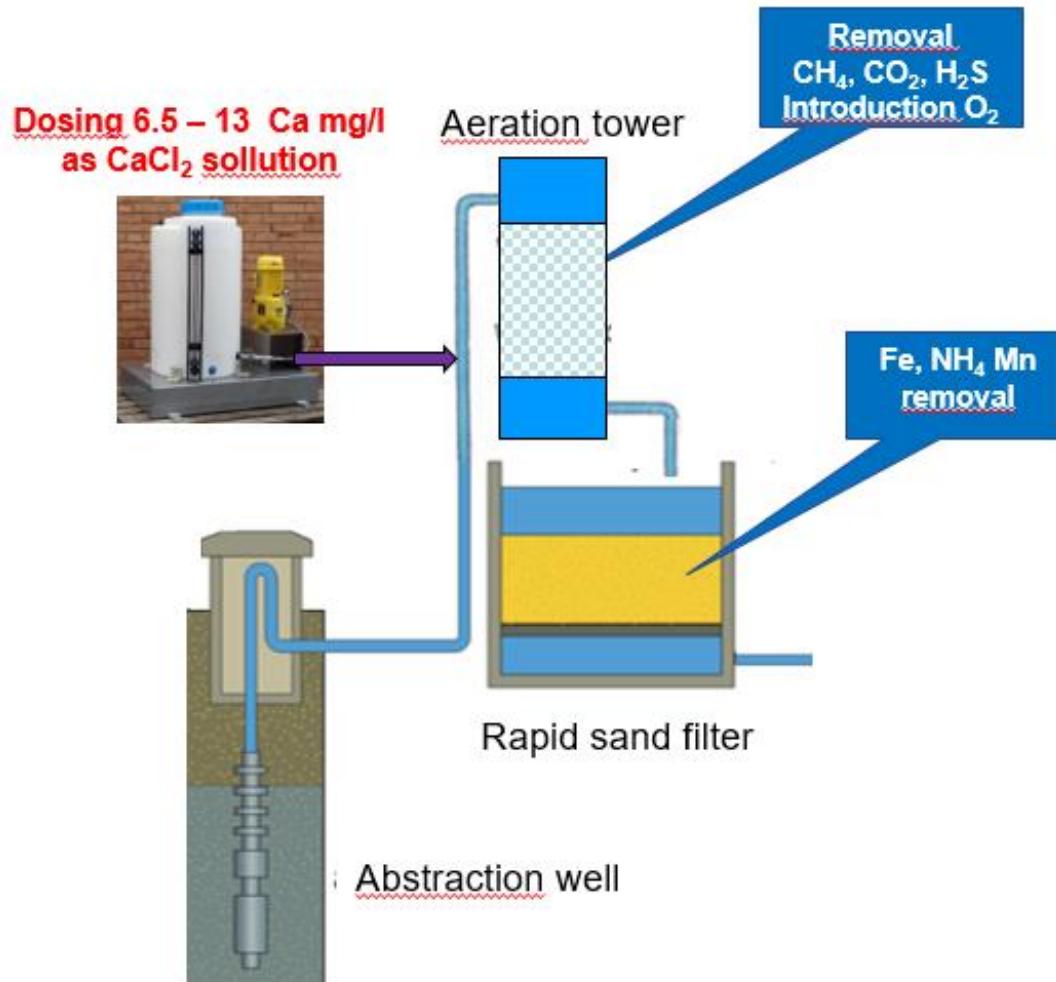
Iron removal as function of water quality

Parameter	Unit	Feed 1	Feed 2
pH		6.96	7.18
Bicarbonate	mg/L HCO ₃	171	139
Electrical Conductivity	µS/cm	260	201
Total iron	mg/L Fe	4392	3456
Dissolved iron	mg/L Fe	4392	3456
Calcium	mg/L Ca	34.4	24.7



