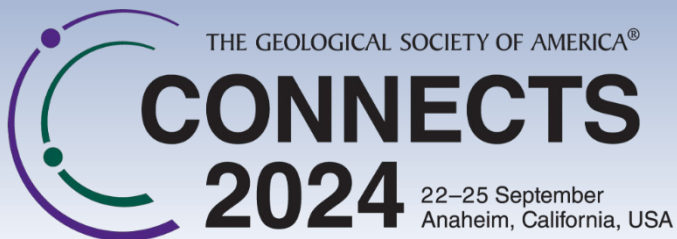


“Visual guides to analyze Microplastics in salts”

María Lería, Vinyet Baqués, Elisabet Playà,
Anna Travé, Irene Cantarero



Drawing to communicate
scientific concepts



MICROPLASTICS IN TABLE SALTS?: How to easily study them (National Geographic Society)

Addressed to society



OBJECTIVES 1/3

- To raise environmental consciousness:
Water and food contamination



Maroc Hebdo



Víctor Barro

OBJECTIVES ^{2/3}

- To raise environmental consciousness:
Water and food contamination
- Promoting Sustainable Development Goals (SDG)



OBJECTIVES 3/3

- Raise environmental consciousness:

Water and food contamination

- Promoting Sustainable Development Goals (SDG)
- To promote and to consolidate collaborations between Art and Geology

METHODOLOGY ^{1/7}

- Protocol- scientific lab



METHODOLOGY ^{2/7}

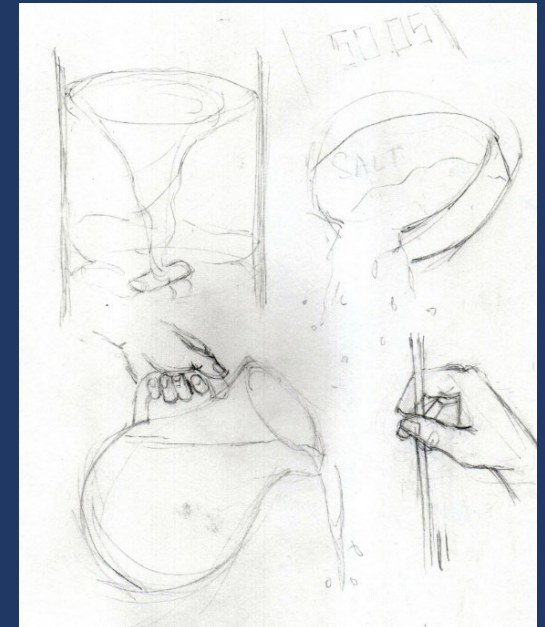
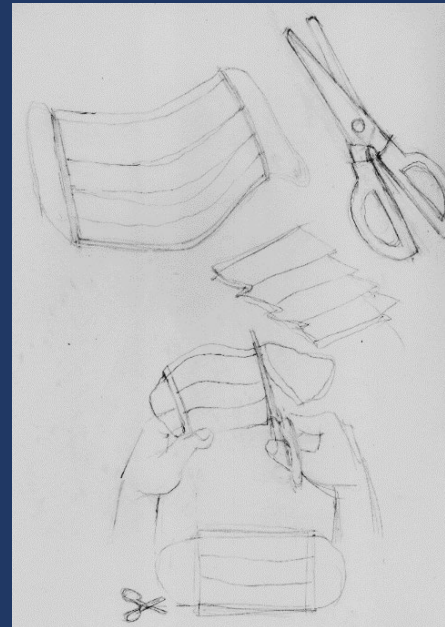
- Protocol- scientific lab
- Drawing protocol
Defining iconicity



