

Geochemical signals of carbonatite-related critical metals in provincial stream sediments



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Abstract

Consisting of at least 30% primary carbonate minerals, carbonatites are rare igneous rocks that have become increasingly important exploration targets, because they are major sources of Nb, rare earth elements (REE), and other critical minerals. Demand for these minerals has rapidly increased as the world transitions to low-carbon technologies. The British Columbia alkaline province, which defines a long (at least 1000 km), narrow (ca. 200 km) orogen-parallel belt along the western flank of Ancestral North America, contains numerous carbonatites and related silica-undersaturated and alkaline silicate rocks that host REE and rare-metal resources. Using multi-element stream-sediment geochemical data collected as part of the Regional Geochemical Survey (RGS) program since 1976, we define a multivariate ‘critical mineral index’ to assess prospectivity for carbonatite-hosted critical metals. Based on discriminant analysis of a training sub-set of the data downstream of known carbonatite occurrences ($n=26$), our carbonatite index, which is validated by a test sub-set of the data ($n=27$), highlights numerous areas prospective for REE in the alkaline province. Stream-sediment data showing carbonatite index scores greater than the 93rd percentile ($n=50$) reveal maximum contrast of REE, Nb, Ta, Ti, Zr, Hf, Th, U, P, K and other carbonatite indicators relative to the median (background) concentrations in stream sediments of the study area. Estimated predicted geochemical resources (in tonnes of metal per 1 m depth), based on productivities of metals in the stream basins, suggest significant potential for REE and other carbonatite-hosted critical metals. Based on data from known carbonatites, we propose a refined prospectivity approach to assess the critical metals potential of underexplored regions that includes detailed stream-sediment, panned heavy mineral concentrate, and soil lithochemical surveys and high-resolution airborne radiometric and magnetic data.

Keywords: Carbonatite, alkaline rocks, regional geochemical survey (RGS), stream sediments, heavy mineral concentrate (HMC), drainage geochemistry, critical minerals, rare earth elements (REE), rare metals, niobium, tantalum, Blue River, Upper Fir, Aley, multivariate statistics, discriminant analysis, carbonatite index, predicted geochemical resources

1. Introduction

Reconnaissance geochemical surveys have a long history of supporting mineral exploration in underexplored regions of British Columbia. Regional sampling of stream sediments and waters has been carried out by mining companies since 1950s and was later adopted by the Geological Survey of Canada, the British Columbia Geological Survey, and Geoscience BC as part of the Regional Geochemical Survey (RGS) programs (Lett and Rukhlov, 2017). Interpretation of these data has led to the discovery of precious and base metal deposits such as the Highland Valley Copper mine and the Galore Creek proposed mine (e.g., Brummer et al., 1987).

Consisting of at least 30% primary carbonate minerals, carbonatites are rare igneous rocks that have become increasingly important exploration targets, because they are major sources of Nb, rare earth elements (REE), and other critical minerals needed as the world transitions to low-carbon technologies (Hickin et al., 2024). Although lithogeochemistry of panned heavy mineral concentrate (HMC) and indicator minerals of stream sediments have become established techniques for rare-metal and REE prospecting (e.g., Rukhlov and Gorham, 2007;

Gorham, 2008; Gorham et al., 2009; Simandl et al., 2017) the application of regional geochemical surveys to carbonatite-hosted minerals has not been evaluated.

The British Columbia alkaline province, which defines a long (at least 1000 km), narrow (ca. 200 km) orogen-parallel belt along the western flank of Ancestral North America, contains numerous carbonatites and related silica-undersaturated and alkaline silicate rocks (Fig. 1). Some of these rocks host REE (e.g., Wicheada; Dalsin et al., 2015; Trofanenko et al., 2016) and Ta-Nb (e.g., Aley; Mäder, 1987; Chakhmouradian et al., 2015; and Upper Fir; Rukhlov et al., 2018). This study evaluates multi-element stream-sediment geochemical data collected as part of the RGS program in the Omineca and Foreland morphotectonic belts for prospectivity indicators of carbonatite-hosted critical metals. We also consider examples of detailed surveys near known carbonatite occurrences in the Blue River area and at the Aley deposit (Fig. 2) to discuss applications of panned heavy-mineral concentrate and soil lithogeochemistry, high-resolution airborne radiometrics and magnetics, and productivities of carbonatite indicators in stream basins. Using these data, we define a multivariate ‘critical mineral index’ that can be used to

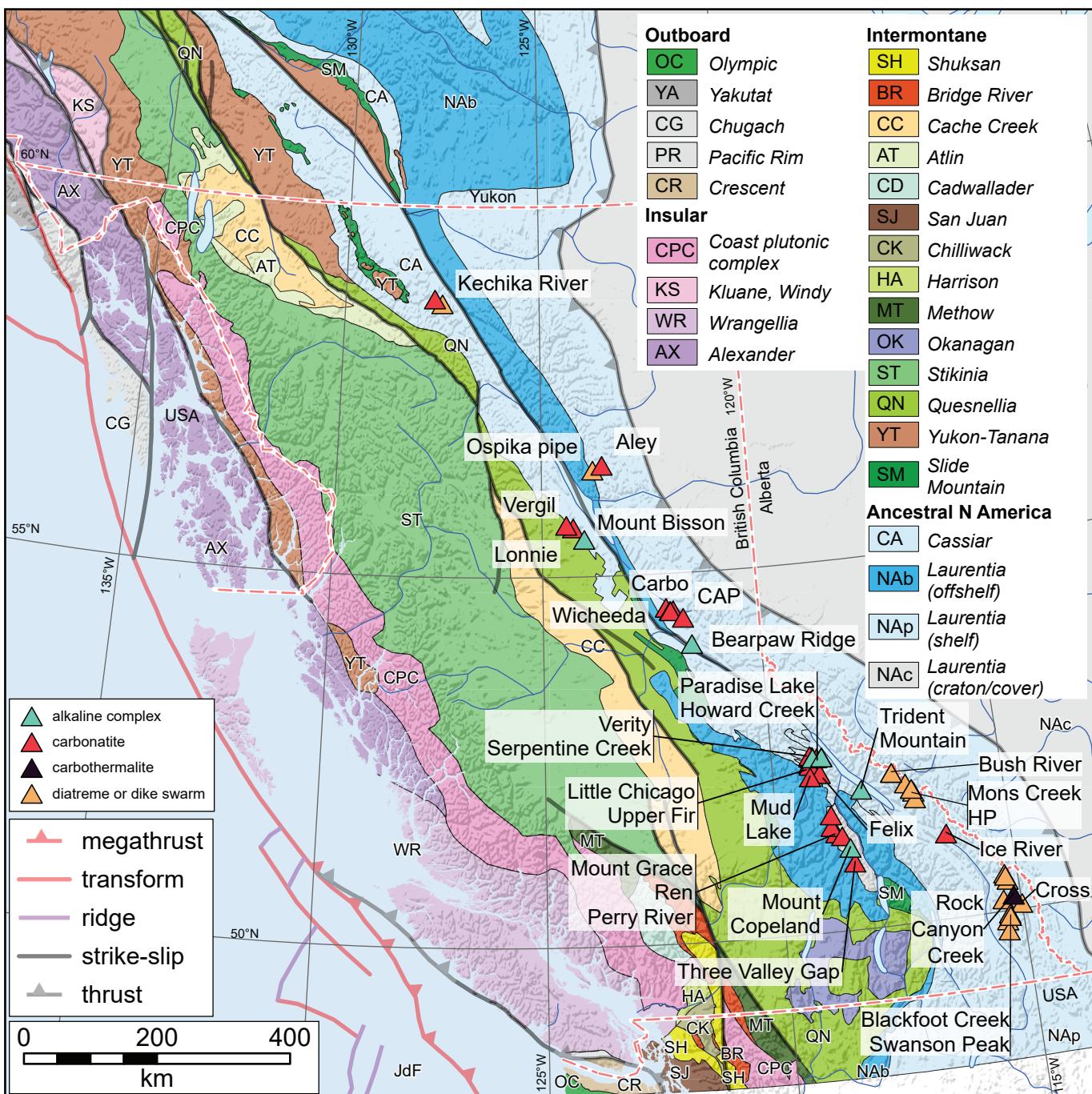


Fig. 1. Carbonatite and related-rock occurrences along the British Columbia alkaline province (after Höy, 1988; Parrish and Scammell, 1988; Pell, 1994; Millonig and Groat, 2013; Rukhlov et al., 2018). Terranes after Colpron (2020).

highlight prospective areas in the alkaline province and regions elsewhere that warrant prospecting for carbonatite-hosted commodities.

2. British Columbia carbonatites and related rocks

In the Canadian Cordillera, carbonatite and related ultramafic, silica-undersaturated and alkaline silicate bodies were emplaced episodically at ca. 810-700 Ma (Mount Copeland, Perry River, Ren), 500-400 Ma (Blackfoot Creek,

Bush River, Felix, HP, Kechika River, Little Chicago, Mons Creek, Swanson Peak), and 360-320 Ma (Aley, Howard Creek, Ice River, Lonnie, Mount Grace, Mud Lake, Ospika, Paradise Lake, Serpentine Creek, Three Valley Gap, Upper Fir, Trident Mountain, Vergil, Verity, Wicheada); the Cross kimberlite is 245 Ma. Collectively, these rocks form part of the British Columbia alkaline province (Fig. 1; Höy, 1988; Parrish and Scammell, 1988; Pell and Höy, 1989; Pell, 1994; Rukhlov and Bell, 2010; Millonig et al., 2012; Millonig and Groat, 2013;

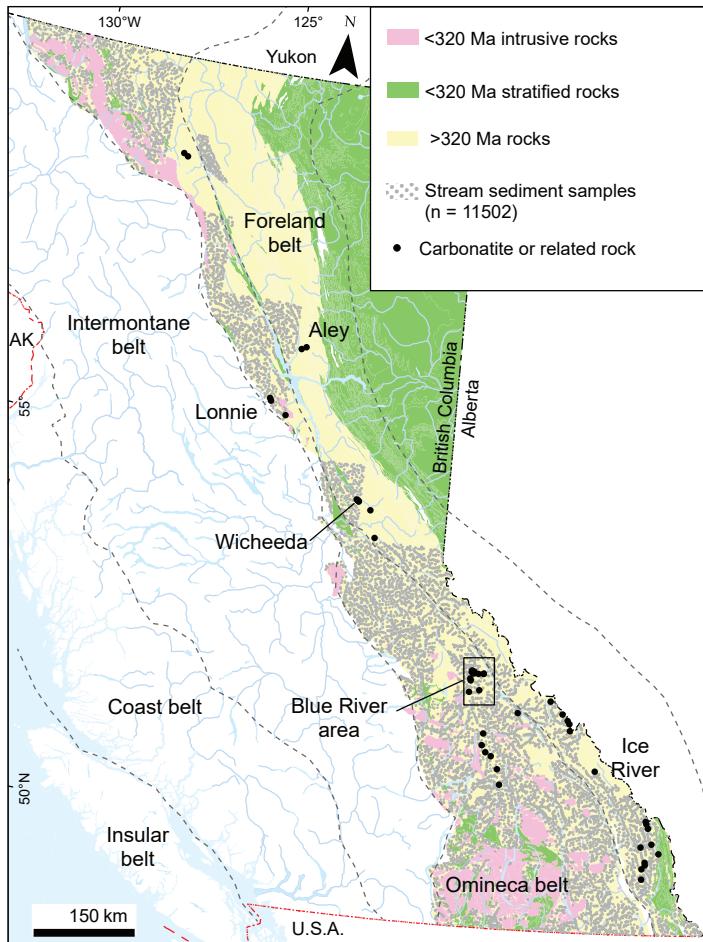


Fig. 2. Selected stream sediment samples in the study area (after Han and Rukhlov, 2017, 2020b). Carbonatite and related-rock occurrences after Höy (1988), Parrish and Scammell (1988), Pell (1994), Millonig and Groat (2013), and Rukhlov et al. (2018). Morphotectonic boundaries after Gabrielse et al. (1991). Geology from BC Digital Geology version 2021-12-19 (Cui et al., 2017).

Chakhmouradian et al., 2015; Mitchell et al., 2017; Rukhlov et al., 2018; McLeish and Johnston, 2019; McLeish et al., 2020; Burgess et al., 2023). The Neoproterozoic and early Paleozoic pulses of carbonatite and alkaline magmatism mark protracted breakup of the supercontinent Rodinia and subsequent passive margin development on the western flank of Laurentia (Li et al., 2008; Bond and Kominz, 1984; Ross, 1991; Colpron et al., 2002). The late Paleozoic carbonatite and alkaline complexes, which host Nb-Ta deposits (e.g., Upper Fir in the Blue River area and Aley) and REE deposits (e.g., Wicheeda), were injected near the continental margin while subduction was taking place to the west (Nelson et al., 2013).

Hosted by the parautochthonous rocks of the Omineca and Foreland belts (Fig. 2), carbonatites and related rocks range from intrusive complexes with a paucity of carbonatites (e.g., Trident Mountain, Mount Copeland) to carbonatite complexes with a paucity of silicate rocks (e.g., Aley, Blue River, Frenchman Cap). Both the carbonatites and host rocks experienced multiple episodes of deformation and metamorphism during Mesozoic and Cenozoic accretionary tectonics while outboard terranes welded to each other and to Laurentia (Scammell, 1987, 1993;

Scammell and Brown, 1990; Pell, 1994; Millonig et al., 2013). Intrusive complexes made up of mainly silica-undersaturated and alkaline silicate rocks, such as the Ice River complex, form small (up to 29 km² at surface), compositionally zoned bodies that are circular to elongate to amoeboid in plan view (Dawson, 1886; Currie, 1975; Peterson and Currie, 1994). Associated ultramafic lamprophyres and REE-Sr-rich carbothermalite dikes are common, with the latter consisting of Mn-calcite, barytocalcite, and zeolite with minor strontianite, Nb-ilmenite, and REE-F-carbonates (Mumford, 2009; Brown, 2013). Carbonatites lacking associated contemporaneous silicate rocks typically form regional swarms of individual occurrences across areas of 1000 km² (e.g., Blue River; Pell, 1994; Mitchell et al., 2017; Rukhlov et al., 2018; Çimen et al., 2019).

3. Blue River area

In the Blue River area (Fig. 3), at least 18 carbonatite and two alkaline, silica-undersaturated-rock bodies are exposed, including at the Upper Fir deposit, one of the largest and best studied Nb-Ta occurrences in the Canadian Cordillera (Chudy, 2013; Rukhlov et al., 2018). Both Cambrian and late

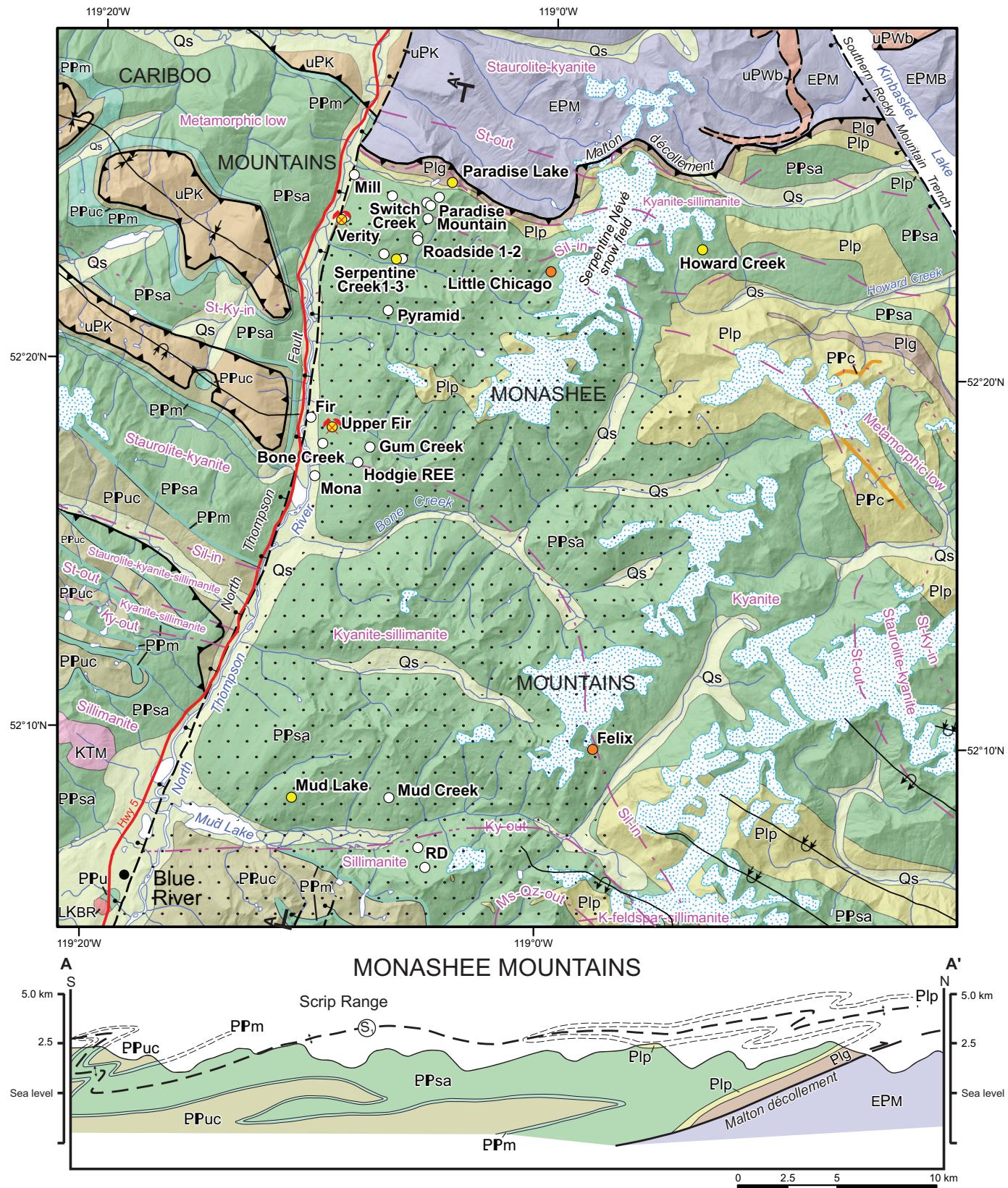


Fig. 3. Carbonatite and related-rock occurrences of the Blue River area (after Pell, 1994; Rukhlov and Bell, 2010; Millonig et al., 2012, 2013; Millonig and Groat, 2013; Rukhlov et al., 2018). Geology and metamorphic isograds after Campbell (1968), Simony et al. (1980), Raeside and Simony (1983), Pell and Simony (1987), McDonough and Murphy (1990), McDonough et al. (1991a, b, 1992), Digel et al. (1998), and Murphy (2007).