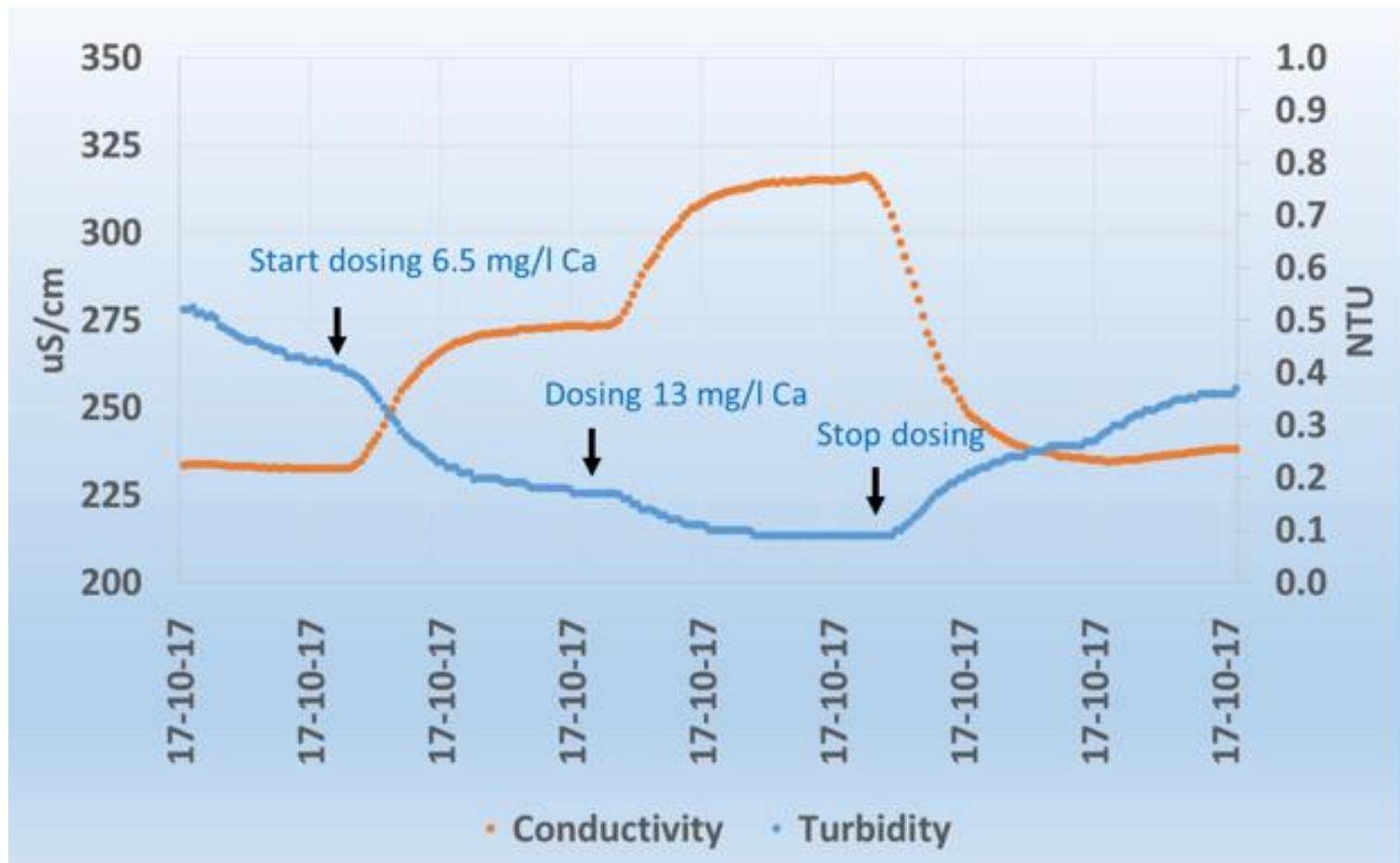


Effect of Calcium dosing



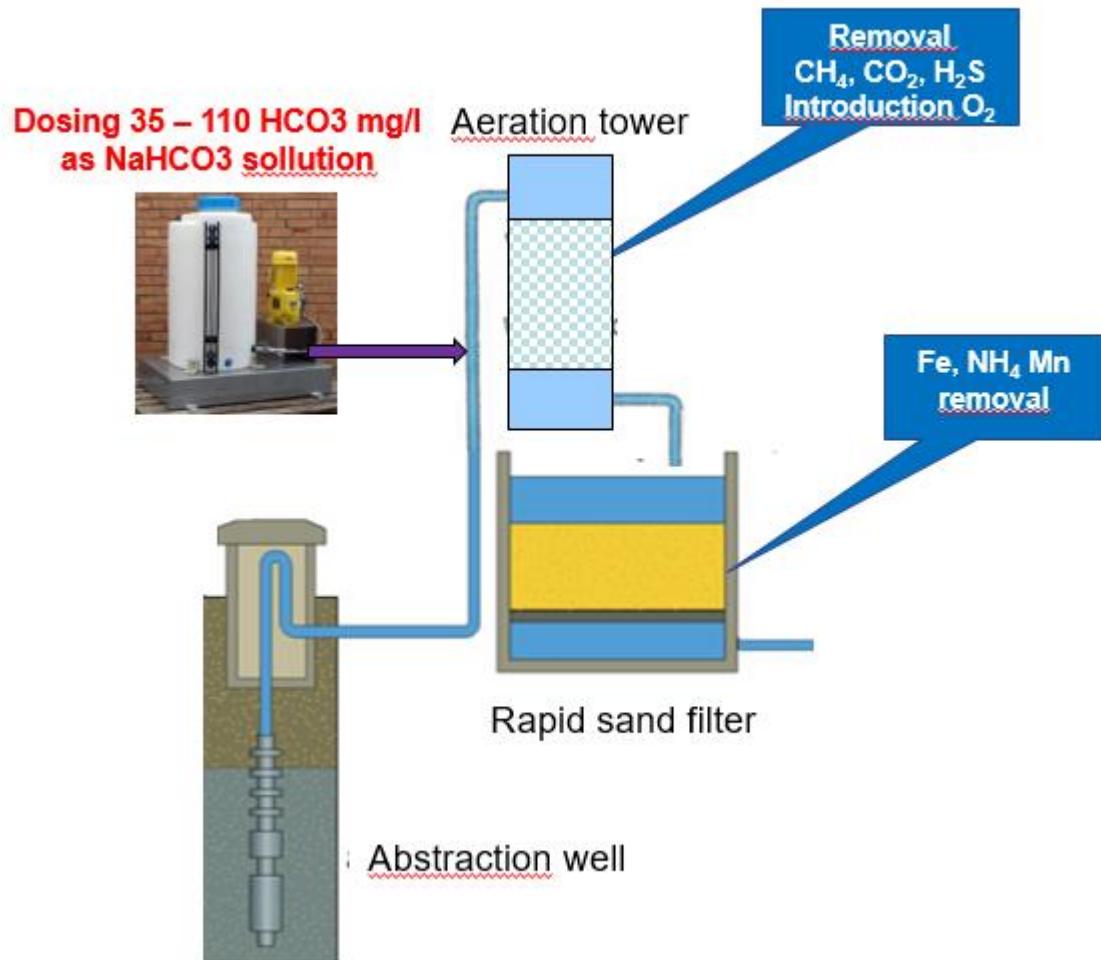