METHODOLOGY ^{3/7}

- Protocol- scientific lab
- Drawing protocol

Defining iconicity
Pencil drawing



METHODOLOGY ^{4/7}

- Protocol- scientific lab
- Drawing protocol

Pencil drawing
Watercolor

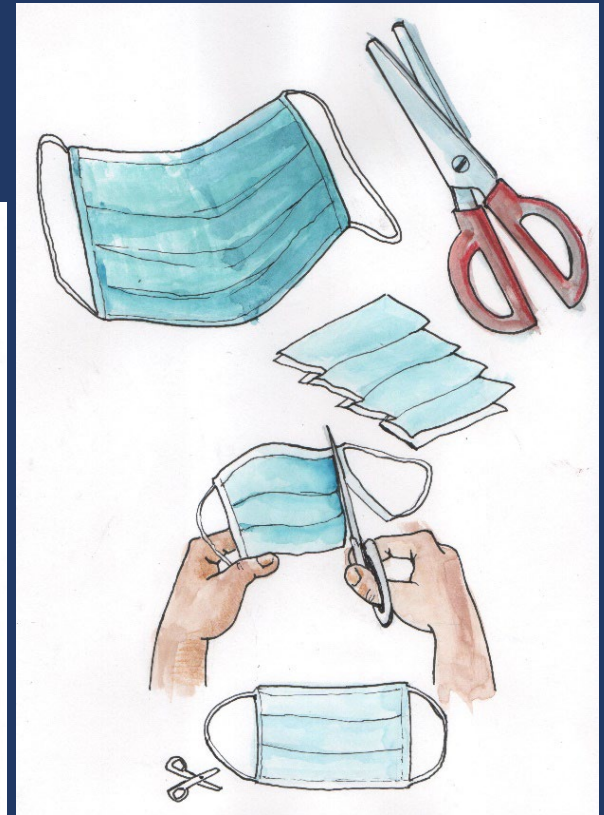


METHODOLOGY ^{5/7}

- Protocol- scientific lab
- Drawing protocol

Watercolor

Defining contours

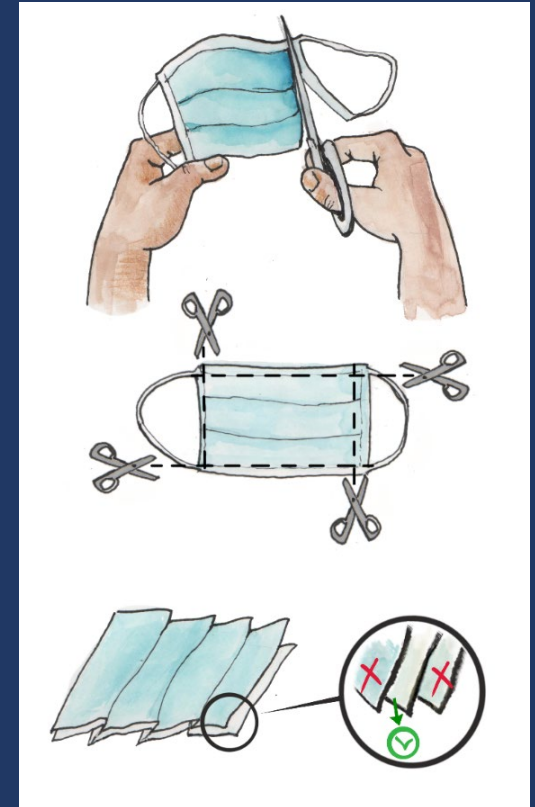


METHODOLOGY ^{6/7}

- Protocol- scientific lab
- Drawing protocol

Defining contours

Digitalizing



METHODOLOGY ^{7/7}

- Protocol- scientific lab
- Drawing protocol
- Control group- test (25), identified tool and process.



María Lería

PROTOCOL (lab)

SCHOOL LABORATORY MATERIAL



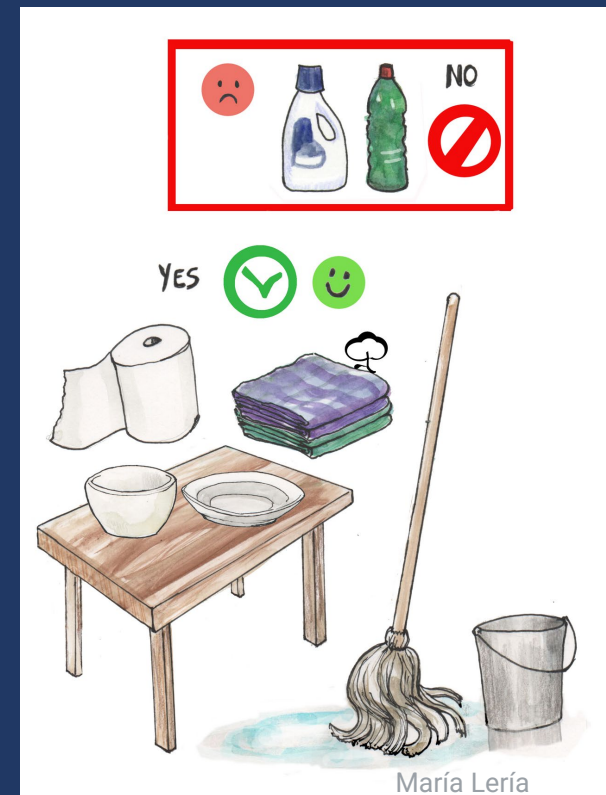
- A. Analytical balance
- B. Petri dishes
- C. Foil
- D. Glass rod
- E. Surgical mask
- F. Magnetic stirrer and magnet
- G. Scissors
- H. Spatula with spoon
- I. Water
- J. Table salt
- K. Beaker
- L. Büchner funnel
- M. Filter
- N. Flask clamps
- O. Water suction pump
- P. Magnifying binocular
- Q. Tweezers
- R. Notebook and pencil
- S. Kitasato flask