UNCONSOLIDATED DEPOSITS

Quaternary

Qs Undifferentiated sand, silt, clay, gravel, till, and colluvium

INTRUSIVE ROCKS

Cretaceous to(?) Paleogene

Murtle pluton

KTM Quartz monzonite and muscovite-biotite granite

Late Cretaceous

Blue River pluton

LKBR Weakly foliated muscovite ±biotite granite

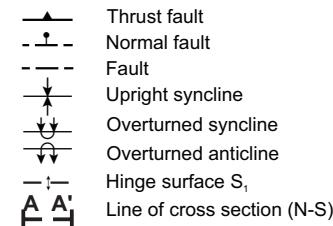
METAMORPHIC ROCKS

Neoproterozoic

Windermere Supergroup

Kaza Group

uPK Undivided psammite, grit, pelitic schist, phyllite, slate, marble



Undivided basal Windermere Supergroup

uPWb Grit, conglomerate or diamictite, psammite, mylonitic quartzite at base, pelitic phyllite or schist, marble and calc-silicate rocks

Proterozoic and(?) Paleozoic

PPu Undivided metamorphic rocks of unknown, probably Proterozoic and possibly Paleozoic age

Upper division of Horsethief Creek Group (equivalent units of Mica Creek succession)

Upper clastic unit

PPuc Quartzofeldspathic psammite and grit, pelitic schist, minor amphibolite

Marble unit

PPm Marble, calcareous pelite, semipelite, schist, discontinuous interbedded pelite, psammite, grit, quartzite, sandy marble

Conglomerate in PPsa (Cariboo) and

Plp (Monashees) units

PPc Conglomerate with clasts of marble, calc-silicate rock, quartzite and granite

REGIONAL METAMORPHISM

Mesozoic isograds

- - - St-Ky-in	Staurolite and kyanite in
- - St-out	Staurolite out
- - Sil-in	Sillimanite in
- - Ky-out	Kyanite out
- - - Ms-Qz-out	Muscovite and quartz out

Paleocene overprint

... . . Pod sillimanite

Semipelite-amphibolite unit

PPsa Quartzose and quartzofeldspathic psammite, grit, pelitic schist, concordant and discordant amphibolite, minor marble, locally marble at base

Proterozoic

Lower division of Horsethief Creek Group (equivalent units of Mica Creek succession)

Lower pelite unit

Plp Pelitic schist, minor quartzofeldspathic psammite, concordant and discordant amphibolite

Lower grit unit

Plg Quartzofeldspathic psammite and grit, minor pelitic schist and amphibolite, locally prominent diamictite-bearing, conglomeratic horizon at base

Paleoproterozoic

Malton gneiss complex

EPM Undivided foliated granitic augen orthogneiss, mafic orthogneiss, paragneiss

Mount Blackman gneiss

EPMB Amphibolitic mafic gneiss, granitic gneiss

Carbonatite and related-rock occurrence

- Unknown age
- ca. 360-330 Ma
- ca. 500 Ma
- ✖ Nb-Ta deposit

Fig. 3. Continued. Legend.

Paleozoic carbonatites made up of dolomite and calcite are hosted by metamorphosed Neoproterozoic pelitic, arenaceous, and amphibolitic rocks of the Mica Creek assemblage. Metamorphosed to amphibolite grade during Mesozoic to Cenozoic orogeny, the carbonatites form isoclinally folded, sill-like tabular bodies up to 72 m thick and display diverse fabrics, including coarse-grained, granoblastic to fine-grained, foliated, and porphyroclastic varieties. They contain 10-15 vol.% fluorapatite, 5-10 vol.% amphiboles, and variable amounts of olivine, chondrodite, clinopyroxene, phlogopite, magnetite, ilmenite, pyrrhotite, pyrite, pyrochlore supergroup, ferrocolumbite, fersmite, nyobiaeschynite, zircon, baddeleyite, zirconolite, and monazite (Rukhlov et al., 2018). The Upper Fir carbonatite contains an NI 43-101-compliant resource of 48.4 million tonnes (Indicated) grading 1610 ppm Nb_2O_5 and 197 ppm Ta_2O_5 plus 5.4 million tonnes (Inferred) averaging 1760 ppm Nb_2O_5 and 191 ppm Ta_2O_5 (Kulla and Hardy, 2015). Pyrochlore and ferrocolumbite are the main hosts of Nb and Ta (Chudy, 2013; Rukhlov et al., 2018). Molybdenite occurs in some carbonatites and related rocks, including alkali-rich metasomatic rocks such as fenites and glimmerites at Fir, Perry River, Mount Grace, Wicheeda, and the Mount Copeland past producer (Currie, 1976; White, 1982; Höy, 1988; Trofanenko et al., 2016; Rukhlov et al., 2018).

4. Rock, soil, and drainage lithogeochemical data from the Blue River area

In addition to distinct physical and mineralogical characteristics, carbonatites are readily distinguished from sedimentary carbonate rocks (Table 1) by their extremely high concentrations (up to 1000 times the average upper continental crust) of REE, rare metals, F, P, Sr, Ba, Th, U, and other elements (Fig. 4). Data from both provincial, reconnaissance-scale drainage surveys and detailed lithogeochemical drainage and soil surveys in the Blue River area (Table 2) highlight the geochemical response from known carbonatite and related rock occurrences (Figs. 5-7). Ratios of the maximum concentrations per element show enrichment of soil samples (residual anomaly) in Ba, Mo, and Th, and the <0.18 mm fraction of stream-sediment samples in Yb and Lu relative to carbonatites (primary anomaly; Table 3). In contrast, panned stream-sediment (<2 mm fraction) heavy mineral concentrate (HMC) samples show up to 192 times enrichment of all carbonatite indicator elements, except Sr, relative to carbonatites, soils, and the <0.18 mm fraction of stream-sediment samples. Panned stream-sediment HMC samples enhance the contrast of carbonatite indicators such as Ta (Fig. 6) and thus are the preferred medium for drainage surveys targeting critical metals (Rukhlov et al., 2020a).

Table 1. Provincial lithogeochemical data from carbonatites and carbonate sedimentary rocks.

Element	Unit	Sedimentary carbonate rocks				Carbonatites			
		Count	Minimum	Maximum	Mean	Count	Minimum	Maximum	Mean
SiO ₂	wt%	32	0.27	67.44	15.62	600	0.09	35.72	5.33
TiO ₂	wt%	34	<0.02	1.42	0.15	594	<0.01	5.96	0.19
Al ₂ O ₃	wt%	34	0.06	10.03	1.97	600	<0.01	12.64	0.68
Fe ₂ O ₃ (T)	wt%	35	0.06	9.00	1.37	600	0.51	83.28	8.56
MnO	wt%	35	<0.01	0.35	0.07	599	0.15	4.11	0.69
MgO	wt%	35	0.15	20.80	4.17	600	0.26	26.02	12.63
CaO	wt%	35	7.77	60.60	40.80	600	3.66	54.50	31.44
Na ₂ O	wt%	35	<0.01	1.82	0.23	600	<0.01	6.84	0.43
K ₂ O	wt%	35	<0.01	1.65	0.38	600	<0.01	4.79	0.38
P ₂ O ₅	wt%	35	<0.01	0.18	0.07	600	<0.01	14.87	3.15
LOI	wt%	31	9.10	44.30	34.91	598	2.30	43.90	34.92
C(T)	wt%	25	2.03	12.26	9.67	516	0.01	12.90	10.35
S(T)	wt%	27	<0.01	0.48	0.06	515	<0.01	3.97	0.30
F	wt%	1	na	na	0.01	14	<0.01	1.45	0.27
Ag	ppm	22	0.03	0.40	0.09	197	<0.1	5.7	0.54
As	ppm	28	<0.5	17	2.2	192	<0.5	77	3.0
Au	ppb	20	<2	8	1.2	110	<0.5	426	6.7
Ba	ppm	29	<5	2379	275	610	15.1	288595	1825
Be	ppm	15	0.06	1.0	0.67	512	<1	11	0.91
Bi	ppm	16	<0.02	0.2	0.09	192	<0.1	3.1	0.24
Br	ppm	4	<0.5	<1	na	1	na	na	2.8
Cd	ppm	21	<0.2	6.9	0.46	110	<0.1	1.5	0.33
Ce	ppm	28	0.8	51	14.7	611	7.1	53200	1169
Co	ppm	30	<1	50	5.1	586	<1	73	15.5
Cr	ppm	30	3.4	506	38.0	556	<0.1	2230	60.8
Cs	ppm	28	<0.5	2.0	0.48	535	<0.1	114	1.73
Cu	ppm	22	0.4	26	6.1	584	<0.1	308	11.3
Dy	ppm	17	0.28	3.3	1.32	606	0.37	106	11.2
Er	ppm	17	0.15	1.9	0.76	606	0.15	17.8	3.54
Eu	ppm	28	0.04	1.1	0.34	607	0.2	179	10.4
Ga	ppm	17	0.1	8.4	2.91	596	<0.5	126	6.1
Gd	ppm	17	0.36	3.5	1.46	604	0.53	404	25.2
Ge	ppm	2	<0.1	0.4	0.23	81	<1	14	2.1
Hf	ppm	28	0.03	4.4	0.93	584	<0.1	53	1.96
Hg	ppb	15	<10	20	7	110	<10	110	8
Ho	ppm	17	<0.1	0.63	0.260	606	0.06	10.8	1.60
In	ppm	2	<0.02	<0.04	na	81	<0.2	<0.3	na
La	ppm	30	<0.5	24	7.4	611	3.2	40500	734
Lu	ppm	28	0.017	0.3	0.097	607	0.03	2.0	0.326
Mo	ppm	29	0.04	3.0	0.470	478	<0.1	125	3.24
Nb	ppm	27	0.12	22	5.25	609	2.8	6532	576
Nd	ppm	28	1.21	23	7.17	610	3.7	11900	347
Ni	ppm	30	0.98	209	15.9	536	<0.1	1237	38.1
Pb	ppm	23	0.54	19	6.56	586	0.5	643	16.2
Pr	ppm	17	0.23	5.94	2.205	606	0.86	4300	90.0
Rb	ppm	28	0.2	59	12.3	594	<0.1	327	12.7
Sb	ppm	28	<0.1	1.1	0.14	191	<0.1	4.7	0.10
Sc	ppm	27	0.20	26	3.84	207	0.24	62	16.7
Se	ppm	19	0.01	0.5	0.23	110	<0.5	7.3	0.56
Sm	ppm	28	0.23	4.2	1.40	607	0.68	947	42.0
Sn	ppm	17	<0.02	2	0.74	344	<1	25	1.6
Sr	ppm	30	47	3882	914	611	360	>50000	4401
Ta	ppm	28	<0.1	1.4	0.24	599	<0.01	646	101
Tb	ppm	28	0.046	0.64	0.176	607	0.07	36.3	2.77
Th	ppm	30	0.06	7.1	1.64	611	<0.1	>10000	71.6
Tl	ppm	15	0.05	1.3	0.16	197	<0.1	1.3	0.10
Tm	ppm	17	0.02	0.31	0.117	606	0.03	2.17	0.433
U	ppm	28	0.2	2.6	0.79	609	<0.1	379	40.0
V	ppm	23	1	139	26.7	585	<5	2713	49.6
W	ppm	28	0.2	1.4	0.34	350	<0.1	22.5	0.89
Y	ppm	28	1.5	22.3	8.43	611	1.7	323	43.0
Yb	ppm	28	<0.2	1.86	0.636	607	0.19	14.1	2.44
Zn	ppm	29	<1	173	24.4	307	2	1949	77.3
Zr	ppm	28	1	173	38.8	610	<0.1	2978	104.8

Data from Chakhmouradian et al. (2015), Chudy (2013), Dalsin et al. (2015), Gorham (2008), Han et al. (2016), Han and Rukhlov (2020a), Locock (1994), Mäder (1987), Mumford (2009), Rukhlov and Gorham (2007), Simandl et al. (2013), Trofanenko (2014), Trofanenko et al. (2016), and Ya'acoby (2014). na - not analyzed.

Table 2. Blue River area lithogeochemical data from rock, soil, RGS stream sediment, and panned stream-sediment heavy mineral concentrate samples.