Effect of Calcium dosing



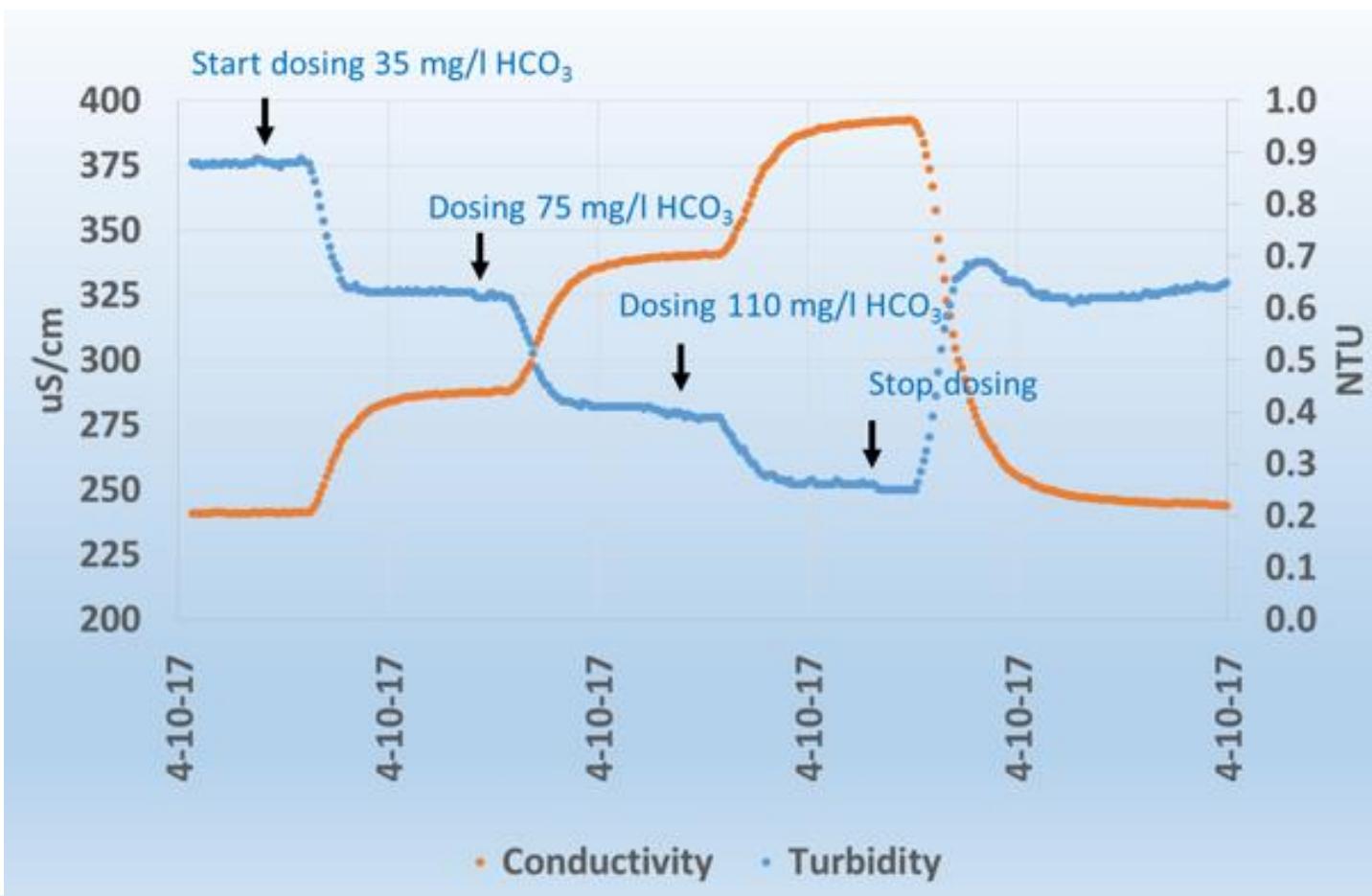


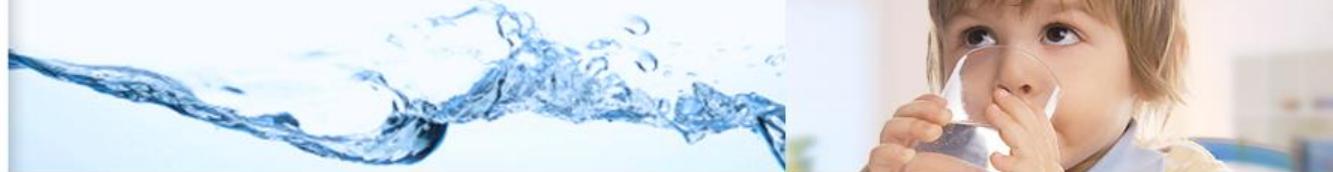