GENERAL CONSIDERATIONS:

- Note the color of the clothes worn during the experiment.
- Write down the type of water used and filter it before hand.
- Note the type of filter used.
- Clothes can be rubbed with a filter to see how fibers come off.

Protocol 1/6

- Cleaning



Protocol 2/6

- Cleaning
- Control



Protocol ^{3/6}

- Cleaning
- Control
- To weigh



Protocol 4/6

- Cleaning
- Control
- To weigh
- To dissolve



Protocol 5/6

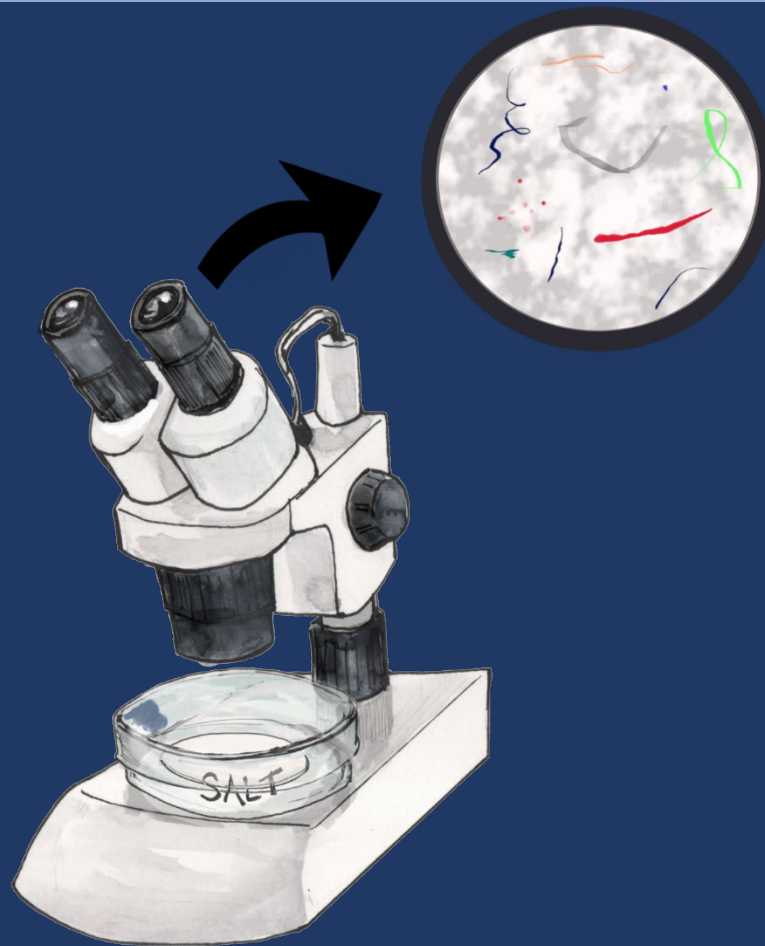
- Filtering



María Lería

Protocol 6/6

- Cleaning
- Control
- To weigh
- To dissolve
- Filtering
- Analyzing and counting



Microplastics in Table Salts?

How to easily study them

SIMPLIFIED INSTRUCTIONS MANUAL FOR CHEMICAL LABORATORY OF EDUCATIONAL CENTER

What is it for?

The aim of this instruction manual is to illustrate the procedure to separate, observe and quantify at home, or in an educational center that has a chemical laboratory, microplastics and other insoluble particles present in table salts. In addition, these activities embrace three of the Sustainable Development Goals (SDGs): 3, 4, 6 and 12.

Who is it for?

This manual is aimed at secondary school students aged 12-16 and high school and also includes a section with a slightly modified protocol, for those educational centers that have a poorly equipped chemical laboratory.

Why to use this manual?

We will use this manual as complementary material to the educational guide and the workbook for the student. In addition, with the help of the instructions we will be able to complete the observations and activities proposed in the educational guide.