Element	Unit	Method	Count	Soil samples				RGS stream sediments				Panned HMC of stream sediments			
				Minimum	Maximum	Mean	Method Count	Minimum	Maximum	Mean	Method Count	Minimum	Maximum	Mean	Method Count
TiO ₂	wt%	T	477	<0.01	5.96	0.143	na	na	na	208	0.02	1.63	0.252	P	24
Al ₂ O ₃	wt%	T	483	<0.01	12.64	0.374	na	na	na	208	0.77	4.85	2.748	P	24
Fe ₂ O ₃ (T)	wt%	T	483	0.51	83.28	8.743	na	na	na	208	2.29	12.53	6.440	T	8
MnO	wt%	T	483	0.15	1.53	0.692	na	na	na	208	<0.01	0.16	0.044	P	24
MgO	wt%	T	483	0.49	26.02	13.204	na	na	na	208	0.22	3.56	1.188	P	24
CaO	wt%	T	483	3.66	52.87	31.759	na	na	na	208	0.11	6.28	0.807	P	24
Na ₂ O	wt%	T	483	<0.01	3.63	0.364	na	na	na	208	0.62	3.10	1.972	T	8
K ₂ O	wt%	T	483	<0.01	4.79	0.176	na	na	na	208	<0.01	1.76	0.640	P	24
P ₂ O ₅	wt%	T	483	0.01	14.87	3.287	na	na	na	208	0.09	2.27	0.310	P	24
SCT	wt%	T	483	<0.01	2.38	0.293	na	na	na	208	<0.01	1.56	0.091	P	24
Ag	ppb	P	97	<100	600	124	P	8082	<100	3100	108	P	208	6	631
As	ppm	P	97	<0.5	5.3	0.93	P	8082	<0.5	4932	1.5	T	P	208	<0.1
Au	ppb	P	97	<0.5	426	7.4	P	7976	<0.5	834	1.2	P	208	<0.2	34
Ba	ppm	T	483	15	5823	218	T	8282	<1	8267	588	T	208	85	1300
Be	ppm	T	418	<1	11	0.62	T	8165	<1	32	2.0	P	na	na	T
Bi	ppm	P	97	<0.1	1.1	0.10	P	8082	<0.1	51	0.26	P	208	0.04	1.8
Cd	ppm	P	97	<0.1	1.5	0.34	P	8082	<0.1	84	0.14	P	208	0.01	0.51
Ce	ppm	T	483	0.37	2418	3.18	T	8282	<0.1	1477	100	T	208	62	1010
Co	ppm	T	483	0.6	73	16	T	8282	<0.2	272	12	T	208	<5	116
Cr	ppm	T	464	<0.1	2230	66.5	na	na	na	208	28	370	109	T	8
Cs	ppm	P	432	<0.1	48	1.5	T	8282	<0.1	23	4.5	T	208	<1	12
Cl	ppm	P	481	0.2	308	9.8	P	8082	1.9	355	20	P	208	3	166
Dy	ppm	T	483	0.37	27	9.2	T	8282	<0.05	33	4.8	na	na	na	T
Er	ppm	T	483	0.15	14	3.2	T	8282	<0.03	13	2.7	na	na	na	T
Fu	ppm	T	483	0.2	22	6.3	T	8282	<0.02	24	1.4	T	180	0.9	8.4
Ga	ppm	T	483	<0.5	28	2.9	T	8282	<0.5	34	1.7	P	208	1.6	11
Gd	ppm	T	483	0.53	52	1.7	T	8282	<0.05	52	5.6	na	na	na	T
Hf	ppm	T	481	<0.1	53	1.5	T	8282	<0.1	49	6.3	T	208	4.0	87
Hg	ppb	P	97	<10	110	7.1	P	7976	<10	380	41	P	208	<5	73
Ho	ppm	T	483	0.06	4.9	1.4	T	8282	<0.02	5.5	0.94	na	na	na	T
La	ppm	T	483	3.2	1695	149	T	8282	<0.01	1269	51	T	208	28	509
Lu	ppm	T	483	0.03	2.0	0.30	T	8282	<0.01	1.5	0.40	T	208	<0.2	3.9
Mo	ppm	P	376	<0.1	21	0.46	P	8082	<1	85	2.3	P	208	0.11	3.8
Nb	ppm	T	483	5	6532	610	T	8282	<0.1	5507	35	na	na	na	T
Nd	ppm	T	483	3.7	6318	141	T	8282	<0.3	483	41	T	180	27	420
Ni	ppm	T, P	436	<0.1	1237	43	P	8082	1.2	1025	18	P	208	2.7	137
Pb	ppm	P	483	<1	30	5.4	T	8282	1.8	547	8.9	P	208	1.93	15
Pr	ppm	T	483	0.86	210	36	T	8282	<0.02	124	11	na	na	na	T
Rb	ppm	T	475	<0.1	249	5.3	T	8282	<0.1	274	83	T	208	28	220
Sb	ppm	P	97	<0.1	0.2	0.056	P	8082	<0.1	42	0.075	P	208	<0.02	0.2
Sc	ppm	T	120	1.0	34	10	na	na	na	208	6.4	35	17	8	19
Se	ppm	P	97	<0.5	1.9	0.47	P	7976	<0.5	12.6	0.19	P	208	0.1	2.2
Sn	ppm	T	483	0.68	78	22	T	8282	<0.05	76	6.8	T	208	5	74
Sin	ppm	T	244	<1	25	1.2	T	8282	<1	35	2.6	na	na	na	T
Sr	ppm	T	483	0.03	10559	4179	T	8282	<0.5	4201	203	P	208	5.5	186
Ta	ppm	T	483	1.1	646	123	T	8282	<0.1	405	2.2	T	208	<1	13
Tb	ppm	T	483	0.07	6.5	2.1	T	8282	<0.01	6.7	0.89	T	208	0.60	9.2
Th	ppm	T	483	<0.1	174	6.7	T	8282	<0.2	484	15	T	208	8.9	162
Tl	ppm	P	97	<0.1	0.3	0.060	P	8220	<0.1	14	0.18	P	208	<0.02	0.72
Yb	ppm	T	483	0.19	13	2.3	T	8282	<0.05	10	2.7	T	208	<2	26
Zn	ppm	P	204	2	288	35	P	8082	5	935	43	P	208	16	201
Zr	ppm	T	483	<0.1	2978	86	T	8282	9.5	1978	220	T	28	>200	2300

Data from Chuday (2013), Dahrouge and Gorham (2001), Dahrouge and Wolbaum (2004), Gorham (2008), Gorham et al. (2009, 2011a, 2011b), Han and Rukhlov (2017, 2020b), Reeder and Dahrouge (2002), Rukhlov and Gorham (2007), Smith and Dahrouge (2003). Method abbreviations: T – total determination via instrumental neutron activation analysis (INAA); RGS stream sediment samples only) or lithium-borate fusion with a combination of inductively coupled plasma atomic emission spectroscopy (ICP-AES) and ICP-MS; na - not analyzed.

Table 3. Blue River area relative enrichment of selected carbonatite indicator elements in different sample media.

Element	S P	Db P	Db S	Dc P	Dc S	Dc Db
Sr	0.4	0.02	0.04	0.4	1.1	24
Ba	1.4	0.2	0.2	3.7	2.6	17
Mo	4.0	0.2	0.05	35	8.7	192
Nb	0.8	na	na	2.6	3.1	na
Ta	0.6	0.02	0.03	3.8	6.1	190
La	0.7	0.3	0.4	8.1	11	27
Ce	0.6	0.4	0.7	11.8	19	28
Eu	1.1	0.4	0.4	9.2	8.7	25
Yb	0.8	2.0	2.6	48	63	24
Lu	0.8	2.0	2.5	48	62	24
Y	1.1	na	na	48	43	na
Th	2.8	0.9	0.3	23	8.4	25
U	0.6	0.1	0.2	6.2	11	65

Enrichment factors calculated as ratios of maximum element concentrations in different sample media (Table 2). **Db** – the <0.18 mm sieved fraction of bulk stream sediment (RGS sample medium), **Dc** – panned heavy mineral concentrate (HMC) of the <2 mm sieved fraction of stream sediment, **P** – rock (carbonatite), and **S** – soil. Values >1 indicate enrichment of the numerator medium relative to the denominator and vice versa. **na** - not analyzed.

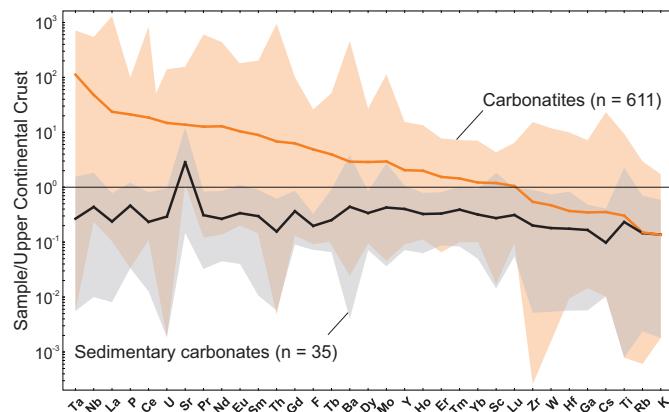


Fig. 4. Compositional ranges and mean compositions (solid lines) of carbonatites and sedimentary carbonate rocks in British Columbia, normalized to the upper continental crust of Rudnick and Gao (2005). Data from Mäder (1987), Locock (1994), Rukhlov and Gorham (2007), Gorham (2008), Mumford (2009), Chudy (2013), Simandl et al. (2013), Trofanenko (2014), Ya'acoby (2014), Chakhmouradian et al. (2015), Dalsin et al. (2015), Han et al. (2016), Trofanenko et al. (2016), and Han and Rukhlov (2020a).

5. Regional geochemical survey data

Collected as part of the Regional Geochemical Survey (RGS) program since 1976, the data for this study (Table 4) include the multi-element determinations for a total of 11502 stream-sediment (<0.18 mm fraction) and water samples at an average sampling density of about 1 site per 10 km² (Han and Rukhlov,

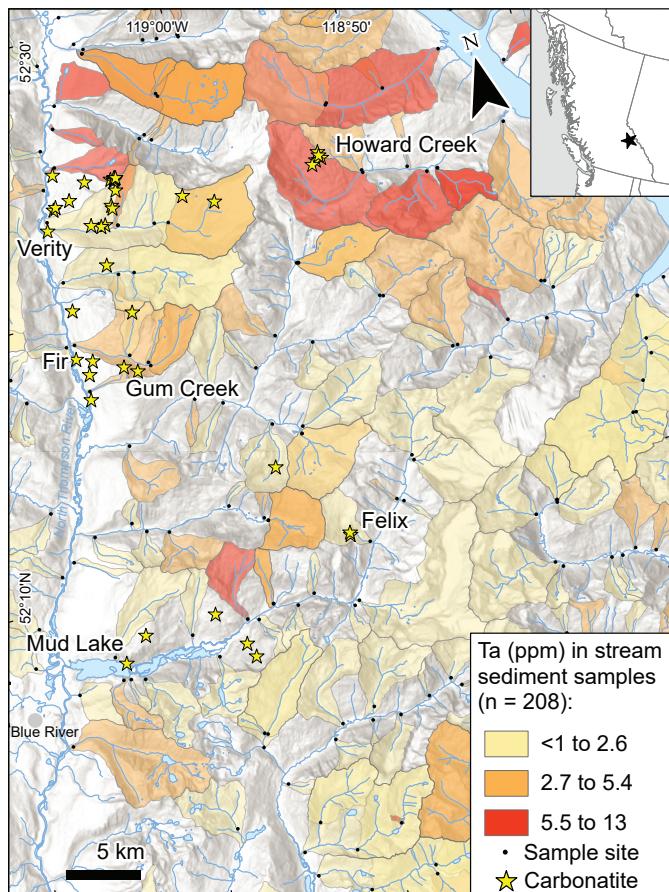


Fig. 5. Blue River area, tantalum-themed catchment basins of regional geochemical survey stream-sediment samples (<0.18 mm fraction; after Han and Rukhlov, 2017, 2020a). Carbonatite occurrences after Rukhlov and Gorham (2007), Gorham (2008), Gorham et al., 2009, 2011a, 2011b, 2013), Millonig and Groat (2013), and Rukhlov et al. (2018).

2017; 2020b; Lett and Rukhlov, 2017). Considering the emplacement ages of carbonatites and related rocks (>320 Ma), the selected stream sediment samples have catchment basins (Cui et al., 2009) that are underlain by the >320 Ma rocks in the study area (Fig. 2). The determinations by aqua-regia digestion with atomic absorption spectrometry (AAS) or a combination of inductively coupled plasma atomic emission spectroscopy (ICP-AES) and inductively coupled plasma mass spectrometry (ICP-MS) are underestimated for elements hosted by silicate and oxide minerals because of only partial dissolution in HCl-HNO₃ acid mixtures. In contrast, instrumental neutron activation analysis (INAA) provides total determinations. Elevated concentrations of carbonatite indicator elements such as La (Fig. 8), Ta (Fig. 9), U (Fig. 10), and P (Fig. 11) in the stream sediments reflect local background variations, including at known carbonatite and related rock occurrences.

6. Discriminant analysis method

Below we evaluate the multi-element drainage geochemical data to identify the carbonatite signature (Fig. 12). To construct

Table 4. Regional geochemical survey (RGS) lithogeochemical data from stream-sediment samples.

Variable	Method	Unit	Count	<MDL ¹	Mean	Minimum	Median	Percentiles			Maximum	Skewness	DPA ²
								87th	93rd	98th			
Area ³		km ²	11502	9.9	0.008	5.5	18	28	54	193	4		
Ag	AAS, ICP-MS	ppb	11073	0.13	122	<2	57	184	278	587	>100000	87	
Al	ICP-AES	wt%	11002	0.00	1.2	0.02	1.07	1.75	2.03	2.53	5.86	1	
As	AAS, ICP-MS	ppm	11467	0.68	9.2	<0.1	4.5	14	21	42	5370	68	
Au	ICP-MS	ppb	11263	11	14	<0.2	1.2	7.6	13	47	40630.4	78	
Ba	ICP-MS	ppm	11002	0.00	136	2.2	72	230	361	856	2769	5	
Ba	INAA	ppm	10104	0.48	835	<50	603	1200	1500	3400	37000	11	Y
Bi	AAS, ICP-MS	ppm	11117	0.97	0.29	<0.02	0.19	0.40	0.51	1.0	125	61	
Ca	ICP-AES	wt%	11002	0.01	2.5	<0.02	0.62	5.87	10.22	18.79	40.00	3	Y
Cd	AAS, ICP-MS	ppm	11298	2.0	0.53	<0.01	0.13	0.5	0.93	3	644.53	83	
Ce	INAA	ppm	10107	0.33	129	<3	100	210	270	430	1450	4	Y
Co	ICP-MS	ppm	11002	0.00	12	0.3	11	18	22	31	173	5	
Co	INAA	ppm	10107	5.4	15	<1	14	24	29	41	217	4	
Cr	ICP-MS	ppm	11002	0.00	27	0.7	21	43	55	90	1051	12	
Cr	INAA	ppm	10107	0.63	115	<5	81	170	230	390	8750	19	
Cs	INAA	ppm	10106	2.6	3.8	<0.5	3.1	6.0	7.6	12	67	5	
Cu	AAS, ICP-MS	ppm	11497	0.00	27	0.71	22	44	54	81	1200	14	
Eu	INAA	ppm	1851	5.0	2.0	<0.2	1.8	3.2	4.0	5.7	20	3	
F	ISE	ppm	4259	0.00	485	50	430	750	890	1189	2040	1	
Fe	ICP-AES	wt%	11002	0.00	2.4	0.04	2.25	3.51	3.95	4.75	26.0	2	Y
Fe	INAA	wt%	10107	0.12	3.6	<0.2	3.5	5.1	5.8	7.3	26.3	1	
Ga	ICP-MS	ppm	11002	0.08	3.5	<0.2	3.3	5.4	6.3	8.2	17	1	
Hf	INAA	ppm	10107	2.3	9.5	<1	8	15	20	31	308	8	
Hg	AAS, ICP-MS	ppb	11000	5.8	40	<5	26	66	92	166	4480	29	
K	ICP-AES	wt%	11002	0.32	0.13	<0.01	0.07	0.25	0.36	0.64	1.46	3	Y
La	ICP-MS	ppm	11002	0.06	20	<0.5	16	29	37	62	1146	19	
La	INAA	ppm	10107	0.48	71	<2	55	119	150	242	1160	5	Y
Li	ICP-MS	ppm	5743	0.00	21	0.5	18	38	46	60	131	1	
Lu	INAA	ppm	10107	28	0.48	<0.05	0.40	0.87	1.1	1.7	21	9	
Mg	ICP-AES	wt%	11002	0.00	0.96	0.03	0.64	1.48	2.18	4.96	21.79	5	Y
Mn	AAS, ICP-AES	ppm	11497	0.00	522	15	402	770	984	1765	>30000	21	Y
Mo	ICP-MS	ppm	11002	0.00	1.1	0.02	0.53	1.6	2.7	6.6	113	17	Y
Na	ICP-AES	wt%	11002	5.4	0.012	<0.001	0.006	0.022	0.032	0.06	1.76	30	
Na	INAA	wt%	10102	1.2	1.1	<0.1	1.0	1.9	2.2	2.7	10.3	1	
Nb	ICP-MS	ppm	5743	3.5	0.51	<0.02	0.27	1.1	1.4	2.3	8.3	4	
Nd	INAA	ppm	1462	0.21	70	<5	59	120	140	207	638	3	
Ni	ICP-MS	ppm	11002	0.00	33	0.3	25	49	62	112	2369	22	Y
P	ICP-AES	wt%	11002	0.00	0.09	0.007	0.074	0.129	0.157	0.234	1.064	8	Y
Pb	AAS, ICP-MS	ppm	11497	0.00	17	0.57	10	21	27	47	>20000	98	Y
Rb	INAA	ppm	10107	0.86	87	<5	84	121	140	170	400	1	
S	ICP-AES	wt%	11002	20	0.059	<0.01	0.03	0.10	0.14	0.31	6.77	21	
Sb	ICP-MS	ppm	11002	3.3	0.39	<0.02	0.15	0.58	0.98	2.3	297	81	
Sb	INAA	ppm	10106	25	0.88	<0.1	0.4	1.4	2.2	4.9	566	82	
Sc	ICP-MS	ppm	11002	0.00	2.6	0.1	2.3	4.0	5.0	7.4	20	2	
Sc	INAA	ppm	10107	0.04	12	<0.5	11	17	20	26	107	2	Y
Se	ICP-MS	ppm	11002	7.0	0.70	<0.1	0.4	1.2	1.7	3.2	58	19	
Sm	INAA	ppm	10107	0.19	10	<0.1	8.2	16	21	34	165	5	Y
Sr	ICP-MS	ppm	11002	0.00	66	0.3	31	127	199	357	1978	7	Y
Ta	INAA	ppm	10107	13	1.7	<0.5	1.5	2.8	3.4	5.7	70	12	
Tb	INAA	ppm	10107	14	1.3	<0.5	1.1	2.1	2.7	4.2	25	6	
Te	ICP-MS	ppm	11002	59	0.02	<0.01	<0.01	0.04	0.05	0.09	10	92	
Th	ICP-MS	ppm	11002	0.16	5.4	<0.1	4.4	9.5	12	16	363	28	
Th	INAA	ppm	10107	0.12	19	<0.2	14	30	41	75	488	7	Y
Ti	ICP-AES	wt%	11002	0.63	0.051	<0.001	0.029	0.112	0.144	0.216	0.990	4	Y
Tl	ICP-MS	ppm	11002	9.7	0.12	<0.02	0.08	0.23	0.31	0.46	3.3	7	
U	ICP-MS	ppm	11002	0.06	2.7	<0.1	1.3	4.1	6.8	15	244	18	
U	INAA	ppm	10105	0.25	6.7	<0.2	4.7	11	15	26	228	10	Y
V	AAS, ICP-MS	ppm	11199	0.32	30	<1	24	53	67	97	500	3	Y
W	ICP-MS	ppm	11002	58	0.49	<0.05	<0.05	0.5	1.2	4.3	101	22	
W	INAA	ppm	10106	78	1.7	<1	<1	3	4	9	1950	92	
Y	ICP-MS	ppm	5743	0.00	9.6	0.27	7.4	13	17	31	647	23	
Yb	INAA	ppm	10107	32	2.8	<0.2	3.0	5.0	6.3	9.5	59	3	
Zn	AAS, ICP-MS	ppm	11497	0.00	99	2.8	59	102	140	360	88000	89	Y
Zr	ICP-MS	ppm	5743	1.9	1.6	<0.1	1.2	2.9	3.7	5.7	103	25	
Zr	INAA	ppm	7035	28	394	<200	330	640	830	1400	15000	10	

¹ Percentage of values less than the minimum detection limit (MDL).² Variables having value of 'Y' (shaded rows) used in discriminant projection analysis (DPA).³ Area of catchment basin in square kilometres.