Effect of bicarbonate dosing





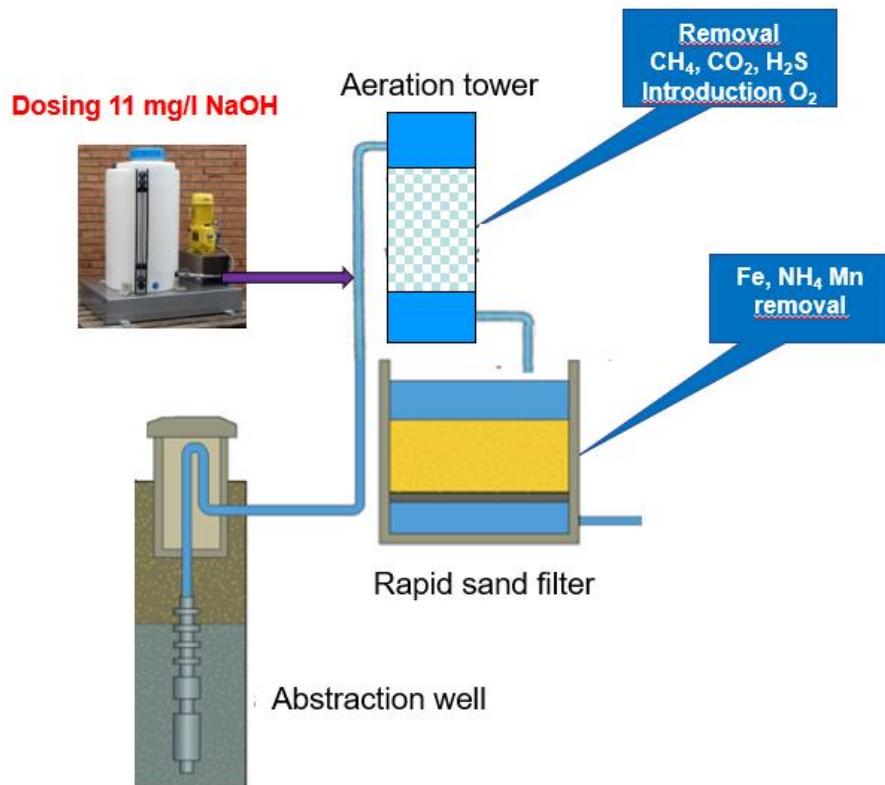
Effect of bicarbonate dosing