How to use the manual?

The manual is structured in 4 parts, where it is illustrated how we should prepare the laboratory, the material that we will need, the procedure and finally the observations and results. To facilitate the development of the activity, additional material can be consulted at:

 <http://www.ub.edu/sedimentary-geology/microplastics-salt>



Authors: **María LERÍA, Vinyet BAQUÉS, Irene CANTARERO, Elisabet PLAYÀ and Anna TRAVÉ**
Supported by a grant from the National Geographic Society's COVID-19 Remote Learning Emergency Fund for Educators and 2020PID-UB/039 Project



Consolidated Teaching Innovation Group GRIMS
(Geochemistry and Igneous, Metamorphic and Sedimentary Rocks)



Consolidated Research Group
of the Generalitat de Catalunya



UNIVERSITAT DE BARCELONA

Faculty of Earth Science
University of Barcelona

LABORATORY PREPARATION

How to avoid sample contamination?



1

- Hair tied back or with a cotton cap.
- Wear brightly colored cotton clothing.
- Wash hands with soap and water.
- Do not use hand cream or makeup.
- Do not wear plastic gloves.

2

- Clean work surfaces and material (inside and outside) with filtered or bottled water.
- Mop the laboratory floor with water.



3

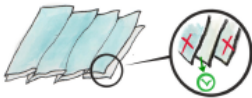
- Use glass, metal or ceramic containers.
- Avoid the use of plastic.



4

Prepare filters and contamination controls:

- Separate and cut the central filter of a mask.



5

- Use a filter to control airborne contamination during laboratory work.
- Uncap and cover the control filter when running salt samples.
- Airborne microplastics will be retained and we will be able to quantify them.



REMEMBER!

Expose the control filter during the experiment and cap after completion. If the salt filter is exposed, it will control microplastic contamination during sample treatment.

HOME LABORATORY MATERIAL



- A. Balance
- B. Foil
- C. Plasticine tablet
- D. Adhesive tape
- E. Domestic vacuum cleaner
- F. Spoon
- G. Plastic hose
- H. Surgical mask
- I. Bottle to collect water
- J. Scissors
- K. Water
- L. Table salt
- M. Italian coffee funnel
- N. Crystal glass
- O. Glass, metal or ceramic bowl
- P. Glass, metal or ceramic plate
- Q. Notebook and pencil
- R. Glass jar with lid
- S. Magnifying glass

GENERAL CONSIDERATIONS:

- Note the color of the clothes worn during the experiment.
- Write down the type of water used and filter it beforehand.
- Note the type of filter used.
- Clothes can be rubbed with a filter to see how fibers come off.

HOME PROCEDURE

PREPARATION OF MATERIALS:

- With the help of the vacuum cleaner, the flexible tube, the adhesive tape, the plasticine and the bottle, **prepare the homemade vacuum pump.**
- Put a surgical mask filter in the funnel of the Italian coffee maker.
- Pierce the bottle and tape the flexible tube so that it is well fixed.



- Choose any water and **filter 1-2L x2 times** using the home vacuum pump.

1 Place the exposed control filter near the salt. Weigh 50 g of salt and record the weight. Cover the salt and control filter.



2 Introduce the salt sample into the glass jar.



3 Dissolve the salt in 500 mL of filtered water. Stir with a spoon until the salt is well dissolved.



4 Cover the glass jar containing the solution and the control filter. If the salt does not dissolve, leave the solution for a few hours or even a day, until it dissolves completely.



5 Put a surgical mask filter in the funnel and pour the solution with the dissolved salt.



6 Cover the funnel and the control filter. Turn on the vacuum cleaner and filter the salt solution with the domestic vacuum pump.



7 Pick up the filter and cover it. Cover the control filter.



REMEMBER!

Always cover and uncover the control filter and the experimental sample at the same time. This way you will know if the sample is being contaminated.

Conclusions

- Cheap and approachable guide
- Contribute to SDG 4
- Awareness: SDG 3, 6, 12
- Adding art is useful



Kaiwikimom: https://en.wikipedia.org/wiki/File:Students_in_lab.jpg

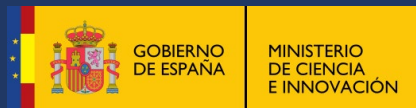


S M R <https://pixahive.com/photo/sea-water-polluted-with-plastic-wastes/>



<http://www.ub.edu/sedimentary-geology/microplastics-salt>

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**Thank you
very much!**



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