Analytical method abbreviations: **AAS** – aqua regia digestion and atomic absorption spectrometry, **ICP-AES** – aqua regia digestion and inductively coupled plasma atomic emission spectroscopy, **ICP-MS** – aqua regia digestion and inductively coupled plasma mass spectrometry, **INAA** – instrumental neutron activation analysis, **ISE** – Na₂CO₃+KNO₃ fusion followed by H₂O leach and ion selective electrode. Data from Han and Rukhlov (2017; 2020b).

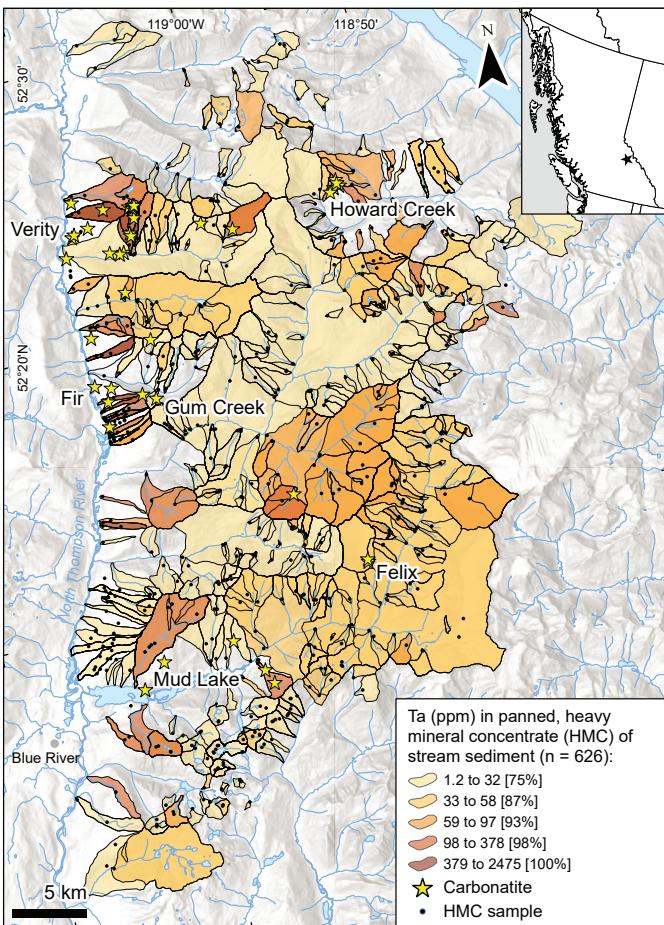


Fig. 6. Blue River area, tantalum-themed, percentile-ranked catchment basins of panned, stream-sediment (<2 mm fraction), heavy mineral concentrate (HMC) samples. Data and carbonatite occurrences from Dahrouge and Reeder (2001), Reeder and Dahrouge (2002), Smith and Dahrouge (2003), Dahrouge and Wolbaum (2004), Rukhlov and Gorham (2007), Gorham (2008), Gorham et al. (2009, 2011a, 2011b, 2013).

the best criteria for discriminating between carbonatite and other signals using multi-element stream-sediment data, we performed a discriminant projection analysis (DPA) on a sub-set of stream-sediment data taken from the regional geochemical survey area (Fig. 13) in ioGAS™ software. DPA is a supervised multivariate statistical technique that determines an optimum projection of multivariate data into a lower dimensional (e.g., bivariate) space to achieve the best separation between user-defined groups (Flury, 1997). The DPA uses an a priori knowledge of the group memberships to define discriminant parameters (DP1, DP2, DPn) that maximize the ratio of the within-groups sum of squares (W) to between-groups sum of squares (B) matrices (W/B). The between-groups matrix is effectively the covariance of the group means and the within-groups matrix is the weighted covariance matrix for the groups. Our group 1 includes stream-sediment data that are downstream of known carbonatite or related rock occurrences ($n=53$); group 2 consists of stream sediments derived mainly from carbonate sedimentary rocks ($n=90$); and group 3 is a random

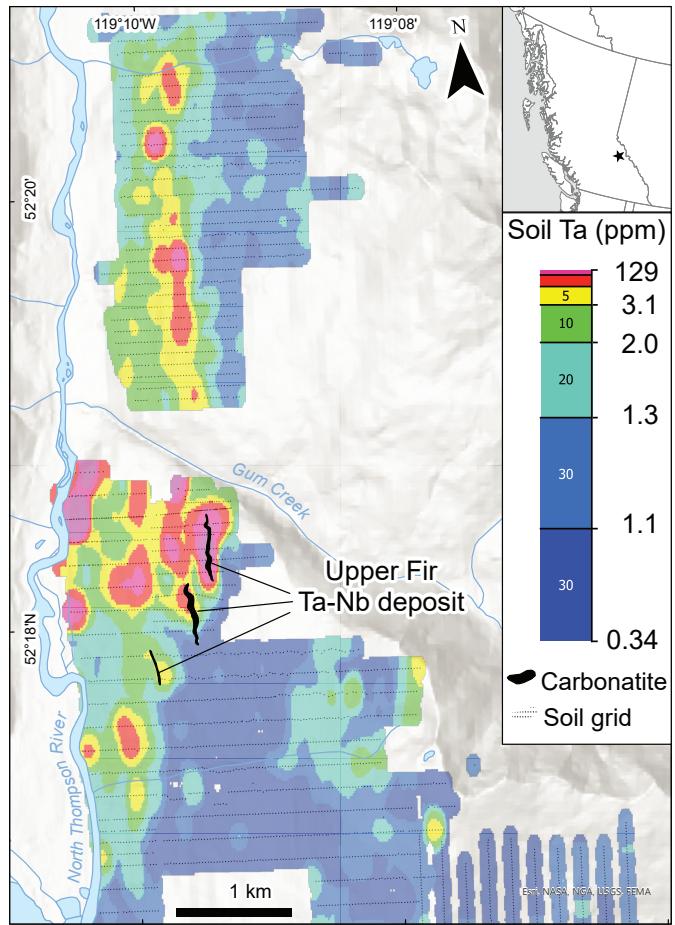


Fig. 7. Blue River area, percentile-gridded, Ta concentrations (ppm) in soil samples (after Reeder and Dahrouge, 2002; Smith and Dahrouge, 2003; Dahrouge and Wolbaum, 2004; Rukhlov and Gorham, 2007; Gorham, 2008; Gorham et al., 2009, 2011a, 2011b). Upper Fir carbonatite-hosted Ta-Nb deposit footprint after Kraft (2011) and Gorham et al. (2013).

20% sub-sample of all stream-sediment data in the study area ($n=1943$; Fig. 13). A ranked element contrast (REC) plot for the average Group 1 stream sediment relative to the median stream sediment in the study area (Fig. 14a) reveals maximum contrast of the carbonatite and related-rock association (Th-K-REE-W-U-Ti-Ta-Hf-Mo-P-Sr). In contrast, the average of a random sub-set of stream sediment samples (Group 3) shows maximum contrast of precious and base metals, along with other pathfinders of hydrothermal ore deposits (Fig. 14b); the ranked element contrast profile for group 2 is similar to that of group 3. Based on the ranked element contrast associations, we selected 20 elements that have <1% of values below the minimum detection limit for the discriminant projection analysis (Table 4). We performed the discriminant projection analysis on a random 50% training sub-set of the group 1 to 3 data (Fig. 12). The constructed discriminant parameters (DP1 and DP2) are linear combinations of the log₁₀-transformed element concentrations that optimally separate the data of groups 1 to 3 (Table 5) as summarized in Equations 1 and 2.

$$\text{DP1} = 0.5125 \cdot \text{Ba} - 0.1132 \cdot \text{Ca} + 0.019 \cdot \text{K} - 0.3508 \cdot \text{Mg} - 0.7108 \cdot \text{Mn} - 0.7341 \cdot \text{Mo} - 0.8466 \cdot \text{Ni} - 0.8647 \cdot \text{P} + 0.8483 \cdot \text{Pb} + 1.108 \cdot \text{Sc} - 0.8055 \cdot \text{Sm} - 0.1415 \cdot \text{Sr} - 2.78 \cdot \text{Th} - 0.1456 \cdot \text{Ti} + 0.2234 \cdot \text{U} - 0.2579 \cdot \text{V} + 0.9749 \cdot \text{Zn} - 2.181 \cdot \text{La} + 2.225 \cdot \text{Fe} + 1.957 \cdot \text{Ce} + 0.038113 \quad (\text{Eqn. 1})$$

$$\text{DP2} = 1.008 \cdot \text{Ba} - 0.6432 \cdot \text{Ca} - 0.834 \cdot \text{K} + 2.034 \cdot \text{Mg} - 0.1111 \cdot \text{Mn} - 0.2682 \cdot \text{Mo} - 0.5129 \cdot \text{Ni} - 0.2222 \cdot \text{P} + 0.3704 \cdot \text{Pb} + 0.01757 \cdot \text{Sc} - 0.02553 \cdot \text{Sm} - 0.7922 \cdot \text{Sr} - 1.464 \cdot \text{Th} + 1.035 \cdot \text{Ti} + 2.46 \cdot \text{U} - 0.8935 \cdot \text{V} + 0.2481 \cdot \text{Zn} + 0.2286 \cdot \text{La} + 0.5293 \cdot \text{Fe} - 0.7624 \cdot \text{Ce} + 1.6581 \quad (\text{Eqn. 2})$$

Calculated contours of constant Mahalanobis distance at $\chi^2 = 0.975$, using robust multivariate estimation (Campbell, 1980), outline most of the data in each group (Fig. 15a). Projecting a random 50% test sub-set of the group 1 to 3 data, which were not used in the discriminant projection analysis, into the DP1 versus DP2 space validates separation of most of the data in each group (Fig. 15c). The DP1 contributes 73% in discriminating the stream-sediment data downstream of known carbonatites and other stream-sediment data, with the calculated Pearson correlation coefficients of the DP1 variables showing significant contributions of Th, La, Sm, Ce, U, K, P, and Ti (Table 5).

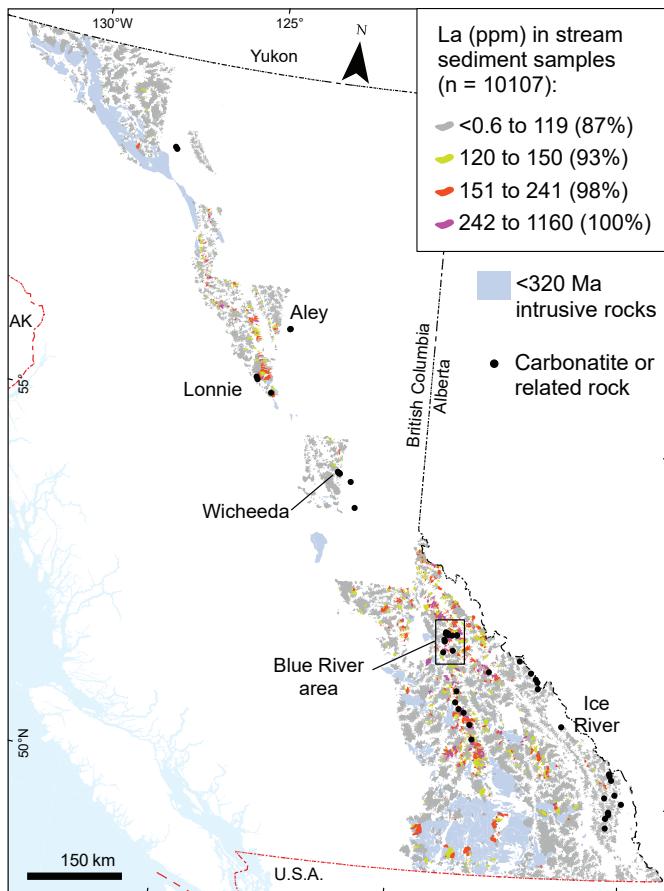


Fig. 8. Regional geochemical survey area, lanthanum-themed, percentile-ranked catchment basins of stream-sediment samples (<0.18 mm fraction; after Han and Rukhlov, 2017, 2020b). Carbonatite and related-rock occurrences after Höy (1988), Parrish and Scammell (1988), Pell (1994), Millonig and Groat (2013), and Rukhlov et al. (2018). The post-320 Ma intrusive rocks from BC Digital Geology version 2021-12-19 (Cui et al., 2017).

We then used the training sub-set of the group 1 to 3 data to construct the DP1 versus DP2 discrimination diagram with a boundary separating most of the data from stream sediments downstream of known carbonatite or related-rock occurrences and other stream-sediment data (combined groups 2 and 3) using the Auto-Domain Classification Diagram tool in ioGAS™, based on the lowest Mahalanobis distance (Fig. 15b). The final discrimination diagram DP1 versus DP2 with the statistically defined boundary separating carbonatite or related rock sources and other rock sources is validated by the test sub-set of the stream-sediment data (Fig. 15d).

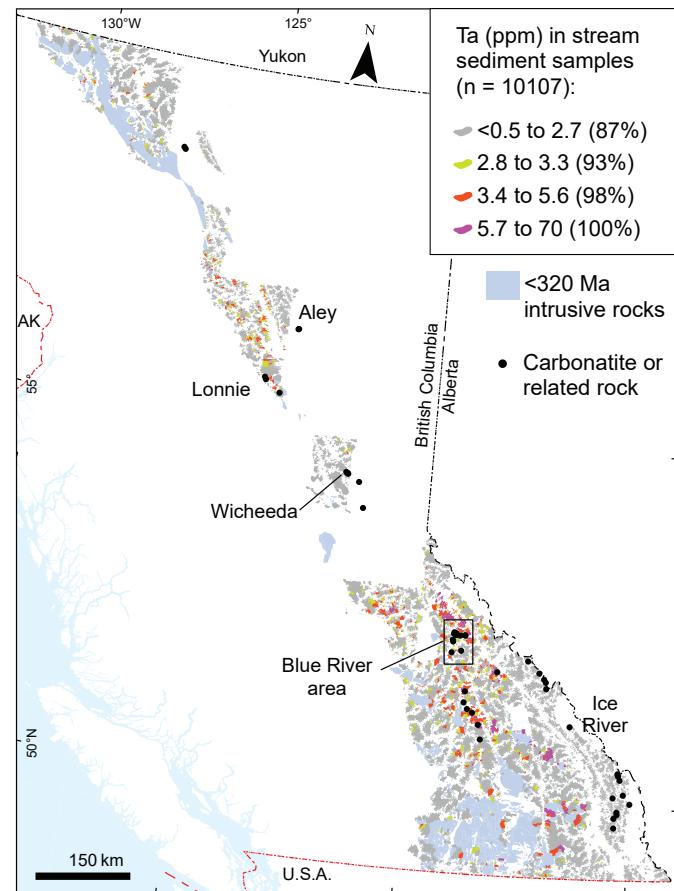


Fig. 9. Regional geochemical survey area, tantalum-themed, percentile-ranked catchment basins of stream-sediment samples (<0.18 mm fraction; after Han and Rukhlov, 2017, 2020b). Carbonatite and related-rock occurrences after Höy (1988), Parrish and Scammell (1988), Pell (1994), Millonig and Groat (2013), and Rukhlov et al. (2018). The post-320 Ma intrusive rocks from BC Digital Geology version 2021-12-19 (Cui et al., 2017).