Final treatment schematic for Eindhoven

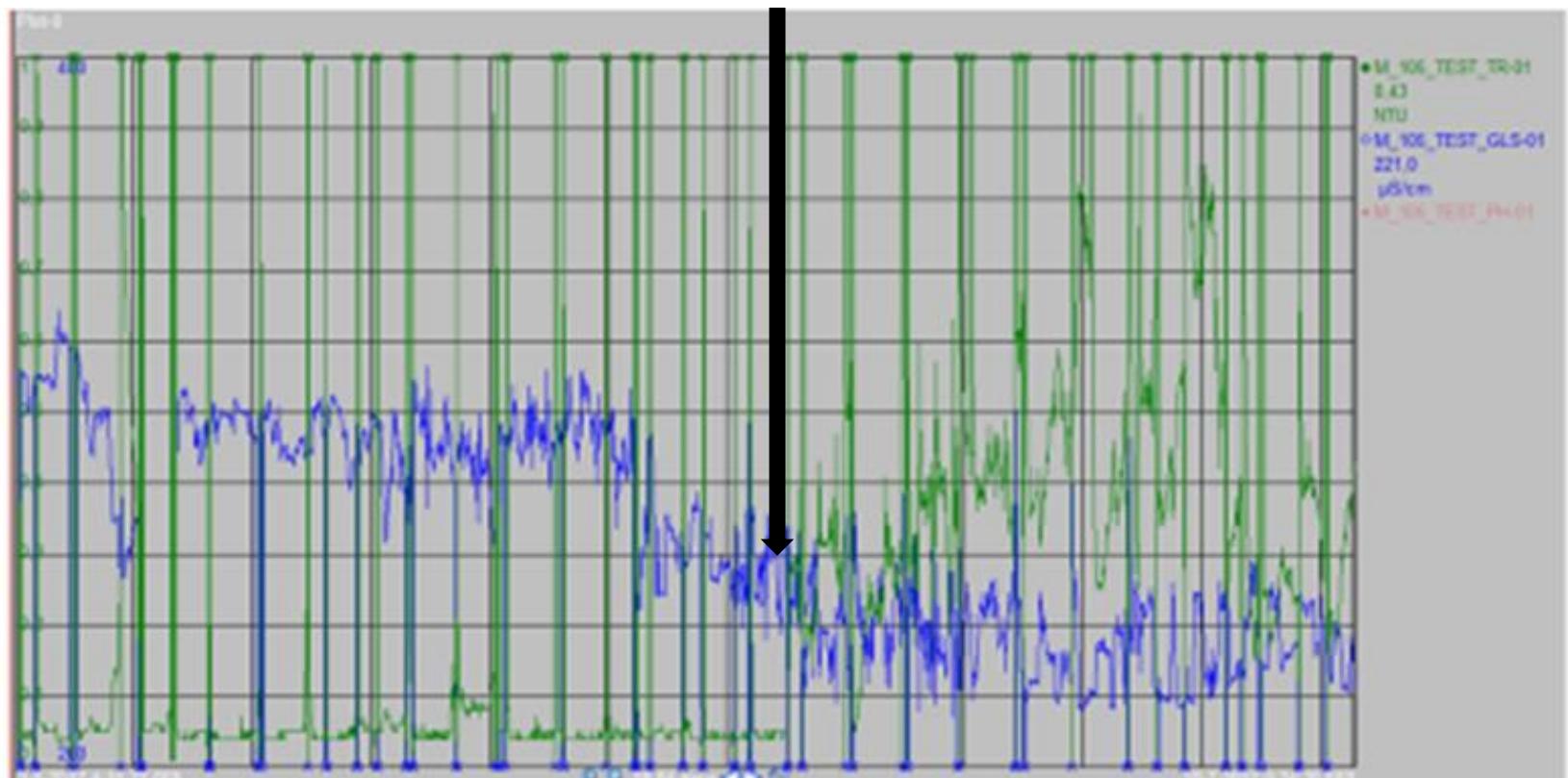
To support iron removal we dosed NaOH instead of CaCl₂ or NaHCO₃. A small increase in HCO₃ was enough for excellent and robust iron removal