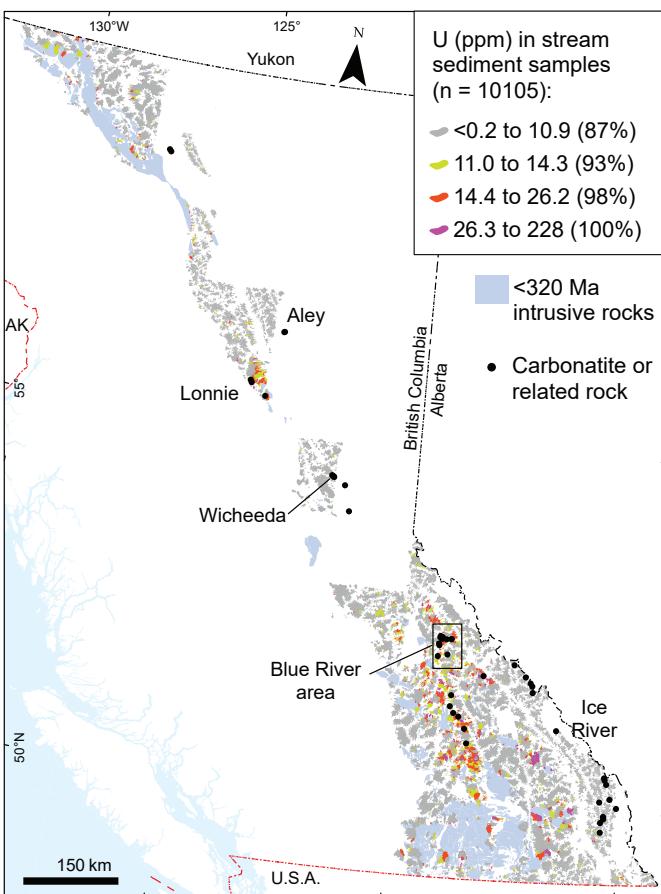


Fig. 10. Regional geochemical survey area, uranium-themed, percentile-ranked catchment basins of stream-sediment samples (<0.18 mm fraction; after Han and Rukhlov, 2017, 2020b). Carbonatite and related-rock occurrences after Höy (1988), Parrish and Scammell (1988), Pell (1994), Millonig and Groat (2013), and Rukhlov et al. (2018). The post-320 Ma intrusive rocks from BC Digital Geology version 2021-12-19 (Cui et al., 2017).

7. Discriminant analysis results

7.1. A ‘carbonatite index’

Applied to all the regional geochemical survey stream-sediment data in the study area, the DP1 versus DP2 discrimination diagram identified a total of 721 stream sediment samples (6.3% of all the data) showing a multivariate carbonatite or related rock signal (Fig. 16). Herein we refer to this signal, which is recast as the DP1 value multiplied by minus one, as the ‘carbonatite index’. The percentile-ranked index highlights prospective stream basins for carbonatite- and related rock-hosted critical metals (Fig. 17). Stream-sediment data showing carbonatite index scores greater than the 93rd percentile (n=50) reveal elevated concentrations of carbonatite indicator elements such as REE, Nb, Ta, Ti, Zr, Hf, Th, U, P, K, and Na, along with indicators of other mineralization-types (e.g., granitoid-related rare metals, precious and base metals), relative to the median (background) concentrations in stream sediments of the study area (Table 6). These geochemical anomalies conspicuously follow a trend of known carbonatite

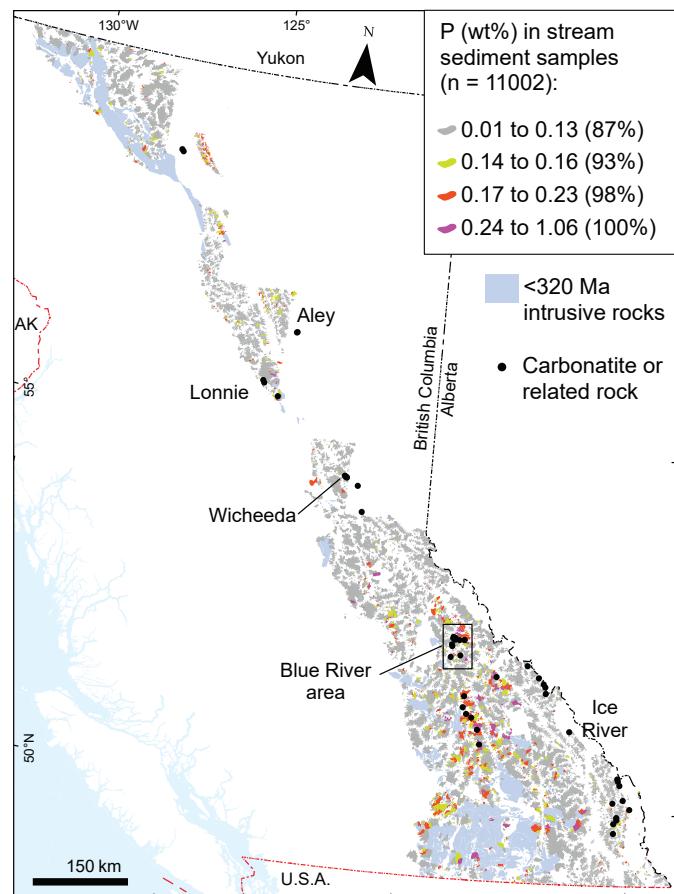


Fig. 11. Regional geochemical survey area, phosphorous-themed, percentile-ranked catchment basins of stream-sediment samples (<0.18 mm fraction; after Han and Rukhlov, 2017, 2020b). Carbonatite and related-rock occurrences after Höy (1988), Parrish and Scammell (1988), Pell (1994), Millonig and Groat (2013), and Rukhlov et al. (2018). The post-320 Ma intrusive rocks from BC Digital Geology version 2021-12-19 (Cui et al., 2017).

occurrences in the Blue River and Frenchman Cap areas, extending it both to the northwest and southeast (Fig. 18).

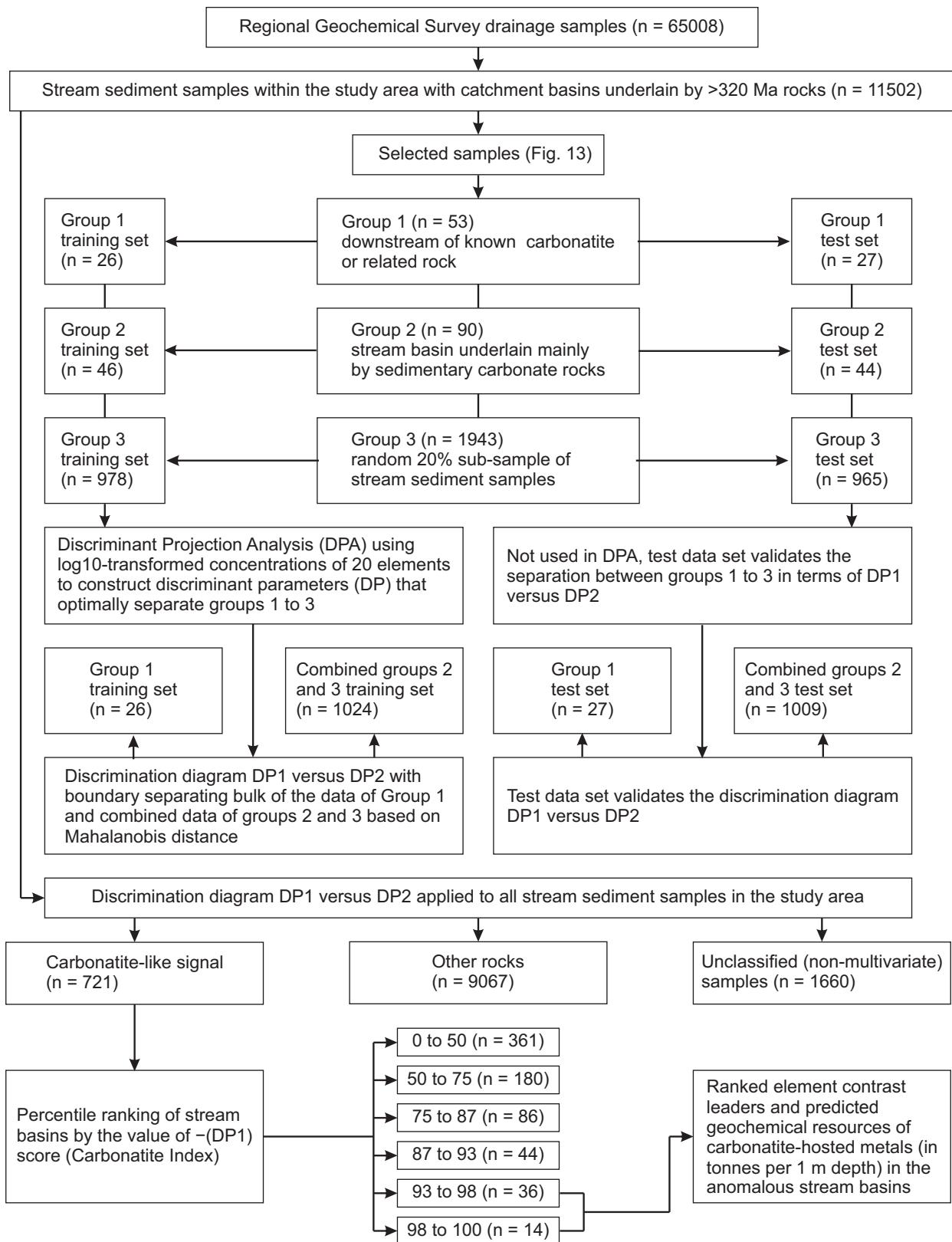
7.2. Measure of geochemical anomaly

The quantity of metal above background (or predicted geochemical resources) based on productivity of an element in a dispersion stream is a parametric and thus an objective measure of a geochemical anomaly, which is the basis of evaluating prospective areas (Rukhlov et al., 2020a, b). The productivity of an element in an ideal dispersion stream, P_x (in $\text{m}^2\%$) is given in Equation 3.

$$P_x = (1/k) \cdot S_x \cdot (C_x - C_b) \quad (\text{Eqn. 3})$$

where

$k < > 1$ is the local proportionality coefficient between the productivity of an element in the dispersion stream and productivity of an element in the secondary or residual dispersion halo (soil anomaly), which depends on hydrography and individual properties of elements resulting in their supergene enrichment or leaching.

**Fig. 12.** Workflow used in this study.

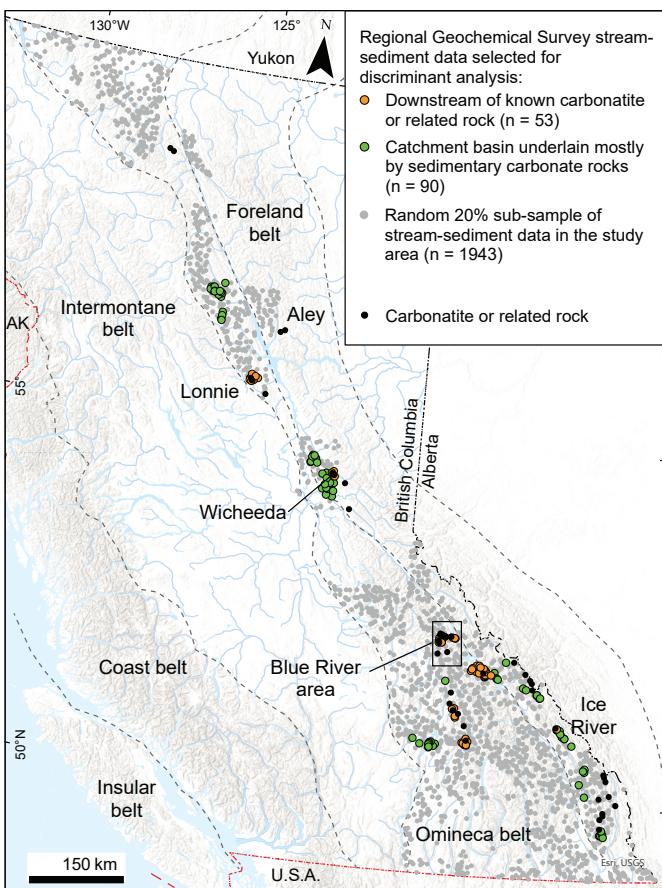


Fig. 13. Selected stream-sediment data from regional geochemical survey area used in training and validating discriminant analysis (after Han and Rukhlov, 2017, 2020b). Carbonatite and related-rock occurrences after Höy (1988), Parrish and Scammell (1988), Pell (1994), Millonig and Groat (2013), and Rukhlov et al. (2018). Morphotectonic boundaries after Gabrielse et al. (1991).

Table 5. Coefficients for discriminant parameters.

Variable	DP1		DP2	
	Projection ¹	Correlation ²	Projection ¹	Correlation ²
log (Ba ppm)	0.5125122	-0.023	1.0080582	0.369
log (Ca wt%)	-0.1132357	0.213	-0.6432403	-0.602
log (K wt%)	0.0190009	-0.458	-0.8340383	0.250
log (Mg wt%)	-0.3508289	0.167	2.0342928	-0.024
log (Mn ppm)	-0.7107679	0.159	-0.1111433	0.119
log (Mo ppm)	-0.7341133	-0.104	-0.2681685	0.191
log (Ni ppm)	-0.8465542	0.004	-0.5128953	0.209
log (P wt%)	-0.8646799	-0.434	-0.2221829	0.034
log (Pb ppm)	0.8482791	0.251	0.3704230	0.073
log (Sc ppm)	1.1082783	-0.233	0.0175719	0.362
log (Sm ppm)	-0.8054578	-0.722	-0.0255307	0.228
log (Sr ppm)	-0.1415442	0.113	-0.7922373	-0.636
log (Th ppm)	-2.7800825	-0.792	-1.4640319	0.140
log (Ti wt%)	-0.1456117	-0.383	1.0353020	0.505
log (U ppm)	0.2233847	-0.648	2.4597713	0.497
log (V ppm)	-0.2578595	-0.102	-0.8935080	0.217
log (Zn ppm)	0.9749035	0.171	0.2481454	0.192
log (La ppm)	-2.1814112	-0.740	0.2285954	0.156
log (Fe wt%)	2.2246271	-0.027	0.5292998	0.188
log (Ce ppm)	1.9567853	-0.718	-0.7624135	0.128
Constant	0.0381130		1.6581413	
Eigenvalue ³	0.286 (73%)		0.103 (27%)	

¹ Scaled eigenvectors or projection coefficients.

² Pearson correlation coefficients between the input data and the projected data indicating contributions of each element in discriminating between groups of samples.

³ Relative significance of discriminant parameter in terms of eigenvalues of the within-groups sum of squares to between-groups sum of squares ratio matrix and their percentage values (in parentheses).

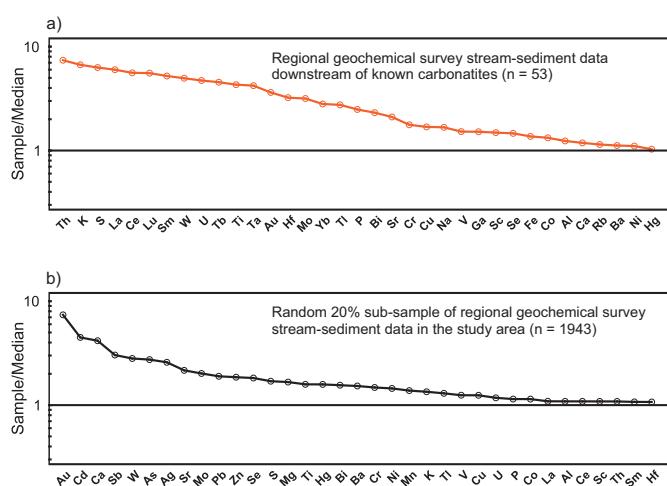


Fig. 14. Ranked element contrast plots relative to the median regional geochemical survey stream-sediment data (Table 4; after Han and Rukhlov, 2017, 2020b). **a)** The average stream-sediment data downstream of known carbonatite or related-rock occurrences. **b)** The average random 20% sub-sample of stream-sediment data.

S_x is the catchment area of stream basin at the sampling site (in m^2)

C_x is the concentration of an element in the stream sediment sample (in wt %).

C_0 is the background concentration of an element (in wt-%).

C_b is the background concentration of an element (in wt. %). We estimate local background as the geometric mean or median concentration of an element in stream sediment samples ($n=20$ to 170) that are close to a geochemical anomaly. Downstream of a metal source (secondary dispersion halo), P_x is constant in an ideal dispersion stream, which may not be true for the second- or higher-order drainages (Rukhlov et al., 2020b).

Assuming $k=1$, the quantity of metal above background or predicted geochemical resources of an element in the stream basin, q (in tonnes per 1 m depth), is calculated as

$$q = H \cdot P_x / 40 = P_x / 40 \quad (\text{Eqn. 4})$$

where

$H=1$ is the calculation depth (in m).

P_x is the productivity of an element in dispersion stream (in m²%). The denominator '40' converts m²% into tonnes. We set the calculation depth, which refers to a probable depth of a mineralized zone, to a constant value of 1 m to simplify

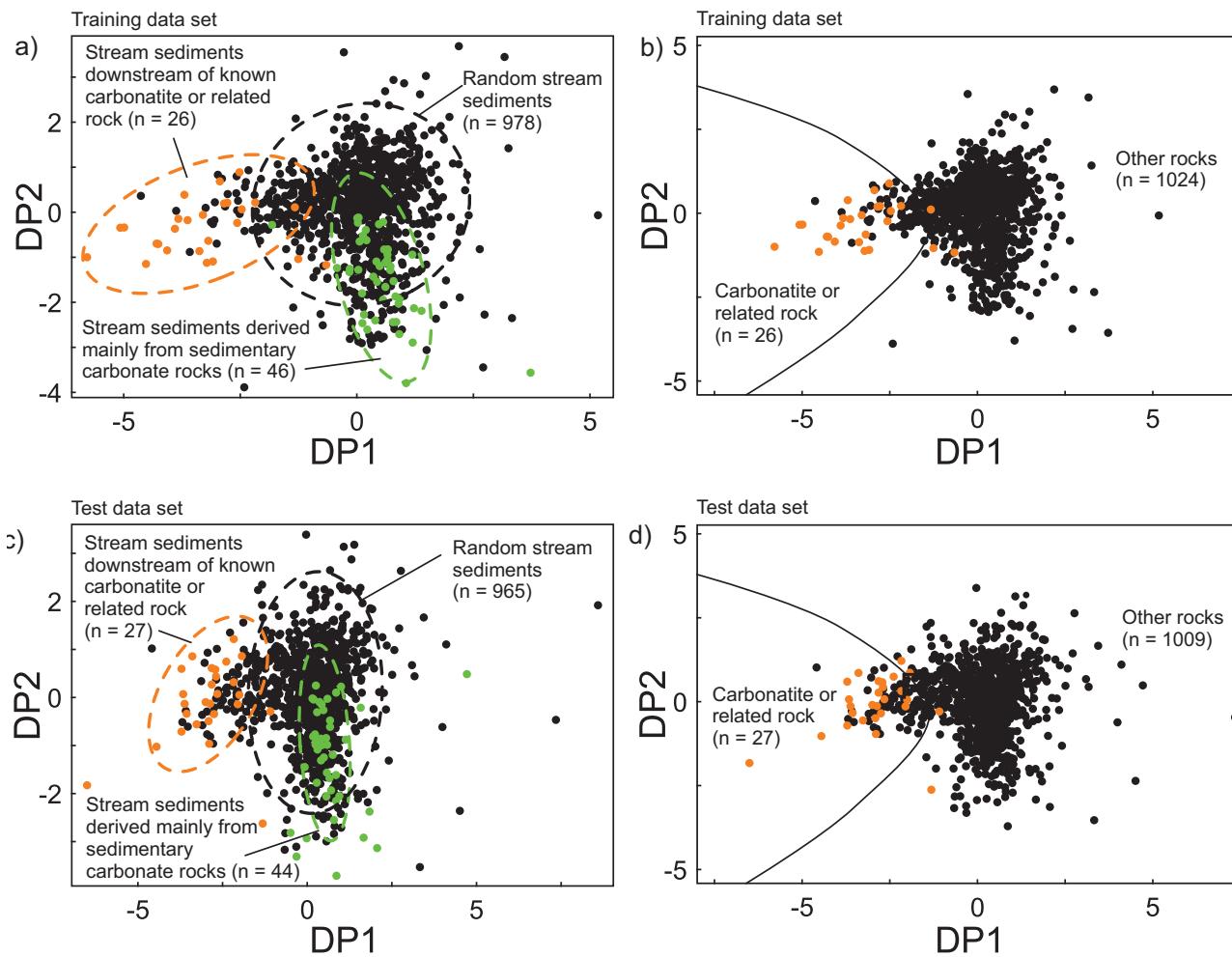


Fig. 15. Discrimination diagram DP1 versus DP2 for regional geochemical survey stream-sediment data. The discriminant-parameter (DP) variables are log₁₀-equivalents of element concentrations in parts per million (ppm) for all elements, except Ca, K, Mg, P, Ti, and Fe which are in weight per cent (wt.%; Table 5). **a)** DP1 versus DP2 for the training data set. **b)** DP1 versus DP2 for the training data set. **c)** DP1 versus DP2 for the test data set (not used in the discriminant analysis). **d)** DP1 versus DP2 for the test data set (not used in the discriminant analysis).

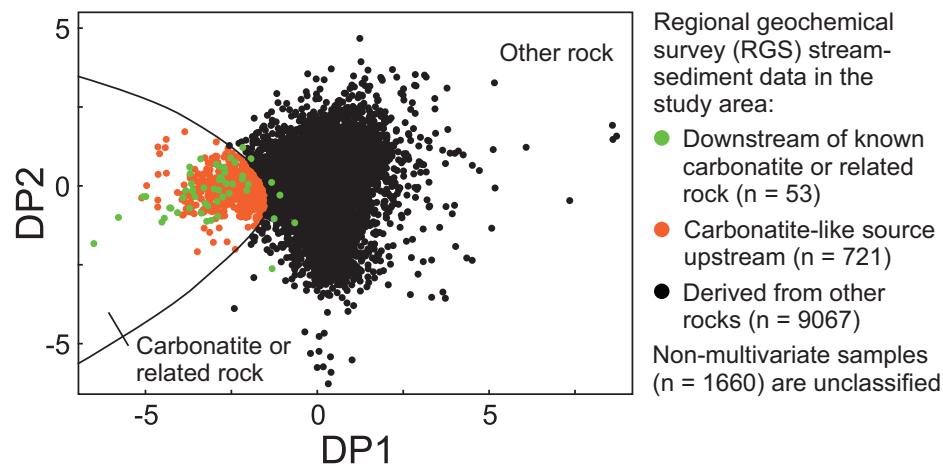


Fig. 16. Discrimination diagram DP1 versus DP2 for all regional geochemical survey stream-sediment data in the study area (after Han and Rukhlov, 2017, 2020b), showing boundary between carbonatite or related-rock sources and other rock sources in the upstream basins. Stream-sediment data downstream of known carbonatite and related-rock occurrences shown separately (green symbols).

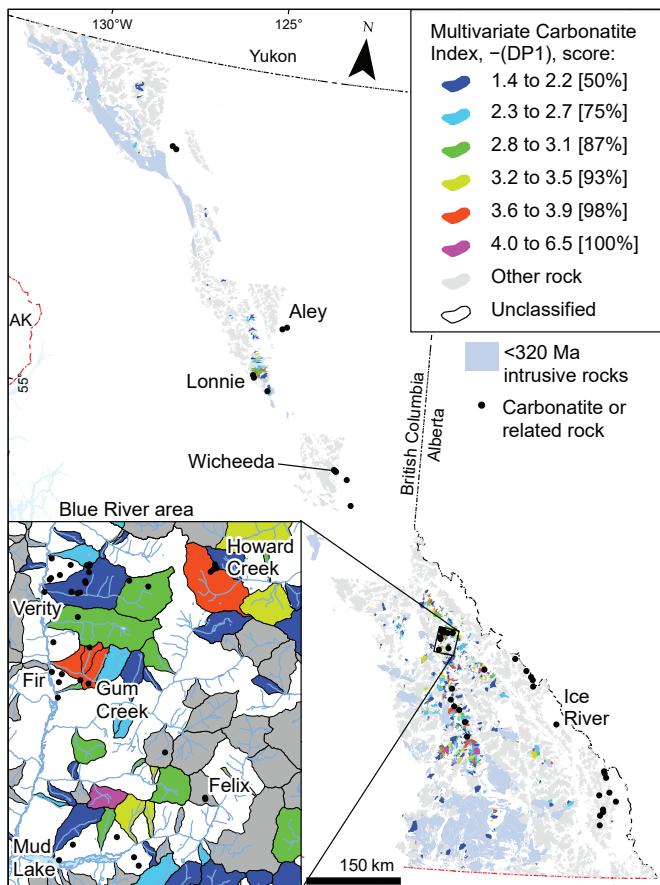


Fig. 17. Carbonatite index-themed, percentile-ranked catchment basins of regional geochemical survey stream-sediment samples (<0.18 mm fraction; after Han and Rukhlov, 2017, 2020b). Carbonatite and related-rock occurrences after Höy (1988), Parrish and Scammell (1988), Pell (1994), Millonig and Groat (2013), and Rukhlov et al. (2018). The post-320 Ma intrusive rocks from BC Digital Geology version 2021-12-19 (Cui et al., 2017).

equation 4 for a regional evaluation of geochemical anomalies (see Rukhlov et al., 2020a, b for details about predicted geochemical resources). The estimated predicted geochemical resources (in tonnes of metal per 1 m depth) suggest significant potential for REE and other carbonatite-hosted critical metals in the identified stream basins (Table 6).

7.3. Application to Aley carbonatite complex

Here we use an example of detailed stream-sediment lithogeochemical data downstream of the Aley carbonatite (Mackay and Simandl, 2014) to evaluate the mechanical dispersion of carbonatite indicators and ultimately the utility of the estimated predicted geochemical resources assuming an ideal dispersion (Table 6). Hosted by siliciclastic and carbonate rocks of the Kechika Group (Cambrian to Ordovician), the Aley carbonatite complex (ca. 370 Ma; Fig. 19) contains an NI 43-101-compliant resource of 286 million tonnes (Measured+Indicated) grading 0.37% Nb₂O₅ plus 144 million tonnes (Inferred) grading 0.32% Nb₂O₅ (Jones et al., 2017), making it the largest Nb deposit of the Cordilleran alkaline

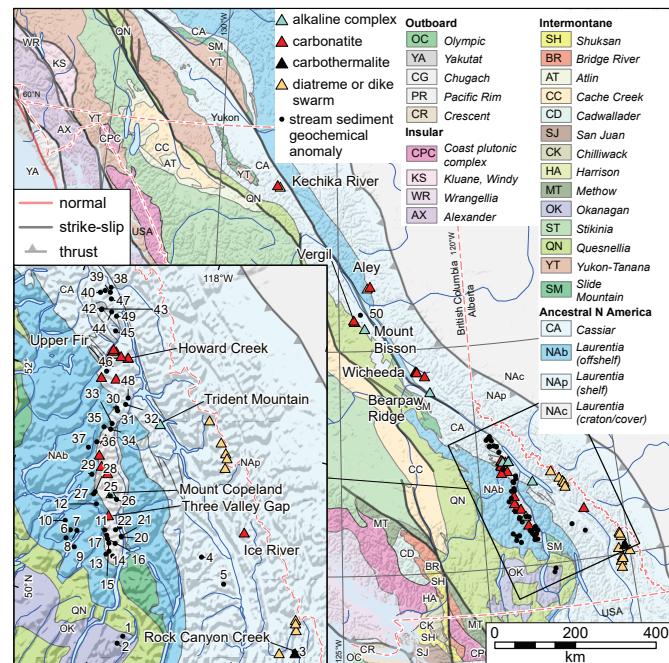


Fig. 18. Stream-sediment geochemical anomalies showing carbonatite index score greater than the 93rd percentile (see Table 6 for locations and other details). Carbonatite and related-rock occurrences after Höy (1988), Parrish and Scammell (1988), Pell (1994), Millonig and Groat (2013), Rukhlov et al. (2018). Terranes after Colpron (2020).

province (Pride, 1983; Mäder, 1987; McLeish, 2013; Chakhmouradian et al., 2015; McLeish and Johnston, 2019; McLeish et al., 2020). Based on determinations using a portable X-ray fluorescence analyzer (Mackay and Simandl, 2014), concentrations of carbonatite indicator elements such as Nb, La, Y, and Sr in the 0.125 to 0.250 mm fraction of stream sediment decrease exponentially downstream of the carbonatite (Fig. 20). Calculated using equation 3 above, productivities of Nb and Y obey an ideal dispersion law (i.e. $P_x \approx \text{constant}$ downstream of the anomaly) even in the second- and higher-order drainage system (Fig. 19). In contrast, productivity of La decreases and productivity of Sr increases downstream of the Aley carbonatite (Fig. 20). The main hosts of Nb and Y at Aley are the pyrochlore supergroup, ferrocolumbite, and other Nb-minerals, including euxenite (Chakhmouradian et al., 2015). These minerals are more resistant relative to monazite and REE ±F-carbonate minerals, which are the main hosts of REE at Aley (Mäder, 1987; Mackay and Simandl, 2014; Simandl et al., 2017). The increasing productivity of Sr (Fig. 20d) probably reflects contribution from the carbonate country rocks of the Kechika and Road River groups and the Skoki Formation downstream of the Aley carbonatite complex (Fig. 19). More detailed surveys are needed to evaluate dispersion of critical metals downstream of other known carbonatite and related-rock occurrences, but the data from Aley indicate that resistant critical minerals generally obey an ideal dispersion law even in the high-order streams, thereby validating the estimated predicted geochemical resources (Table 6).

Table 6. Regional geochemical survey stream-sediment data showing carbonatite index scores >93rd percentile and predicted geochemical resources (q) of carbonatite-hosted metals (in tonnes per 1 m depth) in stream basins.

N ¹	RGS sample	X ²	Y ²	S ³	Cl ⁴	Ranked element contrast leaders ⁵	Predicted geochemical resources (in tonnes per 1 m depth) ⁶	Nb	Ta	La	Ce	Nd	Sm	Eu	Tb	Yb	Lu	Y	Sc	Mo	Th	U	Predominant host rock
1	082F775336	-117.623	49.883	25	3.7	W-41, Nb-14, Au-12, U-11, Th-9, Mo-8, La-7, (Ce, Cd)-5, (Sm, Se, Na, Ti, Th)-3, (K, Te)-2	0	16	13709	19844	na	551	na	16	0	0	0	0	158	4897	983	metamorphic rocks (Harper Ranch assemblage; Carboniferous to Permian)	
2	082F775345	-117.670	49.804	3	3.7	Nb-17, Mo-9, K-8, Th-6, (P, La, Ti, Th)-5, Ce-4, (Y, Hf, Te, W, Sm, U, Cd)-3, (Zr, Ba, V)-2	2.5	0	1019	1528	na	48	na	0.51	0	0	119	0	27	324	na	metamorphic rocks (Harper Ranch assemblage; Carboniferous to Permian)	
3	082I901240	-115.162	50.223	2	3.5	Ca-21, Au-14, (Ba, Mo)-13, Sb-11, Mg-10, La-9, (Te, Cd)-8, Th-7, (Sb, Sr, W)-6, Th-5	2.6	0	936	1749	na	66	na	3.5	0	2.3	86	0	37	185	2.2	limestone, slate, silstone, argillite (Banff and Exshaw formations; Mississippian)	
4	082K775330	-116.879	50.820	19	4.6	Bi-147, Th-82, U-48, Ta-47, Te-29, La-23, Nb-22, W-19, (Hf, V, Zr, Tb)-6, Ce-5, (Fe, V, Na, Y)-3	280	3257	24198	16529	na	712	na	219	0	0	682	0	17	22111	10485	coarse siliciclastic rocks (Horsethief Creek Group; Neoproterozoic)	
5	082K777088	-116.445	50.639	13	3.6	W-158, Nb-22, U-17, Ta-11, Th-9, La-7, U*-6, Mo-5, F*-4, (Ce, P, Y, Na)-3, (Bi, Rb, Cs)-2	192	502	3531	4663	na	55	na	3.3	0	0	447	0	67	1785	1916	quartz arenite (Purcell Supergroup; Mount Nelson and Dutch Creek formations; Mesoproterozoic)	
6	082L763159	-118.876	50.589	15	3.7	Th-13, Ce-8, (Sm, Hf, Zr, Lu, La, Tb)-6, U-5, (W, Na)-4, (Cr, Ti)-3, (Ta, K, Sc, P, S, Yb)-2	na	54	7931	19105	na	1261	na	144	37	59	na	218	0	4979	553	limestone, marble, calcareous sedimentary rocks (Shuswap assemblage; Proterozoic to Paleozoic)	
7	082L763165	-118.855	50.674	23	3.6	(Au, Th)-11, Lu-8, Sm-7, (Ce, La, W)-6, (Tb, Hf, U, Na)-5, (Zr, Ti)-4, (Cr, K, Yb, Sc, Ta)-3	na	125	14252	23051	na	2071	na	205	289	139	na	771	0	6184	757	limestone, marble, calcareous sedimentary rocks (Shuswap assemblage; Proterozoic to Paleozoic)	
8	082L763168	-118.759	50.688	69	5.0	Au-56, W-54, Th-23, Lu-19, Sm-13, La-12, (U, Tb)-11, (Ce, Hf)-9, Zr-8, (Na, Yb)-4, (Ta, Cr)-3	na	562	96127	106998	na	14985	na	1690	1202	1185	na	1743	0	47411	7265	limestone, marble, calcareous sedimentary rocks (Shuswap assemblage; Proterozoic to Paleozoic)	
9	082L763198	-118.717	50.535	20	5.1	Au-50, Th-25, Lu-22, La-17, (Sm, Ce)-15, U-14, Tb-12, (Hf, Zn)-11, W-7, Na-6, Cr-5, Yb-4	na	135	38930	58564	na	4990	na	529	439	381	na	583	0	15229	2753	quartz arenite (Shuswap assemblage; Proterozoic to Paleozoic)	
10	082L765210	-118.973	50.742	14	3.8	Th-9, Zr-8, (Hf, Ce, La, Sm)-6, Lu-5, (U, Tb, Na)-4, (K, Ti, Cd)-3, (Na, P, Ba, Yb, Sc, Ti)-2	na	0	6794	11991	na	941	na	86	34	34	na	52	0	2663	384	metamorphic rocks (Shuswap assemblage; Proterozoic to Paleozoic)	
11	082L765223	-118.372	50.778	46	3.6	W-34, Th-10, (Zr, Ce, La)-8, Sm-7, (Hf, Lu, Na)-6, (U, Tb, Ti, K)-5, (Cr, Yb)-3, (Sc, P, Ti)-2	na	129	30216	49718	na	3647	na	229	228	114	na	1197	25	9061	976	limestone, marble, calcareous sedimentary rocks (Shuswap assemblage; Proterozoic to Paleozoic)	