Without dosing NaOH

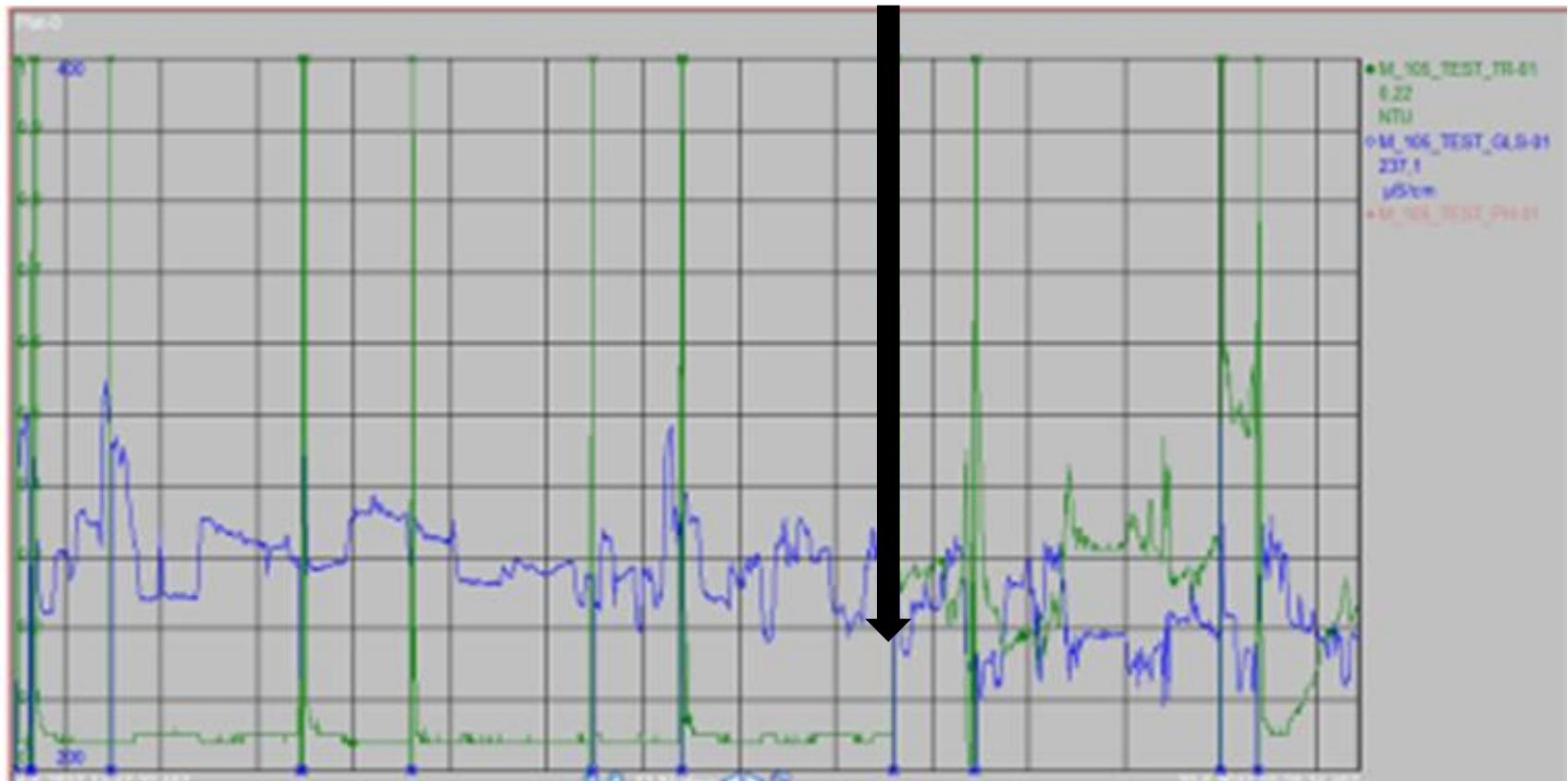
Lower conductivity, higher turbidity

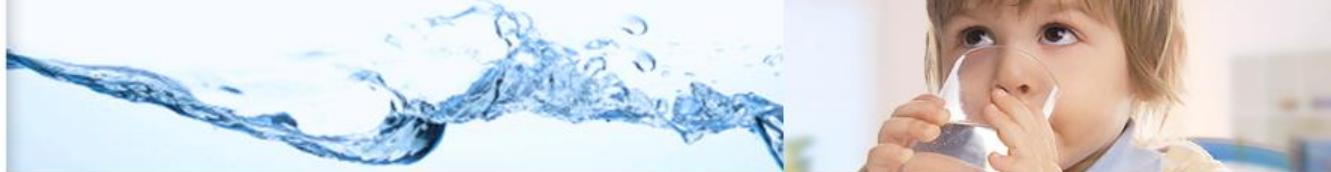




With dosing NaOH

Effect switching of NaOH dosing





Conclusions and outlook

- Major groundwater ions HCO_3 and Ca have a profound impact on the filterability of iron in rapid sand filters. Higher concentrations of these ions in raw water may lead to better removal of iron in rapid sand filters
- These new insights have been already helpful in optimizing iron removal at several other treatment plants of Brabant Water
- The specific role of Ca and HCO_3 during Fe removal in classic rapid sand filtration should be studied further to gain mechanistic insights
- Fundamental research on this topic is started at WETSUS institute
- HCO_3 and Ca as ions are added to the KWR modeling research on iron removal with rapid sand filtration



Thank you for your attention

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This research is published in
Water Matters H₂O:

<http://www.h2o-watermatters.com/>



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