¹ Anomaly number as shown in Fig. 18.² Longitude (X) and latitude (Y) coordinates (in decimal degrees) of regional geochemical survey (RGS) stream sediment sample.³ Area of catchment basin (in km²).⁴ Carbonatite index (Cl) score values: Cl = -DP1 (Table 5).⁵ Ranked element contrast (REC) leaders are sorted (maximum to minimum) element concentrations normalized to the median values of regional geochemical survey stream-sediment data (Table 4). REC leaders reveal associations of elements showing the maximum contrast to background and the magnitude of the contrast (geochemical anomaly), which identify mineralization in the stream basin.⁶ Anomalies 1 to 37 and 50 lack Nd and Eu determinations; anomalies 6 to 50 lack Nb and Y determinations. F* and U* are fluorine and uranium determinations in stream water. na = not analyzed.⁶ Predicted geochemical resources in the stream basin (in tonnes of metal per 1 m depth); see text for details.

Table 6. Continued.

N ¹	RGS sample	X ²	Y ²	S ³	Cl ⁴	Ranked element contrast leaders ⁵	Predicted geochemical resources (in tonnes per 1 m depth) ⁶							Predominant host rock							
12	082L763268	-118.616	50.985	3	3.9	(Th, W)-12, Lu-10, (U*, La, Ce)-8, Sm-7, (U, Tb)-6, Yb-5, (Hf, Ti*)-4, (Ta, W, Na, Ti)-3	na	20	2006	3606	na	236	na	68	24	na	57	0	841	115	
13	082L769008	-118.239	50.566	9	3.6	(K, Th)-10, (W, Lu)-9, (Na, Ce)-7, (Ti, La, Tb, Zr, Sm, U, Ti)-6, Hf-5, Cr-4, (Yb, Sc, P)-3, Ga-2	na	27	4072	6882	na	538	na	76	87	50	na	155	8.0	1600	282
14	082L769010	-118.210	50.602	16	3.5	W-18, Th-8, Lu-7, (Ce, U, Tb, Sm, La)-5, (Yb, Au, K, Zr)-4, (Na, Ti, Hf)-3, (Tl, P, Cr, Na)-2	na	21	4561	8113	na	649	na	101	282	52	na	121	0	2198	394
15	082L769070	-118.252	50.674	7	4.4	(Lu, Th)-16, W-14, U-12, Tb-11, Sm-10, (Ce, La)-9, Yb-7, Al-6, (K, Zr)-5, (Hf, Na)-4, Ti-3	na	9.3	6323	10626	na	934	na	153	263	91	na	35	0	2871	726
16	082L769071	-118.290	50.707	3	4.4	Lu-24, Th-22, U-16, Tb-15, (La, Sm, Ce)-13, Yb-10, Zr-7, (W, Hd)-6, K-5, (Cr, Ti)-4, Na-3	na	12	4920	7943	na	690	na	111	209	71	na	64	0	2103	540
17	082L769075	-118.319	50.738	4	3.9	W-23, Th-17, Tb-12, (Lu, U)-11, Sm-10, (La, Ce)-9, Ti-7, (Zr, Yb)-6, Hf-5, (Au, Rb, Ba)-4	na	30	3495	5656	na	581	na	99	122	33	na	82	7.3	1788	379
18	082L769153	-118.135	50.570	7	3.9	W-30, Lu-17, Yb-12, (Th, U)-11, Tb-9, (U, Ce)-8, (Sm, Na, La)-7, (Zr, Hf)-5, (Ti, K, P)-3	na	5.5	3583	7419	na	530	na	113	488	89	na	153	0	1539	376
19	082L769158	-118.160	50.690	5	3.7	(Th, Lu)-13, W-12, (Tb, Ce)-9, (Sm, La)-8, U-7, (Yb, Au)-6, Hf-5, (Zr, K, Na)-4, (Cr, Ti, Ti)-2	na	2.7	3294	6338	na	524	na	80	155	47	na	36	0	1497	198
20	082L769159	-118.102	50.766	57	4.6	W-19, Th-15, Lu-13, Sm-9, (Tb, Ce, La)-8, U-7, Yb-6, Hf-5, Zr-4, (Na, Ba, Rb, Au)-2	na	61	37810	66635	na	7103	na	857	1569	542	na	0	0	21751	2447
21	082L769160	-118.110	50.771	37	4.6	Th-20, Lu-16, (Tb, W)-14, Sm-12, U-10, (La, Ce)-9, Yb-7, (Zr, S)-5, Hf-4, (Na, K, As)-3	na	143	34167	52331	na	6785	na	1096	1404	477	na	0	0	20635	2963
22	082L769169	-118.220	50.813	10	4.6	Lu-21, Th-20, (La, Sm)-13, (Ce, Hf, Tb)-12, (U, K)-11, Zr-10, Yb-7, Ti-6, Ti-5, Cr-4, (Na, Sc)-3	na	0	13707	22151	na	1944	na	241	372	174	na	171	8.7	5251	933
23	082L769171	-118.174	50.851	34	3.5	Th-6, (Zr, Hf)-5, (Lu, Sm, Tb, Ce)-4, (U, La, K, Na)-3, (W, Ti, P, Yb, Ta)-2	na	46	1126	1048	na	653	na	70	0	26	na	0	0	2303	83
24	082L769172	-118.174	50.851	34	3.6	W-27, Th-6, (Zr, Hf)-5, (Ce, Lu, Sm)-4, (La, Tb, U, Na)-3, (K, Yb, P, Ti)-2	na	28	1985	7064	na	464	na	27	0	17	na	0	0	2157	0
25	082M763128	-118.457	51.098	4	3.6	Mo-159, Na-11, W-8, (K, Ti)-7, (La, F*, Au, U, Zr)-6, (Hf, Ta, Ce, Sr, Th)-4, (P, Ga, Ti)-3	na	44	1083	1748	na	38	na	6.5	20	4.0	na	0	837	48	65

Table 6. Continued.

N ¹	RGS sample	X ²	Y ²	S ³	Cl ⁴ leaders ⁵	Ranked element contrast	Predicted geochemical resources (in tonnes per 1 m depth) ⁶								Predominant host rock						
							Nb	Ta	La	Ce	Nd	Sm	Eu	Tb	Yb	Lu	Y	Sc	Mo	Th	U
26	082M763135	-118.351	51.086	41	3.9	W-45, Mo-18, Zr-11, Hf-8, (Na, Th, La, Ce)-6, (Ta, U)-5, (Lu, F*, Sm, K, Ti, Tb)-4, Sr-3	531	19006	32392	na	1504	na	147	405	111	na	0	861	4263	983	calc-silicate metamorphic rocks (Monashue complex; Proterozoic to early Paleozoic)
27	082M767025	-118.703	51.065	4	4.6	Th-15, La-12, Ce-9, Sm-7, (Hf, P)-6, (Tb, Zr)-5, (U, Ti)-4, (Na, K)-3, (Cr, W, Ca, Ga, Ba, Ta)-2	1.3	4724	5753	na	351	na	31	0	0	na	0	1616	80	coarse siliciclastic rocks (Monashue complex; Proterozoic to early Paleozoic)	
28	082M767028	-118.692	51.089	14	3.9	Th-9, W-8, (La, Na)-7, Ce-6, Sm-5, (Au, Ti, Zr)-4, (Tb, Hf, K, S, U)-3, (P, Sc, Sr, Mo)-2	0	9233	12010	na	671	na	49	0	0	na	120	0	3079	54	calc-silicate metamorphic rocks (Monashue complex; Proterozoic to early Paleozoic)
29	082M769045	-118.823	51.234	59	3.5	W-24, Na-9, (Th, Sm, Tb)-5, (K, Zr, Ti, La, Ce)-4, (Hf, Ta, U, Lu)-3, (Cr, P, Tl, Yb, Mo, Au)-2, (Tb, Ce)-9, (Sm, U)-7, (Tb, Ce)-9, Sm-8, (La, U)-7, (Ta, Yb)-5, (Bi, Au, Cr, Ti, Zr)-4	150	12964	20801	na	3031	na	339	0	67	na	201	0	5753	223	calc-silicate metamorphic rocks (Monashue complex; Proterozoic to early Paleozoic)
30	082M775247	-118.797	51.898	22	3.6	W-198, Na-12, (Lu, Th)-10, (Tb, Ce)-9, Sm-8, (La, U)-7, (Ta, Yb)-5, (Bi, Au, Cr, Ti, Zr)-4	282	13883	32403	na	2392	na	376	485	172	na	386	0	5005	1093	limestone, marble, calcareous sedimentary rocks (Horsethief Creek Group; Neoproterozoic)
31	082M775249	-118.772	51.876	30	3.7	Th-13, Th-12, Tb-10, (Sm, Ce)-9, (W, La)-8, (Na, U)-7, (Zr, Yb)-5, Hf-4, (Ti, Ta, K)-3, Cr-2, Lu-7, (Na, U, La)-6, (Ce, Hf)-5, (Ta, Yb, K, Ti)-4, (Cr, Au, Sc, P)-2	146	21534	49050	na	4092	na	603	832	318	na	414	0	9598	1542	limestone, marble, calcareous sedimentary rocks (Horsethief Creek Group; Neoproterozoic)
32	082M775264	-118.687	51.958	8	3.7	Th-9, (Tb, Sm, W)-8, (Zr, Lu)-7, (Na, U, La)-6, (Ce, Hf)-5, (Ta, Yb, K, Ti)-4, (Cr, Au, Sc, P)-2	78	3439	4349	na	845	na	125	162	34	na	161	0	1619	304	limestone, marble, calcareous sedimentary rocks (Horsethief Creek Group; Neoproterozoic)
33	082M775405	-118.786	51.746	4	3.7	W-12, (Th, Na)-8, Au-6, (Sm, Tb)-5, (La, Ti, Ce)-4, (K, Ta, U, Mo)-3, (Cr, Hf, Zr, Sc, P)-2	15	917	1358	na	167	na	20	0	0	na	28	64	590	19	limestone, marble, calcareous sedimentary rocks (Horsethief Creek Group; Neoproterozoic)
34	082M775408	-118.779	51.692	15	3.8	Th-8, (Na, La, Zr, Sm)-6, (Hf, Ce, Tb, K)-5, (Ti, U, Cr, Lu)-4, Ta-3, (Yb, P, W, Ti, Sc)-2	85	7017	9210	na	915	na	87	73	22	na	109	0	2632	293	limestone, marble, calcareous sedimentary rocks (Horsethief Creek Group; Neoproterozoic)
35	082M775413	-118.887	51.692	20	3.8	Au-14, Th-10, Zr-8, (Sm, Hf, Tb, Lu)-7, (La, Na, Ce)-6, (W, U)-5, (K, Yb)-4, Ti-3, (Cr, Ta)-2	26	10734	14754	na	1855	na	210	294	83	na	308	0	4952	540	limestone, marble, calcareous sedimentary rocks (Horsethief Creek Group; Neoproterozoic)
36	082M775470	-118.916	51.538	5	4.1	W-15, Th-12, (La, Ce)-6, (Sm)-8, Tb-7, Lu-6, (Ti, U)-5, (Na, K, Ta)-4, (Cr, Hf, Zr, Yb, Au)-3	31	3671	6367	na	501	na	57	35	19	na	35	0	1406	129	coarse siliciclastic rocks (Horsethief Creek Group; Neoproterozoic)
37	082M775479	-119.007	51.463	18	3.8	(Th, Lu)-8, W-7, (Sm, Tb)-6, (La, Ce)-5, (Ti, Na)-4, (Yb, S, Cr)-3, (Ti, K, Ta)-2	0	7286	11239	na	1152	na	149	220	92	na	43	0	3205	352	coarse siliciclastic rocks (Horsethief Creek Group; Neoproterozoic)
38	083D051004	-119.499	52.949	0.3	3.5	S-17, Th-6, (La, Ce, K, Hf, Sm, Ce, Eu)-4, (Cu, U, Lu, Ti)-3, (Yb, Na, Ti, P, Bi, Tb)-2	1.2	61	68	0	6.4	1.9	0	0.21	0.07	na	0.08	21	2.7	dolomitic carbonate rocks (Rocky Mountain assemblage; middle Cambrian)	
39	083D051005	-119.484	52.904	7	3.6	Th-8, Ta-7, (La, Ce, K, Sm)-6, (Lu, Yb, U)-5, (Hf, Nd, Tb)-4, (Ti, P, Eu, Ti, W, F*-Na, Sc)-3	89	2892	5332	2006	352	32	26	135	17	na	195	4.2	973	184	mudstone, siltstone, shale (Kaza Group, lower division; Neoproterozoic)

Table 6. Continued.

N ¹	RGS sample	X ²	Y ²	S ³	Cl ⁴	Ranked element contrast leaders ⁵	Predicted geochemical resources (in tonnes per 1 m depth) ⁶										Predominant host rock				
							Nb	Ta	La	Ce	Nd	Sm	Eu	Th	Yb	Lu	U				
40	083D051070	-119.630	52.872	16	4.4	Ti-34, (Hf, Th)-13, (Ce, La, K, Sm)-10, (Au, U)-9, Sm-8, Tb-7, (Nd, Eu)-6, (Ta, Cu, Lu, Yb)-5	na	136	14860	27435	10643	2083	276	211	280	35	na	59	26	5211	1261
41	083D051077	-119.569	52.903	3	3.5	Th-10, (La, Ce)-8, (Lu, Yb, Sm, K)-7, (Tb, Ta)-6, (U, Nd)-5, (Eu, Te, Ti, Hf, Ti)-4, (Sc, Fe)-3	na	24	1631	2936	1132	215	23	25	97	13	na	130	0	561	80
42	083D051144	-119.525	52.723	2	3.8	K-10, (La, Ti, Ta, Th)-5, (Sm, Ce, Eu, P, Ti)-4, (Tb, Na, U, Cu, Yb, Mo)-3, (Hf, Ba, Cr, Nd)-2	na	13	510	399	43	40	14	4.9	5.1	0.43	na	8.5	3.4	74	13
43	083D051145	-119.522	52.729	4	3.7	Au-35, Th-6, La-5, (Ce, Sm, K, Ta)-4, (Tb, Eu, U, Ti, Hf, Ti, W)-3, (Lu, Yb, Se, Nd, Te, Na)-2	na	17	1163	1183	196	99	21	11	1.0	na	na	0	2.9	267	39
44	083D051157	-119.363	52.735	6	3.7	Th-8, (K, La)-7, (Sm, Ce, Ta)-6, U-5, (Tb, Ti, Lu, Yb, Hf, Eu)-4, (Tl, Te, Sc, Na, Co, Nd)-3	na	54	2928	3646	563	364	36	32	72	9.9	na	183	0	960	155
45	083D051371	-119.206	52.579	2	3.7	Th-7, (La, Ce)-6, (Sm, U)-5, (K, Tb, Hf)-4, (Nd, Ti, Eu, Ta, Ti, Lu)-3, (Bi, F*, Yb, Te, Na, Fe)-2	na	0	790	1392	455	108	6.2	11	0	0	na	0	0	266	63
46	083D051424	-119.131	52.193	8	3.9	K-12, (La, Th)-8, (Ti, Co)-7, Sm-6, (Tb, U)-5, (Yb, Lu, Ti, Hf, Na, Te, Mo, Eu)-4, (S, Cr)-3	na	6.0	5528	6940	1248	476	58	33	85	9.2	na	15	12	1180	187
47	083D052034	-119.436	52.848	22	3.5	Th-10, (Ce, La, Bi)-8, Sm-7, (U, K, Nd, Tb)-6, (Ta, Lu)-5, (Hf, S, Yb, Eu)-4, (W, Ti, Ti, Sc)-3	na	181	13570	28871	12285	1996	161	193	285	50	na	580	0	4870	949
48	083D052128	-118.683	52.022	2	3.9	Th-12, Ce-10, (La, Sm)-9, (Na, Tb, Nd, Hf)-7, (U, Lu)-6, (K, Eu, Yb)-5, Ti-4, W-3, (P, Sc)-2	na	0	2225	4410	1718	307	29	53	7.7	na	45	0	692	101	
49	083D053028	-119.270	52.712	15	3.5	Th-11, Ce-10, (La, Sm)-9, U-7, (Ta, Nd, Tb, K, W)-6, (Hf, Lu, Yb)-5, (Eu, Ti)-4, (Tl, Sc, Fe)-3	na	185	12849	25279	9914	1718	145	139	271	38	na	572	0	4202	763
50	093N831597	-124.151	55.812	13	3.8	P-13, Th-9, (Lu, Ce)-6, (Sm, Au, La, U, Nd)-5, (Zr, Tb)-4, (Hf, K)-3, (Ti, Te, Tl, Ta, Yb, Cd)-2	na	20	3805	6842	na	624	na	52	33	50	na	0	0	1911	118

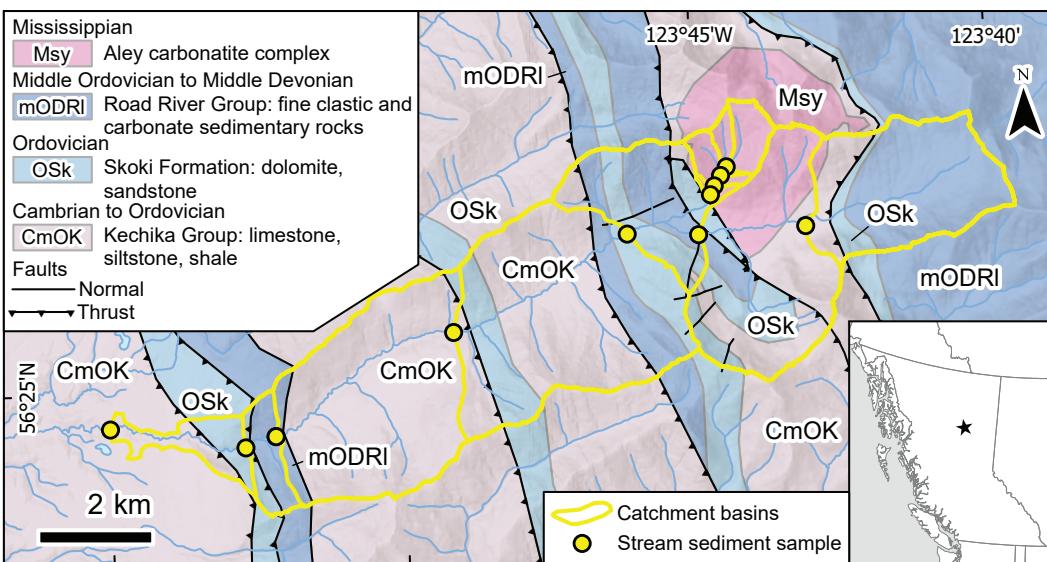


Fig. 19. Sampling sites and catchment basins at Aley carbonatite complex. Sampling sites after Mackay and Simandl (2014). Geology after Pride (1983), Mäder (1987), McLeish (2013), and BC Digital Geology version 2021-12-19 (Cui et al., 2017).

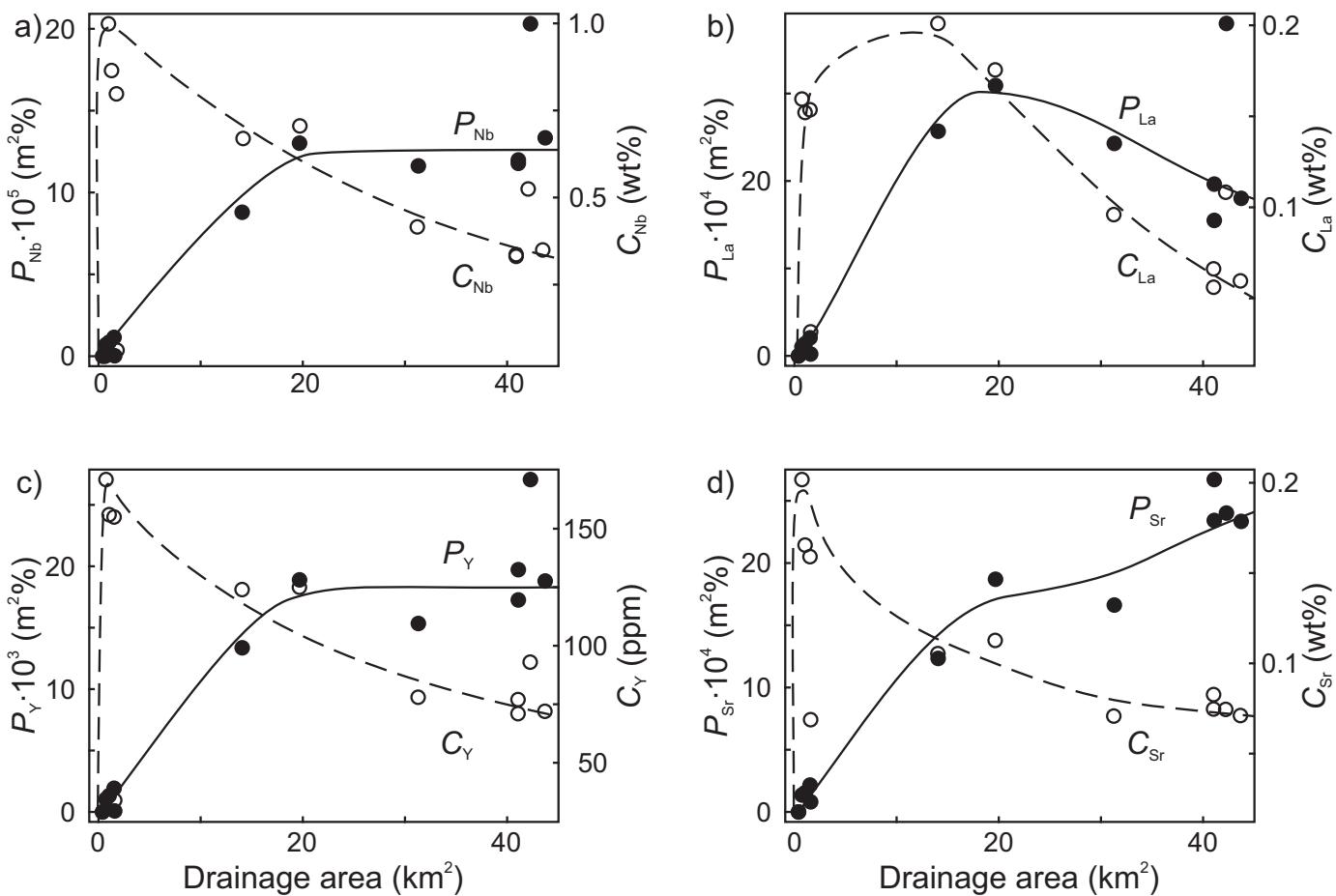


Fig. 20. Aley carbonatite complex, drainage area (km^2) versus concentration, C_x (wt.% or ppm; open symbols; model dashed line), and productivity, P_x ($\text{m}^{20\%}$; solid symbols, model solid line), of selected elements for the 0.125 to 0.250 mm fraction of stream-sediment samples. Original concentrations from Mackay and Simandl (2014). **a)** Drainage area (km^2) versus C_{Nb} (wt.% and $P_{\text{Nb}} \cdot 10^5 (\text{m}^{20\%})$). **b)** Drainage area (km^2) versus C_{La} (wt.% and $P_{\text{La}} \cdot 10^4 (\text{m}^{20\%})$). **c)** Drainage area (km^2) versus C_Y (ppm and $P_Y \cdot 10^3 (\text{m}^{20\%})$). **d)** Drainage area (km^2) versus C_{Sr} (wt.% and $P_{\text{Sr}} \cdot 10^4 (\text{m}^{20\%})$).

8. Discussion

Based on data from known carbonatites, an integrated approach to assess the critical metals potential of underexplored regions might combine detailed stream-sediment sampling and application of the critical mineral index as defined herein, high-resolution airborne radiometric and magnetic data, and panned heavy mineral concentrate and soil lithochemical surveys.

8.1. Geophysical response from carbonatites of the Blue River area

Both airborne radiometric and magnetic surveys are effective prospecting tools for carbonatite-hosted critical minerals (e.g., Simandl and Paradis, 2018), many of which contain U and Th and are associated with abundant magnetic minerals such as magnetite or pyrrhotite and K-rich metasomatic rocks such as glimmerites (carbonate-amphibole-phlogopite rocks). High-resolution, airborne gamma-ray uranium (Fig. 21) and total magnetic intensity (Fig. 22) highs highlight numerous known carbonatite occurrences in the Blue River area, which illustrate the effectiveness of geophysics for critical minerals prospecting, especially in vegetated, low-elevation areas (Gorham, 2008; Shives, 2009). However, difficulty in maintaining constant ground clearance in rugged terrain results in gamma counts and total magnetic intensities that reflect topography, with generally stronger responses from ridges and peaks relative to low-elevation areas (Gorham, 2008).

8.2. Stream-sediment, panned heavy mineral concentrate and soil lithochemical surveys

Concentrations of carbonatite indicator elements such as Ba, Mo, Nb, Ta, REE, Th, and U in stream-sediment heavy mineral concentrate (HMC) samples are up to two orders of magnitude higher relative to those in carbonatites and soil samples, and

up to 192 times higher relative to the bulk stream-sediment <0.18 mm fraction (Table 3). Recovered by panning in the field or by laboratory techniques (Lett and Rukhlov, 2017), HMC lithogeochemistry enhances the contrast of stream-sediment anomalies compared to the conventional, bulk <0.18 mm fraction used in the RGS drainage programs (Rukhlov et al., 2020a, b). Panned, stream-sediment HMC samples in the Blue River area ($n=626$), containing 1.2 to 2475 ppm Ta determined by lithium-fusion ICP-MS (Fig. 6), highlight the known carbonatites in the area, including at the Upper Fir and Verity Ta-Nb deposits (Dahrouge and Reeder, 2001; Reeder and Dahrouge, 2002; Smith and Dahrouge, 2003; Dahrouge and Wolbaum, 2004; Rukhlov and Gorham, 2007; Gorham, 2008; Gorham et al., 2009, 2011a, 2011b). In contrast, the RGS stream-sediment data in the area ($n=208$) show only <1 to 13 ppm Ta by INAA (Fig. 5). In addition, HMC are routinely evaluated for indicator minerals (e.g., Tyson, 2009; Mackay and Simandl, 2014; Mao et al., 2016; Simandl et al., 2017).

As follow-ups to regional stream-sediment and airborne geophysical surveys, grid soil lithogeochemical surveys (Fig. 7), coupled with ground (in situ) gamma-ray and magnetic surveys, effectively delineate secondary (residual) dispersion haloes of carbonatite-related critical metals (Reeder and Dahrouge, 2002; Smith and Dahrouge, 2003; Dahrouge and Wolbaum, 2004; Rukhlov and Gorham, 2007; Gorham, 2008; Gorham et al., 2009, 2011a, 2011b).

9. Conclusion

Carbonatites and related rocks have a very distinct geochemical signature, characterized by extreme concentrations of critical metals such as REE, Nb, Ta and other elements. Because most carbonatites in British Columbia form small sills and dikes, the task of detecting their signal using the

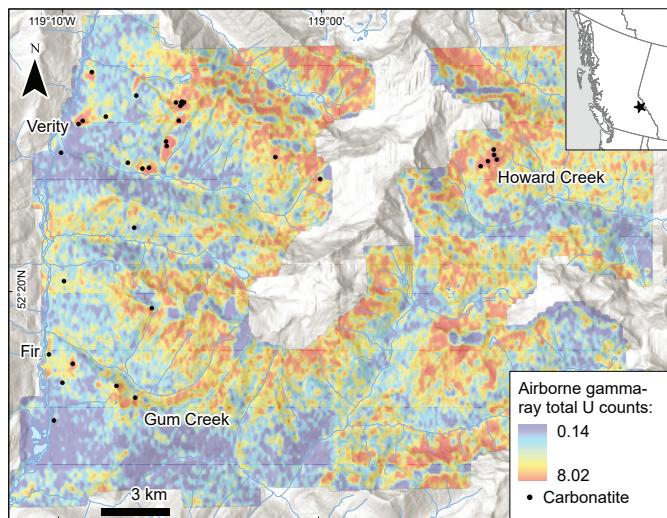


Fig. 21. Blue River area gridded airborne gamma-ray uranium response (total counts). After Gorham (2008) and Shives (2009). Grid cell size is 40 m. Carbonatite occurrences after Rukhlov and Gorham (2007), Gorham (2008), Gorham et al., 2009, 2011a, 2011b, 2013), Millonig and Groat (2013), and Rukhlov et al. (2018).

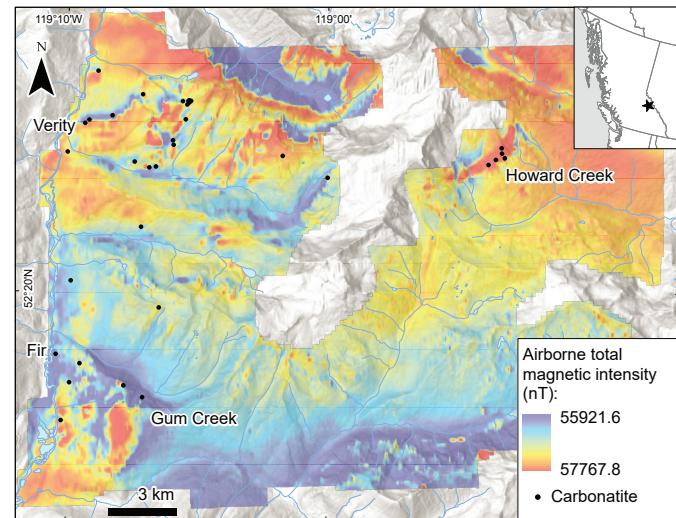


Fig. 22. Blue River area gridded airborne total magnetic intensity response (nT). After Gorham (2008) and Shives (2009). Grid cell size is 40 m. Carbonatite occurrences after Rukhlov and Gorham (2007), Gorham (2008), Gorham et al., 2009, 2011a, 2011b, 2013), Millonig and Groat (2013), and Rukhlov et al. (2018).

multi-element regional geochemical survey stream data with sample catchment areas of up to 193 km² (average 10 km²) is akin to looking for a needle in a haystack. Univariate data reflect background variations, including at known carbonatite and related-rock occurrences. In contrast, our multivariate carbonatite index identifies numerous stream basins that are prospective for carbonatite-related critical minerals. The top 50 anomalies (>93rd percentile) show maximum contrast of carbonatite indicator elements relative to the background in the study area. Estimated predicted geochemical resources suggest significant potential for REE and other carbonatite-related critical metals. We propose a refined prospectivity approach to assess the critical metals potential of underexplored regions that includes stream-sediment, panned heavy mineral concentrate lithochemical and indicator mineral surveys, coupled with high-resolution airborne radiometrics and magnetics, followed up by soil lithochemical and in situ (ground) radiometric and magnetic surveys.

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